



STATE OF CALIFORNIA
DEPARTMENT OF WATER RESOURCES
DIVISION OF RESOURCES PLANNING

Bulletin No. 65-59
QUALITY OF
SURFACE WATERS
IN CALIFORNIA
1959

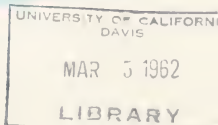
PART I
NORTHERN AND CENTRAL CALIFORNIA

Edmund G. Brown
Governor



William E. Warne
Director of Water Resources

July 1961



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IN CALIFORNIA, 1959

WATER POLLUTION CONTROL REGIONS
NORTHERN AND CENTRAL CALIFORNIA

LEGEND

-  REGION BOUNDARIES
-  REGION 5 SUBAREA BOUNDARIES
-  AREA REPORTED IN PART II
OF BULLETIN 65-59

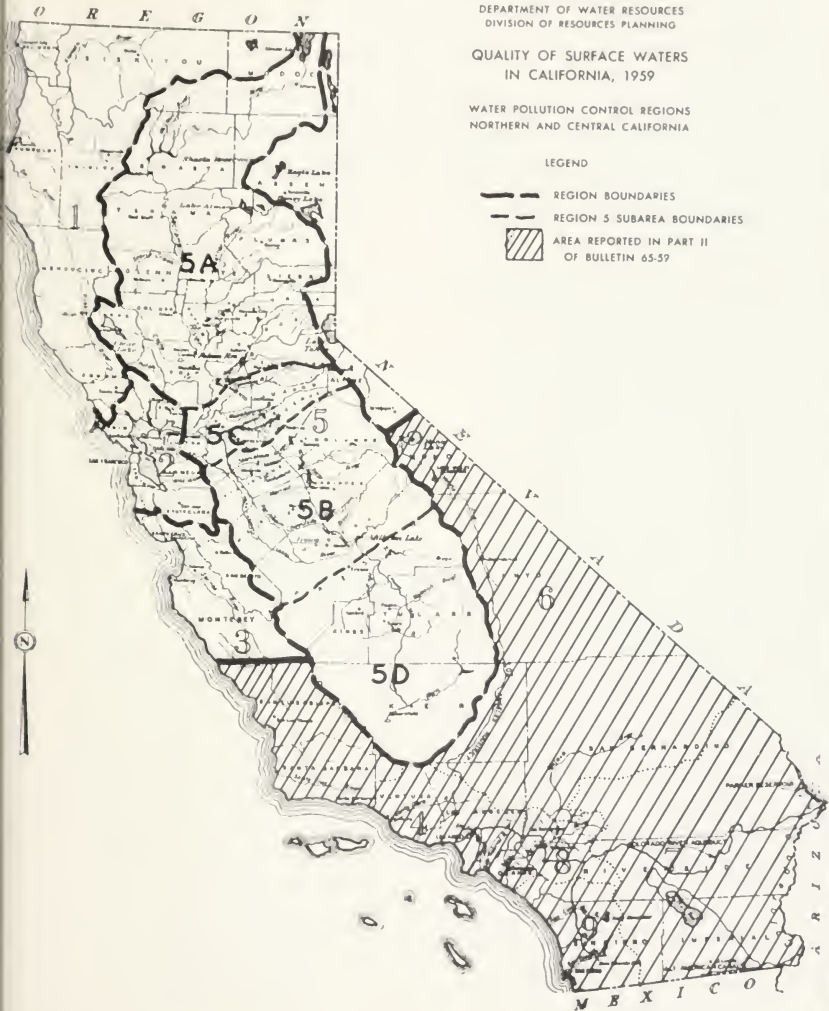


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WILLIAM E. WARNE
DIRECTOR

EDMUND G. BROWN
GOVERNOR

ADDRESS REPLY TO
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STATE OF CALIFORNIA
Department of Water Resources

SACRAMENTO
JUL 10 1961

Honorable Edmund G. Brown, Governor,
and Members of the Legislature of
the State of California

Water Pollution Control Boards

Gentlemen:

I have the honor to transmit Bulletin No. 65-59, entitled "Quality of Surface Waters in California, 1959, Part I, Northern and Central California". The period January through December 1959 is covered in this fifth volume of a continuing chronological series on quality of surface waters in California. Surface waters in Northern and Central California are discussed in Part I; Southern California surface water quality will be reported in Part II.

At the request of the State Water Pollution Control Board, a statewide surface water monitoring program was commenced in April 1951. As authorized by Section 229 of the Water Code, the Department of Water Resources has administered this program in cooperation with the State Department of Public Health, Bureau of Sanitary Engineering; the State Department of Fish and Game; the United States Geological Survey; and various other agencies and individuals. Under the statewide program samples from 210 stations, located on 143 different water sources, are collected and analyzed monthly to maintain surveillance on quality of surface waters in California. This volume reports the results of monitoring at 178 of these stations, located on 110 streams and lakes, in Northern and Central California.

During 1959 quality of surface waters in Northern and Central California was generally excellent, with only insignificant changes from previous years. During late summer and fall, the lower San Joaquin River

and the Sacramento-San Joaquin Delta contained water with mineral concentrations which were the maximum of record. The high mineral concentrations often caused the water to be classed as poor in quality. The poor quality water was attributed to low flows, resulting from below normal precipitation, which afforded only minor dilution to poor quality drainage and effluent ground waters entering the lower reach of the river and the delta.

Part II of this bulletin, which will include an evaluation of surface water quality conditions in Southern California, will be published at a later date.

Sincerely yours,

William F. Wame

Director

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ACKNOWLEDGMENTS

The extensive coverage of the statewide surface water monitoring program is made possible through cooperation of federal, state and local agencies. The helpful cooperation of the following agencies is gratefully acknowledged:

Federal Agencies

Department of the Army

Corps of Engineers

Department of the Interior

Bureau of Reclamation

Geological Survey

Department of Health, Education and Welfare

Public Health Service

State Agencies

California Disaster Office, Radiological Service

Department of Fish and Game

Department of Public Health

Bureau of Sanitary Engineering

Division of Laboratories

State Water Pollution Control Board

Other Public Agencies

City and County of San Francisco

Kern County Land Company

Kings River Water Association

The Department of Water Resources wishes to especially thank the following federal and state agencies who granted permission for inclusion in this report of unpublished water quality data collected under various programs:

United States Department of the Interior

Bureau of Reclamation

Geological Survey

Central Valley Regional Water Pollution Control Board (No. 5)

In addition, the United States Geological Survey performed a substantial portion of the analyses required by this program under a cooperative agreement with the Department of Water Resources. The bacteriological determinations were made by the California State Department of Public Health and the radiological determinations by the California Disaster Office under provisions of an agreement with the State Water Pollution Control Board.

INTRODUCTION

Bulletin No. 65-59 is the fifth volume in a series on surface water quality conditions in California. Data presented were collected by the Department of Water Resources' surface water monitoring program and other public agencies in California. In addition to basic data, this bulletin contains evaluations and interpretations of significant quality variations detected during 1959 and, where possible, an explanation of the causes of these variations.

To disseminate quality data as soon as practicable, the department also publishes, and distributes a monthly report containing data and preliminary evaluations of detected quality variations. These reports are distributed to pollution control, public health, and other agencies and individuals.

An abundant and usable source of water is an essential commodity in man's present environment. To insure that California's rapidly expanding economy and increasing population are provided with a usable supply of water an effective surveillance program must be maintained. The early detection and control of quality impairment is necessary, in order to insure the fullest beneficial use of the State's water resources. Realizing the need for a detection system, the State of California initiated a statewide surface water monitoring program in April 1951. Since that time the monitoring program has been conducted by the Department of Water Resources in cooperation with numerous agencies and individuals. Other agencies have also maintained monitoring stations at various places throughout the State.

Part I of this bulletin presents water quality data and an evaluation of surface water quality conditions in Water Pollution Control Regions 1 and 2, the portion of Region 3 north of the San Antonio-Salinas River drainage boundary, Region 5, and the portion of Region 6 north of the Mono Lake drainage divide. Part II, to be published at a later date, will present data and an evaluation of surface water quality conditions in the southern portion of Region 3 (Santa Ynez, Santa Maria, Nacimiento, and San Antonio Rivers and the portion of Salinas River upstream from the confluence of San Antonio River), all of Region 4, Region 6 south of the northern Mono Lake drainage boundary and all of Regions 7, 8 and 9. The regions and the areas reported on in this volume are shown on the frontispiece map.

The 1959 stream sampling programs reported herein comprised the collection of water samples and analyses from 178 stations on 110 streams and lakes throughout Northern and Central California. Previous quality monitoring data are included in the following report and bulletins:

California Department of Public Works, Division of
Water Resources, Water Quality Investigations,
Report No. 15, "Quality of Surface Waters in
California, 1951-1954"

California Department of Water Resources, Division
of Resources Planning, Bulletin No. 65, "Quality of
Surface Waters in California, 1955-1956"

----. Bulletin No. 65-57, "Quality of Surface Waters
in California, 1957"

----. Bulletin No. 65-58, "Quality of Surface Waters
in California, 1958"

The activities of the department's surface water monitoring program are authorized by Section 229 of the Water Code, which directs that:

"The department, . . . shall investigate conditions of the quality of all waters within the State, including saline waters, coastal and inland, as related to all sources of pollution of whatever nature and shall report thereon to the Legislature and to the appropriate regional water pollution control board annually, and may recommend any steps which might be taken to improve or protect the quality of such waters."

The basic objectives of the department's surface water quality monitoring program are:

- (a) to secure continuous and reliable water quality data, on a monthly basis, from a network of stations which will provide representative data pertaining to the quality of water in the major surface streams and lakes of the State;
- (b) to evaluate and interpret chemical, physical, biological and radiological information collected during the course of the program to develop a comprehensive understanding of the factors which make up and alter the water quality at any station; and
- (c) to detect changes in water quality and to notify the appropriate control agency, (regional water pollution control boards, state and local health departments, State Department of Fish and Game) when warranted.

The discussion of water quality data collected by the Department of Water Resources' surface water monitoring program is presented in this bulletin, in successive order, by water pollution control regions which are numbered and named substantially in accordance with the major surface drainage basins with which they are coterminous (see Frontispiece). For convenience in presentation, the Central Valley Region (No. 5), has been divided into four separate areas, 5a, 5b, 5c, and 5d. Area 5a embraces

the Sacramento Valley, 5b the San Joaquin Valley, 5c the Sacramento-San Joaquin Delta, and 5d the Tulare Lake Basin. Within each region, the discussion is presented by basins or stream groups. In each basin or stream group, the main stream is discussed first, followed by a discussion and summary of data, in downstream order, of all monitoring stations. The discussion for each monitoring station includes a detailed location description of the sampling point, period of quality record, a detailed discussion of water quality characteristics, and an analysis of significant water quality changes in 1959. For each station a presentation is given for the maximum and minimum concentrations of the mineral constituents in the water for the total period of record and for 1959; curves depicting the monthly variation, for the period of record, of stream flow, specific conductance, and, where applicable, pertinent problem mineral constituent concentrations.

Following the discussion and analysis of the Department of Water Resources monitoring program, a listing of water quality monitoring stations maintained by other agencies during 1959 is presented. This listing includes the name and number of the station, a description of the sampling point, the agency responsible for the station operation, and where known, the period of water quality record. No attempt is made in this bulletin to present an evaluation of quality monitoring data collected by other agencies.

Results of bacteriological and radiological determinations presented in this bulletin should be considered as only qualitative indicators and undue weight should not be given to quantitative values. The indicators contribute to long-term environmental studies.

Results of bacteriological examinations are expressed as the most probable number (MPN) of coliform bacteria per milliliter (ml) of sample. In view of the rapidity and frequency of change in the density of coliform organisms, frequent and lengthy sampling is necessary before a truly reliable evaluation can be made.

Results of radiological determinations are expressed in terms of activity, measured in micro-micro curies per liter ($\mu\mu\text{c/l}$). No well-defined limits have been established for maximum safe concentrations of unknown alpha and beta emitters in domestic water supplies. The International Commission on Radiological Protection has recommended provisional criteria for permissible concentrations of radioactivity in water. Even though evaluation criteria have been recommended by this commission, this bulletin does not attempt to evaluate the specific safety conditions. Pertinent features of these criteria are given in Appendix A.

Appendix A of this bulletin contains a discussion of field and laboratory procedures and methods, and the criteria utilized by the Department of Water Resources in evaluating the quality of water. Appendix B contains the physical, mineral, bacteriological and radiological data for samples collected during 1959.

SURFACE WATER QUALITY, DEPARTMENT OF WATER RESOURCES
MONITORING PROGRAM

Summary

During 1959 quality of surface waters in Northern and Central California was generally excellent, with only insignificant changes from previous years. During late summer and fall, the lower San Joaquin River and the Sacramento-San Joaquin Delta contained water with mineral concentrations which were the maximum of record. The high mineral concentrations often caused the water to be classed as poor in quality. The poor quality water was attributed to low flows, resulting from below normal precipitation, which afforded only minor dilution to poor quality drainage and effluent ground waters entering the lower reach of the rivers and the delta.

North Coastal Region (No. 1)

The North Coastal Region extends southward from the Oregon border 270 miles, to the northern boundary of Lagunitas Creek Basin in Marin County, and ranges in width from 180 miles at the Oregon boundary to 30 miles in the southern portion.

Terrain of this region is largely mountainous, with cliffs often several hundred feet high along the coast line, and steep canyons and numerous ridges with many peaks inland. Valley and mesa land, easily adaptable to agricultural development, comprises about 15 percent of the 19,586 square miles in this region. A fairly thick absorptive soil mantle covers much of the area and helps sustain stream flow through drier portions of the year.

Natural mean seasonal surface runoff is estimated to exceed 28,800,000 acre-feet. Principal hydrographic units in this region include the drainage basins of the Smith, Klamath, Mad, Eel, and Russian Rivers. Thirty-two sampling stations shown on Plate 1, "Surface Water Monitoring Program Stream Sampling Stations North Coastal Region (No. 1)", are being monitored to obtain information and to provide a continuing check on the quality of surface water resources in the North Coastal Region. Monitored streams are listed below with the number of sampling stations along each in parentheses.

Klamath River (5)	Outlet Creek (1)
Antelope Creek (1)	Eel River, Middle Fork (1)
Butte Creek (1)	Eel River, South Fork (1)
Shasta River (1)	Van Duzen River (1)
Scott River (1)	Mattole River (1)
Salmon River (1)	Noyo River (1)
Trinity River (3)	Big River (1)
Smith River (1)	Navarro River (1)
Redwood Creek (1)	Gualala River (1)
Mad River (1)	Russian River (3)
Eel River (3)	Russian River, East Fork (1)

A review of quality data revealed surface water in the northern portion of this region to be predominantly calcium-magnesium bicarbonate, while streams in the remaining portions were generally calcium bicarbonate in character. Excellent quality water for all but the most exacting requirements is found in North Coastal streams. During 1959 there was no appreciable change in the mineral quality of streams in the North Coastal Region.

Klamath River Basin

The California portion of the Klamath River Basin is located in the northern section and comprises over one-half the North Coastal Area. The watershed includes all tributaries downstream from the boundary between Oregon and California as well as those portions of Butte Valley (a basin of interior drainage), Lost River and Tule Lake Basins that lie in California. The Klamath's main tributaries in California are Trinity, Salmon, Scott, Shasta, and Lost Rivers. The Klamath River Basin encompasses 15,715 square miles of which approximately 10,020 square miles are in California. The average seasonal flow of the Klamath River into the Pacific Ocean is about 12,500,000 acre-feet.

Land classification surveys indicate approximately 405,000 acres of land in this basin are irrigable of which 182,000 acres are presently irrigated. The approximately 6,000,000 remaining acres are comprised of a series of mountain ranges separated by long, narrow river valleys. The mountainous areas and undeveloped valley lands are used extensively for livestock range, timber production, mining, and recreation. Support of fish and wildlife is of major importance to the welfare of this basin.

Numerous lumbermill operations and small communities discharge waste into the Klamath River. Most of these wastes are minor in quantity and do not result in a discernible quality impairment problem. Irrigation return causes some mineral impairment of tributaries to Klamath River; however, the overall effect is not significant.

Thirteen surface water monitoring stations are located in the Klamath River Basin. The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Klamath River near Copco	12
Klamath River above Hamburg Reservoir Site	14
Klamath River near Seiad Valley	16
Klamath River at Somesbar	18
Klamath River near Klamath	20
Antelope Creek near Tennant	22
Butte Creek near MacDoel	24
Shasta River near Yreka	26
Scott River near Fort Jones	28
Salmon River at Somesbar	30
Trinity River at Lewiston	32
Trinity River near Burnt Ranch	34
Trinity River near Hoopa	36

KLAMATH RIVER NEAR COPCO (STA. 1)

Sampling Point The monitoring station is located in Section 36 of Township 48 North, Range 5 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage 1 mile downstream from Copco No. 2 power plant of the California-Oregon Power Company, 500 feet downstream from Fall Creek.

Period of Record April 1951 through December 1959.

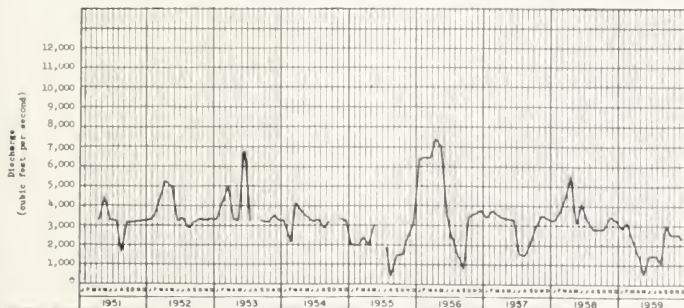
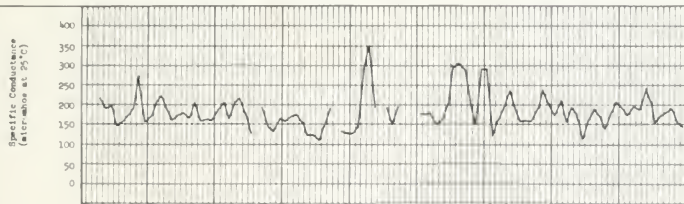
Water Quality Characteristics Klamath River near Copco is excellent, a bicarbonate type with sodium as its most predominant cation, class 1 for irrigation, soft to slightly hard, and meets drinking water standards for mineral content.

Significant Water Quality Changes Concentrations of dissolved oxygen were generally lower throughout 1959 with the minimum of record, 4.7 ppm, being reported in June. This apparent deoxygenation of the river was possibly caused by algal bloom die-off, resulting in an increased B.O.D. along this reach.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	71	171	70	147
Temperature in °F	74	4	7	39
Dissolved oxygen in parts per million	11.4	5.7	8.7	4.7
Percent saturation	111	59	77	59
pH	8.2	6.5	7.8	7.2
Mineral constituents in parts per million				
Calcium (Ca)	25	5.1	13	10
Magnesium (Mg)	11	3.6	10	6.1
Sodium (Na)	29	0.9	7	12
Potassium (K)	4.5	1.3	3.5	2.5
Carbonate (CO ₃)	55.0	8.5	46.5	36.5
Bicarbonate (HCO ₃)	117	54	98	72
Sulfate (SO ₄)	50	8.7	29	8.7
Chloride (Cl)	9.4	0.0	9.4	1.2
Nitrate (NO ₃)	5.4	1.2	3.9	2.8
Fluoride (F)	0.6	0.0	0.3	0.2
Boron (B)	0.30	0.0	0.1	0.0
Silica (SiO ₂)	42	5.3	41	22
Total dissolved solids in parts per million	250	84	153	106
Percent sodium	55	19	41	26
Hardness as CaCO ₃ in parts per million				
Total	115	18	74	19
Noncarbonate	28	14	3	0.0
Turbidity	50	0.0	20	0.0
Coliforma in most probable number per milliliter	>7,000.	5.11	7,000.	5.28
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.90	0.30	0.90	0.00
Solid alpha	0.52	0.00	0.37	0.36
Dissolved beta	22.9	0.00	7.93	6.18
Solid beta	5.9	0.00	2.05	0.73

WATER QUALITY VARIATIONS



KLAMATH RIVER NEAR COPCO (STA. 1)

KLAMATH RIVER ABOVE HAMBURG RESERVOIR SITE (STA. 1c)

Sampling Point Klamath River monitoring Station 1c is located in Section 14 of Township 46 North, Range 10 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a bridge on State Highway 96, in the center of the channel of flow, about 6 miles upstream from the mouth of Scott River, about 7 miles northeast of the town of Hamburg.

Period of Record December 1958 through December 1959.

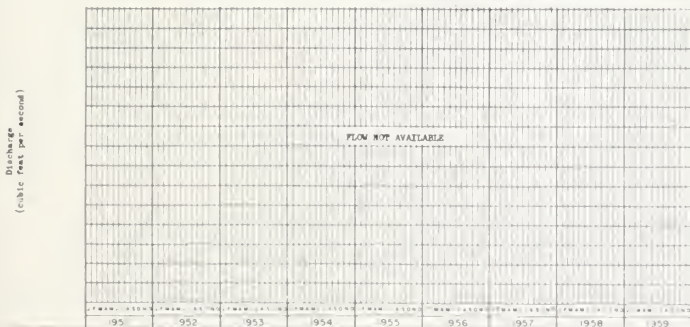
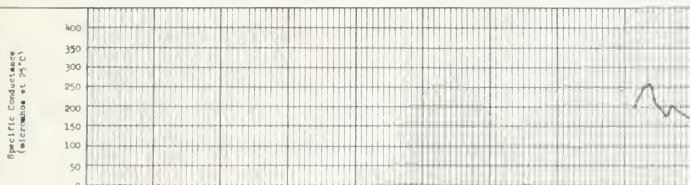
Water Quality Characteristics Mineral classification of analyses of samples from this station show the water to be a bicarbonate type with no major cation. Qualitatively, this river is class 1 for irrigation, soft to slightly hard, and meets drinking water standards for mineral content. The concentration of most mineral constituents in Klamath River between Station 1 near Copco and Station 1c are fairly comparable. Based upon limited data, it appears that the Shasta River at times may slightly degrade water quality of the Klamath in the reach immediately above Station 1c.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	259	170
Temperature in °F			74	30
Dissolved oxygen in parts per million			12.3	7.7
Percent saturation			108	86
pH			7.9	7.4
Mineral constituents in parts per million				
Calcium (Ca)			19	12
Magnesium (Mg)			10	6.6
Sodium (Na)			20	12.6
Potassium (K)			1.2	1.6
Carbonate (CO ₃)			0.7	0.0
Bicarbonate (HCO ₃)			117	86
Sulfate (SO ₄)			31	5.0
Chloride (Cl)			12	5.0
Nitrate (NO ₃)			3.0	0.0
Fluoride (F)			0.2	0.0
Boron (B)			0.2	0.1
Silica (SiO ₂)			42	19
Total dissolved solids in parts per million			168	125
Percent sodium			35	26
Hardness as CaCO ₃ in parts per million				
Total			88	57
Noncarbonate			4	0.0
Turbidity (Not Measured)				
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.10	0.00
Solid alpha			0.27	0.00
Dissolved beta			8.96	3.59
Solid beta			2.15	1.78

WATER QUALITY VARIATIONS



KLAMATH RIVER ABOVE HAMBURG RESERVOIR SITE (STA. 1c)

KLAMATH RIVER NEAR SEIAD VALLEY (STA. 2b)

Sampling Point Station 2b is located in Section 3 of Township 46 North, Range 12 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-depth, from the right bank, at the USGS gaging station, 0.4 mile upstream from Bittenbender Creek, about 14 miles downstream from the mouth of Scott River, and 2.2 miles west of the town of Seiad Valley.

Period of Record December 1958 through December 1959.

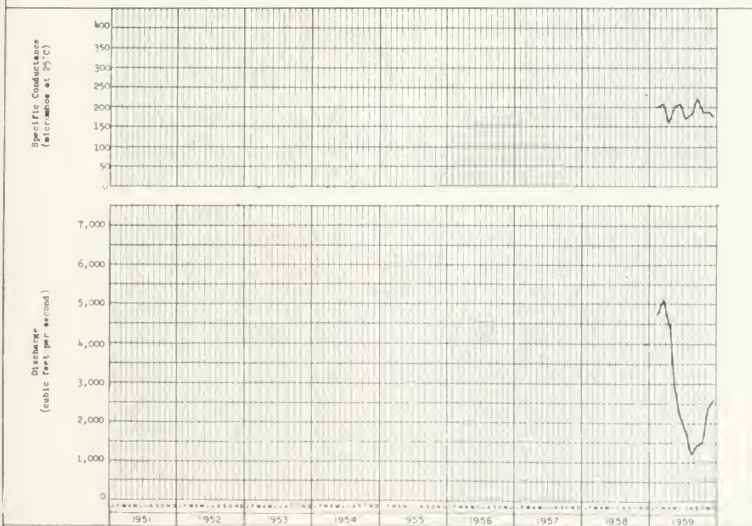
Water Quality Characteristics Since inception of monitoring of the Klamath River at this station, the water has been excellent in quality, calcium-magnesium bicarbonate in character, class 1 for irrigation, slightly hard, and has met drinking water standards for mineral content. There is normally no significant difference between the mineral content of the Klamath River at this station and above Hamburg Reservoir Site (Station 1c). However, it is noted that during periods of high inflow from the Scott River, the major tributary to the Klamath River between Stations 1c and 2b, regardless of flow conditions on the Klamath, a significant decrease in the mineral content of the Klamath River occurs. This condition was shown by conductivity values in April 1959, which decreased from approximately 250 micromhos at Station 1c to approximately 160 micromhos at Station 2b. This phenomenon indicates mineral content of Scott River is sufficiently low to more than offset the degradation of Klamath River caused by Shasta River.

Significant Water Quality Changes The maximum radioactivity found in the Klamath River Basin during 1959 was the 35.3 $\mu\text{uc}/\text{l}$ total activity recorded at Station 2b in May. The activity decreased appreciably during the year to 8.3 $\mu\text{uc}/\text{l}$ (micro-micro curies per liter) in September.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1977	See 1977	224	181
Temperature in °F			71	71
Dissolved oxygen in parts per million			8.9	8.9
Percent saturation			83.2	86
pH			7.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)			7	47
Magnesium (Mg)			12	1
Sodium (Na)			7	1
Potassium (K)			7.8	1.2
Carbonate (CO ₃)			4	2
Bicarbonate (HCO ₃)			118	85
Sulfate (SO ₄)			29	1.5
Chloride (Cl)			18	2.5
Nitrate (NO ₃)			2.4	1.5
Fluoride (F)			2	1.5
Boron (B)			0.2	0.2
Silica (SiO ₂)			40	37
Total dissolved solids in parts per million			166	100
Percent sodium			77	17
Hardness as CaCO ₃ in parts per million				
Total			20	82
Noncarbonate			2	1.5
Turbidity (Not Measured)				
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.20	0.08
Solid alpha			0.56	0.44
Dissolved beta			15.01	7.71
Solid beta			19.00	0.00

WATER QUALITY VARIATIONS



KLAMATH RIVER NEAR SEIAD VALLEY (STA. 2b)

KLAMATH RIVER AT SOMESBAR (STA. 2)

Sampling Point The Somesbar station is located on the Klamath River in Section 4 of Township 11 North, Range 6 East, Humboldt Base and Meridian. Monthly grab samples were collected at mid-depth, from the left bank, 100 feet downstream from the USGS gage, 1 mile west of Somesbar post office and 300 feet downstream from Salmon River.

Period of Record April 1951 through December 1959.

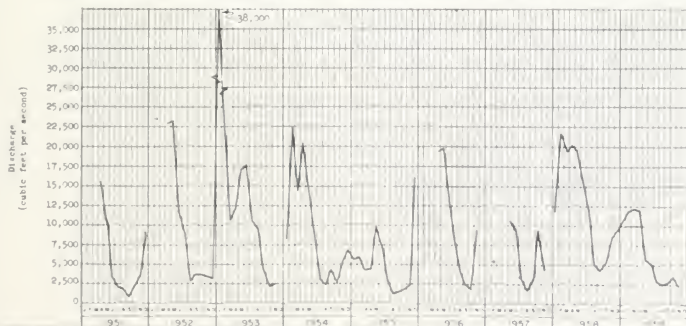
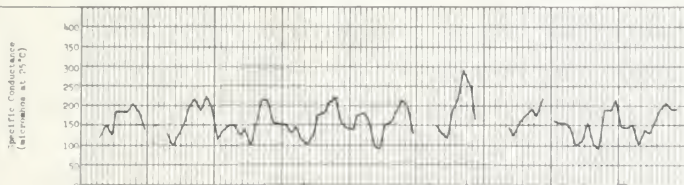
Water Quality Characteristics Since inception of the monitoring program, flow in Klamath River at Station 2 has been excellent in quality, calcium-magnesium bicarbonate in character, class 1 for irrigation, with a range from soft to slightly hard and has consistently met drinking water standards for mineral content. Review of data reveals a general improvement, averaging about 30 micromhos, in the mineral quality of Klamath River flow between Station 2b above Hamburg Reservoir Site and Station 2. This improvement is attributed to dilution by better quality tributary waters between the two stations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1957
Specific conductance (micromhos at 25°C)	280	91.3	208	108
Temperature in °F	81	39	78	41
Dissolved oxygen in parts per million	14.2	7.2	13.1	8.1
Percent saturation	124	50	100	97
pH	8.8	6.3	8.3	7.3
Mineral constituents in parts per million				
Calcium (Ca)	23	9.2	17	9
Magnesium (Mg)	13	2.6	8.1	7.2
Sodium (Na)	22	1.8	18	1.8
Potassium (K)	1.4	0.7	2.1	0.9
Carbonate (CO ₃)	2	0.7	2	0.7
Bicarbonate (HCO ₃)	124	50	118	46
Sulfate (SO ₄)	35	1.5	11	0.6
Chloride (Cl)	2.0	0.0	2.0	0.0
Nitrate (NO ₃)	2.4	0.0	2.4	0.0
Fluoride (F)	0.4	0.0	0.4	0.0
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	36	5.3	30	15
Total dissolved solids in parts per million	200	57	140	63
Percent sodium	34		34	
Hardness as CaCO ₃ in parts per million				
Total	98	48	86	44
Noncarbonate	18	0.0	0	0.0
Turbidity	300	0.0	300	0.0
Coliform in most probable number per milliliter	2,400	0.04	690	0.04
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.27	0.09	0.27	0.09
Solid alpha	0.72	0.00	0.36	0.00
Dissolved beta	13.30	0.00	13.44	0.00
Solid beta	22.5	0.00	19.96	0.00

WATER QUALITY VARIATIONS



KLAMATH RIVER AT SOMESBAR (STA. 2)

KLAMATH RIVER NEAR KLAMATH (STA. 3)

Sampling Point Station 3 is located in Section 17 of Township 13 North, Range 2 East, Humboldt Base and Meridian. Monthly grab samples were collected at mid-depth from the right bank at the USGS gaging station, about 6 miles upstream from the mouth, 3.3 miles east of Klamath (on Highway 101) and 0.4 mile upstream from Klamath Glen Road.

Period of Record April 1951 through December 1959.

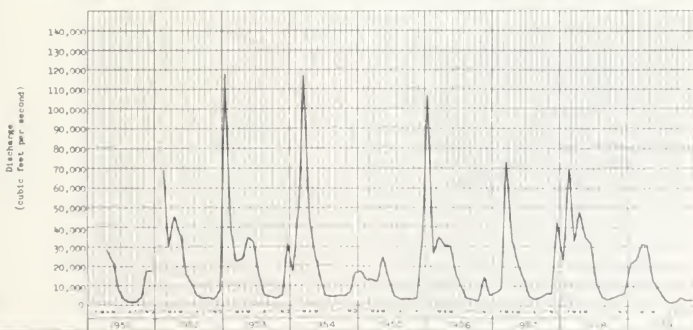
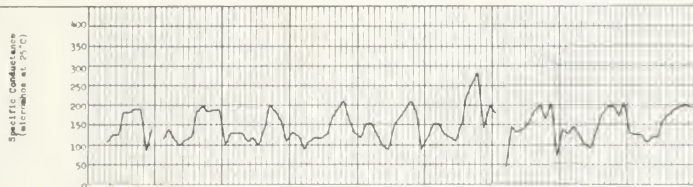
Water Quality Characteristics Antecedent data reveal Klamath River water, at this station, to be excellent in quality, calcium bicarbonate in character and class 1 for irrigation. It consistently ranges from soft to slightly hard and meets drinking water standards for mineral content. In past years, Klamath River water, due to tributary inflow, has consistently undergone a gradual change in character from bicarbonate type with no predominant cation at Station 1 (near Copco) to a generally calcium bicarbonate type water at Station 3 (near Klamath). A slight decrease in the concentration of constituents, on the order of 25 micromhos from the upstream station to the downstream station, has also been noted. Analyses of water samples collected from the Klamath River at Stations 1c and 2b disclose that the concentration of mineral constituents in the river normally increase to a maximum in the vicinity of these two stations. Mineral content again decreases, as tributary inflows dilute mineral concentration, to a minimum at Station 3.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	280	12.1	200	100
Temperature in °F	74	40	53	32
Dissolved oxygen in parts per million	14.5	7.4	12.4	7.7
Percent saturation	118	81	108	71
pH	8.3	7.7	7.7	7.3
Mineral constituents in parts per million				
Calcium (Ca)	31	4.0	18	1.8
Magnesium (Mg)	18	1.0	12	0.8
Sodium (Na)	17	2.6	12	0.8
Potassium (K)	12	4	2.2	0.2
Carbonate (CO ₃)	130	10	100	10
Bicarbonate (HCO ₃)	29	7.4	21	1.6
Sulfate (SO ₄)	17	6.7	6.7	1.6
Chloride (Cl)	1.5	0.3	1.2	0.3
Nitrate (NO ₃)	1	0.2	0.8	0.2
Fluoride (F)	0.4	0.1	0.3	0.1
Boron (B)	0.4	0.1	0.3	0.1
Silica (SiO ₂)	29	5.7	24	10
Total dissolved solids in parts per million	182	30	151	40
Percent sodium	32	9	25	9
Hardness as CaCO ₃ in parts per million				
Total	64	18	46	18
Noncarbonate	8	0.2	8	0.2
Turbidity	100	0.0	80	0
Coliform in most probable number per milliliter	>7,000	0.06	>7,000	0.06
Radioactivity in micro-curies per liter				
Dissolved alpha	0.24	0.00	0.20	0.10
Solid alpha	1.43	0.00	1.43	0.00
Dissolved beta	16.00	0.00	16.00	0.00
Solid beta	16.00	0.00	16.00	0.00

WATER QUALITY VARIATIONS



KLAMATH RIVER NEAR KLAMATH (STA. 3)

ANTELOPE CREEK NEAR TENNANT (STA. 1e)

Sampling Point The station is located in Section 25 of Township 43 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank. The sampling point is 4 miles downstream from Frog Lake, 17 miles southeast of the town of Mount Hebron, and 2.5 miles south of Tennant.

Period of Record March 1959 through December 1959.

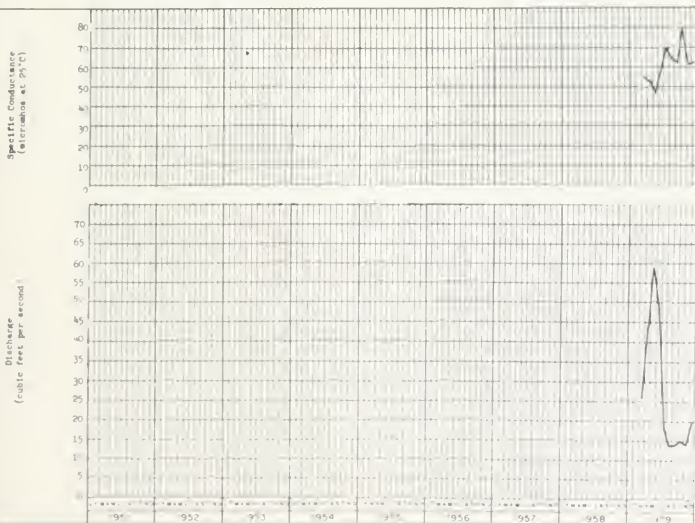
Water Quality Characteristics Past analyses of samples of Antelope Creek show it to be calcium bicarbonate in character, class 1 for irrigation, soft, and meets drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	See 1049	See 1049	72.4	47.8
Temperature in °F			60	32
Dissolved oxygen in parts per million			15.8	7.7
Percent saturation			77	73
pH			7.4	7.1
Mineral constituents in parts per million				
Calcium (Ca)			8.8	4.2
Magnesium (Mg)			1.9	1.7
Sodium (Na)			5.0	1.9
Potassium (K)			2.6	0.3
Carbonate (CO ₃)			0.0	0.0
Bicarbonate (HCO ₃)			47	23
Sulfate (SO ₄)			0.6	1.0
Chloride (Cl)			6.0	0.5
Nitrate (NO ₃)			0.5	0.0
Fluoride (F)			0.3	0.0
Boron (B)			0.0	0.0
Silica (SiO ₂)			17	24
Total dissolved solids in parts per million			88	45
Percent sodium			28	13
Hardness as CaCO ₃ in parts per million				
Total			36	20
Noncarbonate			7	0.3
Turbidity	(Not Measured)			
Coliforms in most probable number per milliliter	(Not Measured)			
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.41	17
Solid alpha			3.54	0.00
Dissolved beta			3.36	0.00
Solid beta			13.16	0.00

WATER QUALITY VARIATIONS



ANTELOPE CREEK NEAR TENNANT (STA. 1e)

BUTTE CREEK NEAR MACDOEL (STA. 1d)

Sampling Point Station 1d is located in Section 30 of Township 45 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank 7.5 miles downstream from Little Antelope Creek and 7 miles south of Macdoel.

Period of Record March 1959 through December 1959.

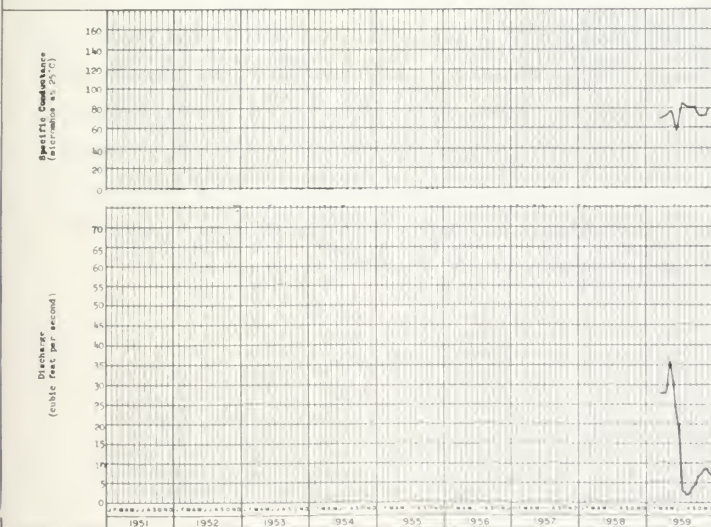
Water Quality Characteristics Samples of water from Butte Creek are a bicarbonate type with calcium and magnesium as major cations. This water is excellent in quality, class 1 for irrigation, soft, and has a mineral content within the limits for drinking water.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (micromhos at 25°C)	See p. 2	See p. 2	75	25
Temperature in °F			75	50
Dissolved oxygen in parts per million			12	5
Percent saturation				
pH			7.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)			150	50
Magnesium (Mg)			40	10
Sodium (Na)			150	10
Potassium (K)			20	5
Carbonate (CO ₃)			150	50
Bicarbonate (HCO ₃)			150	50
Sulfate (SO ₄)			150	50
Chloride (Cl)			150	50
Nitrate (NO ₃)			150	50
Fluoride (F)			150	50
Boron (B)			150	50
Silica (SiO ₂)			150	50
Total dissolved solids in parts per million			250	60
Percent sodium			25	10
Hardness as CaCO ₃ in parts per million				
Total			250	60
Noncarbonate			50	10
Turbidity: (Not Measured)				
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			75	25
Solid alpha			150	50
Dissolved beta			150	50
Solid beta			150	50

WATER QUALITY VARIATIONS



BUTTE CREEK NEAR MACDOEL (STA. 1d)

SHASTA RIVER NEAR YREKA (STA. 1a)

Sampling Point Station 1a is located in Section 24 of Township 46 North, Range 7 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the right bank 0.5 mile upstream from the mouth of the Shasta River and 7 miles north of Yreka.

Period of Record December 1958 through December 1959.

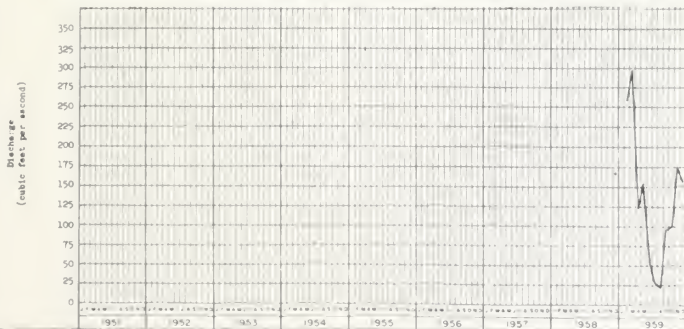
Water Quality Characteristics Since inception of a monitoring station on this river concentrations of mineral constituents, with the exception of boron, have been within the acceptable limits for nearly all beneficial uses. A good quality sodium-bicarbonate type water, moderate to very hard, is characteristic of samples from Shasta River. At times boron is detected in excess of 0.5 ppm, the upper limit for a class 1 irrigation water.

Significant Water Quality Changes During six months of 1959 boron exceeded 0.5 ppm in Shasta River. Boron in concentrations of this magnitude was generally characteristic of samples collected during late spring, the summer and early fall. The quantity of discharge in the river directly affected boron concentrations, with the lower flows being associated with higher concentrations. The source of boron in this river is believed to stem from irrigation return and mineralized spring waters.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See [unclear]	See [unclear]	500	50
Temperature in °F			6	4
Dissolved oxygen in parts per million			1.8	0.8
Percent saturation			100	20
pH			7.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)			40	24
Magnesium (Mg)			4.5	2.8
Sodium (Na)			28	28
Potassium (K)			1.4	0.1
Carbonate (CO ₃)			1	0.1
Bicarbonate (HCO ₃)			186	247
Sulfate (SO ₄)			23	4.8
Chloride (Cl)			21	1.8
Nitrate (NO ₃)			2.8	0.1
Fluoride (F)			0.4	0.1
Boron (B)			0.7	0.1
Silica (SiO ₂)			62	4
Total dissolved solids in parts per million			435	277
Percent sodium			30	24
Hardness as CaCO ₃ in parts per million				
Total			269	119
Noncarbonate			0.8	0.1
Turbidity			40	0
Coliform in most probable number per milliliter			17,000	1,000
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.30	0.00
Solid alpha			0.45	0.28
Dissolved beta			14.07	0.00
Solid beta			0.96	0.00

WATER QUALITY VARIATIONS



SHASTA RIVER NEAR YREKA (STA. 1a)

SCOTT RIVER NEAR FORT JONES (STA. 1b)

Sampling Point Scott River sampling station is located in Section 28 of Township 44 North, Range 10 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-depth, from the right bank, 150 feet south of the Fort Jones-Scotts Bar road, about 20 miles upstream from the mouth and 10.5 miles downstream from Fort Jones.

Period of Record December 1958 through December 1959.

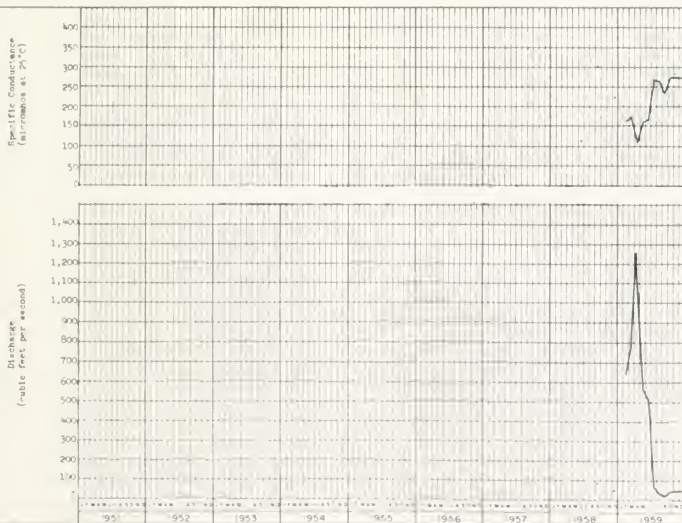
Water Quality Characteristics A review of analyses reveals Scott River to be excellent in quality, magnesium-calcium bicarbonate in character and class 1 for irrigation. It ranges from slightly to moderately hard, and does not exceed the drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	331	116	273	116
Temperature in °F	89	37	69	37
Dissolved oxygen in parts per million	12.5	8.2	12.5	8.2
Percent saturation	107	81	118	81
pH	8.1	7.3	8.1	7.3
Mineral constituents in parts per million				
Calcium (Ca)	33	30	31	30
Magnesium (Mg)	17	6.9	17	6.9
Sodium (Na)	5.2	1.8	5.2	1.8
Potassium (K)	2.2	0.1	2.2	0.1
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	179	70	171	70
Sulfate (SO ₄)	18	0.6	10	0.6
Chloride (Cl)	0.0	0.4	0.0	0.4
Nitrate (NO ₃)	1.2	0.3	2.2	0.3
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	22	14	21	14
Total dissolved solids in parts per million	190	74	176	74
Percent sodium	9	5	9	5
Hardness as CaCO ₃ in parts per million				
Total	150	56	147	59
Noncarbonate	0	0.0	8	0.0
Turbidity	See 1959	See 1959	30	10
Coliform in most probable number per milliliter	620	0.62	620	0.62
Radioactivity in micro-curies per liter	See 1959	See 1959		
Dissolved alpha			0.00	0.10
Solid alpha			0.45	0.00
Dissolved beta			1.73	0.00
Solid beta			1.63	0.00

WATER QUALITY VARIATIONS



SCOTT RIVER NEAR FORT JONES (STA. 1b)

SALMON RIVER AT SOMESBAR (STA. 2a)

Sampling Point Station 2a is located in Section 1 of Township 11 North, Range 6 East, Humboldt Base and Meridian. Monthly water samples were collected at mid-depth, from the right bank, 0.5 mile east of Somesbar post office and 3 miles upstream from the confluence with the Klamath River.

Period of Record November 1958 through December 1959.

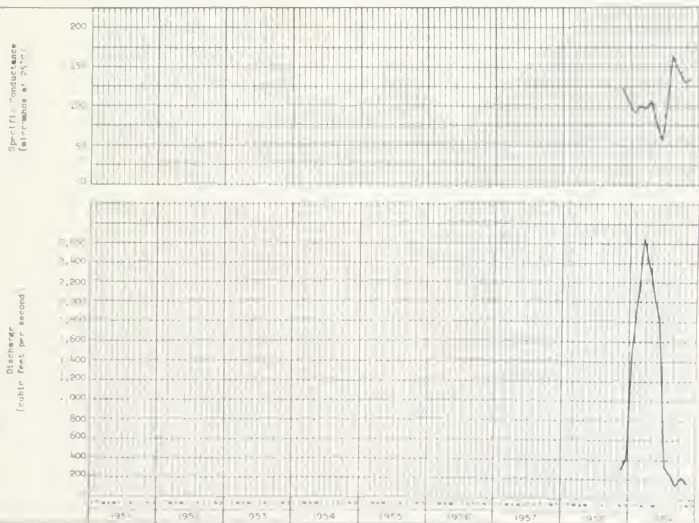
Water Quality Characteristics Antecedent data classify flow in Salmon River as excellent in quality, bicarbonate in character, soft to slightly hard, class 1 for irrigation and well within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	176	43.2	176	43.2
Temperature in °F	73	19	73	19
Dissolved oxygen in parts per million	11.1	8.1	11.1	8.1
Percent saturation	100	75	100	75
pH	8.1	7.3	8.1	7.3
Mineral constituents in parts per million				
Calcium (Ca)	150	40	150	40
Magnesium (Mg)	20	5.6	20	5.6
Sodium (Na)	10	1.5	10	1.5
Potassium (K)	1	0.2	1	0.2
Carbonate (CO ₃)	30	8	30	8
Bicarbonate (HCO ₃)	70	11	70	11
Sulfate (SO ₄)	10	1	10	1
Chloride (Cl)	5	1.5	5	1.5
Nitrate (NO ₃)	1	0.2	1	0.2
Fluoride (F)	0.5	0.05	0.5	0.05
Boron (B)	0.5	0.1	0.5	0.1
Silica (SiO ₂)	10	0.5	10	0.5
Total dissolved solids in parts per million	109	41	109	41
Percent sodium	24	7	24	7
Hardness as CaCO ₃ in parts per million				
Total	68	28	68	28
Noncarbonate	11	0.0	11	0.0
Turbidity			15	5
Coliform in most probable number per milliliter	2,500	0.06	2,500	0.06
Radioactivity in micro-curies per liter	See 1959	See 1959		
Dissolved alpha		0.10	0.10	0.00
Solid alpha		0.00	0.00	0.00
Dissolved beta		1.50	1.50	0.00
Solid beta		17.67	17.67	0.14

WATER QUALITY VARIATIONS



SALMON RIVER AT SOMESBAR (STA. 2a)

TRINITY RIVER AT LEWISTON (STA. 4a)

Sampling Point Station 4a is located in Section 19 of Township 33 North, Range 8 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-depth, from the left bank, at the USGS gaging station at Lewiston, and 0.8 mile downstream from Deadwood Creek.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Historical records at this station show the water to be excellent in quality, generally magnesium bicarbonate in character, class 1 for irrigation, with a range from soft to slightly hard, and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1955	Minimum 1955
Specific conductance (micromhos at 25°C)	180	60	180	60
Temperature in °F	85	60	75	60
Dissolved oxygen in parts per million	14.5	5.9	12.0	7.0
Percent saturation	100	70	100	70
pH	8.4	7.0	7.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	20	10	20	10
Magnesium (Mg)	15	5	15	5
Sodium (Na)	10	5	10	5
Potassium (K)	5	2	5	2
Carbonate (CO ₃)	10	5	10	5
Bicarbonate (HCO ₃)	77	30	77	30
Sulfate (SO ₄)	20	10	20	10
Chloride (Cl)	5	2	5	2
Nitrate (NO ₃)	5	2	5	2
Fluoride (F)	5	2	5	2
Boron (B)	5	2	5	2
Silica (SiO ₂)	20	10	20	10
Total dissolved solids in parts per million	100	40	125	40
Percent sodium	80	5	21	5
Hardness as CaCO ₃ in parts per million				
Total	88	27	180	40
Noncarbonate	5	5	7	5
Turbidity	1.00	0.00	60	0.00
Coliform in most probable number per milliliter	7,000	0	10	0
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.50	0.00	0.50	0.00
Solid alpha	0.50	0.00	0.50	0.00
Dissolved beta	0.50	0.00	0.50	0.00
Solid beta	16.77	0.00	0.27	0.00

WATER QUALITY VARIATIONS



TRINITY RIVER AT LEWISTON (STA. 4a)

TRINITY RIVER NEAR BURNT RANCH (STA. 4b)

Sampling Point The Burnt Ranch station is located in Section 19 of Township 5 North, Range 7 East, Humboldt Base and Meridian. Monthly grab samples were collected from mid-depth, from the left bank 500 feet upstream from Cedar Flat Creek, 700 feet upstream from Highway 299 bridge at Cedar Flat, and 2.3 miles southeast of the town of Burnt Ranch.

Period of Record April 1958 through December 1959.

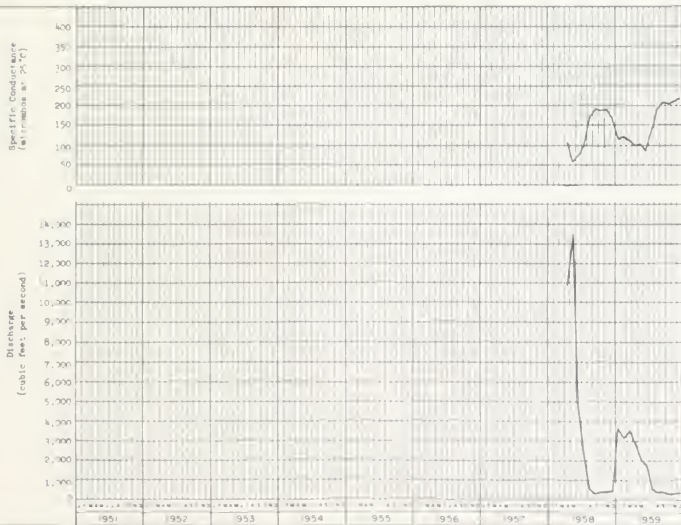
Water Quality Characteristics Past water analyses from Station 4b show the water to be excellent in quality, calcium-magnesium bicarbonate in character, class 1 for irrigation, soft to slightly hard, and within drinking water standards for mineral content. A study of analyses of Trinity River waters reveals a slight increase in mineral concentrations, on the order of 30 micromhos, occurs between Station 4a (Lewiston) and 4b.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1954	Minimum - 1959
Specific conductance (micromhos at 25°C)	850	70	300	20
Temperature in °F	80	40	70	40
Dissolved oxygen in parts per million	20.0	6.0	12.0	4.0
Percent saturation	100	40	100	40
pH	8.1	7.2	7.6	7.1
Mineral constituents in parts per million				
Calcium (Ca)	90	10	20	10
Magnesium (Mg)	10	0.5	10	0.5
Sodium (Na)	10	0.5	10	0.5
Potassium (K)	10	0.5	10	0.5
Carbonate (CO ₃)	10	0.5	10	0.5
Bicarbonate (HCO ₃)	10	0.5	10	0.5
Sulfate (SO ₄)	10	0.5	10	0.5
Chloride (Cl)	10	0.5	10	0.5
Nitrate (NO ₃)	10	0.5	10	0.5
Fluoride (F)	10	0.5	10	0.5
Boron (B)	10	0.5	10	0.5
Silica (SiO ₂)	10	0.5	10	0.5
Total dissolved solids in parts per million	100	40	100	40
Percent sodium	20	0	20	0
Hardness as CaCO ₃ in parts per million				
Total	100	30	100	30
Noncarbonate	10	0	10	0
Turbidity	See 1959	See 1959	See 1959	See 1959
Coliform in most probable number per milliliter (Not Measured)	See 1959	See 1959	See 1959	See 1959
Radioactivity in micro-micro curies per liter	See 1959	See 1959	See 1959	See 1959
Dissolved alpha			0.01	0.01
Solid alpha			0.00	0.00
Dissolved beta			15.56	4.14
Solid beta			6.09	5.04

WATER QUALITY VARIATIONS



TRINITY RIVER NEAR BURNT RANCH (STA. 4b)

TRINITY RIVER NEAR HOOPA (STA. 4)

Sampling Point Station 4 is located in Section 31 of Township 8 North, Range 5 East, Humboldt Base and Meridian. Monthly water samples were collected from the left bank at the USGS gage 2 miles southeast of Hoopa, 0.5 mile downstream from Campbell Creek on the Hoopa Indian Reservation, and 12 miles upstream from its confluence with Klamath River.

Period of Record April 1951 through December 1959.

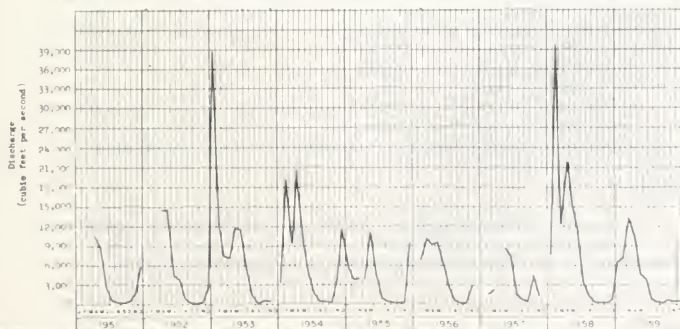
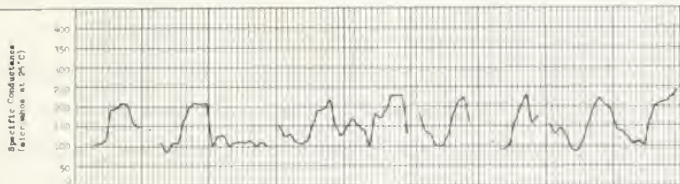
Water Quality Characteristics Trinity River at Station 4 has historically been excellent in quality, magnesium-calcium bicarbonate in character, class 1 for irrigation, soft to moderately hard, and within drinking water standards for mineral content. Past records indicate a gradual increase of about 15 micromhos in the concentration of most dissolved minerals between Station 4b, Burnt Ranch, and Station 4.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	241	84	241	100
Temperature in °F	80	41	76	41
Dissolved oxygen in parts per million	14	7.7	12.6	7.6
Percent saturation	128	66	100	88
pH	10.0	5.2	8.1	7.4
Mineral constituents in parts per million				
Calcium (Ca)	26	7	18	12
Magnesium (Mg)	15	2.1	15	7.3
Sodium (Na)	8.4	1.4	8.4	2.8
Potassium (K)	1.6	0.1	1.1	0.1
Carbonate (CO ₃)	2	0.0	2	0.0
Bicarbonate (HCO ₃)	126	47	126	68
Sulfate (SO ₄)	12	0.6	11	4.8
Chloride (Cl)	12	0.0	12	2.0
Nitrate (NO ₃)	1.2	0.0	0.5	0.2
Fluoride (F)	0.3	0.0	0.1	0.0
Boron (B)	0.21	0.0	0.2	0.0
Silica (SiO ₂)	21	11	15	14
Total dissolved solids in parts per million	148	56	148	64
Percent sodium	18	6	13	7
Hardness as CaCO ₃ in parts per million				
Total	120	40	120	50
Noncarbonate	17	0.0	17	1
Turbidity	180	0.2	70	1
Coliform in most probable number per milliliter	7,000	<0.045	7,000	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.02	0.00	0.02	0.00
Solid alpha	0.54	0.00	0.54	0.00
Dissolved beta	36.46	0.00	1.09	0.11
Solid beta	2.73	0.00	2.73	0.00

WATER QUALITY VARIATIONS



TRINITY RIVER NEAR HOOPA (STA. 4)

Smith River Basin

The California portion of the Smith River Basin occupies approximately 780 square miles in the extreme northwest portion of the North Coastal Region. The major portion of the area is drained by the Smith River whose Middle and South Forks originate on the western slope of the Siskiyou Mountains, and whose North Fork has its headwaters in Curry County, Oregon. The basin is bounded by the Pacific Ocean on the west, the California-Oregon state line to the north, the Del Norte-Siskiyou County line to the east, and the Klamath River watershed divide to the south.

Topography of the area is generally mountainous though interrupted with numerous steep-walled canyons and stream valleys. Elevation varies from sea level to heights of over 6,000 feet. Total average annual runoff in the Smith River Basin is on the order of 2,900,000 acre-feet.

Rough but relatively low mountains cover approximately 95 percent of this river unit. The Smith River Plain which lies along the coast covers about 50 square miles of agriculturally adaptable land. Logging and forest products constitute the largest source of income, followed in order of their importance by agriculture (dairying and bulb raising), mineral production, recreation and commercial fishing.

Waste discharges constitute only a minor source of inflow to the Smith River watershed and have not created a water impairment problem.

A surface water sampling station is maintained on Smith River near Crescent City to monitor quality of runoff from this basin.

SMITH RIVER NEAR CRESCENT CITY (STA. 3a)

Sampling Point Station 3a is located in Section 10 of Township 16 North, Range 1 East, Humboldt Base and Meridian. Monthly grab samples were collected from the left bank at the USGS gage, 8 miles east of Crescent City, 0.5 mile downstream from the south fork of the Smith River, and about 12 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

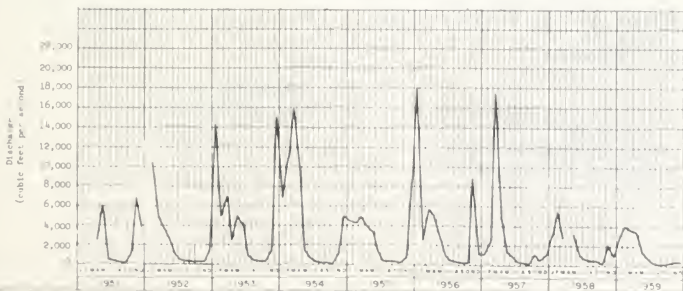
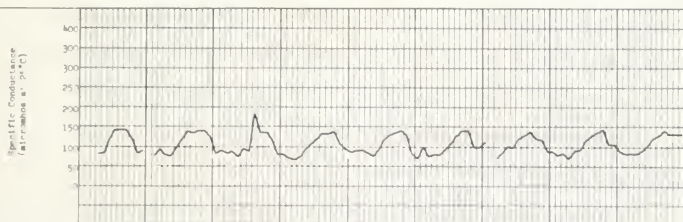
Water Quality Characteristics Past analyses of this water have shown it to be excellent in quality, magnesium bicarbonate in character, class 1 for irrigation, and soft to slightly hard. It has consistently met drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (microhos at 25°C)	147	64	141	77.7
Temperature in °F	78	19	70	41
Dissolved oxygen in parts per million	14.1	6.4	12.4	8.6
Percent saturation	114	59	103	89
pH	8.6	6.3	7.3	7.2
Mineral constituents in parts per million				
Calcium (Ca)	11	2.6	2.6	1.2
Magnesium (Mg)	13	1.4	1.1	0.6
Sodium (Na)	6.4	1.0	1.4	1.2
Potassium (K)	7	0.6	0.4	0.2
Carbonate (CO ₃)	8	0.6	0.0	0.2
Bicarbonate (HCO ₃)	80	10	80	41
Sulfate (SO ₄)	7.7	1.4	4.0	4.8
Chloride (Cl)	8	0.6	4.8	2.2
Nitrate (NO ₃)	1.3	0.6	0.1	0.6
Fluoride (F)	0.2	0.6	0.1	0.6
Boron (B)	0.18	0.6	0.1	0.6
Silice (SiO ₂)	27	11	13	1.2
Total dissolved solids in parts per million	94	41	94	40
Percent sodium	20	6	12	6
Hardness as CaCO ₃ in parts per million				
Total	76	12	76	18
Noncarbonate	11	0.6	11	1
Turbidity	53	0.3	20	1
Coliform in most probable number per milliliter	230	0.044	230	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	8.2	0.00	0.20	0.18
Solid alpha	9	0.00	0.54	0.20
Dissolved beta	10.00	0.00	1.87	0.46
Solid beta	23.27	0.00	2.67	2.48

WATER QUALITY VARIATIONS



SMITH RIVER NEAR CRESCENT CITY (STA. 3a)

Redwood Creek and Mad River Unit

Mad River is a large stream, draining a total of 496 square miles in Humboldt and Trinity Counties. Redwood Creek drains an area of about 279 square miles, north of Mad River Basin in Humboldt County. Both of these streams enter the Pacific Ocean and estimated mean annual runoffs of Mad River and Redwood Creek are 925,500 acre-feet and 823,500 acre-feet, respectively. Like other streams in the North Coastal Region, precipitation and runoff are high during the winter months and generally quite low in the late summer and fall.

In both of these stream basins a total of only 21 square miles is classed as valley and mesa land, the remaining area being a rugged mountainous terrain. Lumbering activities comprise the major users of surface waters in these basins; however, water is diverted from Mad River for use as a municipal supply for the communities of Arcata and Eureka. Both of these streams support runs of anadromous fish and resident trout and are considered to have significant value as recreational areas.

Waste discharges entering these streams are insignificant and do not cause an impairment problem.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Redwood Creek at Orick	44
Mad River at Arcata	46

REDWOOD CREEK AT ORICK (STA. 3b)

Sampling Point Redwood Creek sampling station is located in Section 4 of Township 10 North, Range 1 East, Humboldt Base and Meridian. Monthly grab samples were collected from the left bank on the downstream side of the U. S. Highway 101 bridge at Orick and about 2 miles upstream from the mouth.

Period of Record November 1958 through December 1959.

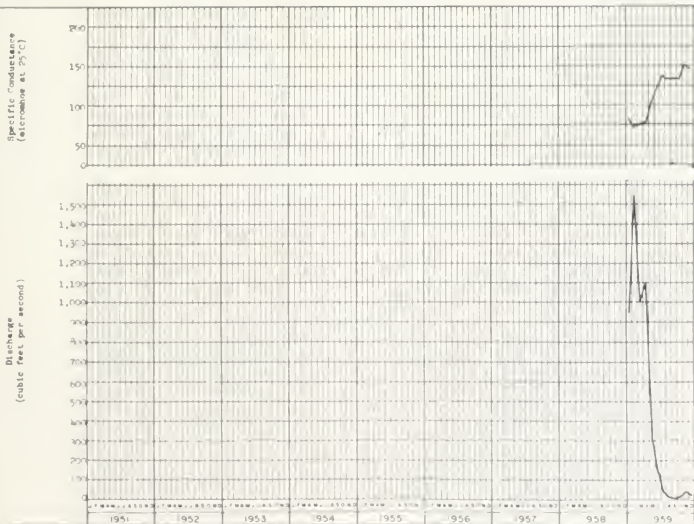
Water Quality Characteristics Past analyses show the water at Station 3b to be excellent in quality, calcium bicarbonate in character, class 1 for irrigation and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhos at 25°C)	150	77	150	77
Temperature in °F	88	47	88	47
Dissolved oxygen in parts per million	11.7	7.7	11.7	7.7
Percent saturation	100	77	100	77
pH	7.8	6.9	7.8	6.9
Mineral constituents in parts per million				
Calcium (Ca)	22	8.8	22	8.8
Magnesium (Mg)	4.3	1.6	4.3	1.6
Sodium (Na)	7.3	2.3	7.3	2.3
Potassium (K)	1.0	0.8	1.0	0.8
Carbonates (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonates (HCO ₃)	7.1	1.0	7.1	1.0
Sulfate (SO ₄)	1.1	1.0	1.1	1.0
Chloride (Cl)	1.8	1.2	1.8	1.2
Nitrate (NO ₃)	1.0	0.0	1.0	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	12	4.7	12	4.7
Total dissolved solids in parts per million	78	42	78	42
Percent sodium	21	9	21	9
Hardness as CaCO ₃ in parts per million				
Total	68	29	68	29
Noncarbonate	16	0.0	16	0.0
Turbidity	See 1959	See 1959	88	1
Coliform in most probable number per milliliter	7,000	0.62	7,000	0.62
Radioactivity in micro-curies per liter	See 1959	See 1959		
Dissolved alpha			0.20	0.00
Solid alpha			0.27	0.00
Dissolved beta			0.00	0.00
Solid beta			4.51	1.88

WATER QUALITY VARIATIONS



REDWOOD CREEK AT ORICK (STA. 3b)

MAD RIVER NEAR ARCATA (STA. 6a)

Sampling Point Station 6a is located in Section 15 of Township 6 North, Range 1 East, Humboldt Base and Meridian. Monthly water samples are collected in center of stream from Highway 299 bridge, about 4.5 miles upstream from the mouth, and 3 miles northeast of Arcata.

Period of Record November 1958 through December 1959.

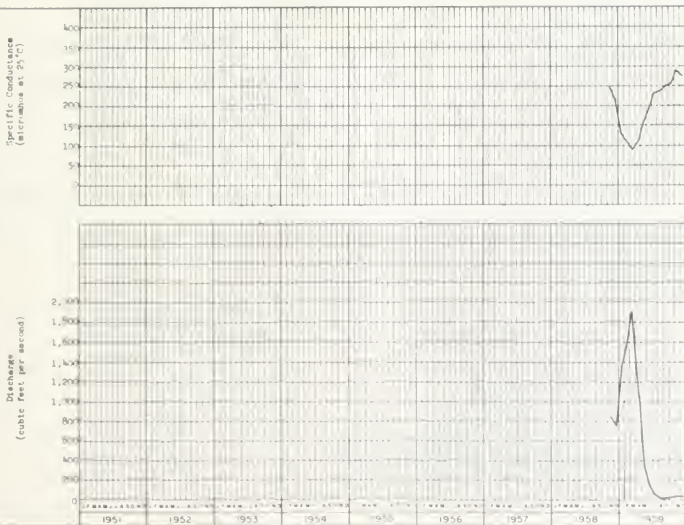
Water Quality Characteristics Runoff in Mad River is excellent in quality, calcium bicarbonate in character, class 1 for irrigation, soft to moderately hard, and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum	Minimum
Specific conductance (micromhos at 25°C)	100	25.1	100	25.1
Temperature in °F	79	48	79	48
Dissolved oxygen in parts per million	12.5	2.8	12.5	2.8
Percent saturation	100	20	100	20
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	15	10	15	10
Magnesium (Mg)	10	5	10	5
Sodium (Na)	10	5	10	5
Potassium (K)	10	5	10	5
Carbonate (CO ₃)	10	5	10	5
Bicarbonate (HCO ₃)	10	5	10	5
Sulfate (SO ₄)	10	5	10	5
Chloride (Cl)	10	5	10	5
Nitrate (NO ₃)	10	5	10	5
Fluoride (F)	10	5	10	5
Boron (B)	10	5	10	5
Silica (SiO ₂)	10	5	10	5
Total dissolved solids in parts per million	174	64	174	64
Percent sodium	12	7	12	7
Hardness as CaCO ₃ in parts per million				
Total	136	40	136	40
Noncarbonate	16	14	16	14
Turbidity	See 1950	See 1950	See 1950	See 1950
Coliforms in most probable number per milliliter	7,000	100	7,000	100
Radioactivity in micro-micro curies per liter	See 1950	See 1950	See 1950	See 1950
Dissolved alpha			0.00	0.00
Solid alpha			0.00	0.00
Dissolved beta			0.01	0.00
Solid beta			0.02	0.00

WATER QUALITY VARIATIONS



MAD RIVER NEAR ARCATA (STA. 6a)

Eel River Basin

The Eel River watershed traverses the south-central portion of the North Coastal Region. It drains an area of 3,701 square miles, of which 3,574 square miles are rugged mountains, scarred by numerous landslides and narrow, steep stream canyons. Several small river terraces and a broad coastal plain constitute the remaining 127 square miles in the basin. The Eel River has an average annual discharge of about 6,273,000 acre-feet.

Eel River water is used for irrigation, power development, industry, recreation, and public and domestic water supplies. Except for power diversions which discharge to Russian River Basin, these users divert extremely small quantities and the abundant water resources of this basin are largely undeveloped. Lumber by-product industries and irrigation are considered the most probable future users of significant quantities of water within the basin.

Waste discharges and irrigation return entering the Eel River at the present time are small in quantity and do not significantly impair the receiving waters.

The following tabulation presents the names of stations maintained to monitor surface water quality in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Eel River near Dos Rios	50
Eel River near McCann	52
Eel River at Scotia	54
Outlet Creek near Longvale	56
Eel River, Middle Fork at Dos Rios	58
Eel River, South Fork near Miranda	60
Van Duzen River near Bridgeville	62

EEL RIVER NEAR DOS RIOS (STA. 5d)

Sampling Point Station 5d is located in Section 31 of Township 21 North, Range 13 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected in the center of the channel from the highway bridge 250 feet upstream from the confluence of Outlet Creek, 7.5 miles northeast of Longvale and 8.5 miles south of Dos Rios.

Period of Record April 1958 through December 1959.

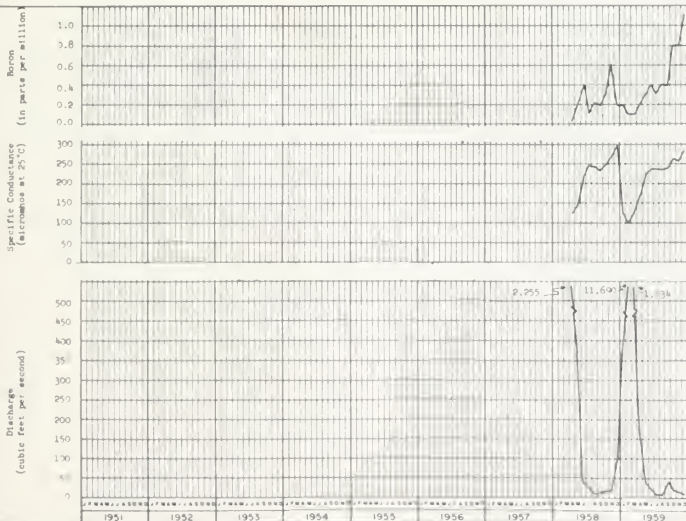
Water Quality Characteristics Past analyses identify the water at Station 5d as good in quality, calcium bicarbonate in character, soft to moderately hard, and within drinking water standards for mineral content. At times boron concentrations in excess of 0.5 ppm are found, placing this water in class 2 for irrigation. Although the source of boron in this river has not yet been identified, evidence indicates the boron originates from geologic formations existing in the watershed upstream from and in the vicinity of this station. Runoff from numerous mineralized springs, probably of deep-seated origin, also enter the waterway of Eel River upstream from this station. It has been established that springs high in boron exist throughout much of the Clear Lake area which is coterminous with the upper watershed of this basin.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	36	10	30	10
Temperature in °F	8	4	8	4
Dissolved oxygen in parts per million	11	5.1	11	5.1
Percent saturation	11	5.1	11	5.1
pH	10	7.5	10	7.5
Mineral constituents in parts per million				
Calcium (Ca)	12	2.9	12	2.9
Magnesium (Mg)	3.9	1.1	3.9	1.1
Sodium (Na)	5.1	1.1	5.1	1.1
Potassium (K)	7	1.1	7	1.1
Carbonate (CO ₃)	1.1	1.1	1.1	1.1
Bicarbonate (HCO ₃)	1.1	1.1	1.1	1.1
Sulfate (SO ₄)	1.1	1.1	1.1	1.1
Chloride (Cl)	1.1	1.1	1.1	1.1
Nitrate (NO ₃)	1.1	1.1	1.1	1.1
Fluoride (F)	1.1	1.1	1.1	1.1
Boron (B)	1.1	1.1	1.1	1.1
Silica (SiO ₂)	1.1	1.1	1.1	1.1
Total dissolved solids in parts per million	169	71	168	71
Percent sodium	11	11	11	11
Hardness as CaCO ₃ in parts per million				
Total	136	46	136	46
Noncarbonate	16	16	16	16
Turbidity	1	1	1	1
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



EEL RIVER NEAR DOS RIOS (STA. 5d)

EEL RIVER NEAR MCCANN (STA. 5)

Sampling Point The McCann station is located in Section 3 of Township 2 South, Range 3 East, Humboldt Base and Meridian. Monthly water samples were collected from the center of the channel, from the McCann Bridge 46.5 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Eel River water near McCann is calcium bicarbonate in character, class 1 for irrigation uses, with a range from soft to moderately hard, and consistently meets drinking water standards for mineral content. An increase in conductivity, averaging about 25 micromhos, usually occurs from Stations 5d to 5. Boron, however, decreases significantly from the upstream station to the downstream station. Boron concentrations at Station 5 range from 0 to 0.3 ppm. This decrease in boron between the two stations is attributable to low boron content waters tributary to the Eel River between the two stations.

Significant Water Quality Changes During 1959, samples of water from Station 5 ranged from slightly to very hard, reaching the maximum for the period of record in December when 204 ppm hardness was reported. The lack of dilution waters during the extremely low flow period, occurring during the latter part of the year, was probably the cause of the high concentrations of hardness found at this station.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micro-mhos at 25°C)	199	101	199	114
Temperature in °F	80	42	71	44
Dissolved oxygen in parts per million	15.4	6.6	11.8	8.6
Percent saturation	150	54	109	59
pH	8.6	6.8	7.8	7.3
Mineral constituents in parts per million				
Calcium (Ca)	40	9.8	37	25
Magnesium (Mg)	11	2.9	9.6	5.1
Sodium (Na)	16	2.6	11	2.9
Potassium (K)	2.7	0.5	1.4	0.5
Carbonate (CO ₃)	4	0.0	4	0.0
Bicarbonate (HCO ₃)	230	53	230	70
Sulfate (SO ₄)	26	7.7	22	12
Chloride (Cl)	20	10.0	13	2.2
Nitrate (NO ₃)	0.7	0.0	0.7	0.1
Fluoride (F)	0.3	0.0	0.1	0.0
Boron (B)	0.30	0.0	0.30	0.0
Silica (SiO ₂)	14	7	10	7
Total dissolved solids in parts per million	214	54	214	72
Percent sodium	26	9	15	8
Hardness as CaCO ₃ in parts per million				
Total	204	41	204	65
Noncarbonate	25	0.0	25	7
Turbidity	1,100	0.0	85	7
Coliform in most probable number per milliliter	>7,000	<0.045	23	0.046
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.43	0.00	0.72	0.00
Solid alpha	0.59	0.00	0.00	0.00
Dissolved beta	45.1	0.00	2.08	0.00
Solid beta	14.8	3.30	1.22	1.58

WATER QUALITY VARIATIONS



EEL RIVER NEAR McCANN (STA. 5)

EEL RIVER AT SCOTIA (STA. 6)

Sampling Point The station on Eel River at Scotia is located in Section 5 of Township 1 North, Range 1 East, Humboldt Base and Meridian. Monthly grab samples were collected from the left bank approximately 0.6 mile downstream from Highway 101 bridge between Scotia and Rio Dell at the foot of Painter Street, and about 12 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

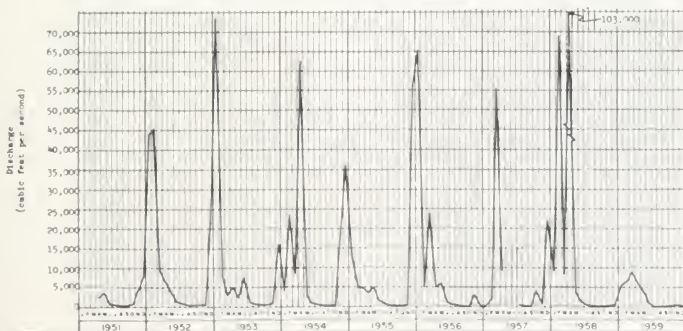
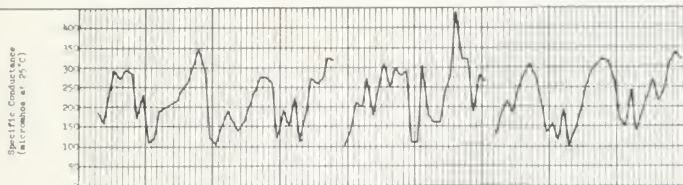
Water Quality Characteristics Antecedent data reveal the water at Station 6 to be excellent in quality, generally calcium bicarbonate in character, soft to moderately hard, and within the recommended limits for mineral content in drinking water. Only minor increases in mineral content occur between this station and Station 5 and on occasions, when tributary inflow rates are high, the mineral content of Eel River has decreased slightly in this reach.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum, 1959	Minimum, 1959
Specific conductance (micromhos at 25°C)	443	87	303	133
Temperature in °F	88	48	86	37
Dissolved oxygen in parts per million	10.8	6.3	10.3	6.0
Percent saturation	100	50	100	50
pH	8.8	7.1	8.3	7.0
Mineral constituents in parts per million				
Calcium (Ca)	98	38	94	28
Magnesium (Mg)	28	8	23	8
Sodium (Na)	10	3	11	4
Potassium (K)	1.5	.6	1.4	.4
Carbonate (CO ₃)	4	1	4	1
Bicarbonate (HCO ₃)	280	80	180	70
Sulfate (SO ₄)	11	3	15	12
Chloride (Cl)	20	5	15	3
Nitrate (NO ₃)	0	0	1.5	0
Fluoride (F)	0	0	0.0	0
Boron (B)	1.0	0.0	1.0	0.0
Silica (SiO ₂)	15	4	10	4
Total dissolved solids in parts per million	254	67	186	69
Percent sodium	26	8	20	11
Hardness as CaCO ₃ in parts per million				
Total	212	63	140	66
Noncarbonate	83	10	23	4
Turbidity	1,100	50	60	1
Coliform in most probable number per milliliter	>7,000	10	600	10
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.73	0.00	0.04	0.00
Solid alpha	1.22	0.00	0.13	0.17
Dissolved beta	17.30	0.00	7.00	5.15
Solid beta	13.90	0.00	3.04	0.00

WATER QUALITY VARIATIONS



EEL RIVER AT SCOTIA (STA. 6)

OUTLET CREEK NEAR LONGVALE (STA. 5b)

Sampling Point The station is located in Section 31 of Township 21 North, Range 13 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, 300 feet downstream from the railroad bridge, 200 feet upstream from the confluence with the Eel River, 7.5 miles northeast of Longvale and 8.5 miles south of Dos Rios.

Period of Record May 1958 through December 1959.

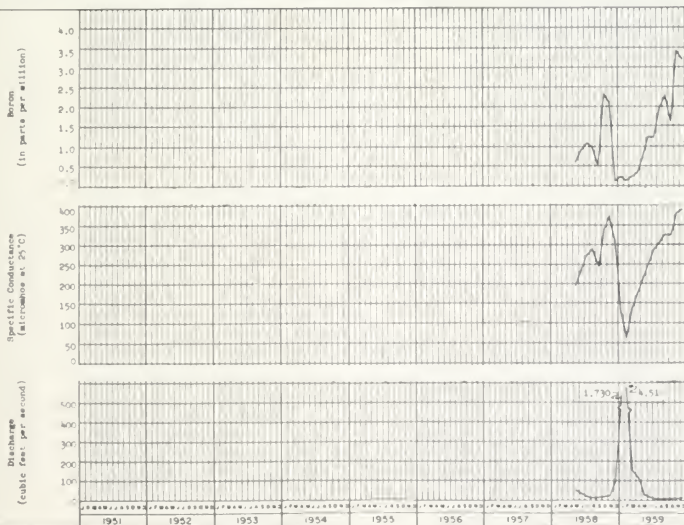
Water Quality Characteristics Since inception of a monitoring station on Outlet Creek, waters have been calcium bicarbonate in character, slightly to moderately hard and within drinking water standards for mineral content. The water in Outlet Creek, because of boron concentrations, ranges from class 1 to class 3 for irrigation use. Boron usually exceeds 0.5 ppm and periodically reaches values in excess of 2.0 ppm.

Significant Water Quality Changes During 1959, boron concentrations in Outlet Creek reached a maximum of 3.4 ppm in November and were sufficiently high during the last eight months of the year to cause the water to be class 2 or 3 for irrigation. The high concentrations of boron is attributed to the lack of dilution waters in Outlet Creek. The source of boron degradation to Outlet Creek has not as yet been ascertained. However, it is believed that the source, as in the headwaters of the Eel River, is geologic formations and springs.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	100	10	100	10
Temperature in °F	70	40	70	40
Dissolved oxygen in parts per million	12.8	6.0	12.8	6.0
Percent saturation	110	50	110	50
pH	8.1	7.1	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	100	20	100	20
Magnesium (Mg)	20	10	20	10
Sodium (Na)	100	10	100	10
Potassium (K)	2.5	1.0	2.5	1.0
Carbonate (CO ₃)	100	10	100	10
Bicarbonate (HCO ₃)	100	10	100	10
Sulfate (SO ₄)	100	10	100	10
Chloride (Cl)	100	10	100	10
Nitrate (NO ₃)	100	10	100	10
Fluoride (F)	100	10	100	10
Boron (B)	100	10	100	10
Silica (SiO ₂)	100	10	100	10
Total dissolved solids in parts per million	234	40	234	40
Percent sodium	81	24	81	24
Hardness as CaCO ₃ in parts per million				
Total	100	20	100	20
Noncarbonate	100	20	100	20
Turbidity	40	10	40	10
Coliforms in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



OUTLET CREEK NEAR LONGVALE (STA. 5b)

EEL RIVER, MIDDLE FORK AT DOS RIOS (STA. 5c)

Sampling Point Station 5c is located in Section 6 of Township 21 North, Range 13 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the channel from the highway bridge 0.5 mile southeast of Dos Rios and 0.2 mile upstream from the confluence with Eel River.

Period of Record April 1958 through December 1959.

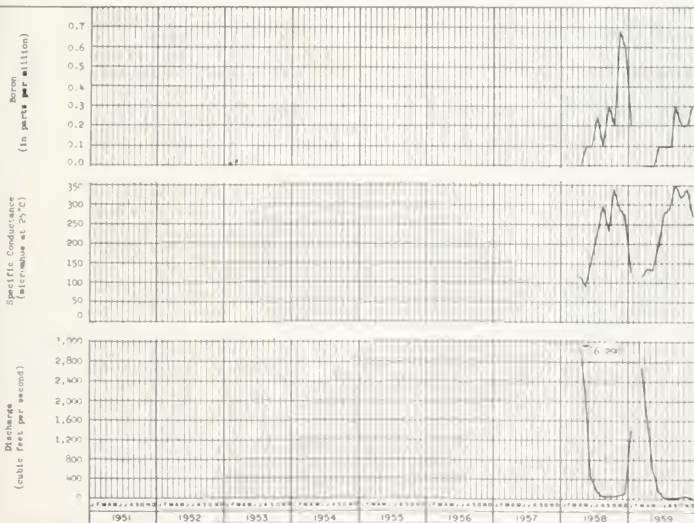
Water Quality Characteristics A review of analyses of samples reveals water at this station to be calcium bicarbonate in character, soft to moderately hard, and to consistently meet drinking water standards for mineral content. Boron concentrations at times place this water in class 2 for irrigation.

Significant Water Quality Changes In respect to boron, waters at this station were class 1 during the entire 1959 year.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	374	374	374	100
Temperature in °F	72	40	72	40
Dissolved oxygen in parts per million	11.5	4.2	11.5	7.4
Percent saturation	100	70	100	70
pH	8.2	7.0	8.2	7.0
Mineral constituents in parts per million				
Calcium (Ca)	40	10	40	10
Magnesium (Mg)	20	5	20	5
Sodium (Na)	10	2	10	2
Potassium (K)	1.0	0.2	1.0	0.2
Carbonate (CO ₃)	1.0	0.2	1.0	0.2
Bicarbonate (HCO ₃)	1.0	0.2	1.0	0.2
Sulfate (SO ₄)	1.0	0.2	1.0	0.2
Chloride (Cl)	1.0	0.2	1.0	0.2
Nitrate (NO ₃)	1.0	0.2	1.0	0.2
Fluoride (F)	1.0	0.2	1.0	0.2
Boron (B)	10	2	10	2
Silica (SiO ₂)	10	2	10	2
Total dissolved solids in parts per million	207	54	207	56
Percent sodium	21	7	21	7
Hardness as CaCO ₃ in parts per million				
Total	207	54	207	56
Noncarbonate	207	54	207	56
Turbidity	10	1	10	1
Coliform in most probable number per milliliter (No. Measured)				
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



EEL RIVER, MIDDLE FORK AT DOS RIOS (STA. 5c)

EEL RIVER, SOUTH FORK NEAR MIRANDA (STA. 7)

Sampling Point Station 7 is located in Section 30 of Township 3 South, Range 4 East, Humboldt Base and Meridian. Monthly water samples were collected from the right bank, at the USGS gage at Sylvandale camp grounds on U. S. Highway 101, 6 miles south of Miranda and about 12 miles upstream from the confluence with Eel River.

Period of Record April 1951 through December 1959.

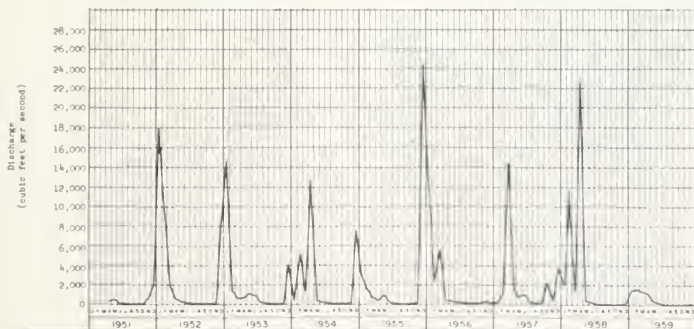
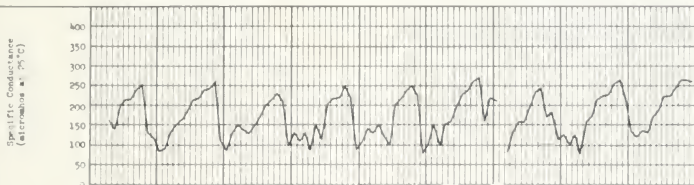
Water Quality Characteristics Past records show South Fork Eel River water to be excellent in quality, calcium bicarbonate in character, class 1 for irrigation, soft to moderately hard, and within the recommended limits for minerals in drinking water.

Significant Water Quality Changes Boron reached 0.5 ppm, the upper limit for class 1 irrigation water, in December 1959. Boron possibly occurs in higher concentrations in the upstream reaches of the South Fork since this fork of the Eel River originates in geological formations which are known to contribute boron to surface runoff.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	270	75	200	100
Temperature in °F	70	45	75	35
Dissolved oxygen in parts per million	15.0	2.0	11.0	2.0
Percent saturation	100	20	100	20
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	60	10	50	10
Magnesium (Mg)	20	5	15	5
Sodium (Na)	25	5	20	5
Potassium (K)	5	1	4	1
Carbonate (CO ₃)	5	1	4	1
Bicarbonate (HCO ₃)	10	2	8	2
Sulfate (SO ₄)	0.6	0.2	0.4	0.2
Chloride (Cl)	0.5	0.1	0.4	0.1
Nitrate (NO ₃)	0.5	0.1	0.4	0.1
Fluoride (F)	0.5	0.1	0.4	0.1
Boron (B)	5	1	4	1
Silice (SiO ₂)	10	5	5	5
Total dissolved solids in parts per million	100	40	60	30
Percent sodium	25	5	20	5
Hardness as CaCO ₃ in parts per million				
Total	100	40	60	30
Noncarbonate	10	5	5	5
Turbidity	1,000	50	950	40
Coliform in most probable number per milliliter	2,000	100	1,900	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.20	0.00	1.20	0.00
Solid alpha	0.70	0.00	0.70	0.00
Dissolved beta	10.5	0.00	10.5	0.00
Solid beta	13.5	0.00	13.5	0.00

WATER QUALITY VARIATIONS



EEL RIVER, SOUTH FORK NEAR MIRANDA (STA. 7)

VAN DUZEN RIVER NEAR BRIDGEVILLE (STA. 5a)

Sampling Point The station is located in Section 17 of Township 1 North, Range 3 East, Humboldt Base and Meridian. Monthly water samples were collected at the USGS gage, from the center of the channel from the bridge on Highway 36, 0.3 mile downstream from Pip Creek, 0.5 mile upstream from Rogers Creek, 4 miles west of Bridgeville and about 20 miles upstream from its confluence with Eel River.

Period of Record April 1958 through December 1959.

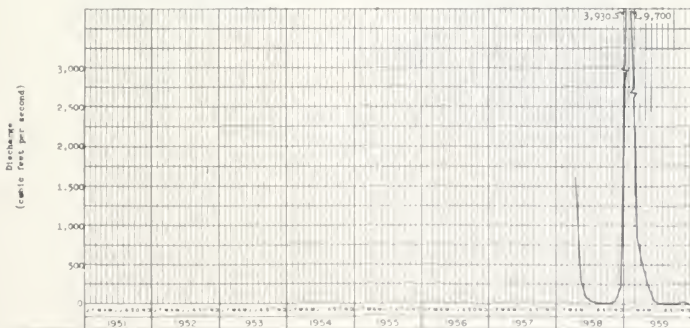
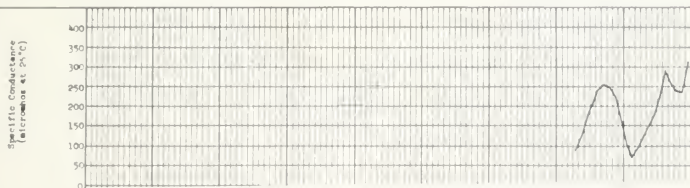
Water Quality Characteristics Water at Station 5a is calcium bicarbonate in character, class 1 for irrigation, ranging from soft to moderately hard. It meets drinking water standards for mineral content. The quality of this water does not differ significantly from the quality of Eel River water at Scotia.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhos at 25°C)	311	71.7	311	71.7
Temperature in °F	51	43	51	43
Dissolved oxygen in parts per million	11.8	7.6	11.8	7.6
Percent saturation	103	81	103	80
pH	8.1	7.7	8.1	7.7
Mineral constituents in parts per million				
Calcium (Ca)	93	5.8	93	5.8
Magnesium (Mg)	35	1.1	35	1.1
Sodium (Na)	31	3.7	31	3.7
Potassium (K)	1.5	0.3	1.5	0.3
Carbonate (CO ₃)	5	0.5	5	0.5
Bicarbonate (HCO ₃)	180	17	180	17
Sulfate (SO ₄)	5	1.4	5	1.4
Chloride (Cl)	1.5	0.5	1.5	0.5
Nitrate (NO ₃)	1.8	0.8	1.8	0.8
Fluoride (F)	2.5	0.5	2.5	0.5
Boron (B)	2	0.5	2	0.5
Silica (SiO ₂)	1.2	0.2	1.2	0.2
Total dissolved solids in parts per million	207	47	207	47
Percent sodium	81	5	81	5
Hardness as CaCO ₃ in parts per million				
Total	180	20	180	20
Noncarbonate	25	0.5	25	0.5
Turbidity	See 1959	See 1959	15	1
Coliform in most probable number per milliliter	See 1959	See 1959	60	15 (14)
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.41	0.006
Solid alpha			3.63	0.19
Dissolved beta			7.38	0.87
Solid beta			1.8	1.20

WATER QUALITY VARIATIONS



VAN DUZEN RIVER NEAR BRIDGEVILLE (STA. 5a)

Mattole River-Mendocino Coast Unit

The unit is comprised of several noncontiguous watersheds draining the south coastal portion of Region 1 and includes the following rivers: Mattole, Noyo, Big, Navarro and Gualala. These rivers drain approximately 1,290 square miles of predominately mountainous coast land with less than one percent of the area being valley and mesa lands. The combined annual mean seasonal runoff of these rivers is estimated to exceed 2,430,000 acre-feet.

Present development in this area is dependent on the lumber industry and to a limited extent on stock raising. Water development is largely on an individual basis with a few small public agencies formed to develop and distribute domestic and municipal supplies. Waste discharges from lumber industries and small communities have not created any significant water quality impairment problems in these basins.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this unit and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Mattole River near Petrolia	66
Noyo River near Fort Bragg	68
Big River near mouth	70
Navarro River near Navarro	72
Gualala River, South Fork near Annapolis	74

MATTOLE RIVER NEAR PETROLIA (STA. 7a)

Sampling Point Station 7a is located in Section 11 of Township 2 South, Range 2 West, Humboldt Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage 0.2 mile downstream from Clear Creek, 1.3 miles upstream from North Fork, 1.2 miles southeast of Petrolia, Humboldt County, and about 5 miles upstream from the mouth.

Period of Record January 1959 through December 1959.

Water Quality Characteristics Water at Station 7a is calcium bicarbonate in character, excellent in quality, class 1 for irrigation, soft to moderately hard, and within mineral standards for drinking water.

Significant Water Quality Changes Radioactivity decreased from 15.7 $\mu\text{c}/\text{l}$ total activity in May to 0.3 $\mu\text{c}/\text{l}$ in September.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1273	See 1259	200	12
Temperature in °F			76	47
Dissolved oxygen in parts per million			12.7	8.0
Percent saturation			16	80
pH			8.0	7.4
Mineral constituents in parts per million				
Calcium (Ca)			40	21
Magnesium (Mg)			1.5	0.6
Sodium (Na)			11.8	0.0
Potassium (K)			1.4	0.0
Carbonate (CO ₃)			1	0
Bicarbonate (HCO ₃)			17	0
Sulfate (SO ₄)			40	0.6
Chloride (Cl)			7	1.8
Nitrate (NO ₃)			2	0
Fluoride (F)			0.2	0
Boron (B)			1	0
Silica (SiO ₂)			15	0.0
Total dissolved solids in parts per million			94	26
Percent sodium			18	1
Hardness as CaCO ₃ in parts per million				
Total			114	41
Noncarbonate			27	2
Turbidity			30	1
Coliform in most probable number per milliliter			2,400	0
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.27	0.01
Solid alpha			0.00	0.00
Dissolved beta			0.48	0.01
Solid beta			0.31	0.00

WATER QUALITY VARIATIONS



MATTOLE RIVER NEAR PETROLIA (STA. 7a)

NOYO RIVER NEAR FORT BRAGG (STA. 10c)

Sampling Point Station 10c is situated in Section 10 of Township 18 North, Range 17 West, Mt. Diablo Base and Meridian. Prior to November 1959, monthly grab samples were collected from the right bank, 3.5 miles east of Fort Bragg, and about 4 miles upstream from the mouth. In November the station was moved upstream approximately one mile to its present site at the proposed Fort Bragg Municipal Water Supply intake. This relocation was made because of the occurrence of unusually high concentrations of most mineral constituents caused by sea-water incursion due to tidal action.

Period of Record January 1959 through December 1959.

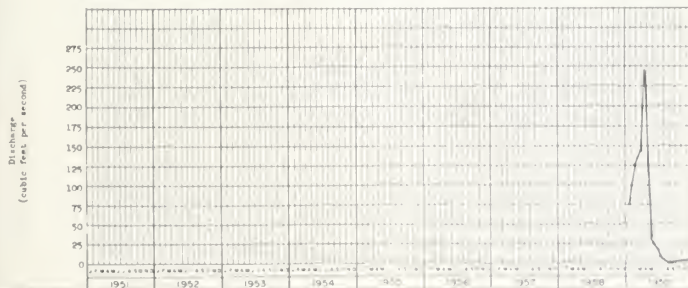
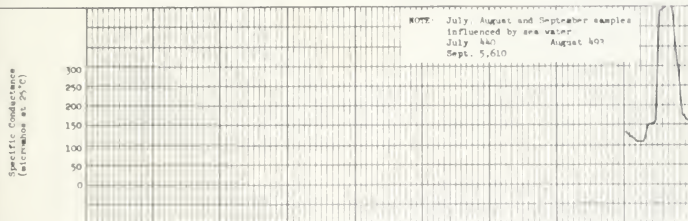
Water Quality Characteristics Noyo River water at Station 10c is excellent in quality. It is a bicarbonate type with calcium as its major cation. Mineral concentrations place this water in class 1 for irrigation, soft to slightly hard and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	174	113
Temperature in °F			66	41
Dissolved oxygen in parts per million			11	8
Percent saturation			99	84
pH			7.2	7.1
Mineral constituents in parts per million				
Calcium (Ca)			16	7.6
Magnesium (Mg)			5.6	2.9
Sodium (Na)			1.5	7
Potassium (K)			1.2	0.6
Carbonate (CO ₃)			82	47
Bicarbonate (HCO ₃)			5.8	1.8
Sulfate (SO ₄)			11	5.5
Chloride (Cl)			6	2
Nitrate (NO ₃)			2	1
Fluoride (F)			1	0
Boron (B)			21	12
Silica (SiO ₂)				
Total dissolved solids in parts per million			113	80
Percent sodium			71	80
Hardness as CaCO ₃ in parts per million				
Total			67	47
Noncarbonate				
Turbidity			18	2
Coliform in most probable number per milliliter			2,500	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha			3.35	2.00
Solid alpha			12.75	7.50
Dissolved beta			8.10	3.36
Solid beta				

WATER QUALITY VARIATIONS



NOYO RIVER NEAR FORT BRAGG (STA. 10c)

BIG RIVER NEAR MOUTH (STA. 8c)

Sampling Point Station 8c is located in Section 24, Township 17 North, Range 17 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the right bank approximately 12 miles upstream from the mouth about 9 miles east of Mendocino.

Period of Record January 1959 through December 1959.

Water Quality Characteristics Water at Station 8c is excellent in quality, calcium bicarbonate in character, soft to moderately hard and within drinking water standards for mineral content. Although it is class 1 for irrigation throughout the year, boron has reached the maximum recommended concentration of 0.5 ppm at various times. The source of the boron in this stream has not as yet been ascertained.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	See 1949	See 1949	211	126
Temperature in °F			71	39
Dissolved oxygen in parts per million			14.7	8.1
Percent saturation			100	96
pH			7.6	7.1
Mineral constituents in parts per million				
Calcium (Ca)			30	12
Magnesium (Mg)			8.8	3.4
Sodium (Na)			16	7.5
Potassium (K)			2.8	0.7
Carbonate (CO ₃)			0	0
Bicarbonate (HCO ₃)			127	54
Sulfate (SO ₄)			14	1
Chloride (Cl)			16	6.5
Nitrate (NO ₃)			3.7	0.2
Fluoride (F)			0.5	0.4
Boron (B)				
Silica (SiO ₂)			24	16
Total dissolved solids in parts per million			154	80
Percent sodium			28	23
Hardness as CaCO ₃ in parts per million				
Total			98	44
Noncarbonate			7	0.9
Turbidity			3	0.9
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.20	0.17
Solid alpha			0.61	0.09
Dissolved beta			3.59	0.01
Solid beta			8.82	0.00

WATER QUALITY VARIATIONS



BIG RIVER NEAR MOUTH (STA. 8c)

NAVARRO RIVER NEAR NAVARRO (STA. 8b)

Sampling Point Navarro River sampling Station 8b is located in Section 7 of Township 15 North, Range 16 West, Mt. Diablo Base and Meridian.

Monthly grab samples were collected from the left bank at the USGS gage 2.7 miles downstream from North Fork, 5.4 miles upstream from the mouth and 6.6 miles west of Navarro.

Period of Record January 1959 through December 1959.

Water Quality Characteristics Past analyses show water at this station to be excellent in quality, calcium bicarbonate to calcium-magnesium bicarbonate in character, class 1 for irrigation, slightly to moderately hard, and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	119	88
Temperature in °F			--	71
Dissolved oxygen in parts per million			12.0	8.8
Percent saturation			128	91
pH			7.7	7.3
Mineral constituents in parts per million				
Calcium (Ca)			30	18
Magnesium (Mg)			12	7.2
Sodium (Na)			24	8.5
Potassium (K)			2.0	1.0
Carbonate (CO ₃)			0.4	0.2
Bicarbonate (HCO ₃)			155	46
Sulfate (SO ₄)			25	7.7
Chloride (Cl)			20	7.5
Nitrate (NO ₃)			1.3	0.6
Fluoride (F)			0.2	0.1
Boron (B)			0.3	0.2
Silica (SiO ₂)			21	13
Total dissolved solids in parts per million			188	121
Percent sodium			12	19
Hardness as CaCO ₃ in parts per million				
Total			126	77
Noncarbonate			3	0.0
Turbidity			4	1
Coliforms in most probable number per milliliter			620	0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.09	0.00
Solid alpha			0.42	0.09
Dissolved beta			2.16	0.00
Solid beta			6.71	0.00

WATER QUALITY VARIATIONS



NAVARRO RIVER NEAR NAVARRO (STA. 8b)

GUALALA RIVER, SOUTH FORK NEAR ANNAPOLIS (STA. 9a)

Sampling Point Station 9a is located in Section 21 of Township 10 North, Range 14 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage, 1,000 feet downstream from Wheatfield Fork Gualala River, 4.8 miles west of Annapolis, and about 8 miles upstream from the mouth.

Period of Record January 1959 through December 1959.

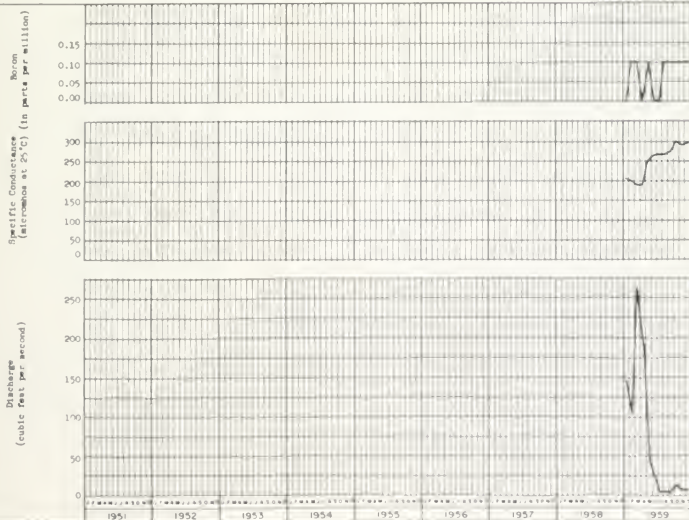
Water Quality Characteristics Gualala River is calcium-magnesium bicarbonate in character, class 1 for irrigation, slightly to moderately hard, and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	10	10
Temperature in °F			70	60
Dissolved oxygen in parts per million			4.2	2.2
Percent saturation			121	65
pH			8.0	6.5
Mineral constituents in parts per million				
Calcium (Ca)			10	10
Magnesium (Mg)			10	7.0
Sodium (Na)			7	7.0
Potassium (K)			1.0	0
Carbonate (CO ₃)			1.0	0
Bicarbonate (HCO ₃)			1.0	0
Sulfate (SO ₄)			1.0	0
Chloride (Cl)			1.0	0
Nitrate (NO ₃)			1.0	0
Fluoride (F)			1.0	0
Boron (B)			1.0	0
Silica (SiO ₂)			1.0	0
Total dissolved solids in parts per million			120	11
Percent sodium			22	17
Hardness as CaCO ₃ in parts per million				
Total			140	79
Noncarbonate			8	0
Turbidity			2	0.3
Coliform in most probable number per milliliter			200	145
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.51	0.00
Solid alpha			0.63	0.00
Dissolved beta			0.70	0.00
Solid beta			4.37	0.00

WATER QUALITY VARIATIONS



GUALALA RIVER, SOUTH FORK NEAR ANNAPOLIS (STA. 9a)

Russian River Basin

The Russian River Basin lies in the southern end of the North Coastal Region (No. 1) and covers about 1,500 square miles, of which approximately 1,200 are mountains and foothills and the remainder valley and mesa lands. The watershed is bounded on the east by the Cow Mountain Range and on the west by the Coastal Range. Waters draining from the watershed flow into the Pacific Ocean at Jenner, approximately 15 miles downstream from Guerneville. The Russian River has a total annual flow of approximately 1,500,000 acre-feet. Approximately 180,000 acre-feet of Eel River water is imported annually from Lake Pillsbury for power generation at Potter Valley Powerhouse within the Russian River watershed.

The most prominent uses of surface waters in this basin are recreational and industrial. The Russian River valley area contains a large number of recreational facilities for boating, swimming, and fishing. Logging and lumber operations and food processing comprise the major industrial uses of water. Approximately 300 square miles of the Russian River drainage basin are potential agricultural lands.

Most water users in the Russian River Basin discharge wastes to the river in quantities less than 0.5 mgd (million gallons per day). Three users, the Masonite Corporation, the City of Ukiah, and the City of Santa Rosa, discharge wastes in quantities over 0.5 mgd to the Russian River or its tributaries. During the nine-year period of quality record on the Russian River, none of these wastes discharges has seriously impaired the quality of surface waters.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Russian River near Hopland	78
Russian River near Healdsburg	80
Russian River at Guerneville	82
Russian River, East Fork at Potter Valley Powerhouse	84

RUSSIAN RIVER NEAR HOPLAND (STA. 8a)

Sampling Point Station 8a is located in Section 36 of Township 14 North, Range 12 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at Largo Road bridge site, 0.6 mile east of Highway 101, and 3.8 miles north of Hopland.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Past analyses show Russian River at Station 8a to be generally calcium bicarbonate in character, soft to moderately hard, and chemically suitable for drinking water. The quality of water at Station 8a does not differ significantly from the quality at Potter Valley Powerhouse (Station 10a). Boron frequently causes the water at Station 8a to be class 2 for irrigation. Highly mineralized spring runoff and solution of minerals from geologic formations in tributary streams are the source of the boron in this river.

Significant Water Quality Changes For the first year since 1953, boron, which reached the maximum of 0.5 ppm in December 1959, did not exceed the limit for class 1 irrigation water.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohms at 25°C)	276	100	197	168
Temperature in °F	84	48	71	48
Dissolved oxygen in parts per million	16	7.0	11	7.0
Percent saturation	150	68	113	88
pH	8.4	6.4	7.9	7.1
Mineral constituents in parts per million				
Calcium (Ca)	11	11	20	18
Magnesium (Mg)	36	4.5	7.9	7.7
Sodium (Na)	14	4.1	8.7	4.3
Potassium (K)	2.0	0.5	1.7	1.4
Carbonate (CO ₃)	11	0.0	4.0	0.0
Bicarbonate (HCO ₃)	142	62	118	78
Sulfate (SO ₄)	11	1.0	5.6	4.4
Chloride (Cl)	9.6	1.1	9.4	3.2
Nitrate (NO ₃)	1.6	2.3	1.2	2.6
Fluoride (F)	2.3	0.0	0.1	0.0
Boron (B)	0.60	0.2	0.5	0.2
Silica (SiO ₂)	17	7.3	12	10
Total dissolved solids in parts per million	161	71	116	99
Percent sodium	20	10	19	10
Hardness as CaCO ₃ in parts per million				
Total	116	52	88	70
Noncarbonate	12	0.0	8	0.0
Turbidity	600	0.0	30	
Coliform in most probable number per milliliter	>7,000	<0.005	2,400	<0.005
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.12	0.30	0.51	0.19
Solid alpha	0.54	0.1	0.4	0.71
Dissolved beta	11.91	0.30	2.0	1.0
Solid beta	14.71	0.30	8.4	0.1

WATER QUALITY VARIATIONS



RUSSIAN RIVER NEAR HOPLAND (STA. 8a)

RUSSIAN RIVER NEAR HEALDSBURG (STA. 9)

Sampling Point Healdsburg station is located in Section 22 of Township 9 North, Range 9 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank at the USGS gage, 2 miles east of Healdsburg and 3.5 miles upstream from Dry Creek.

Period of Record April 1951 through December 1959.

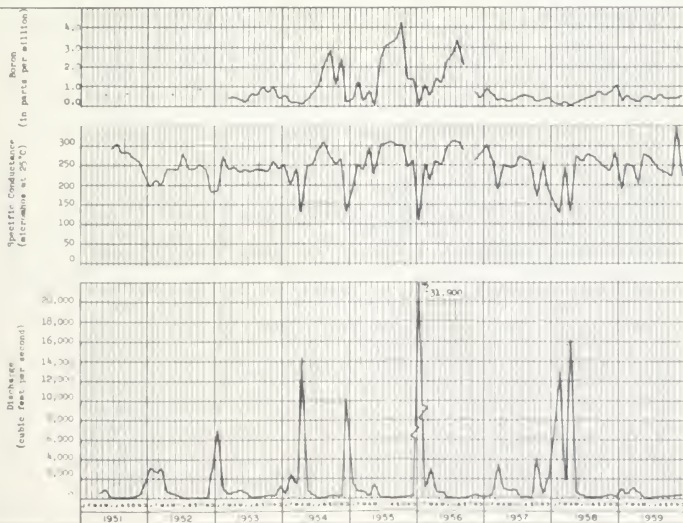
Water Quality Characteristics Russian River water at Station 9 is, with the exception of boron, good to excellent in quality, calcium bicarbonate to magnesium bicarbonate in character, ranges from soft to moderately hard and meets drinking water standards for mineral content. Prior to 1956, boron concentrations often exceeded the limit for class 1, and at times class 2, irrigation water. The major source of excess boron was detected to be an industrial discharge, which was discontinued in September 1956. Following its removal, boron concentrations have remained below 1.0 ppm. Dissolved minerals are found in slightly higher concentrations (averaging about 70 micromhos) at Station 9 than at the upstream Station 8a.

Significant Water Quality Changes During 1959 the water improved in quality with respect to boron. The boron limit for class 1 irrigation water was exceeded only twice, in February and August, with 0.6 ppm reported each month. The continuation of boron concentrations of less than 1.0 ppm during 1959 indicates that the boron content in this water has been stabilizing since the discontinuance of the degrading industrial waste discharge.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	N/A	108	N/A	190
Temperature in °F	80	45	77	53
Dissolved oxygen in parts per million	11.2	7.0	11.6	7.6
Percent saturation	119	70	107	83
pH	8.6	6.3	7.9	7.3
Mineral constituents in parts per million				
Calcium (Ca)	11	10	24	23
Magnesium (Mg)	16	3.1	16	12
Sodium (Na)	19	3.2	19	6.9
Potassium (K)	3.2	0.8	1.5	1.4
Carbonate (CO ₃)	6	0.1	3.1	0.0
Bicarbonate (HCO ₃)	179	58	179	75
Sulfate (SO ₄)	14	2.9	8.6	7.2
Chloride (Cl)	4	1.5	14	4.2
Nitrate (NO ₃)	1.6	0.1	1.4	0.0
Fluoride (F)	0.3	0.1	0.3	0.0
Boron (B)	4.3	0.36	0.6	0.06
Silica (SiO ₂)	98	5.6	13	11
Total dissolved solids in parts per million	234	64	204	119
Percent sodium	36	11	23	12
Hardness as CaCO ₃ in parts per million				
Total	142	46	142	78
Noncarbonate	16	0.0	16	0.0
Turbidity	1,000	0.1	40	3
Coliform in most probable number per milliliter	>7,000	3.36	2,100	0.13
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.94	0.30	0.94	0.09
Solid alpha	0.30	0.00	0.42	0.09
Dissolved beta	1.00	0.30	0.40	0.00
Solid beta	1.00	0.30	0.40	0.00

WATER QUALITY VARIATIONS



RUSSIAN RIVER AT GUERNEVILLE (STA. 10)

Sampling Point Station 10 is located in Section 32 of Township 8 North, Range 10 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at the State Highway 12 bridge in Guerneville, and about 13 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

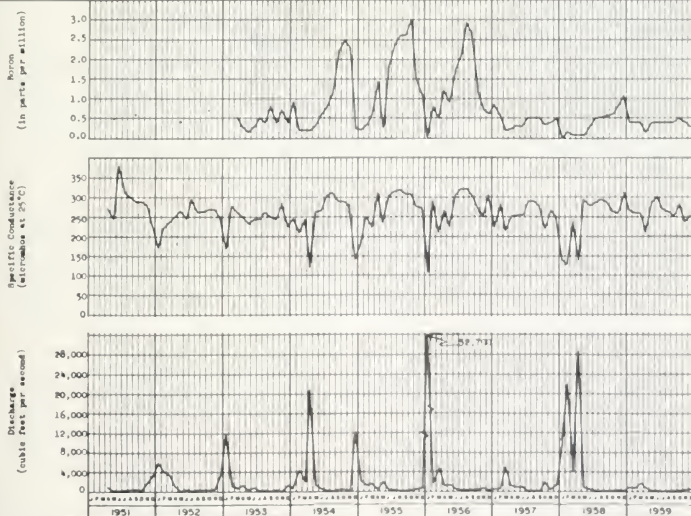
Water Quality Characteristics Water at Station 10 is calcium-magnesium bicarbonate in character, soft to very hard, and within drinking water standards for mineral content. As at all stations in the Russian River Basin, boron concentrations have often been in excess of class 1 irrigation limits. Prior to 1957, boron was frequently found in excess of class 2 requirements. After source of excess boron mentioned in the discussion of Station 9 was removed in September 1956, boron concentrations decreased significantly. During 1957 and 1958 the maximum concentration reported was 1.1 ppm, as contrasted to the maximum for the period of record of 3.0 ppm reported in October 1955. An average increase in conductivity of about 15 micromhos occurs between Stations 9 and 10 indicating only a slight increase in the amount of mineral constituents.

Significant Water Quality Changes During 1959 boron was not detected in excess of the 0.5 ppm limit for class 1 irrigation water at Station 10. The low boron concentrations at this station substantiate the conclusion, as stated in the discussion of Station 9, that the boron content of Russian River waters is approaching a steady state.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1967	Minimum 1967
Specific conductance (micromhos at 25°C)	300	10	100	10
Temperature in °F	80	40	75	50
Dissolved oxygen in parts per million	10.0	1.0	10.0	0.5
Percent saturation	100	20	100	20
pH	9.0	6.0	7.0	6.0
Mineral constituents in parts per million				
Calcium (Ca)	50	10	50	10
Magnesium (Mg)	10	5	10	5
Sodium (Na)	10	5	10	5
Potassium (K)	5	1	5	1
Carbonate (CO ₃)	10	5	10	5
Bicarbonate (HCO ₃)	10	5	10	5
Sulfate (SO ₄)	10	5	10	5
Chloride (Cl)	10	5	10	5
Nitrate (NO ₃)	10	5	10	5
Fluoride (F)	10	5	10	5
Boron (B)	10	5	10	5
Silica (SiO ₂)	10	5	10	5
Total dissolved solids in parts per million	200	50	100	10
Percent sodium	70	10	10	10
Hardness as CaCO ₃ in parts per million				
Total	30	5	10	5
Noncarbonate	10	5	10	5
Turbidity	1,000	0.0	10	2
Coliform in most probable number per milliliter	7,000	3,000	600	3,000
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.01	0.0	0.00	0.0
Solid alpha	0.01	0.0	0.01	0.0
Dissolved beta	23.27	0.00	0.00	0.0
Solid beta	9.00	0.00	0.00	0.00

WATER QUALITY VARIATIONS



RUSSIAN RIVER AT GUERNEVILLE (STA. 10)

RUSSIAN RIVER, EAST FORK AT POTTER VALLEY POWERHOUSE (STA. 10a)

Sampling Point Station 10a is located in Section 6 of Township 17 North, Range 11 West, Mt. Diablo Base and Meridian. Monthly water samples for quality analyses were collected from the tailrace of the PG&E powerhouse, 3 miles northeast of the town of Potter Valley.

Period of Record June 1951 through December 1959.

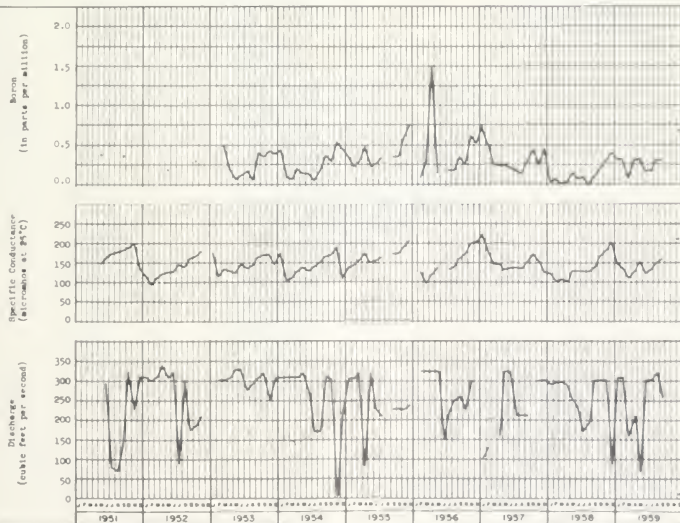
Water Quality Characteristics Water at Station 10a is calcium bicarbonate in character, soft to moderately hard and within drinking water standards for mineral content. As at Station 8a, boron frequently causes the water to be class 2 for irrigation use. Water at this station is comprised of water exported from the Eel River Basin. Boron in waters at this station originates from geologic formations and mineralized springs along the upper reaches of Eel River.

Significant Water Quality Changes Radioactivity was significantly higher during 1959 than during past years of record with the exception of May 1953 when 34.6 $\mu\text{c}/\text{l}$ total activity were reported. The total activity in May 1959 was reported as 24.3 $\mu\text{c}/\text{l}$ and 24.7 $\mu\text{c}/\text{l}$ in September.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	175	75	105	115
Temperature in °F	65	41	75	47
Dissolved oxygen in parts per million	11.5	5.5	11.5	5.5
Percent saturation	100	51	95	51
pH	8.5	7.0	8.5	7.0
Mineral constituents in parts per million				
Calcium (Ca)	105	35	105	19
Magnesium (Mg)	115	35	105	45
Sodium (Na)	105	105	105	105
Potassium (K)	105	105	105	105
Carbonate (CO ₃)	105	105	105	105
Bicarbonate (HCO ₃)	105	105	105	105
Sulfate (SO ₄)	105	105	105	105
Chloride (Cl)	105	105	105	105
Nitrate (NO ₃)	105	105	105	105
Fluoride (F)	105	105	105	105
Boron (B)	105	105	105	105
Silica (SiO ₂)	105	105	105	105
Total dissolved solids in parts per million	135	75	125	68
Percent sodium	17	5	16	11
Hardness as CaCO ₃ in parts per million				
Total	102	42	95	54
Noncarbonate	8	0	8	0
Turbidity	190	5	75	5
Coliform in most probable number per milliliter	10,000	0.001	6000	0.001
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.1	0.00	1.1	0.00
Solid alpha	1.0	0.00	1.0	0.00
Dissolved beta	1.1	0.00	1.1	0.00
Solid beta	1.1	0.00	1.1	0.00

WATER QUALITY VARIATIONS



RUSSIAN RIVER, EAST FORK AT POTTER VALLEY POWERHOUSE (STA. 10a)

Station
Number

- | | |
|-----|------------|
| 1 | Klamath |
| 1a | Shasta R. |
| 1b | Scott R. |
| 1c | Klamath |
| 1d | Butte Cr. |
| 1e | Antelope |
| 2 | Klamath |
| 2a | Salmon R. |
| 2b | Klamath |
| 3 | Klamath |
| 3a | Smith R. |
| 3b | Redwood |
| 4 | Trinity |
| 4a | Trinity |
| 4b | Trinity |
| 5 | Eel River |
| 5a | Van Duzee |
| 5b | Outlet C. |
| 5c | Eel River |
| 5d | Eel River |
| 6 | Eel River |
| 6a | Mad River |
| 7 | Eel River |
| 7a | Mattole |
| 8a | Russian |
| 8b | Navarro |
| 8c | Big River |
| 9 | Russian |
| 9a | Gualala |
| 10 | Russian |
| 10a | Russian |
| | Valley |
| 10c | Noyo River |

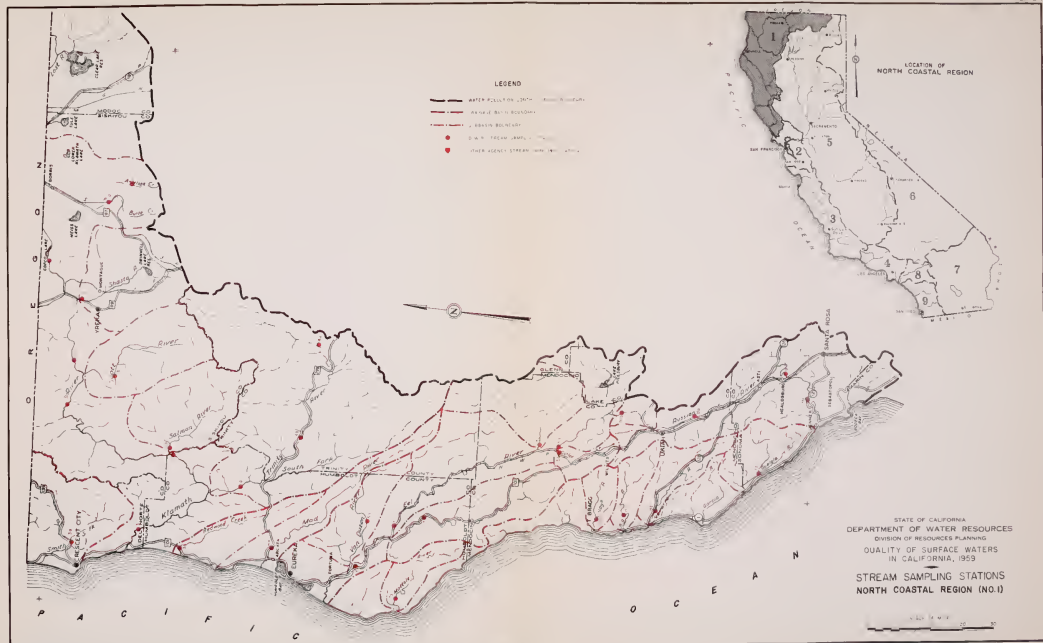


STATE OF CALIFORNIA
DEPARTMENT OF WATER RESOURCES
DIVISION OF RESOURCES PLANNING
QUALITY OF SURFACE WATERS
IN CALIFORNIA, 1959
STREAM SAMPLING STATIONS
NORTH COASTAL REGION (NO.1)

STREAM SAMPLING STATIONS
NORTH COASTAL REGION (NO. 1)

Station Number	Station Name
1	Klamath River near Copco
1a	Shasta River near Yreka
1b	Scott River near Fort Jones
1c	Klamath River above Sanburg Reservoir Site
1d	Butte Creek near MacDoel
1e	Antelope Creek near Tennent
2	Klamath River at Somebar
2a	Salmon River at Somebar
2b	Klamath River near Selad Valley
3	Klamath River near Klamath
3a	Smith River near Crescent City
3b	Redwood Creek at Orick
4	Trinity River near Hoopa
4a	Trinity River at Lewiston
4b	Trinity River near Burnt Ranch
5	Eel River near McCann
5a	Van Duzen River near Bridgeville
5b	Outlet Creek near Longvale
5c	Eel River, Middle Fork at Dos Rios
5d	Eel River near Dos Rios
6	Eel River at Scotia
6a	Mad River near Arcata
7	Eel River, South Fork near Miranda
7a	Mattole River near Petrolia
8a	Russian River near Hopland
8b	Navarro River near Navarro
8c	Big River near Mouth
9	Russian River near Healdsburg
9a	Gualala River, South Fork near Annapolis
10	Russian River at Guerneville
10a	Russian River, East Fork at Potter Valley Powerhouse
10c	Noyo River near Fort Bragg

44460



San Francisco Bay Region (No. 2)

One of the most highly industrialized regions of California is encompassed by the boundaries of the San Francisco Bay Region. This region contains approximately 4,400 square miles in the north central coastal portion of California and includes the industrial and municipal complexes of the City of San Francisco, the Peninsula, and East Bay communities.

Prominent among the physical features of the region is the outstanding natural harbor consisting of San Francisco Bay, San Pablo Bay and that portion of Suisun Bay below Antioch. This harbor is the focal point of numerous valley basins drained by the watercourses tributary to the bay. These valleys are interspersed and parallel the mountains and foothills of the Coast Range, which rise from sea level to elevations of over 4,000 feet and cover two-thirds of the bay region.

Estimated mean annual surface runoff is 1,245,000 acre-feet in this region. To maintain a surveillance on quality of surface runoff in this area, five monitor stations are maintained on five streams. The monitored streams and the number of the station on each (in parentheses) are as follows:

Napa River (1)	Coyote Creek (1)
Alameda Creek (1)	Los Gatos Creek (1)
Arroyo del Valle (1)	

Analyses of samples collected from streams in the San Francisco Bay Region indicate bicarbonate type waters with generally no predominant cation. These waters are suitable for domestic and most industrial uses and range from class 1 to 2 for irrigation. Although precipitation during 1959 was generally below normal in this region, only minor changes in quality were detected by the surface water monitoring program.

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Napa River Basin

Napa River drains a watershed area of 417 square miles located at the north end of San Francisco Bay Region. The river flows southward through Napa Valley and discharges into San Pablo Bay. Average annual discharge of Napa River is estimated to be 186,300 acre-feet.

Napa River Basin includes approximately 157 square miles of fertile valley and mesa land. Agricultural pursuits are the major users of water and are the dominant economic enterprises of the basin. However, industrial and urban development has accelerated rapidly in the last decade and these are playing a proportionately larger role in the economy of the valley.

Numerous wastes from individual domestic, industrial and agricultural sources, and several community collection systems discharge into Napa River. None of these waste discharges individually exceed 0.3 mgd, except for the Napa County Sanitation District discharge of 4.1 mgd.

A surface water sampling station is maintained on Napa River near St. Helena to monitor quality of runoff from this basin.

NAPA RIVER NEAR ST. HELENA (STA. 72)

Sampling Point Station 72 is located in Section 32 of Township 8 North, Range 5 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at the center of the stream, from the highway bridge 0.2 mile downstream from the USGS water stage recorder. This gage is located 1.0 mile east of Highway 128 and 2.5 miles southeast of St. Helena.

Period of Record December 1951 through December 1959.

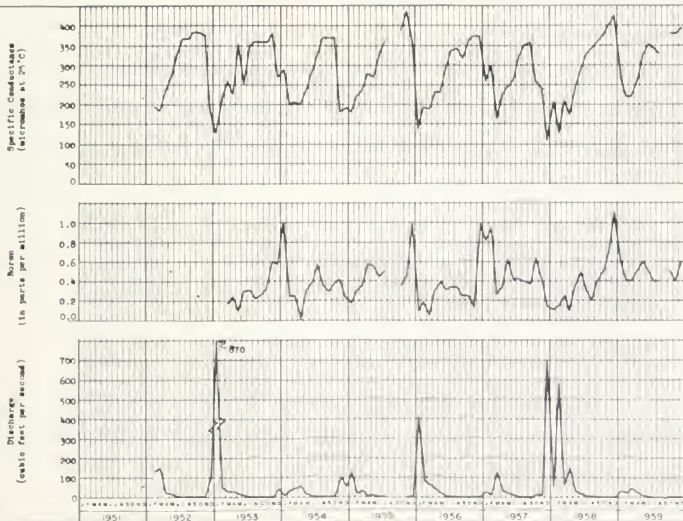
Water Quality Characteristics Chemical classification of past analyses show Napa River, at this station, to vary in character from calcium bicarbonate to calcium-sodium bicarbonate. Flow past Station 72 has met the criteria for class 1 irrigation supply, except for boron which generally ranges from 0.1 ppm to 1.0 ppm. Boron in waters entering Napa River is probably derived from the geologic formations comprising the earth's mantle in this watershed. Hardness ranges from soft to moderately hard and concentrations of minerals in this water are within the limits for drinking water.

Significant Water Quality Changes During September 1959 the stream was dry and comprehensive analysis usually performed on the sample collected that month was omitted.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	413	108	305	218
Temperature in °F	80	45	80	50
Dissolved oxygen in parts per million	15.0	5.5	13.2	6.2
Percent saturation	175	60	162	69
pH	8.2	6.8	7.2	7.1
Mineral constituents in parts per million				
Calcium (Ca)	38	11	26	
Magnesium (Mg)	19	4.0	11	
Sodium (Na)	12	6.4	26	1.6
Potassium (K)	8.1	1.8	3.4	
Carbonate (CO ₃)	26	0.0	2	0.0
Bicarbonate (HCO ₃)	210	44	196	80
Sulfate (SO ₄)	44	4	17	
Chloride (Cl)	45	5.0	34	11
Nitrate (NO ₃)	6.2	0.5	2.1	
Fluoride (F)	0.5	0.1	0.5	
Boron (B)	1.1	0.05	0.6	0.4
Silica (SiO ₂)	42	14	38	
Total dissolved solids in parts per million	288	71	263	145
Percent sodium	41	15	38	21
Hardness as CaCO ₃ in parts per million				
Total	169	37	158	70
Noncarbonate	30	0.0	20	0.0
Turbidity	70	0.0	60	0.0
Coliform in most probable number per milliliter	>7,000.	<0.045	7,000.	0.62
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.38	0.00	0.21	
Solid alpha	0.48	0.00	0.20	
Dissolved beta	17.4	0.00	1.88	
Solid beta	8.82	0.00	5.27	

WATER QUALITY VARIATIONS



NAPA RIVER NEAR SAINT HELENA (STA. 72)

Alameda Creek Basin

Alameda Creek watershed is located east of and drains into the southern arm of San Francisco Bay. The drainage basin encompasses about 272 square miles of mountains and foothills of the Diablo Range and 157 square miles of valley and mesa lands. Mean seasonal natural runoff for Alameda Creek Basin is about 130,700 acre-feet. There are numerous water supply developments in this watershed which greatly affect the runoff characteristics of Alameda Creek.

Agricultural development is still significant in the valley areas of the basin; however, urban, industrial, and commercial growth has been given considerable impetus by the expanding East Bay economy. Surface water in the basin is insufficient to meet present demands, and additional imported water supplies will be needed to sustain the present rate of growth.

Numerous waste discharges, originating from industrial and municipal developments, are discharged into the Alameda Creek watershed waterways. A list of the major waste discharges in this watershed and their daily outflows are:

City of Livermore	1.0 mgd
City of Pleasanton	0.6 mgd
Holly Sugar Company at Alvarado, California	1.6 mgd
Pacific States Steel, Niles, California	70.5 mgd
Parks Air Force Base	0.5 mgd
Rickenbacker Dairy	0.32-1.29 mgd
Union Sanitary District, Fremont, California	70.5 mgd
West Vaco Chemical Company, Newark, California	1.34 mgd

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Alameda Creek near Niles	94
Arroyo del Valle at V.A. Hospital	96

ALAMEDA CREEK NEAR NILES (STA. 73)

Sampling Point The sampling point for this station is located in Section 15, Township 4 South, Range 1 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the right bank at the concrete control structure of the USGS gaging station located 0.2 mile downstream from the railroad bridge and 1.2 miles northeast of Niles.

Period of Record December 1951 through December 1959. The stream is dry a portion of each year; consequently, data are not available for all months.

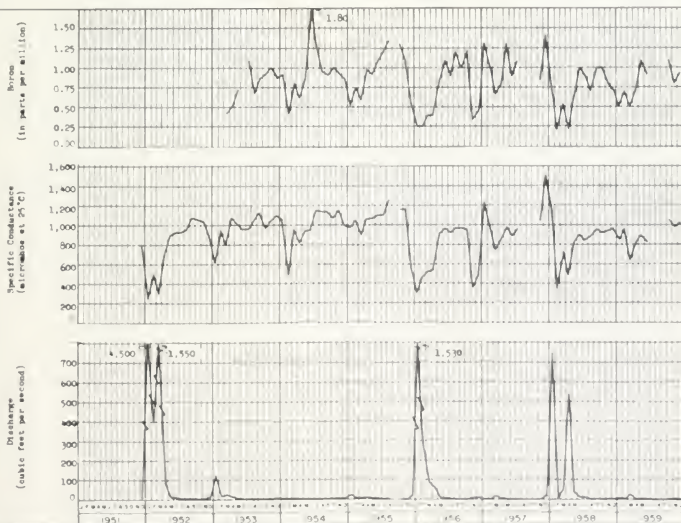
Water Quality Characteristics Since inception of a monitoring station on this stream the water has been bicarbonate in character with none of the major cations, calcium, magnesium, or sodium, being predominant. Due to fluctuation of electrical conductivity, concentrations of total dissolved solids, or boron, singly or in combination, this water ranges from class 1 to class 2 for irrigation use. The source of boron originates from springs in the north and western portions of the watershed. The water is moderate to very hard. From the standpoint of mineral constituents, this water meets the criteria for domestic use.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,500	246	1,060	640
Temperature in °F	76	41	74	44
Dissolved oxygen in parts per million	17.0	6.3	12.5	8.4
Percent saturation	158	54	108	88
pH	8.5	7.0	8.1	7.5
Mineral constituents in parts per million				
Calcium (Ca)	96	5.7	63	
Magnesium (Mg)	70	18	42	
Sodium (Na)	136	14	22	36
Potassium (K)	16	1.7	3.4	
Carbonate (CO ₃)	23	0.0	0.0	0.0
Bicarbonate (HCO ₃)	479	110	364	241
Sulfate (SO ₄)	150	90	140	
Chloride (Cl)	210	9	82	40
Nitrate (NO ₃)	4.1	0.8	0.0	
Fluoride (F)	0.4	0.0	0.2	
Boron (B)	1.80	0.21	1.1	0.5
Silica (SiO ₂)	21	2.9	2.9	
Total dissolved solids in parts per million	915	150	646	340
Percent sodium	38	20	35	24
Hardness as CaCO ₃ in parts per million				
Total	474	56	306	242
Noncarbonate	364	5	125	36
Turbidity	550	0.3	30	1
Coliform in most probable number per milliliter	7,000	0.045	230	0.06
Radioactivity in micro-curies per liter				
Dissolved alpha	0.42	0.00	0.41	
Solid alpha	2.82	0.00	0.62	
Dissolved beta	13.47	0.00	2.24	
Solid beta	12.44	0.00	0.31	

WATER QUALITY VARIATIONS



ARROYO DEL VALLE AT VETERANS ADMINISTRATION HOSPITAL (STA. 71)

Sampling Point The location of this station is Section 4, Township 4 South, Range 2 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage, adjacent to Arroyo Road. The gage is located immediately upstream from the Veterans Administration Hospital bridge, approximately 4.5 miles south of Livermore.

Period of Record July 1958 through December 1959. Arroyo del Valle is dry during a portion of each year and water quality data are not available for all months.

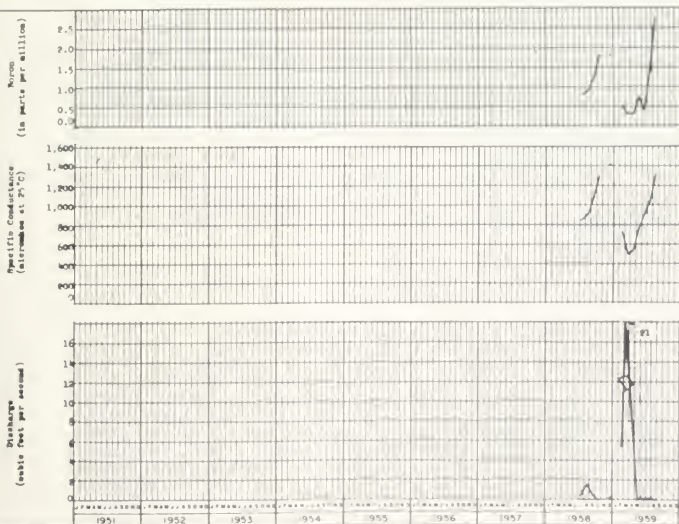
Water Quality Characteristics A review of available analyses show this water to consistently exhibit a bicarbonate characteristic; however, no specific cation is predominant. Calcium, magnesium, and sodium are present in significant amounts; and in approximately equal equivalents per million. With the exception of a single boron determination of 2.7 ppm in August 1959, this water has met class 2 irrigation criteria. Although the water is very hard, it meets the limits for mineral constituents in drinking water.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	8,413	474	1,120	477
Temperature in °F	84	59	68	52
Dissolved oxygen in parts per million	10.0	4.1	10.0	4.5
Percent saturation	119	41	100	44
pH	8.1	7.3	8.1	7.4
Mineral constituents in parts per million				
Calcium (Ca)	115	20	83	20
Magnesium (Mg)	24	22	24	22
Sodium (Na)	140	20	140	20
Potassium (K)	8.6	1.2	8.6	1.7
Carbonate (CO ₃)	8	0.2	8	0.2
Bicarbonate (HCO ₃)	450	291	450	291
Sulfate (SO ₄)	169	55	130	55
Chloride (Cl)	159	15	159	15
Nitrate (NO ₃)	3.5	0.0	3.5	0.0
Fluoride (F)	0.4	0.0	0.4	0.0
Boron (B)	2.7	0.3	2.7	0.3
Silica (SiO ₂)	34	13	34	13
Total dissolved solids in parts per million	870	268	820	268
Percent sodium	43	17	43	17
Hardness as CaCO ₃ in parts per million				
Total	526	210	526	210
Noncarbonate	150	0.0	68	0.0
Turbidity	See 1959	See 1959	0.5	
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.52	0.05	0.52	
Solid alpha	0.62	0.00	0.62	
Dissolved beta	6.18	5.10	5.10	
Solid beta	3.74	0.71	1.96	

WATER QUALITY VARIATIONS



ARROYO DEL VALLE AT VETFRANS HOSPITAL (STA. 71)

Coyote Creek Basin

Coyote Creek originates in the Diablo Range in the southeast corner of the San Francisco Region. It flows northeasterly through a portion of Santa Clara Valley and discharges into San Francisco Bay. Coyote Creek Basin drains 404 square miles of which approximately one-fourth is valley and mesa lands. The creek has a total annual flow of about 89,000 acre-feet.

Land use in the valley areas of this basin is devoted to intensive agricultural production. There has been considerable recent industrial development in the vicinities of San Jose and Milpitas, accompanied by a large population growth.

Several waste discharges are tributary to Coyote Creek. Most of these discharges are of minor quantities. The significant waste discharges are the outflow of the primary treatment plant receiving industrial and domestic waste from the City of San Jose which exceeds 20 million gallons per day (mgd) and the Milpitas Sanitary District which discharges about 1.0 mgd from its secondary treatment plant.

A surface water sampling station is maintained on Coyote Creek near Madrone to monitor quality of runoff from this basin.

COYOTE CREEK NEAR MADRONE (STA. 82)

Sampling Point The station is located in Section 9, Township 9 South, Range 3 East, Mt. Diablo Base and Meridian, and lies in the northwest corner of the San Jose Grant. Monthly water samples were obtained from the right bank at the USGS gaging station, 0.2 mile downstream from the county road bridge, 2.8 miles northeast of Madrone.

Period of Record January 1952 through December 1959.

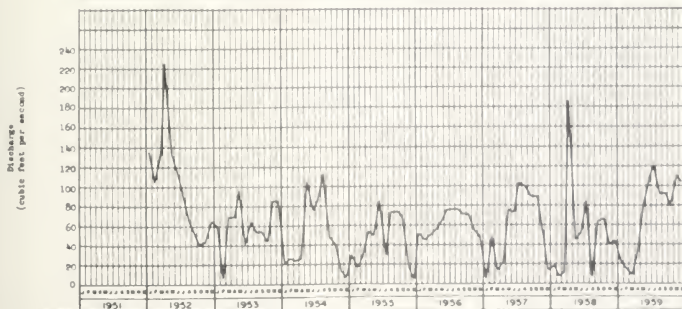
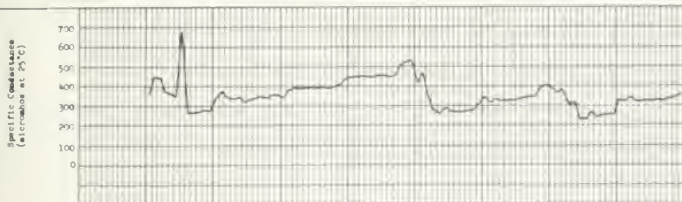
Water Quality Characteristics Past chemical analyses of Coyote Creek near Madrone show it to be characteristically a calcium-magnesium type water. This water consistently meets the criteria for a class 1 irrigation supply and the mineral constituent standards for drinking water. Water at this station ranges from slightly hard to very hard with quality fluctuations throughout the year being relatively small.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum 1959
Specific conductance (microhms at 25°C)	270	200	160	191
Temperature in °F	72	40	55	31
Dissolved oxygen in parts per million	11.5	8.1	11.4	9.7
Percent saturation	104	76	100	98
pH	8.0	7.	8.1	7.5
Minerals: constituents in parts per million				
Calcium (Ca)	53	24	34	30
Magnesium (Mg)	26	10	17	14
Sodium (Na)	24	10	17	12
Potassium (K)	12	5.8	2.4	1.2
Carbonate (CO ₃)	18	0.0	0.8	0.0
Bicarbonate (HCO ₃)	140	104	184	140
Sulfate (SO ₄)	45	20	32	27
Chloride (Cl)	25	5.5	25	8.2
Nitrate (NO ₃)	4.4	4.5	1.0	0.8
Fluoride (F)	0.9	0.0	0.1	0.0
Boron (B)	0.20	0.0	0.2	0.0
Silica (SiO ₂)	16	4.5	11	5.8
Total dissolved solids in parts per million	294	133	211	186
Percent sodium	21	14	20	16
Hardness as CaCO ₃ in parts per million				
Total	322	90	161	137
Noncarbonate	76	0.0	30	10
Turbidity	330	30.0	35	1
Coliforms in most probable number per milliliter	27,000	10,045	600	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.60	0.00	0.60	0.19
Solid alpha	8.16	0.00	0.69	0.19
Dissolved beta	17.44	0.00	8.56	0.00
Solid beta	11.46	0.00	11.46	9.05

WATER QUALITY VARIATIONS



COYOTE CREEK NEAR MADRONE (STA. 82)

Los Gatos Creek Basin

Los Gatos Creek watershed encompasses approximately 65 square miles in the southwestern portion of the San Francisco Bay Region. Los Gatos Creek originates in the Santa Cruz Mountains and flows northeasterly a distance of about 20 miles to join Guadalupe River at the City of San Jose. Mean seasonal runoff from this basin is about 35,800 acre-feet.

Due to mountainous terrain along the upper reaches of Los Gatos Creek, development is almost exclusively confined to the drainage area tributary to its lower ten-mile reach. Land use is largely urban, interspersed with light industry. Land devoted to agriculture in this watershed has diminished rapidly in the past decade and only scattered orchards and vineyards remain.

Numerous waste discharges enter Los Gatos Creek in minor quantities. There are no waste discharges in excess of 0.5 mgd being disposed of directly into the waterway of Los Gatos Creek.

A surface water monitoring station is maintained on Los Gatos Creek immediately above the community of Los Gatos.

LOS GATOS CREEK AT LOS GATOS (STA. 74)

Sampling Point Station 74 is located in Section 29, Township 8 South, Range 1 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank at the USGS gage about 0.75 mile upstream from Los Gatos, approximately 0.25 mile below Lexington Dam. This point is 10.5 miles above the mouth of the creek.

Period of Record December 1951 through December 1959.

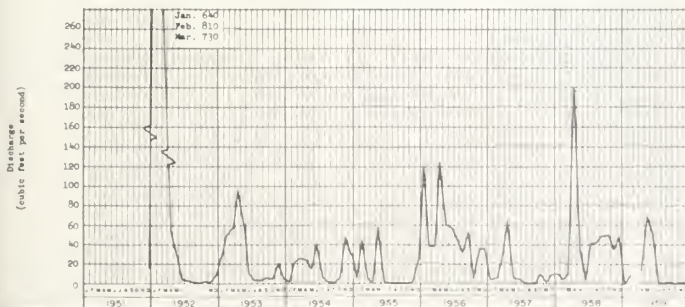
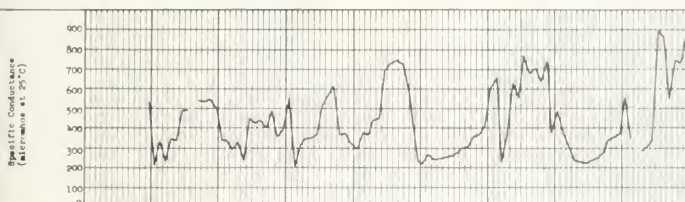
Water Quality Characteristics A review of past analyses reveals that this stream has a bicarbonate characteristic, and calcium and magnesium are its principal cations with neither consistently predominant. Los Gatos Creek water at this station meets the requirements for class 1 irrigation water (except for one boron determination of 1.0 ppm in 1956), ranges from slightly hard to very hard, and contains mineral concentrations within the limits for drinking water.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	812	201	896	896
Temperature in °F	79	41	79	41
Dissolved oxygen in parts per million	11.4	7.6	12	7.6
Percent saturation	118	78	80	81
pH	8.3	6.8	7	7.1
Mineral constituents in parts per million				
Calcium (Ca)	85	18	89	18
Magnesium (Mg)	39	7.7	81	12
Sodium (Na)	39	6.8	8	9.7
Potassium (K)	3.9	0.8	2.1	0.8
Carbonate (CO ₃)	13	0.1	2	0.2
Bicarbonate (HCO ₃)	428	69	428	101
Sulfate (SO ₄)	103	13	71	49
Chloride (Cl)	67	1.6	23	1.6
Nitrate (NO ₃)	4.4	0.2	2.7	0.5
Fluoride (F)	0.4	0.1	0.1	0.1
Boron (B)	0.25	0.1	0.1	0.1
Silica (SiO ₂)	38	11	15	11
Total dissolved solids in parts per million	546	124	546	176
Percent sodium	29	12	1	12
Hardness as CaCO ₃ in parts per million				
Total	471	69	471	194
Noncarbonate	120	0.0	120	41
Turbidity	900	0.4	180	1
Coliforms in most probable number per milliliter	77,000	0.06	2,400	0.21
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.52	0.00	0.52	0.00
Solid alpha	1.30	0.00	0.69	0.10
Dissolved beta	17.44	0.00	3.17	0.60
Solid beta	18.24	0.00	6.92	3.46

WATER QUALITY VARIATIONS



LOS GATOS CREEK AT LOS GATOS (STA. 74)

STREA

SAN FRANC

Station
Number

15c	Sacram
28a	Carquin
71	Arroyo
	Hosp
72	Napa R
73	Alameda
74	Los Ga
82	Coyote

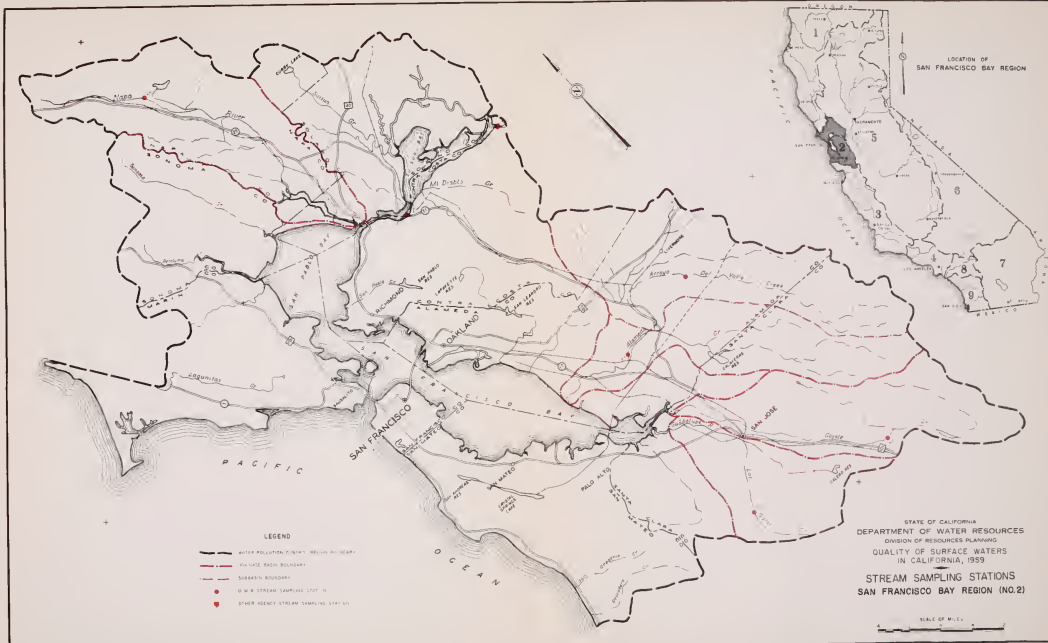
STREAM SAMPLING STATIONS
SAN FRANCISCO BAY REGION (NO. 2)

Station
Number

Station Name

15c	Sacramento River at Mallard Slough
28a	Carquinez Strait at Martinez
71	Arroyo del Valle at Veterans Administration Hospital
72	Napa River near Saint Helena
73	Alameda Creek near Niles
74	Los Gatos Creek at Los Gatos
82	Coyote Creek near Madrone

44460



Central Coastal Region (No. 3)

The Central Coastal Region contains approximately 11,000 square miles of coastal valleys and mountain ranges. The region extends 220 miles north-south from the southern boundary of Pescadero Creek Basin (about 35 miles south of the City of San Francisco) to the northeastern boundary of Rincon Creek Basin (approximately 70 miles north of Los Angeles).

Valley and mesa areas cover over 2,000 square miles of this region, with the valley fill along Salinas River comprising over 40 percent of these lands. The coast line is rocky and rugged except for a few river deltas. Mountain peaks in excess of 5,000 feet elevation exist in most of the ranges with Sawmill Mountain at the head of Santa Maria River reaching 8,750 feet.

Mean seasonal runoff from this region is 2,447,600 acre-feet. Principal hydrographic units in the Central Coastal Region include the San Lorenzo, Pajaro, Salinas, Carmel, Santa Maria, and Santa Ynez. In the Central Coastal Region (No. 3) 14 sampling stations are being monitored to maintain a surveillance on quality of surface waters. The monitored streams and the number of stations on each (in parentheses) are presented in the following tabulation:

San Lorenzo River (1)	Nacimiento River (1)
Soquel Creek (1)	San Antonio Creek (1)
Pajaro River (1)	Carmel River (1)
Uvas Creek (1)	Cuyama River (1)
San Benito River (1)	Santa Ynez River (2)
Salinas River (3)	

The upper reaches of the Salinas, Nacimiento, Cuyama and Santa Ynez Rivers, and San Antonio Creek are in Southern California and will be discussed in Part II of this bulletin.

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Salinas River (3)	

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San Lorenzo River Basin

The San Lorenzo River Basin drains an area of 137 square miles in the northwest corner of the Central Coastal Region. The river flows north to south and discharges into Monterey Bay at the City of Santa Cruz.

The San Lorenzo River watershed is generally mountainous with only seven square miles being identified as valley or mesa lands. Urban and light industrial development are prominent in the Santa Cruz area. Along the upper reaches of the river, recreation, a few lumber mills, and resort facilities support the economy of the area. Natural mean seasonal runoff is estimated to be 125,100 acre-feet in this basin.

Waste discharges entering San Lorenzo Basin waterways are not of significant quantity. Several gravel wash discharges of about 0.10 mgd constitute the only notable source of possible impairment under present development.

A surface water sampling station is maintained on San Lorenzo River at Big Trees to monitor quality of runoff from this basin.

SAN LORENZO RIVER AT BIG TREES NEAR FELTON (STA. 75)

Sampling Point The sampling point for this station is located in Section 26, Township 10 South, Range 2 West, Mt. Diablo Base and Meridian, Canada del Rincon Grant. Monthly grab samples were collected from the right bank at Sequoia Gardens Resort, 1.7 miles south of Felton and east of State Highway 9.

Period of Record December 1951 through December 1959.

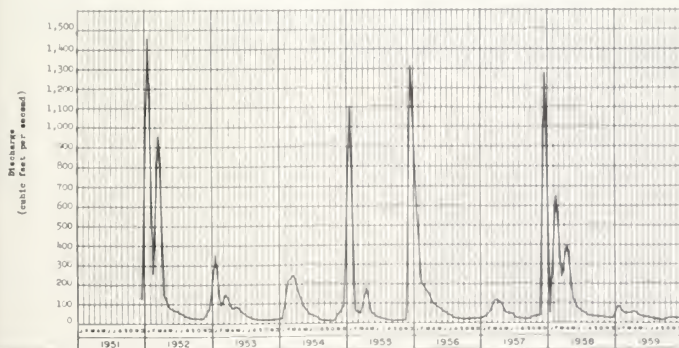
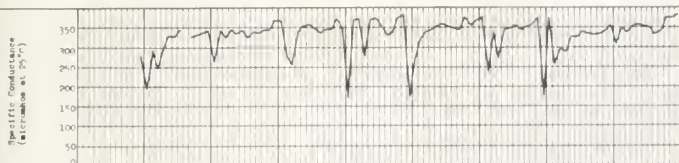
Water Quality Characteristics A review of past analyses of San Lorenzo River show it to be characterized by a calcium bicarbonate type water, relatively low in total dissolved solids, and slightly to moderately hard. During the period of record these waters have consistently qualified as a class 1 irrigation water. Likewise, concentrations of mineral constituents meet the criteria for domestic uses. The City of Santa Cruz uses San Lorenzo River water as a source of municipal supply.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1949	Minimum - 1959
Specific conductance (microhmoh at 25°C)	18	168	360	111
Temperature in °F	71	43	67	43
Dissolved oxygen in parts per million	1.1	0.5	11.8	9.2
Percent saturation	126	80	112	93
pH	8.2	6.8	7.9	7.3
Mineral constituents in parts per million				
Calcium (Ca)	43	19	16	19
Magnesium (Mg)	12	1.3	8.5	6.9
Sodium (Na)	28	14	28	17
Potassium (K)	2.7	1.4	2.1	1.8
Carbonate (CO ₃)	6	0.2	-	0.2
Bicarbonate (HCO ₃)	170	18	150	88
Sulfate (SO ₄)	28	28	40	28
Chloride (Cl)	32	6.5	32	16
Nitrate (NO ₃)	0.7	0.0	0.7	0.4
Fluoride (F)	0.3	0.09	0.1	0.1
Boron (B)	0.21	0.0	0.1	0.0
Silica (SiO ₂)	80	20	27	20
Total dissolved solids in parts per million	246	102	231	189
Percent sodium	31	20	31	22
Hardness as CaCO ₃ in parts per million				
Total	165	59	160	114
Noncarbonate	42	8	42	14
Turbidity	2,400	0.6	50	1
Coliform in most probable number per milliliter	>7,000	0.19	7,000	1.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.70	0.0	0.20	0.17
Solid alpha	1.73	0.00	7.20	0.17
Dissolved beta	15.31	0.0	7.74	3.13
Solid beta	7.32	0.0	5.80	0.00

WATER QUALITY VARIATIONS



SAN LORENZO RIVER AT BIG TREES (STA. 75)

Soquel Creek Basin

Soquel Creek drains an area of 91 square miles immediately south of the San Lorenzo River Basin in the northwest portion of the Central Coastal Region. Soquel Creek watershed has a mean annual runoff of about 63,500 acre-feet. Topography in the creek changes along its entire reach. The river originates in rugged mountains and flows southward through a gradual transition to rolling hills and finally, at the edge of the Pacific Ocean, it traverses a marine terrace.

Development in the river basin is primarily concentrated along the marine terrace bordering the lower reaches. Urban, agricultural, and light industrial development combine to sustain the growing population of the area.

Waste discharges from present development are minor and have not created notable impairment problems. A sand and gravel borrow area is located in the stream bed upstream from the town of Soquel and occasionally causes considerable turbidity in Soquel Creek.

A surface water sampling station is maintained on Soquel Creek at Soquel to monitor quality of runoff from this basin.

SOQUEL CREEK AT SOQUEL (STA. 76)

Sampling Point Soquel Creek is sampled in Section 10, Township 11 South, Range 1 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank at the USGS gage, which is located 0.25 mile upstream from the bridge on old Santa Cruz highway and about 1.2 miles from the mouth.

Period of Record December 1951 through December 1959.

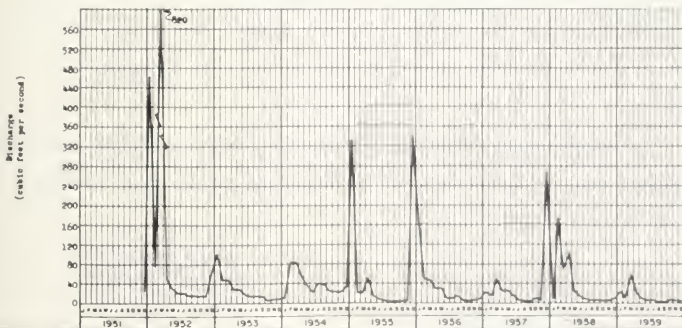
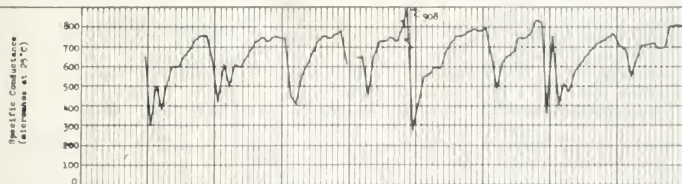
Water Quality Characteristics Water samples collected at this station exhibit a bicarbonate characteristic with no specific cation being consistently predominant, although the calcium or calcium and magnesium ions have displayed a predominance. This water qualifies as class 1 for irrigation, has mineral concentrations meeting the drinking water standards and a hardness ranging from slightly hard to very hard.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1972	Minimum - 1972
Specific conductance (micromhos at 25°C)	908	378	81	554
Temperature in °F	84	48	68	35
Dissolved oxygen in parts per million	17.6	7.1	11.5	7.1
Percent saturation	146	71	18	66
pH	8.5	7.1	8.1	7.2
Mineral constituents in parts per million				
Calcium (Ca)	84	28	76	28
Magnesium (Mg)	35	2.7	29	2.7
Sodium (Na)	74	15	67	27
Potassium (K)	6.8	1.3	5.5	1.3
Carbonate (CO ₃)	14	4	12	4
Bicarbonate (HCO ₃)	277	78	262	18
Sulfate (SO ₄)	187	75	188	71
Chloride (Cl)	14.9	1	92	2
Nitrate (NO ₃)	8.9	0.6	7.7	0.6
Fluoride (F)	8.4	0.3	7.4	0.2
Boron (B)	7.13	0.3	6.2	0.3
Silica (SiO ₂)	60	20	17	27
Total dissolved solids in parts per million	573	175	411	350
Percent sodium	34	20	32	21
Hardness as CaCO ₃ in parts per million				
Total	321	72	309	215
Noncarbonate	118	0	118	19
Turbidity	260	8.0	8	0.7
Coliform in most probable number per milliliter	7,000	0.62	2,400	1.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.41	0.00	0.11	0.00
Solid alpha	1.06	0.00	1.04	0.15
Dissolved beta	14.1	0.00	5.56	0.00
Solid beta	7.09	0.00	7.09	1.38

WATER QUALITY VARIATIONS



SOQUEL CREEK AT SOQUEL (STA. 76)

Pajaro River Basin

The Pajaro River drainage basin encompasses 1,303 square miles in the northwestern portion of the Central Coastal Region. Llagas Creek, Pacheco Creek and several other smaller streams combine to form the Pajaro River in the lower end of South Santa Clara Valley. The confluence of the San Benito and Pajaro Rivers is located just east of Pajaro Gap and less than ten miles downstream from Pajaro River formation point. Only 116 square miles of drainage area contribute to the runoff to the Pajaro River below Pajaro Gap. The Pajaro River Basin has an average annual runoff of about 222,500 acre-feet.

Mountains and foothills cover almost 80 percent of the land area in this basin. The three valley fill areas, Pajaro, South Santa Clara and Hollister, comprise 280 square miles of potential or already developed agricultural lands. Agriculture is the predominant user of water and is the major factor in economic development in the area.

Waste discharges entering waterways of this basin originate from urban, light industrial, and individual domestic sources. Excepting for the waste outflows from the Cities of Gilroy (2.75 mgd) and Hollister (0.50 mgd) these discharges do not exceed 0.50 mgd. Irrigation return flows are the major source of quality impairment in the Pajaro River Basin.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussed</u>
Pajaro River near Chittenden	118
Uvas Creek near Morgan Hill	120
San Benito River near Bear Valley Fire Station	122

PAJARO RIVER NEAR CHITTENDEN (STA. 77)

Sampling Point Station 77 is located in Section 12, Township 12 South, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the bridge crossing on Chittenden Road at the Santa Cruz-San Benito County Line. The sampling point is located 1 mile southeast of Chittenden and 2.5 miles downstream from the San Benito River confluence.

Period of Record December 1951 through December 1959.

Water Quality Characteristics Since inception of a monitoring station on Pajaro River analysis of the water reveals bicarbonate to be the predominant anion and calcium, magnesium, and sodium cations to be nearly equal in equivalents per million. The water at this station is generally class 2 for irrigation. In 1954, a boron concentration of 2.0 ppm was found, which is the upper limit for a class 2 water. In 1957 a chloride concentration of 374 ppm made waters from Pajaro River class 3. Pajaro River water also has, on occasion, exceeded the mineral criteria for drinking water and generally ranges from moderately hard to very hard.

Significant Water Quality Changes During 1959, in May, September, and December, standard mineral and heavy metals analyses, concentrations of certain mineral constituents were found to exceed the criteria for drinking water. In particular, manganese was found to be 6.5 ppm, which greatly exceeded the previously recorded maximum of 0.02 ppm. The cause of this extreme has not been ascertained. The water, during 1959, was consistently very hard, substantiating an apparent trend that Pajaro River water is becoming harder each year.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum 1959
Specific conductance (micromhos at 25°C)	2,251	268	1,480	64
Temperature in °F	77	45	76	50
Dissolved oxygen in parts per million	14.7	9.4	9.9	6.8
Percent saturation	121	88	100	77
pH	8.4	5.1	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	11.9	28	10.0	6.4
Magnesium (Mg)	9.6	17	7.5	4.8
Sodium (Na)	27.1	29	17.7	9.6
Potassium (K)	17	1.7	12	3
Carbonate (CO ₃)	36	8.1	1.1	0.0
Bicarbonate (HCO ₃)	620	172	620	207
Sulfate (SO ₄)	304	67	298	67
Chloride (Cl)	174	8	132	37
Nitrate (NO ₃)	15	0.1	15	0.8
Fluoride (F)	0.6	0.0	0.1	0.1
Boron (B)	2.0	0.18	0.7	0.1
Silica (SiO ₂)	12	10	32	21
Total dissolved solids in parts per million	1,400	149	935	400
Percent sodium	75	21	48	21
Hardness as CaCO ₃ in parts per million				
Total	625	106	625	251
Noncarbonate	310	0.0	310	0.0
Turbidity	1,600	0.4	35	1
Coliform in most probable number per milliliter	97,000	0.21	2,400	0.62
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.86	0.00	0.00	0.00
Solid alpha	0.59	0.00	0.41	0.18
Dissolved beta	11.71	0.00	5.55	4.00
Solid beta	9.69	0.00	9.29	1.6

WATER QUALITY VARIATIONS



PAJARO RIVER NEAR CHITTENDEN (STA. 77)

UVAS CREEK NEAR MORGAN HILL (STA. 96)

Sampling Point Uvas Creek is sampled in Section 18, Township 10 South, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected immediately below Uvas Dam at the outlet, about 0.6 mile downstream from Eastman Canyon and 4.8 miles southwest of Morgan Hill.

Period of Record July 1952 through December 1959.

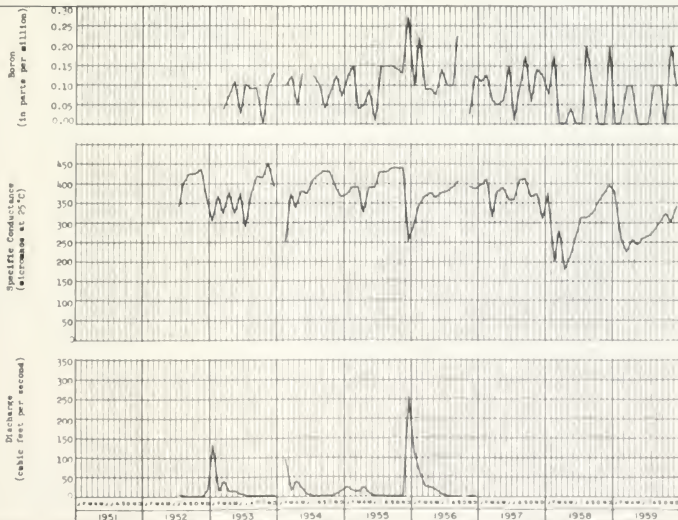
Water Quality Characteristics Chemical classification of past analyses of this water revealed a predominant bicarbonate anion and the principal cations to be calcium and magnesium, neither of which is consistently predominant. Uvas Creek water is class 1 for irrigation, meets the drinking water mineral limits, and ranges from slightly hard to very hard. This water is a source of municipal supply for the City of Gilroy.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	451	181	374	280
Temperature in °F	80	45	71	47
Dissolved oxygen in parts per million	12.0	5.6	11.9	7.3
Percent saturation	110	66	104	70
pH	8.5	7.0	7.7	7.1
Mineral constituents in parts per million				
Calcium (Ca)	52	13	36	25
Magnesium (Mg)	42	10	13	12
Sodium (Na)	41	5.6	11	6.7
Potassium (K)	1.7	0.6	2.4	1.1
Carbonate (CO ₃)	17	0.0	2	0.0
Bicarbonate (HCO ₃)	238	95	183	115
Sulfate (SO ₄)	44	9.6	21	9.6
Chloride (Cl)	14	4.5	9.0	5.0
Nitrate (NO ₃)	1.6	0.0	1.5	0.1
Fluoride (F)	0.2	0.0	0.0	0.0
Boron (B)	0.27	0.0	0.2	0.0
Silica (SiO ₂)	26	12	13	12
Total dissolved solids in parts per million	378	113	231	140
Percent sodium	32	9	15	12
Hardness as CaCO ₃ in parts per million				
Total	222	82	172	104
Noncarbonate	28	4	22	7
Turbidity	140	0.0	95	2
Coliform in most probable number per milliliter	>7,000.	<0.065	2,400.	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.90	0.00	0.90	0.10
Solid alpha	0.50	0.00	0.00	0.00
Dissolved beta	6.40	0.00	4.40	0.00
Solid beta	6.65	0.00	6.65	0.00

WATER QUALITY VARIATIONS



UVAS CREEK NEAR MORGAN HILL (STA. 96)

SAN BENITO RIVER NEAR BEAR VALLEY FIRE STATION (STA. 77a)

Sampling Point The location of this sampling station is in Section 28, Township 15 South, Range 7 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank about 1.7 miles downstream from Willow Creek, 10.4 miles northwest of San Benito, and 3.0 miles north of Bear Valley Fire Station.

Period of Record July 1958 through December 1959.

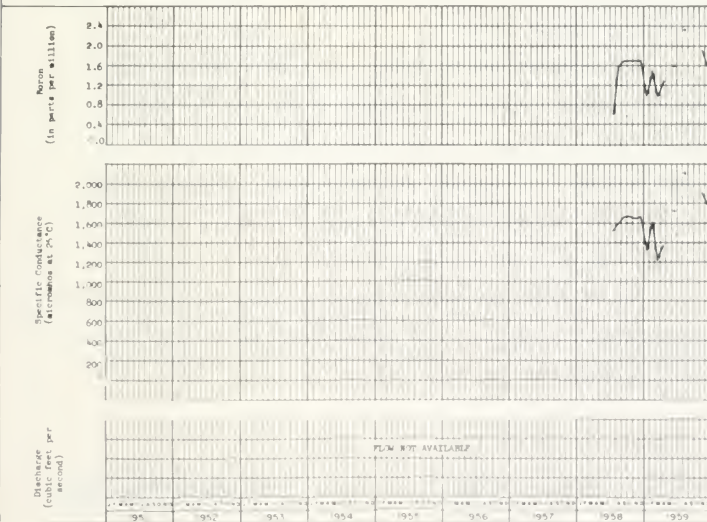
Water Quality Characteristics A review of past analyses shows the principal cations in San Benito River to be magnesium and sodium, and the principal anions to be bicarbonate and sulfate. This water is usually class 2 for irrigation except for one boron determination of 2.3 ppm which occurred in August 1959, making it class 3 at that time. Sulfate and total dissolved solids concentrations make this water unsatisfactory for domestic use. This water is extremely hard with a range from 476 ppm to 596 ppm total hardness.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	2,130	1,220	2,130	1,220
Temperature in °F	86	49	85	50
Dissolved oxygen in parts per million	19.6	8.0	11.2	8.0
Percent aeration	126	86	126	94
pH	8.4	8.0	8.1	8.1
Mineral constituents in parts per million				
Calcium (Ca)	52	2.4	52	2.4
Magnesium (Mg)	114	0.1	114	0.1
Sodium (Na)	292	114	292	114
Potassium (K)	7.2	2.7	7.2	2.7
Carbonate (CO ₃)	29	0.0	29	0.0
Bicarbonate (HCO ₃)	526	417	526	417
Sulfate (SO ₄)	464	192	464	192
Chloride (Cl)	195	64	195	64
Nitrate (NO ₃)	1.9	0.0	1.9	0.0
Fluoride (F)	0.5	0.2	0.5	0.2
Boron (B)	2.3	0.6	2.3	1.0
Silica (SiO ₂)	16	4.0	16	4.0
Total dissolved solids in parts per million	1,390	756	1,390	756
Percent sodium	52	33	52	33
Hardness as CaCO ₃ in parts per million				
Total	596	476	596	476
Noncarbonate	233	64	147	64
Turbidity	9	1	9	1
Coliform in most probable number per milliliter	230.	0.23	230.	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.00			
Solid alpha	0.00			
Dissolved beta	0.20			
Solid beta	1.33			

WATER QUALITY VARIATIONS



SAN BENITO RIVER NEAR BEAR VALLEY FIRE STATION (STA. 77a)

Salinas River Basin

The Salinas River system drains an area of about 4,400 square miles which cover the central portion of the Central Coastal Region. Mountains and foothills cover approximately 3,480 square miles and valley and mesa land occupy the remaining 220 square miles. From its coast line along Monterey Bay the basin extends southeasterly about 150 miles. Mean seasonal runoff in the Salinas River watershed is 713,800 acre-feet.

The Salinas River meanders along its course through Salinas Valley for nearly 100 miles. The economy of the basin is based on the agricultural complex which has developed on the fertile valley floor. Water requirements of the basin for domestic, municipal, industrial and irrigation are supplied mostly from ground water. In recent years several small dams on tributaries to the Salinas River have provided surface water supplies to the upper end of the valley.

Waste disposal and irrigation return water have not created serious deleterious effects on the quality of water in the basin. Significant waste discharges entering this stream system are for the most part outflows from community sewerage systems. Prominent among these are Alisal Sanitary District (1.3 mgd), King City (0.4 mgd), City of Salinas (2.93 mgd), and Soledad Prison (0.5 mgd).

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

Monitoring Station

Page Number of
Station Discussed

Salinas River at Paso Robles*
Salinas River near Bradley*
Salinas River near Spreckels
Nacimiento River near San Miguel*
San Antonio River at Pleyto*

126

* Monitoring stations are in Southern California and will be discussed in Part II of this bulletin.

SALINAS RIVER NEAR SPRECKELS (STA. 43)

Sampling Point Station 43 is located in Section 8, Township 15 South, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gaging station 50 yards upstream from the bridge 4 miles south of Salinas and 2.0 miles west of Spreckels.

Period of Record April 1951 through May 1957 and April 1958 through December 1959. Salinas River at this station is dry during a portion of most years and data are not available for all months.

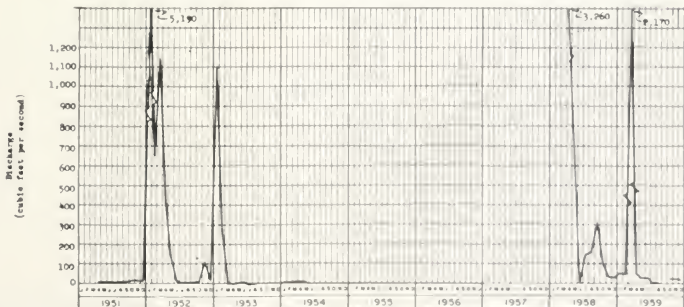
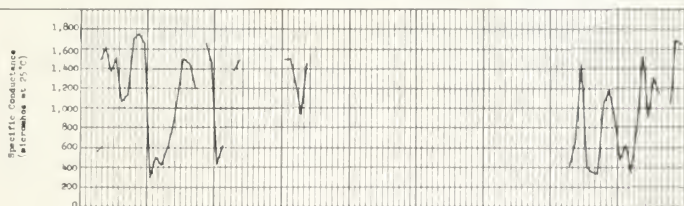
Water Quality Characteristics Antecedent data reveal water at this station to be characteristically a bicarbonate type with the cations nearly evenly divided between calcium, magnesium and sodium. From an irrigation standpoint Salinas River water is class 2 due to boron and dissolved solids concentrations. Total dissolved solids also exceed the limits recommended for a domestic supply. This water ranges from moderately hard to very hard. Effluent from Alisal Sanitary District Sewage Treatment Plant is discharged to the river about 100 yards upstream.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	1,773	107	1,666	152
Temperature in °F	81	48	33	28
Dissolved oxygen in parts per million	12.2	7.3	11.9	7.7
Percent saturation	136	77	136	73
pH	8.1	6.6	8.0	7.1
Mineral constituents in parts per million				
Calcium (Ca)	112	12	110	
Magnesium (Mg)	48	13	40	
Sodium (Na)	176	13	180	12
Potassium (K)	34	1.8	10	
Carbonate (CO ₃)	21	0.8	3	0.0
Bicarbonate (HCO ₃)	910	134	872	143
Sulfate (SO ₄)	206	40	190	
Chloride (Cl)	12	14	190	14
Nitrate (NO ₃)	19	0.0	0.2	
Fluoride (F)	2.0	0.2	0.3	
Boron (B)	0.6	0.0	0.4	0.1
Silica (SiO ₂)	44	22	33	
Total dissolved solids in parts per million	1,030	180	987	298
Percent sodium	56	19	56	19
Hardness as CaCO ₃ in parts per million				
Total	650	112	694	146
Noncarbonate	263	0.0	263	0.0
Turbidity	400	0.0	390	0.5
Coliform in most probable number per milliliter	62,000	0.06	7,000	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.72	0.00	0.20	
Solid alpha	0.47	0.00	0.41	
Dissolved beta	27.48	0.00	0.00	
Solid beta	9.40	0.00	2.39	

WATER QUALITY VARIATIONS



SALINAS RIVER NEAR SPRECKELS (STA. 43)

Carmel River Basin

In the north central portion of the Central Coastal Region and immediately south of Monterey Bay lies the Carmel River watershed. The area of this basin is 254 square miles, of which nearly 249 square miles comprise foothills and mountains. The Carmel River Basin has a mean annual discharge of about 142,300 acre-feet.

Development in this area is centered around Carmel Valley and in the widely known resort area on Monterey Peninsula. Irrigated lands in the valley, and urban and domestic requirements of the peninsula, are the major water users in the basin.

Wastes in this basin, for the most part, are discharged to the Pacific Ocean. Wastes entering Carmel River are minor in quantity and do not create serious impairment problems.

A surface water sampling station is maintained on Carmel River at Robles del Rio to monitor quality of runoff from this basin.

CARMEL RIVER AT ROBLES DEL RIO (STA. 83)

Sampling Point Station 83 is located in Section 17, Township 16 South, Range 1 East, Mt. Diablo Base and Meridian. The station was sampled monthly from mid-stream from Robles del Rio bridge in the town of Robles del Rio.

Period of Record January 1952 through December 1959. Carmel River has been dry on several occasions during this period, therefore, data are not available for all months.

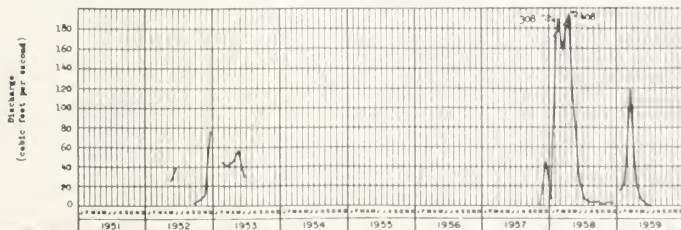
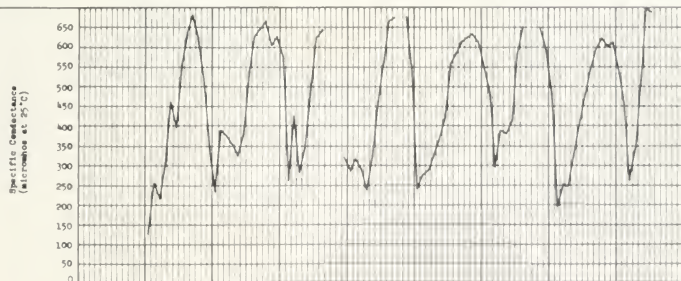
Water Quality Characteristics A review of past analyses reveals that the water at Station 83 exhibits no predominant cation or anion. Calcium, magnesium and sodium are all present in significant proportions, while the principal anions are bicarbonate and sulfate. The water consistently qualifies as class 1 for irrigation and meets the mineral standards for drinking water. Waters at this station have ranged from slightly hard to very hard.

Significant Water Quality Changes During 1959, samples of Carmel River water revealed the previous recorded maximum concentrations of conductivity, total dissolved solids and hardness were exceeded. Extremely low flows were encountered in June and July, possibly accounting for the excessive mineral concentrations found in samples collected during these two months. The river was dry during the last five months of 1959.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (microhm at 25°C)	188	1	187	1
Temperature in °F	78	32	46	32
Dissolved oxygen in parts per million	16.4	0.2	16.2	0.2
Percent saturation	188	1	187	1
pH	8.1	7.0	1.1	1.1
Mineral constituents in parts per million				
Calcium (Ca)	61	2	59	2
Magnesium (Mg)	11	1	10	1
Sodium (Na)	5	11	6	11
Potassium (K)	1.4	1.5	0.1	0.1
Carbonate (CO ₃)	29	0.8	28.2	0.8
Bicarbonate (HCO ₃)	174	16	158	16
Sulfate (SO ₄)	62	6	56	6
Chloride (Cl)	1.7	0.5	1.2	0.5
Nitrate (NO ₃)	0.5	0.2	0.3	0.2
Fluoride (F)	0.24	0.0	0.24	0.0
Silice (SiO ₂)	27	21	6	6
Total dissolved solids in parts per million	431	41	390	41
Percent sodium	43	22	21	22
Hardness as CaCO ₃ in parts per million				
Total	256	44	212	44
Noncarbonate	95	7	88	7
Turbidity	90	0.0	90	0.0
Coliforms in most probable number per milliliter	7,000	<0.005	7,000	<0.005
Radioactivity in micro-micro curies per liter				
Dissolved alpha	2.42	0.00	2.42	0.00
Solid alpha	0.17	0.00	0.17	0.00
Dissolved beta	7.0	0.00	7.0	0.00
Solid beta	7.22	0.00	7.22	0.00

WATER QUALITY VARIATIONS



CARMEL RIVER AT ROBLES DEL RIO (STA. 83)

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STREAM SAMPLING STATIONS
CENTRAL COASTAL REGION (NO. 3)

Station
Number

Station Name

43	Salinas River near Spreckels
75	San Lorenzo River at Big Trees near Felton
76	Soquel Creek at Soquel
77	Pajaro River near Chittenden
77a	San Benito River near Bear Valley Fire Station
83	Carmel River at Robles del Rio
96	Uvas Creek near Morgan Hill

44460



Central Valley Region (No. 5)

The Central Valley Region occupies about one-third of the total area of California and is comprised of all stream basins which drain into the Sacramento and San Joaquin Valleys. The basin extends from the Oregon-California border in the northeastern corner of the State to the crest of the Tehachapi Mountains 60 miles north of the City of Los Angeles. In order to facilitate discussion of this region it was subdivided into four separate areas as listed below:

<u>Name of Areas</u>	<u>Numerical Designation</u>
Sacramento River Valley	5a
San Joaquin River Valley	5b
Sacramento-San Joaquin Delta	5c
Tulare Lake Drainage	5d

To monitor quality of water in this basin samples are collected from 119 stations on 50 separate watercourses and 1 lake as shown on Plate 1. The four areas and their respective watercourses are as listed:

Sacramento River Valley (5a)

Sacramento River	Stony Creek
Colusa Trough	Clear Lake
McCloud River	Cache Creek
Pit River	Putah Creek
Clear Creek	Antelope Creek
Cow Creek	Mill Creek
Cottonwood Creek	Big Chico Creek
Battle Creek	Butte Creek
Paynes Creek	Feather River
Redbank Creek	Indian Creek
Elder Creek	Yuba River
Thomes Creek	Bear River
	American River

San Joaquin River Valley (5b)

San Joaquin River	Bear Creek
Salt Slough	Merced River
Fresno River	Tuolumne River
Chowchilla River	Stanislaus River

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Paynes Creek	Feather River
Redbank Creek	Indian Creek
Elder Creek	Yuba River
Thomes Creek	Bear River
	American River

San Joaquin River Valley (5b)

San Joaquin River	Bear Creek
Salt Slough	Merced River
Fresno River	Tuolumne River
Chowchilla River	Stanislaus River

Sacramento-San Joaquin Delta (5c)

Lindsey Slough	Delta-Mendota Canal
Sacramento River	Italian Slough
Delta Cross Channel	Indian Slough
Little Potato Slough	Rock Slough
San Joaquin River	Cosumnes River
Stockton Ship Channel	Mokelumne River
Old River	Calaveras River
Grant Line Canal	

Tulare Lake Drainage (5d)

Kings River	Tule River
Kaweah River	Kern River

Five new stations were added to the surface water monitoring program in Region 5 during 1959. Sampling was commenced in January on Big Chico Creek at Chico (85a), Redbank Creek at Foothills (88d), Elder Creek at Gerber (95a), Thomes Creek near mouth (95b), and Bear Creek at Merced (111a). These stations were established to determine base level quality conditions at proposed water conservation project sites and to provide water quality monitoring on streams where coverage was deficient.

Waters in the Central Valley Region vary in quality from excellent to poor, depending on locality, flow and degradents encountered. Waters emanating from the Sierra Nevada, Cascade and Trinity Mountains were generally of excellent quality, while surface runoff from the Tehachapi Mountains in the south and the coastal ranges along the western perimeter varied from excellent to poor. Waters in the Sacramento Valley and foothill slopes of the San Joaquin, Sacramento-San Joaquin Delta, and Tulare Lake drainage were generally calcium bicarbonate in character. Waters in the San Joaquin Valley floor and in the Sacramento-San Joaquin Delta proper were usually sodium chloride in character due to the effect of such degradents as irrigation returns, ground water accretions, and sea-water incursion.

Sparse precipitation during 1959 resulted in an increase in most chemical constituents in waters of the Central Valley Region. However, with the exception of the delta area, the increase was usually not significant. In the delta, the paucity of tributary inflow and the continued heavy diversions for irrigation use in the delta uplands area resulted in significant degradation to surface waters from ground water accretions, irrigation returns, and sea-water incursion.

Sacramento River Valley (5a)

The Sacramento River Valley embraces all of the watersheds tributary to Sacramento River upstream from the southern drainage boundaries of Putah Creek and the American River hydrographic units. The basin extends north-south approximately 270 miles and contains over 26,000 square miles of highly variable terrain.

Mountains and foothills cover about 65 percent of the area, with the Sierra Nevada dominating the eastern portion, the Coastal Range to the west, and the Klamath Mountains and Cascade Range on the north. Bountiful valley and mesa lands exist in this area, with the extensive agricultural lands of the Sacramento Valley being predominant in this land class.

Mean seasonal surface runoff exceeds 22,300,000 acre-feet in the area. To facilitate discussion of the numerous drainage areas in this area, they are segregated into the following units with the number of sampling station of each in parentheses:

- Sacramento River Unit (10)
- McCloud River Basin (1)
- Pit River Basin (4)
- Redding Stream Unit (7)
- West Side Stream Unit (12)
- Sacramento Valley Northeast Stream Unit (6)
- Feather River Basin (4)
- Yuba-Bear Rivers Unit (4)
- American River Basin (4)

Sacramento River Unit. The Sacramento River Unit extends from the northwesternmost corner of the Central Valley Region through the entire length of the Sacramento River Valley. The unit includes the drainage area of the Sacramento River above Shasta Reservoir, and the valley floor area of Sacramento Valley below Red Bluff. Mountainous terrain occupies all but a few of the 618 square miles along the reach of the river above Shasta Reservoir; while along its course below Red Bluff only Sutter Buttes breaks the 4,946 square miles of flat, gently rolling valley floor. Mean seasonal natural runoff for this unit is about 1,220,000 acre-feet.

Development along the upper reach is primarily associated with recreation or lumbering. Along the valley floor, agriculture and its allied food-processing industries are the primary economic endeavors. Mining, production of natural gas, recreation, and in recent years development related to military and aircraft programs, provide additional economic stability to the valley. These developments use considerable quantities of surface and ground water in their operations. Continued growth of the industrial and urban complex centers as well as irrigated agricultural expansion depend upon controlling, quantitatively and qualitatively, the water supply of the unit.

Waste discharges originating from industrial and municipal developments enter this major waterway along its entire length. In the upper reaches lumbermill effluent and sanitary sewage from resort communities constitute the major waste sources. In the valley floor area, lumber by-product industries, cities and towns, light industries, food product plants, and a considerable volume of irrigation return flow

all combine to impose a significant waste load on the Sacramento River. A study is presently being conducted by the Department of Water Resources to evaluate the present effect of waste discharges and to determine the waste assimilating capacity of the Sacramento River. The major discharges entering the river and their approximate quantities in million gallons per day (mgd) are listed:

City of Redding	2.0 mgd
City of Red Bluff	1.0 mgd
Diamond Gardner International Corporation	5.0 mgd
City of Corning (Intermittently)	0.3 mgd
Natomas Drain (McClellan Air Force Base, City of Rio Linda, and North Sacramento)	6.3 mgd
City of Sacramento	50.0 mgd
City of Mountain View	6.0 mgd
City of Rio Linda	0.05 mgd
City of West Sacramento	2.4 mgd
American Crystal Sugar Refining (Clarksburg)	5.0 mgd

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed.

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Sacramento River at Delta	140
Sacramento River at Keswick	142
Sacramento River at Bend (Red Bluff)	144
Sacramento River near Hamilton City	146
Sacramento River at Butte City	148
Sacramento River at Colusa	150
Sacramento River at Knights Landing	152
Sacramento River at Sacramento	154
Colusa Trough near Colusa	156
Sacramento Slough near Knights Landing	158

SACRAMENTO RIVER AT DELTA (STA. 11)

Sampling Point Station 11 is located in Section 35 of Township 36 North, Range 5 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage 0.2 mile downstream from Dog Creek and 0.6 mile southeast of Delta.

Period of Record April 1951 through December 1959.

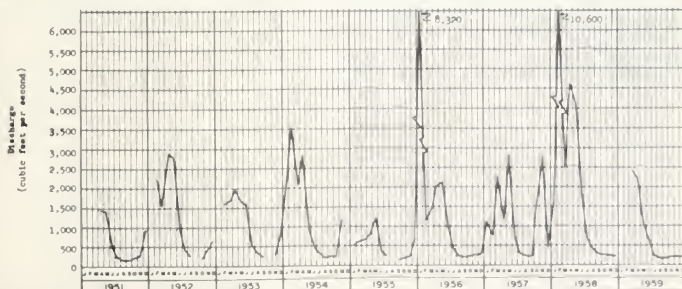
Water Quality Characteristics During periods of higher surface runoff, the water is magnesium bicarbonate in character. In summer months, however, when flows are comparatively low, calcium, magnesium, and sodium are about equal in importance. Very little variation in quality is noted at this point, and the water is of excellent quality for all uses. On rare occasions the water is slightly hard; however, it is soft the majority of the time. Total radioactivity reached 25.2 micro-micro curies per liter in May 1956, which is the highest value noted during the period of record.

Significant Water Quality Changes During 1959 total hardness reached 62 ppm, and noncarbonate hardness 6 ppm. These values are significant only inasmuch as they are the extremes for the period of record.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	289	55.7	169	81.7
Temperature in °F	80	56	74	40
Dissolved oxygen in parts per million	15.3	6.9	19.4	8.9
Percent saturation	124	68	107	86
pH	8.4	7.1	8.3	7.3
Mineral constituents in parts per million				
Calcium (Ca)	11	3.4	7.6	5.8
Magnesium (Mg)	8.9	4.0	8.5	6.0
Sodium (Na)	15	1.4	19	9.9
Potassium (K)	2.2	0.1	1.7	0.3
Carbonate (CO ₃)	3	0.0	3	0.0
Bicarbonate (HCO ₃)	87	39	83	49
Sulfate (SO ₄)	5.0	1.0	9.9	9.0
Chloride (Cl)	19	0.0	11	9.5
Nitrate (NO ₃)	0.6	0.0	0.9	0.0
Fluoride (F)	0.3	0.0	0.1	0.0
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	41	19	34	20
Total dissolved solids in parts per million	170	40	117	58
Percent sodium	38	9	33	11
Hardness as CaCO ₃ in parts per million				
Total	69	26	69	14
Noncarbonate	6	0.0	6	0.0
Turbidity	40	0.0	30	1
Coliforms in most probable number per milliliter	7,000.	0.62	230.	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.32	0.00	0.58	0.36
Solid alpha	2.81	0.00	0.59	0.18
Dissolved beta	19.6	0.00	1.36	0.00
Solid beta	9.18	0.00	6.73	2.46

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT DELTA (STA. 11)

SACRAMENTO RIVER AT KESWICK (STA. 12)

Sampling Point Station 12 is situated in Section 28 of Township 32 North, Range 5 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at the USGS gage, 0.6 mile downstream from Keswick Dam, 0.6 mile upstream from Middle Creek and 10 miles downstream from Shasta Dam.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Antecedent data show the water at Station 12 to be of excellent quality, soft to slightly hard, and to vary only slightly in mineral content. However, on several occasions during recent years, comparatively large concentrations of heavy metals coming from Spring Creek have been sufficient to kill fish in the vicinity of this station. Streams draining the Spring Creek watershed frequently are acidic and have undesirable heavy metal concentrations and other toxic salts leached from tailings of both operating and abandoned mines. This situation has been partially corrected through increased releases from Shasta Reservoir coincident with increases of surface runoff in Spring Creek. The water at Station 12 is bicarbonate in type with calcium slightly dominant over other cations. This water is class 1 for irrigation, and excellent for domestic and industrial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	139	86.5	134	107
Temperature in °F	60	41	59	50
Dissolved oxygen in parts per million	11.4	6.8	11.2	8.0
Percent saturation	119	57	99	79
pH	7.7	6.5	7.7	7.0
Mineral constituents in parts per million				
Calcium (Ca)	18	8.3	11	8.8
Magnesium (Mg)	6.3	1.9	6.3	4.3
Sodium (Na)	9.8	3.9	8.5	3.9
Potassium (K)	1.9	0.7	1.4	1.1
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	76	47	76	52
Sulfate (SO ₄)	11.0	2.9	9.6	4.8
Chloride (Cl)	6.2	0.0	4.0	2.2
Nitrate (NO ₃)	1.0	0.0	1.0	0.5
Fluoride (F)	0.3	0.0	0.3	0.1
Boron (B)	0.18	0.0	0.1	0.0
Silica (SiO ₂)	29	15	29	25
Total dissolved solids in parts per million	102	67	102	79
Percent sodium	30	15	26	15
Hardness as CaCO ₃ in parts per million				
Total	66	36	53	40
Noncarbonate	6	0.0	5	0.0
Turbidity	80	0.0	35	3
Coliform in most probable number per milliliter	7,000	<0.045	23	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.33	0.00	0.58	0.00
Solid alpha	0.96	0.00	0.47	0.29
Dissolved beta	13.8	0.00	8.26	2.42
Solid beta	24.9	0.00	13.30	0.70

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT KESWICK (STA. 12)

SACRAMENTO RIVER AT BEND (STA. 12c)

Sampling Point Bend station is located in Section 20 of Township 28 North, Range 3 West, Mt. Diablo Base and Meridian. Daily composite and monthly grab samples were collected from the left bank 100 yards downstream from Bend Road bridge, 4.0 miles upstream from the mouth of Paynes Creek and approximately 6.0 miles north of Red Bluff.

Period of Record May 1955 through December 1959.

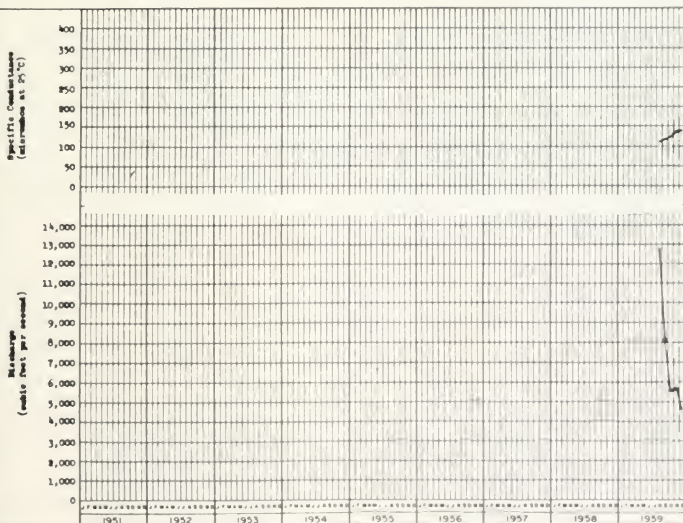
Water Quality Characteristics Sacramento River at Station 12c is bicarbonate in type with calcium the predominant cation, soft to slightly hard, class 1 for irrigation use and suitable for domestic and industrial use. Only minor variations in quality have been noted at this point during the period of record. There is no significant difference in quality of Sacramento River water between Station 12 near Redding and the Bend station.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	179	71.2	145	71.2
Temperature in °F	60	54	60	44
Dissolved oxygen in parts per million	10.4	9.6	10.4	9.6
Percent saturation	101	94	101	94
pH	8.0	6.8	7.8	6.2
Mineral constituents in parts per million				
Calcium (Ca)	15	7.6	13	8.0
Magnesium (Mg)	7.1	2.4	6.7	1.6
Sodium (Na)	11	3.2	8.7	1.2
Potassium (K)	3.0	0.6	2.1	0.7
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	44	28	79	28
Sulfate (SO ₄)	16	1.9	16	1.0
Chloride (Cl)	8.0	0.1	6.0	1.8
Silicate (SiO ₂)	4.2	0.0	1.7	0.0
Fluoride (F)	0.3	0.0	0.2	0.0
Boron (B)	0.46	0.0	0.1	0.0
Silica (SiO ₂)	45	14	37	16
Total dissolved solids in parts per million	143	48	120	48
Percent sodium	28	16	27	18
Hardness as CaCO ₃ in parts per million				
Total	66	28	60	30
Noncarbonate	7	0.0	7	0.0
Turbidity (Not Measured)				
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.0		0.0	
Solid alpha	2.27		2.27	
Dissolved beta	8.26		8.26	
Solid beta	4.81		4.81	

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT BEND (STA. 12c)

SACRAMENTO RIVER NEAR HAMILTON CITY (STA. 13)

Sampling Point Station 13 is located in Section 20 of Township 22 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a state highway bridge 10.5 miles west of Chico, 1.2 miles northeast of Hamilton City, and about 6.0 miles upstream from the mouth of Stony Creek.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Past analyses show the water at the sampling point to be bicarbonate in type with calcium usually predominant. The water is soft to slightly hard, class 1 for irrigation, meets drinking water requirements for mineral content, and is excellent for industrial use. There have been no significant variations noted at this point during the period of record.

Significant Water Quality Changes There were no significant changes in water quality at this point during 1959 with the exception of total radioactivity. In the September sample, 18.4 micro-micro curies per liter were present, which represents the highest value reported at this point for the period of record. This value, however, is well below the recommended safe limit.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1957	Minimum 1959
Specific Conductance (micromhos at 25°C)	161	37	180	111
Temperature in °F	71	41	71	37
Dissolved oxygen in parts per million	12.2	1.0	11.4	5.1
Percent saturation	120.7	33	109	51
pH	8.2	6.8	7.5	7.2
Mineral constituents in parts per million				
Calcium (Ca)	11	2.8	11	8.0
Magnesium (Mg)	7	1.1	7	5.2
Sodium (Na)	117	4	117	4.1
Potassium (K)	2.1	0.4	1.7	1.1
Carbonate (CO ₃)	12.0	0	12	0
Bicarbonate (HCO ₃)	12.0	0	12	0
Sulfate (SO ₄)	4.2	0.2	4.2	0.2
Chloride (Cl)	0.2	0	0.2	0
Nitrate (NO ₃)	1.1	0	1.1	0
Fluoride (F)	1	0	1	0
Boron (B)	0.1	0	0.1	0
Silica (SiO ₂)	20	20	20	20
Total dissolved solids in parts per million	126	48	177	86
Percent sodium	75	14	68	16
Hardness as CaCO ₃ in parts per million				
Total	48	17	60	41
Noncarbonate	6	0	0	0
Turbidity	350	0.0	30	0
Coliform in most probable number per milliliter	>7,000	0.06	7,000	6.2
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.13	0.00	0.00	0.00
Solid alpha	1.43	0.00	0.00	0.00
Dissolved beta	12.00	0.00	0.00	0.00
Solid beta	14.08	0.00	0.00	0.00

WATER QUALITY VARIATIONS



SACRAMENTO RIVER NEAR HAMILTON CITY (STA. 13)

SACRAMENTO RIVER AT BUTTE CITY (STA. 87a)

Sampling Point Station 87a is located in Section 32 of Township 19 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab and daily composite samples were collected at the highway bridge just ~~downstream~~ from the gaging station and 0.5 mile south of Butte City.

Period of Record May 1955 through December 1959.

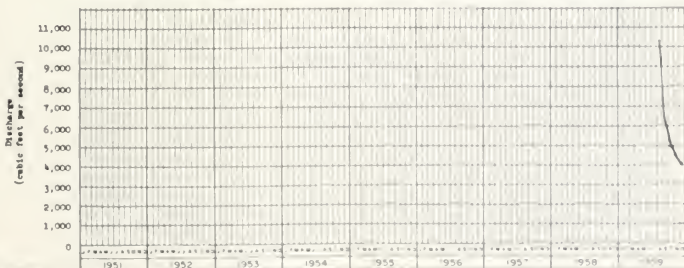
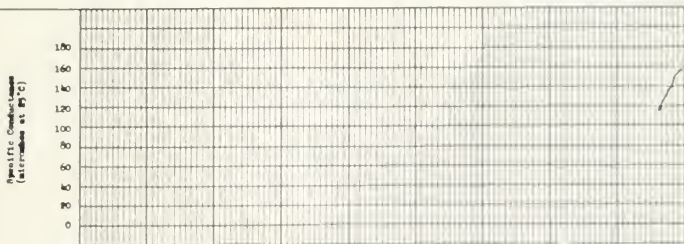
Water Quality Characteristics Analyses of daily composite samples show the water at Station 87a to be bicarbonate in type with calcium the major cation, soft to slightly hard, class 1 for irrigation use, and within drinking water requirements for mineral content. Comparison of analyses of samples from this station with those from the Sacramento River at Hamilton City (Station 13) reveal no significant difference in mineral quality.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	174	88.9	164	88.9
Temperature in °F				
Dissolved oxygen in parts per million				
Percent saturation				
pH	8.1	6.6	8.1	6.8
Mineral constituents in parts per million				
Calcium (Ca)	17	8.0	15	11
Magnesium (Mg)	8.6	1.9	7.7	1.2
Sodium (Na)	8.0	4.1	9.3	5.1
Potassium (K)	2.0	0.8	2.0	1.3
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	69	40	84	40
Sulfate (SO ₄)	16	1.9	16	1.3
Chloride (Cl)	8.5	1.0	8.5	2.6
Nitrate (NO ₃)	4.5	0.0	3.1	0.0
Fluoride (F)	0.7	0.0	0.3	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	35	14	35	20
Total dissolved solids in parts per million	126	65	120	72
Percent sodium	27	18	27	21
Hardness as CaCO ₃ in parts per million				
Total	70	35	64	38
Noncarbonate	11	0.0	11	0.0
Turbidity				
Coliform in most probable number per milliliter				
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT BUTTE CITY (STA. 87a)

SACRAMENTO RIVER AT COLUSA (STA. 13b)

Sampling Point Colusa station is located within Section 32 of Township 19 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the River Road bridge at Colusa.

Period of Record October 1958 through December 1959.

Water Quality Characteristics Water at this station is excellent in quality, bicarbonate in type, with calcium somewhat dominant over the cations. It is soft to slightly hard, class 1 for irrigation, and meets drinking water requirements. Comparison of data shows a minor increase in mineral concentration (10 to 50 micromhos) in the 157 miles between Keswick Dam (Station 12) and Colusa (Station 13b).

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micro-mhos at 25°C)	160	111	160	111
Temperature in °F	68	47	68	47
Dissolved oxygen in parts per million	10.6	8.8	10.6	8.8
Percent saturation	97	81	97	81
pH	7.8	7.1	7.8	7.2
Mineral constituents in parts per million				
Calcium (Ca)	14	9.6	14	9.6
Magnesium (Mg)	7.8	1.1	7.8	1.1
Sodium (Na)	11	4.9	11	4.9
Potassium (K)	1.9	0.9	1.9	0.9
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	4	4.1	4	4.1
Sulfate (SO ₄)	15	2.9	15	1.0
Chloride (Cl)	8.0	2.2	8.0	2.2
Nitrate (NO ₃)	2.0	0.0	2.0	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	14	1.8	14	1.8
Total dissolved solids in parts per million	123	76	123	76
Percent sodium	27	17	27	17
Hardness as CaCO ₃ in parts per million				
Total	67	49	67	49
Noncarbonate	7	0.0	7	0.0
Turbidity	9	1	9	1
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.68	0.00	0.68	0.00
Solid alpha	0.20	0.00	0.20	0.00
Dissolved beta	3.44	1.08	3.44	1.08
Solid beta	7.86	2.18	7.86	2.18

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT COLUSA (STA. 13b)

SACRAMENTO RIVER AT KNIGHTS LANDING (STA. 14)

Sampling Point Knights Landing station is located in Section 14 of Township 11 North, Range 2 East, Mt. Diablo Base and Meridian. Daily composites and monthly grab samples were collected at the Southern Pacific Railroad bridge, at Knights Landing, just downstream from the gaging station and about 34 miles upstream from Sacramento.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Water in Sacramento River at Knights Landing is bicarbonate in type with calcium and magnesium the predominant cations. The water is slightly hard to moderately hard, meets drinking water requirements for mineral content, and is generally class 1 for irrigation. In August 1954 boron reached 0.86 ppm placing this water in class 2 for irrigation. Comparison of analyses of samples from Station 14 with those of water from the Colusa station show that mineral concentrations become considerably higher (on the order of 160 micromhos) in Sacramento River at Knights Landing. Numerous irrigation drainage waters entering the river between these two stations accounts for the higher mineral concentrations at Station 14.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	475	55	375	115
Temperature in °F	81	48	80	47
Dissolved oxygen in parts per million	12.8	4.8	12.7	4.8
Percent saturation	100	42	98	71
pH	8.3	6.8	7.3	5.6
Mineral constituents in parts per million				
Calcium (Ca)	38	8.8	14	14
Magnesium (Mg)	20	3.4	8.3	6.8
Sodium (Na)	86	2.5	34	4.8
Potassium (K)	6.8	0.65	1.4	1.2
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	263	36	131	96
Sulfate (SO ₄)	52	2.4	15	9.0
Chloride (Cl)	57	2.0	17	3.8
Nitrate (NO ₃)	1.2	0.0	5.3	0.0
Fluoride (F)	0.5	0.0	0.1	0.0
Boron (B)	0.86	0.0	0.20	0.0
Silica (SiO ₂)	31	15	27	24
Total dissolved solids in parts per million	423	59	202	74
Percent sodium	52	13	44	21
Hardness as CaCO ₃ in parts per million				
Total	169	36	96	42
Noncarbonate	16	0.0	12	0.0
Turbidity	600	0.0	85	0.0
Coliform in most probable number per milliliter	>7,000	0.046	230	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.42	0.00	0.38	0.18
Solid alpha	0.67	0.00	0.40	0.09
Dissolved beta	6.86	0.00	6.86	0.00
Solid beta	14.75	0.00	3.83	0.00

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT KNIGHTS LANDING (STA. 14)

SACRAMENTO RIVER AT SACRAMENTO (STA. 15)

Sampling Point Station 15 is located in Section 35 of Township 9 North, Range 4 East, Mt. Diablo Base and Meridian. Daily composites and monthly grab samples were collected at Tower Bridge, 0.4 mile downstream from the gaging station at Sacramento, and about 1.3 miles downstream from the confluence of the American River.

Period of Record April 1951 through December 1959.

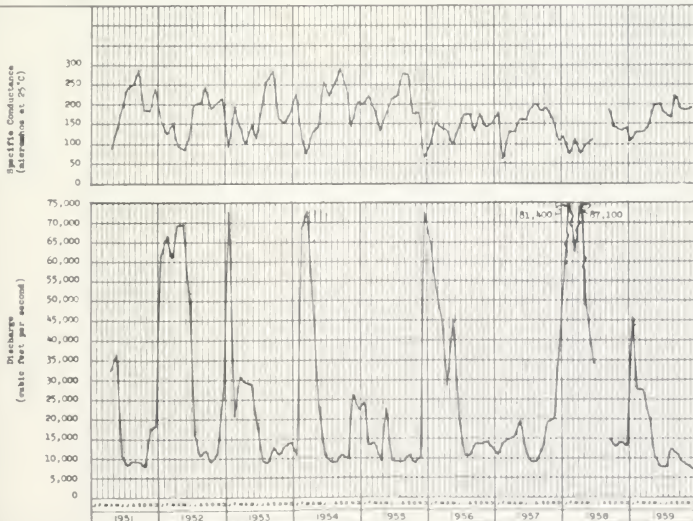
Water Quality Characteristics Past analyses show the water to be bicarbonate in type with calcium and magnesium about equal in predominance as the major cations, soft to slightly hard, class 1 for irrigation, suitable for all but the most exacting industrial uses, and within drinking water requirements for mineral content. Mineral concentrations are lower (30 to 60 micromhos) at Sacramento as compared to Knights Landing due to the influence of the American and Feather Rivers inflow.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	29	7.1	29	7.1
Temperature in °F	75	41	74	47
Dissolved oxygen in parts per million	11.6	7.7	11.6	7.7
Percent saturation	116	79	104	88
pH	8.2	6.4	8	6.7
Mineral constituents in parts per million				
Calcium (Ca)	19	9.4	18	7.6
Magnesium (Mg)	12	1.8	11	1.7
Sodium (Na)	25	1.9	24	4.4
Potassium (K)	2.1	0.4	2.1	0.4
Carbonate (CO ₃)	0.4	0.0	0.4	0.0
Bicarbonate (HCO ₃)	139	26	139	40
Sulfate (SO ₄)	20	1.8	19	4.4
Chloride (Cl)	34	7.0	34	1.2
Nitrate (NO ₃)	4.8	0.0	3.7	0.0
Fluoride (F)	0.1	0.0	0.2	0.0
Boron (B)	0.39	0.0	0.1	0.0
Silica (SiO ₂)	34	10	28	10
Total dissolved solids in parts per million	179	41	165	60
Percent sodium	37	14	34	29
Hardness as CaCO ₃ in parts per million				
Total	97	22	88	17
Noncarbonate	11	0.0	11	0.0
Turbidity	170	4	70	5
Coliform in most probable number per milliliter	7,000	0.21	7,000	0.21
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.67	0.00	0.09	0.00
Solid alpha	0.56	0.00	0.30	0.00
Dissolved beta	6.5	0.00	4.40	0.41
Solid beta	12.56	0.00	12.56	3.64

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT SACRAMENTO (STA 15)

COLUSA TROUGH NEAR COLUSA (STA. 87)

Sampling Point Station 87 is located in Section 34, Township 16 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected near the surface, along the right bank, from State Highway 120 bridge 3 miles west of Colusa.

Period of Record July 1952 through December 1959.

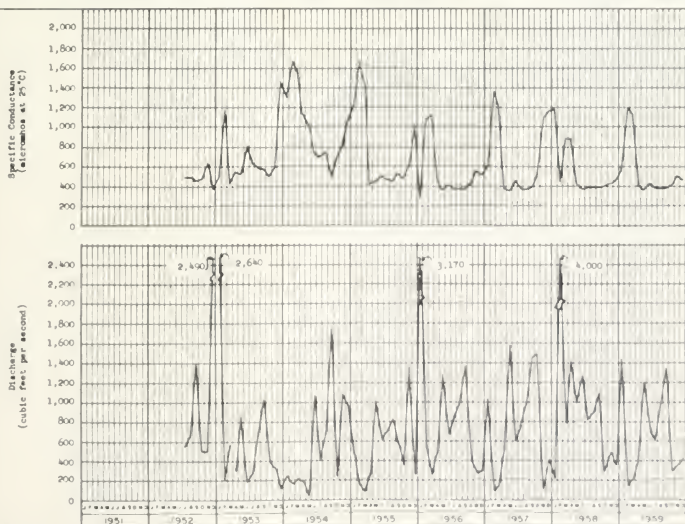
Water Quality Characteristics Past analyses show Colusa Trough water to be predominantly a mixed sodium-magnesium-calcium bicarbonate-sulfate type with concentrations of dissolved solids approaching the upper limit of class 1 for irrigation. Hardness ranged from moderately hard to very hard, limiting some domestic and industrial use. During the irrigation season this water is chiefly return flow from Colusa Basin and reflects the mineralized conditions of waters used and reused for agricultural purposes.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,670	269	1,200	159
Temperature in °F	81	43	79	40
Dissolved oxygen in parts per million	12.4	4.9	10.5	6.8
Percent saturation	112	46	87	78
pH	8.4	6.8	8.1	7.4
Mineral constituents in parts per million				
Calcium (Ca)	70	15	54	19
Magnesium (Mg)	73	9.4	43	14
Sodium (Na)	224	26	145	35
Potassium (K)	5.4	1.1	3.3	1.5
Carbonate (CO ₃)	14	0.0	0	0.8
Bicarbonate (HCO ₃)	363	96	327	142
Sulfate (SO ₄)	226	21	226	21
Chloride (Cl)	172	11	93	14
Nitrate (NO ₃)	5.8	0.0	5.8	0.0
Fluoride (F)	0.6	0.8	0.5	0.1
Boron (B)	0.37	0.0	0.3	0.1
Silica (SiO ₂)	30	9.9	30	13
Total dissolved solids in parts per million	990	160	741	218
Percent sodium	58	34	50	35
Hardness as CaCO ₃ in parts per million				
Total	418	76	312	104
Noncarbonate	129	0.8	44	10
Turbidity	520	9	60	2
Coliforms in most probable number per milliliter	2,400	5	2,400	1
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha			5.63	0.26
Solid alpha			0.10	0.00
Dissolved beta			5.35	3.87
Solid beta			1.88	0.30

WATER QUALITY VARIATIONS



COLUSA TROUGH NEAR COLUSA (STA. 87)

SACRAMENTO SLOUGH NEAR KNIGHTS LANDING (STA. 14a)

Sampling Point The station is within Section 20, Township 11 North, Range 2 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected near the surface from the center of the stream, from a bridge crossing Sutter Bypass, near the discharge pipes from Reclamation District 1500 pumping plant about 4 miles east of Knights Landing.

Period of Record June 1951 through December 1959.

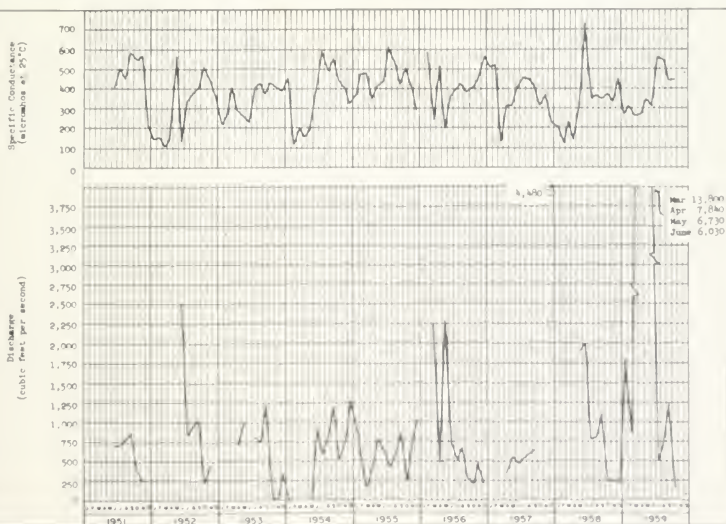
Water Quality Characteristics Past analyses show this water to be predominantly a mixed magnesium-calcium-sodium bicarbonate type, with low to moderate concentrations of dissolved solids, and class 1 for irrigation use. Water from Sacramento Slough is moderately hard and of limited use for some domestic and industrial uses. Flow in this slough is chiefly irrigation return and local drainage from Reclamation District 1500.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1959
Specific conductance (micromhos at 25°C)	790	106	569	948
Temperature in °F	82	41	78	46
Dissolved oxygen in parts per million	12.4	5.8	10.1	6.1
Percent saturation	107	64	96	79
pH	8.9	6.9	8.0	7.0
Mineral constituents in parts per million				
Calcium (Ca)	47	11	16	18
Magnesium (Mg)	30	4.9	26	19
Sodium (Na)	66	5.0	50	11
Potassium (K)	3.2	0.9	2.0	1.2
Carbonate (CO ₃)	4.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	272	51	216	120
Sulfate (SO ₄)	13	1.8	26	1.8
Chloride (Cl)	114	3.2	59	9.2
Nitrate (NO ₃)	1.4	0.0	1.8	0.0
Fluoride (F)	0.4	0.0	0.2	0.0
Boron (B)	0.19	0.0	0.1	0.0
Silica (SiO ₂)	46	15	16	19
Total dissolved solids in parts per million	440	64	177	154
Percent sodium	48	18	37	21
Hardness as CaCO ₃ in parts per million				
Total	218	44	109	98
Noncarbonate	37	0.0	1	0.0
Turbidity	310	5	16	11
Coliforms in most probable number per milliliter	>7,000	0.69	7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.48	0.00	0.48	0.00
Solid alpha	0.60	0.00	0.60	0.00
Dissolved beta	3.64	0.00	1.64	0.00
Solid beta	9.23	0.00	2.27	2.36

WATER QUALITY VARIATIONS



SACRAMENTO SLOUGH NEAR KNIGHTS LANDING (STA. 14a)

McCloud River Basin. McCloud River watershed lies immediately north of Shasta Reservoir in the northern portion of the Central Valley Region. The river basin drains a heavily forested, predominantly mountainous terrain. Valley and mesa lands cover about 15 percent of the 685 square miles in the McCloud River Basin. Estimated mean seasonal runoff is 1,403,000 acre-feet.

Timber production provides the major stimulus for the economy of this basin. Livestock raising and recreation provide a supplement to the economic development. Water use and waste discharges of these developments are comparatively minor and have not created problems of any consequence.

To maintain surveillance on quality of runoff from this basin a monitoring station is located on McCloud River above Lake Shasta.

McCLOUD RIVER ABOVE SHASTA LAKE (STA. 18)

Sampling Point Station 18 is located in Section 31 of Township 36 North, Range 3 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, just above the backwater of Shasta Lake, 11 miles east of the town of Delta.

Period of Record April 1951 through December 1959.

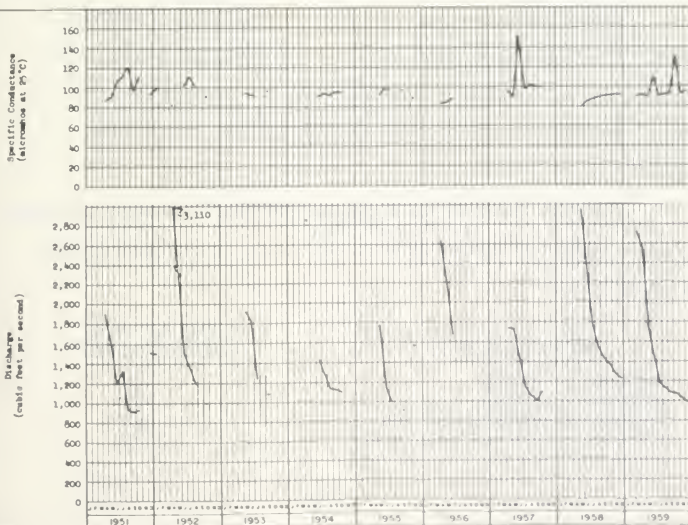
Water Quality Characteristics Past analyses show the water to be generally calcium bicarbonate in character, class 1 for irrigation, soft and within drinking water standards for mineral content. The station is generally inaccessible during the winter months and, therefore, samples are usually collected only during the spring, summer and fall months.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhms at 25°C)	150	79	199	88.4
Temperature in °F	60	42	57	42
Dissolved oxygen in parts per million	14.1	7.5	11.9	9.9
Percent saturation	124	61	108	94
pH	8.1	7.3	7.7	7.3
Mineral constituents in parts per million				
Calcium (Ca)	13	8	8.8	8.0
Magnesium (Mg)	5.1	2.1	3.6	1.5
Sodium (Na)	9.9	3.1	8.6	3.1
Potassium (K)	2.5	0.6	1.5	0.8
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	86	14	70	47
Sulfate (SO ₄)	2.5	0.0	1.5	0.6
Chloride (Cl)	8	0.0	4.8	1.0
Nitrate (NO ₃)	1.0	0.0	1.0	0.4
Fluoride (F)	0.2	0.0	0.1	0.0
Boron (B)	0.72	0.0	0.0	0.0
Silica (SiO ₂)	41	25	38	29
Total dissolved solids in parts per million	134	66	105	71
Percent sodium	28	14	27	14
Hardness as CaCO ₃ in parts per million				
Total	54	31	51	35
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	35	0.0	35	0.0
Coliforms in most probable number per milliliter	>7,000	<0.045	230	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.51	0.00	0.51	0.00
Solid alpha	0.55	0.17	0.55	0.22
Dissolved beta	10.80	0.00	3.77	2.10
Solid beta	5.96	0.00	0.58	0.11

WATER QUALITY VARIATIONS



McCLOUD RIVER ABOVE SHASTA LAKE (STA. 18)

Pit River Basin. Runoff from 5,758 square miles in California (including Goose Lake Basin) drains into the Pit River. Goose Lake Basin in California comprises 412 square miles which, during extremely wet years, spills over into the Pit River. Included with the Pit River Unit Basin are some 2,270 square miles of valley and mesa lands. Prominent among the valley fill areas are South Fork Pit River, Big Valley, Goose Lake, and Fall River. Estimated mean annual runoff of Pit River Basin is 3,426,000 acre-feet.

Topography of the area is characterized by several large upland valley areas, ranging in elevation from 2,500 to 5,000 feet, surrounded by rugged, volcanic peaks of the Cascade Range. These mountain and valley lands are used extensively for dry range for livestock and also support irrigated agriculture, timber production, mining, and recreation.

Waste discharges from several small communities, lumbermills, and local light industries enter the Pit River along its course. These wastes are minor, the largest being less than 0.5 mgd from the City of Alturas, and do not create any significant pollution or impairment problems:

The following tabulation presents the names of station maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Pit River near Canby	166
Pit River near Bieber	168
Pit River near Montgomery Creek	170
Pit River, South Fork near Likely	172

PIT RIVER NEAR CANBY (STA. 17a)

Sampling Point The Canby station is situated in Section 10 of Township 41 North, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the channel of flow from the Highway 299 bridge 4.5 miles southwest of Canby.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Pit River at Station 17a is sodium-calcium bicarbonate, class 1 for irrigation and soft to slightly hard. Iron occasionally exceeds the recommended maximum for iron and manganese combined in drinking water. All other minerals are within drinking water standards. A significant increase (70 to 140 ppm) in the concentration of minerals occurs between the South Fork Pit River near Likely station and the Canby station.

Significant Water Quality Changes The iron concentration of 0.33 ppm reported in September 1959 exceeded the recommended maximum for iron and manganese together in drinking water.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1957
Specific conductance (microhos at 25°C)	176	127	176	88
Temperature in °F	79	33	78	33
Dissolved oxygen in parts per million	12.5	6.0	12.5	7.2
Percent saturation	98	68	98	77
pH	8.4	6.8	8.3	6.7
Mineral constituents in parts per million				
Calcium (Ca)	24	11	24	20
Magnesium (Mg)	11	4.4	11	8.3
Sodium (Na)	47	8.4	47	20
Potassium (K)	7.7	2.1	7.0	5.0
Carbonates (CO ₃)	6	0.0	6	0.0
Bicarbonates (HCO ₃)	190	76	190	113
Sulfate (SO ₄)	24	4.9	24	9.0
Chloride (Cl)	20	0.0	20	4.5
Nitrate (NO ₃)	9.2	0.0	1.1	1.0
Fluoride (F)	0.6	0.0	0.3	0.0
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	38	29	33	30
Total dissolved solids in parts per million	263	89	263	142
Percent sodium	51	24	51	35
Hardness as CaCO ₃ in parts per million				
Total	106	40	106	67
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	140	3	120	8
Coliform in most probable number per milliliter	>7,000	0.25	2,400	0.5
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.19	0.00	0.19	0.17
Solid alpha	1.43	0.00	0.40	0.09
Dissolved beta	10.83	0.00	6.72	4.64
Solid beta	8.37	0.00	3.25	0.00

WATER QUALITY VARIATIONS



PIT RIVER NEAR CANBY (STA. 17a)

PIT RIVER NEAR BIEBER (STA. 17e)

Sampling Point Station 17e is located within Section 34 of Township 37 North, Range 7 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the right bank, at the USGS gage 1.5 miles upstream from Spring Gulch and 8 miles south of Bieber.

Period of Record October 1958 through December 1959.

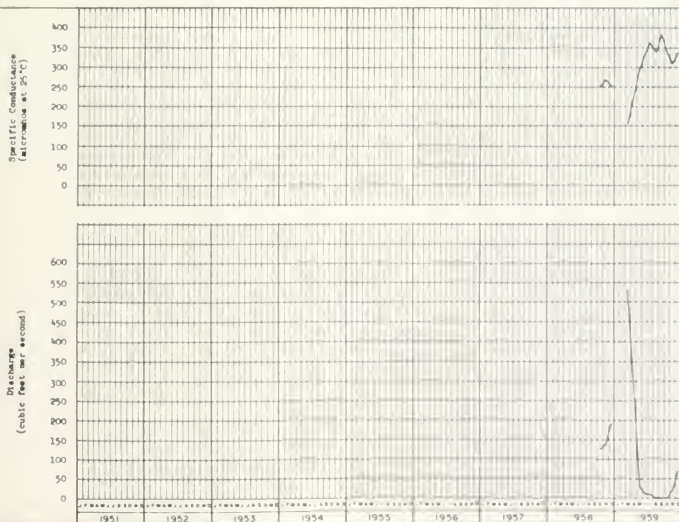
Water Quality Characteristics Water at this station is sodium-calcium bicarbonate in character, class 1 for irrigation, soft and within mineral standards for drinking water. There is no significant difference in conductivity of Pit River between the Canby station and the Bieber station.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum 1959
Specific conductance (microhms at 25°C)	181	155	181	155
Temperature in °F	77	33	77	33
Dissolved oxygen in parts per million	14.6	7.3	14.6	7.3
Percent saturation	174	76	174	76
pH	8.3	7.3	8.3	7.4
Mineral constituents in parts per million				
Calcium (Ca)	28	13	28	13
Magnesium (Mg)	12	4.9	12	4.9
Sodium (Na)	68	14	68	14
Potassium (K)	8.8	2.6	8.8	2.6
Carbonate (CO ₃)	53	0.0	53	0.0
Bicarbonate (HCO ₃)	120	80	120	80
Sulfate (SO ₄)	40	7.7	40	13
Chloride (Cl)	23	3.5	23	3.5
Nitrate (NO ₃)	1.3	0.0	1.3	0.0
Fluoride (F)	0.5	0.2	0.5	0.2
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	35	10	34	10
Total dissolved solids in parts per million	246	122	246	122
Percent sodium	68	35	68	35
Hardness as CaCO ₃ in parts per million				
Total	110	53	110	53
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	See 1959	See 1959	100	13
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha			0.00	0.00
Solid alpha			0.10	0.00
Dissolved beta			13.33	6.15
Solid beta			7.16	2.66

WATER QUALITY VARIATIONS



PIT RIVER NEAR BIEBER (STA. 17e)

PIT RIVER NEAR MONTGOMERY CREEK (STA. 17)

Sampling Point Station 17 is located in Section 32 of Township 35 North, Range 1 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage 1 mile upstream from Cow Creek and 3.5 miles west of the town of Montgomery Creek.

Period of Record April 1951 through December 1959.

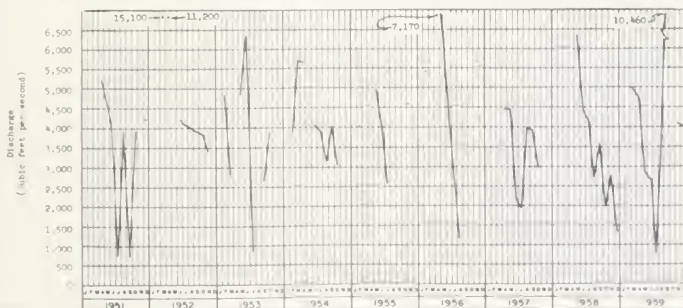
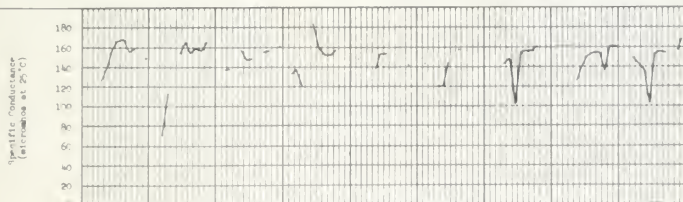
Water Quality Characteristics Past analyses show the character of the water at Station 17 to be calcium-magnesium bicarbonate, class 1 for irrigation, soft to slightly hard, and within the recommended limits for mineral content in drinking water. The concentration of most dissolved minerals in the Pit River normally decrease significantly (about 100 micromhos) between the Bieber station and Montgomery Creek station because of tributary inflow of better quality water.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	181	70	169	111
Temperature in °F	70	56	66	41
Dissolved oxygen in parts per million	11.8	6.9	11.9	8.8
Percent saturation	130	73	99	94
pH	8.4	7.1	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	14	6.9	11	11
Magnesium (Mg)	7.1	2.1	8.9	6.4
Sodium (Na)	14	2.6	14	7.4
Potassium (K)	3.2	0.8	2.1	1.4
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	98	44	91	41
Sulfate (SO ₄)	4.8	0.9	4.8	1.0
Chloride (Cl)	8	0.8	4.4	1.4
Nitrate (NO ₃)	1.0	0.0	1.0	0.4
Fluoride (F)	0.2	0.0	0.2	0.1
Boron (B)	0.3	0.0	0.2	0.0
Silica (SiO ₂)	18	17	14	20
Total dissolved solids in parts per million	187	58	187	86
Percent sodium	35	14	35	24
Hardness as CaCO ₃ in parts per million				
Total	74	39	69	41
Noncarbonate	3	0.0	0.0	0.0
Turbidity	60	0.5	50	1
Coliform in most probable number per milliliter	>7,000	<0.045	130	<0.045
Radioactivity in micro-curie per liter				
Dissolved alpha	0.09	0.00	0.09	0.09
Solid alpha	0.78	0.00	0.50	0.17
Dissolved beta	8.89	0.00	8.89	0.00
Solid beta	1.66	0.00	0.64	0.04

WATER QUALITY VARIATIONS



PIT RIVER NEAR MONTGOMERY CREEK (STA. 17)

PIT RIVER, SOUTH FORK NEAR LIKELY (STA. 18a)

Sampling Point Station 18a is the upstream station on the Pit River Basin and is located in Section 11 of Township 39 North, Range 13 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the USGS gage 1.3 miles downstream from West Valley Creek and 3.5 miles east of Likely.

Period of Record August 1958 through December 1959.

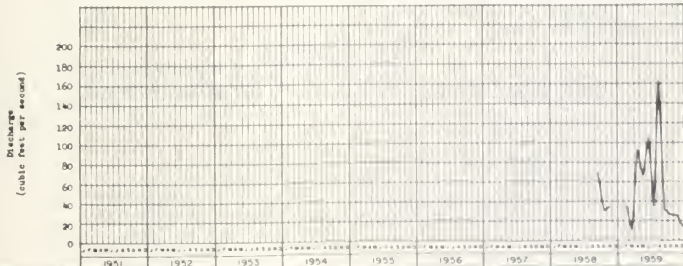
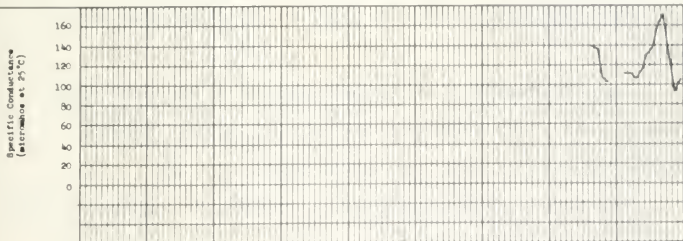
Water Quality Characteristics South Fork Pit River near Likely is calcium-magnesium bicarbonate in character, class 1 for irrigation and soft. Although it meets drinking water standards for mineral content, iron concentrations occasionally exceed the recommended limit of 0.3 ppm for iron and manganese combined.

Significant Water Quality Changes During September 1959, iron exceeded the maximum recommended limit for iron and manganese combined in drinking water when 0.74 ppm was reported. The source of the excessive iron concentrations has not been determined.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	172	91.2	172	91.2
Temperature in °F	78	32	78	32
Dissolved oxygen in parts per million	12.2	7.1	12.2	7.1
Percent saturation	97	79	97	79
pH	8.3	7.1	8.3	7.1
Mineral constituents in parts per million				
Calcium (Ca)	15	8.4	15	8.4
Magnesium (Mg)	6.2	1.5	6.1	1.4
Sodium (Na)	12	4.8	12	4.8
Potassium (K)	4.7	1.9	4.7	1.9
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	98	44	98	44
Sulfate (SO ₄)	8.6	0.0	8.6	0.0
Chloride (Cl)	7.5	0.5	7.5	1.0
Nitrate (NO ₃)	1.5	0.0	1.5	0.0
Fluoride (F)	0.3	0.0	0.2	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	40	11	40	11
Total dissolved solids in parts per million	177	81	177	81
Percent sodium	28	17	28	17
Hardness as CaCO ₃ in parts per million				
Total	63	16	63	16
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	45	1	45	1
Coliform in most probable number per milliliter	See 1959	See 1959	>7,000	0.004
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.51	0.27	0.51	0.27
Solid alpha	0.44	0.27	0.44	0.27
Dissolved beta	9.12	6.35	9.12	6.35
Solid beta	0.47	0.00	0.50	0.00

WATER QUALITY VARIATIONS



PIT RIVER, SOUTH FORK NEAR LIKELY (STA. 18a)

Redding Stream Unit. The Redding stream unit is located on the northern extremity of Sacramento Valley and includes all major streams tributary to Sacramento River between Keswick Dam and Red Bluff. To the west of the Sacramento River, Cottonwood and Clear Creeks are the major tributaries, and Cow, Bear, Battle and Paynes Creeks contribute from the east. The unit drains an area of about 2,610 square miles of which 780 square miles is valley and mesa land. Mean annual runoff in the unit totals 2,740,000 acre-feet.

The terrain of the unit is comprised of a fertile valley floor, rolling grass-covered foothills, and rugged mountains at the eastern and western boundaries. Developments in this area are centered around agriculture and lumbering activities. Livestock raising, recreation, and light industry are also prevalent in the unit. The Sacramento River and the underlying ground water basin provide most of the water used in the unit.

Waste discharges of significant quantity in this unit include outflows from United States Plywood Corporation (.34 mgd), Anderson Sanitation District (.75 mgd), and City of Redding (2.5 mgd).

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Clear Creek near Igo	176
Cow Creek near Millville	178
Cottonwood Creek below North Fork	
Cottonwood Creek	180
Cottonwood Creek near Cottonwood	182
Cottonwood Creek, South Fork above	
Cottonwood Creek	184
Battle Creek near Cottonwood	186
Paynes Creek near Red Bluff	188

CLEAR CREEK NEAR IGO (STA. 12d)

Sampling Point Station 12d is located in Section 27, Township 31 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at the Redding-Igo road bridge, 1.0 mile northeast of Igo, 8 miles southwest of Redding, and 10.5 miles upstream from the mouth of the creek.

Period of Record April 1958 through December 1959.

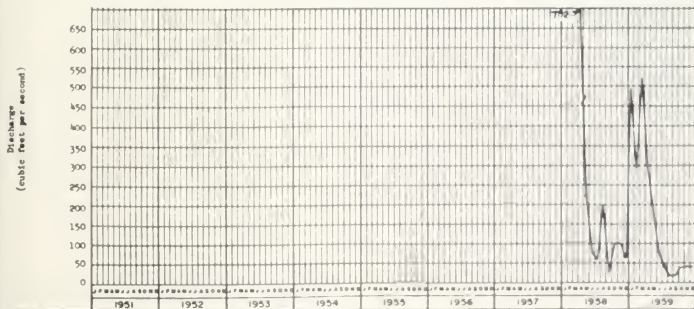
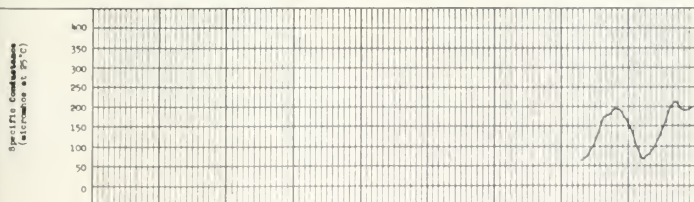
Water Quality Characteristics A review of past analyses show the water at Station 12d to be bicarbonate in type with no predominant cation, soft to slightly hard and meets the drinking water standards for mineral content. Mineral concentrations in this water identify it as class 1 for irrigation.

Significant Water Quality Changes Total radioactivity reached 28.4 micro-micro curies per liter in September 1959, which is a little higher than that normally found in streams of this unit.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (microhm/cm at 25°C)	215	65	215	65
Temperature in °F	80	60	77	60
Dissolved oxygen in parts per million	11.1	6.5	11.1	7.0
Percent saturation	106	74	106	90
pH	9.1	7.1	7.9	7.1
Mineral constituents in parts per million				
Calcium (Ca)	19	6.4	19	6.4
Magnesium (Mg)	5.1	1.8	5.1	2.1
Sodium (Na)	18	1.1	18	1.1
Potassium (K)	1.1	0.3	1.1	0.3
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	70	26	70	26
Sulfate (SO ₄)	17	1.9	17	1.9
Chloride (Cl)	26	1.0	26	1.0
Nitrate (NO ₃)	1.7	0.0	1.7	0.0
Fluoride (F)	0.2	0.0	0.1	0.0
Boron (B)	0.13	0.0	0.1	0.0
Silica (SiO ₂)	21	12	21	12
Total dissolved solids in parts per million	130	40	130	45
Percent sodium	37	21	37	21
Hardness as CaCO ₃ in parts per million				
Total	65	24	65	26
Noncarbonate	21	0.0	21	1
Turbidity	10	1	10	1
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.58	0.10	0.58	0.10
Solid alpha	0.47	0.20	0.47	0.20
Dissolved beta	21.21	4.44	21.21	4.44
Solid beta	6.54	4.62	6.54	4.62

WATER QUALITY VARIATIONS



CLEAR CREEK NEAR IGO (STA. 12d)

COW CREEK NEAR MILLVILLE (STA. 88a)

Sampling Point The sampling station is located in Section 32 of Township 31 North, Range 3 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage, 4.2 miles southwest of Millville, and 4.3 miles downstream from Little Cow Creek.

Period of Record April 1958 through December 1959.

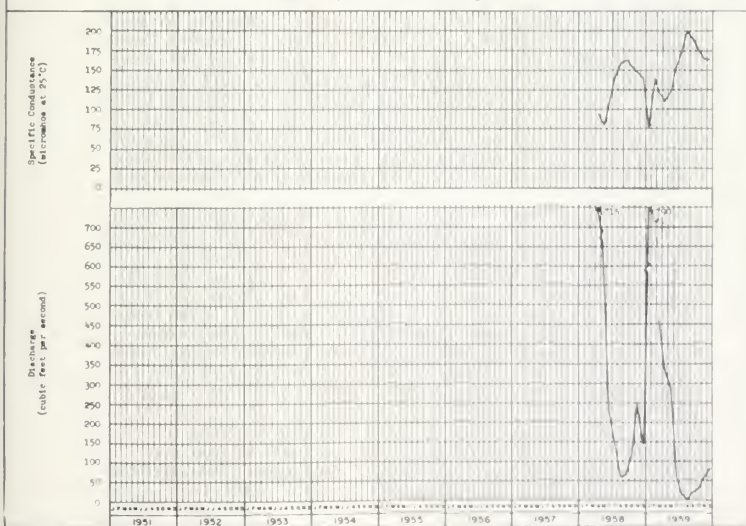
Water Quality Characteristics Analyses show water at Station 88a to be a calcium bicarbonate type, soft to slightly hard, class 1 for irrigation and meets the drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum 1959
Specific conductance (micromhos at 25°C)	261	76.6	261	76.6
Temperature in °F	87	43	87	43
Dissolved oxygen in parts per million	12.7	6.1	12.7	6.1
Percent saturation	106	72	106	72
pH	8.1	6.8	8.1	6.8
Mineral constituents in parts per million				
Calcium (Ca)	20	7.2	20	7.2
Magnesium (Mg)	7.8	2.9	7.8	2.9
Sodium (Na)	12	3.5	12	3.5
Potassium (K)	2.8	0.6	2	0.6
Carbonate (CO ₃)	5.0	0.0	5.0	0.0
Bicarbonate (HCO ₃)	108	34	108	34
Sulfate (SO ₄)	12	8.6	12	8.6
Chloride (Cl)	9.8	3.2	9.8	3.2
Nitrate (NO ₃)	0.9	0.0	0.9	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.30	0.0	0.30	0.0
Silica (SiO ₂)	37	14	37	14
Total dissolved solids in parts per million	146	60	146	60
Percent sodium	25	18	25	18
Hardness as CaCO ₃ in parts per million				
Total	80	30	80	30
Noncarbonate	9	0.0	9	0.0
Turbidity	See 1959	See 1959	15	0
Coliforms in most probable number per milliliter (Not Measured)	See 1959	See 1959		
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha			0.000000	0.000000
Solid alpha			0.000000	0.000000
Dissolved beta			0.000000	0.000000
Solid beta			0.000000	0.000000

WATER QUALITY VARIATIONS



COW CREEK NEAR MILLVILLE (STA 880)

COTTONWOOD CREEK BELOW NORTH FORK COTTONWOOD CREEK (STA. 11a)

Sampling Point The monitoring station is located in Section 2, Township 29 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank 13.5 miles west of the town of Cottonwood along Gas Point Road about 1.0 mile downstream from the mouth of North Fork Cottonwood Creek.

Period of Record October 1958 through December 1959.

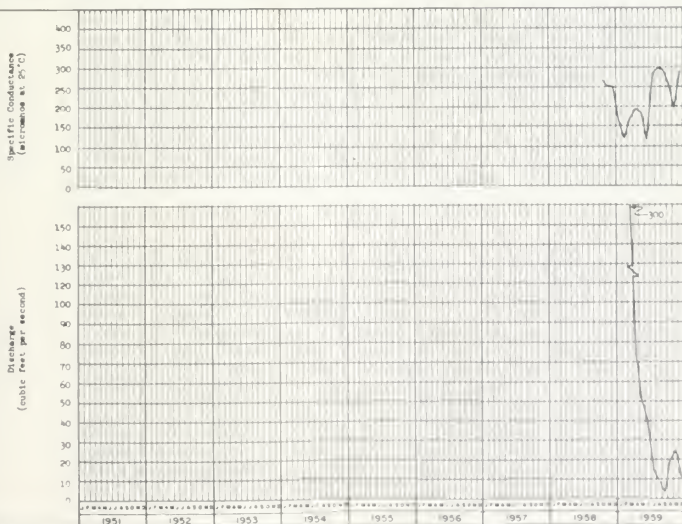
Water Quality Characteristics Water at Station 11a is bicarbonate in type, with no predominant cation, moderately hard, class 1 for irrigation, and meets drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	200	116	200	116
Temperature in °F	88	33	88	33
Dissolved oxygen in parts per million	13.1	6.6	13.1	6.6
Percent saturation	106	77	106	77
pH	8.1	7.1	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	30	11	30	11
Magnesium (Mg)	15	3.9	15	3.9
Sodium (Na)	14	4.1	14	4.1
Potassium (K)	2.6	0.4	2.6	0.4
Carbonate (CO ₃)	5	0.0	5	0.0
Bicarbonate (HCO ₃)	161	11	161	11
Sulfate (SO ₄)	27	4.8	27	4.8
Chloride (Cl)	21	4.0	21	4.0
Nitrate (NO ₃)	3.5	0	3.5	0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	24	11	24	11
Total dissolved solids in parts per million	183	80	183	80
Percent sodium	30	9	30	9
Hardness as CaCO ₃ in parts per million				
Total	135	48	135	48
Noncarbonate	20	0	20	0
Turbidity	2	1	2	1
Coliforms in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.34	0.00	0.34	0.00
Solid alpha	0.89	0.69	0.89	0.69
Dissolved beta	6.24	4.27	6.24	4.27
Solid beta	4.99	3.41	4.99	3.41

WATER QUALITY VARIATIONS



COTTONWOOD CREEK BELOW NORTH FORK COTTONWOOD CREEK (STA. 110)

COTTONWOOD CREEK NEAR COTTONWOOD (STA. 12b)

Sampling Point Station 12b is located in Section 7 of Township 29 North, Range 3 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage 2 miles east of the town of Cottonwood, and approximately 2.5 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

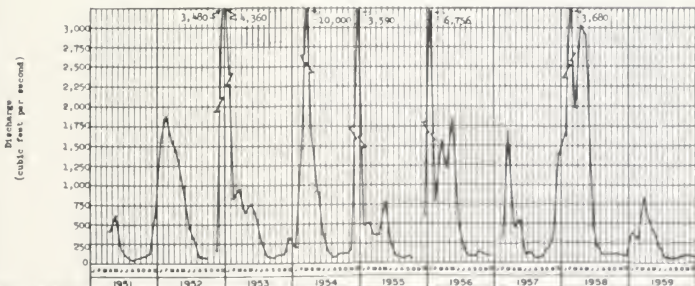
Water Quality Characteristics Water at Station 12b is bicarbonate in type with generally no predominant cation. However, a tendency has been noted for calcium to become the dominant cation during periods of high flow. Concentrations of dissolved minerals vary only slightly at this point and depend chiefly on the rate of surface runoff. Samples of water from this station are class 1 for irrigation, are slightly to moderately hard, meet drinking water standards for mineral content, and are suitable for nearly all industrial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	169	89.1	246	133
Temperature in °F	84	40	84	40
Dissolved oxygen in parts per million	11.9	6.7	11.9	8.0
Percent saturation	147	71	147	93
pH	8.2	6.8	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	37	8.1	15	15
Magnesium (Mg)	14	5.1	9	8.4
Sodium (Na)	18	2.9	11	6
Potassium (K)	2.7	0.7	1.7	1
Carbonate (CO ₃)	1	0	0	0
Bicarbonate (HCO ₃)	148	47	113	66
Sulfate (SO ₄)	13	4.0	5.8	4.0
Chloride (Cl)	37	0.6	18	1.4
Nitrate (NO ₃)	9.1	0.1	0.9	0.1
Fluoride (F)	0.3	0.0	0.1	0.0
Boron (B)	0.20	0.0	0.1	0.0
Silica (SiO ₂)	26	18	27	24
Total dissolved solids in parts per million	228	52	140	104
Percent sodium	28	10	28	10
Hardness as CaCO ₃ in parts per million				
Total	150	41	100	72
Noncarbonate	29	0.0	12	8
Turbidity	264	0.0	25	1
Coliform in most probable number per milliliter	>7,000	0.046	2,400	0.046
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.18	0.00	0.97	0.00
Solid alpha	0.89	0.00	0.89	0.00
Dissolved beta	2.09	0.00	0.33	0.00
Solid beta	10.7	0.00	2.87	1.89

WATER QUALITY VARIATIONS



COTTONWOOD CREEK NEAR COTTONWOOD (STA. 12b)

COTTONWOOD CREEK, SOUTH FORK ABOVE COTTONWOOD CREEK (STA. 11b)

Sampling Point The station is located in Section 17, Township 29 North, Range 4 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-stream from the Evergreen Road bridge, approximately 3.2 miles west of State Highway 99 and 1 mile upstream from the mouth.

Period of Record November 1958 through December 1959.

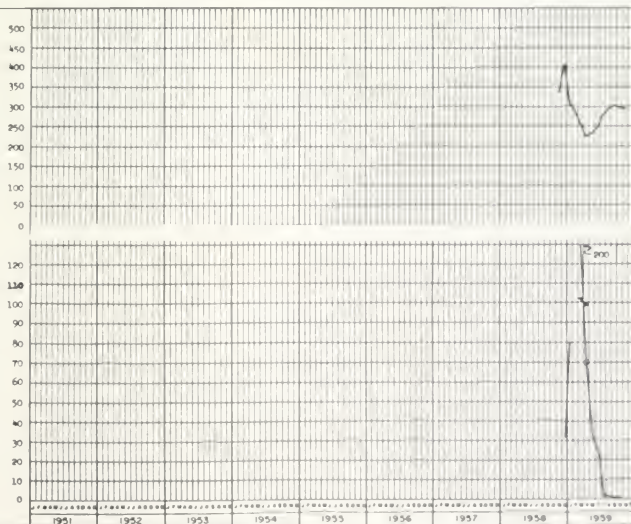
Water Quality Characteristics Water at this station is calcium bicarbonate in character, moderately hard, and class 1 for irrigation. The water is suitable for most industrial purposes and meets drinking water requirements for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	500	200	500	200
Temperature in °F	80	4	80	40
Dissolved oxygen in parts per million	12.8	5.1	12.8	5.1
Percent saturation	100	43	100	61
pH	7.9	7.2	7.9	7.2
Mineral constituents in parts per million				
Calcium (Ca)	43	23	36	21
Magnesium (Mg)	15	7.7	11	7.7
Sodium (Na)	17	8.4	14	8.4
Potassium (K)	2.0	3.4	2.0	3.4
Carbonate (CO ₃)	2	2	2	2
Bicarbonate (HCO ₃)	163	100	163	100
Sulfate (SO ₄)	23	7.9	18	7.9
Chloride (Cl)	49	8.9	34	8.9
Nitrate (NO ₃)	8.5	0.0	8.5	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.2	0.0	0.2	0.0
Silica (SiO ₂)	22	10	22	10
Total dissolved solids in parts per million	214	146	200	146
Percent sodium	20	16	20	16
Hardness as CaCO ₃ in parts per million				
Total	168	97	147	97
Noncarbonate	53	0.0	13	0.0
Turbidity	4	1	4	1
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.11	0.00	0.11	0.00
Solid alpha	0.47	0.00	0.47	0.00
Dissolved beta	3.30	2.18	3.30	2.18
Solid beta	2.41	0.96	2.41	0.96

WATER QUALITY VARIATIONS



COTTONWOOD CREEK, SOUTH FORK ABOVE COTTONWOOD CREEK (STA. 11b)

BATTLE CREEK NEAR COTTONWOOD (STA. 88b)

Sampling Point Station 88b is located in Section 6, Township 29 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected on the right bank, at the USGS gaging station, 6.3 miles upstream from the mouth, and 7.6 miles east of Cottonwood.

Period of Record April 1958 through December 1959.

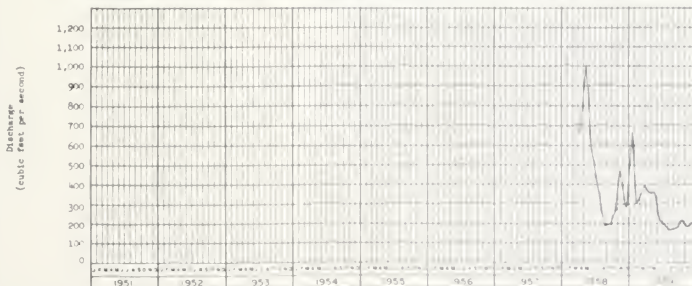
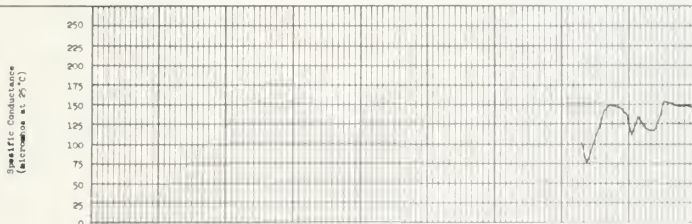
Water Quality Characteristics The water at Station 88b is bicarbonate in type with no predominant cation, excellent in quality, class 1 for irrigation, soft, and meets the requirements for drinking water. Mineral concentrations in water at Station 88b do not vary appreciably due to the effects of controlled flow resulting from upstream power developments.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	154	75.7	104	111
Temperature in °F	72	42	64	42
Dissolved oxygen in parts per million	12.6	6.3	12.6	8.7
Percent saturation	108	59	107	50
pH	8.2	7.0	8.2	7.4
Mineral constituents in parts per million				
Calcium (Ca)	12	6.8	12	7.6
Magnesium (Mg)	7.5	3.6	7.5	3.6
Sodium (Na)	10	4.1	10	6.0
Potassium (K)	2.6	1.3	2.6	1.4
Carbonate (CO ₃)	1	0.5	1	0.5
Bicarbonate (HCO ₃)	92	42	92	47
Sulfate (SO ₄)	6.7	3.2	6.7	3.2
Chloride (Cl)	4.0	1.9	4.0	1.9
Nitrate (NO ₃)	1.3	0.8	1.3	0.8
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.20	0.0	0.2	0.0
Silica (SiO ₂)	5.3	3.0	5.3	3.0
Total dissolved solids in parts per million	132	71	132	71
Percent sodium	30	19	30	20
Hardness as CaCO ₃ in parts per million				
Total	82	30	82	44
Noncarbonate	10.0	3.3	10.0	3.3
Turbidity	See 1959	See 1959	20	5
Coliform in most probable number per milliliter (Not Measured)	See 1959	See 1959		
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha			1.7	0.0
Solid alpha			2.0	0.0
Dissolved beta			4.04	2.79
Solid beta			2	1.4

WATER QUALITY VARIATIONS



BATTLE CREEK NEAR COTTONWOOD (STA. 88b)

PAYNES CREEK NEAR RED BLUFF (STA. 88g)

Sampling Point Red Bluff station is located in Section 3 of Township 28 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, 100 yards upstream from Long Road bridge at Dales station, approximately 14 miles east of Red Bluff, and 7 miles upstream from the USGS gage, which is located 0.4 mile upstream from the mouth.

Period of Record October 1958 through December 1959.

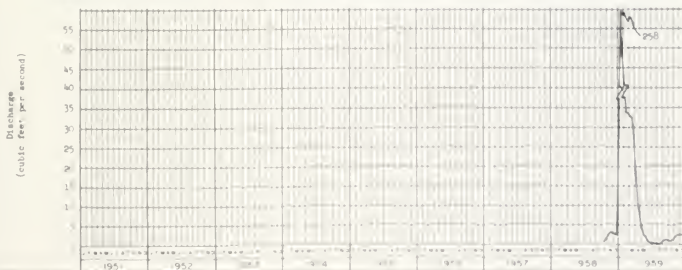
Water Quality Characteristics Past analyses show water at this point to be magnesium bicarbonate in type, slightly hard and within drinking water standards. Boron concentrations occasionally cause the water to be class 2 for irrigation. Boron in this stream is attributable to the geologic formations inherent in the drainage basin.

Significant Water Quality Changes For several months during 1959 boron concentrations exceeded the limits of a class 1 irrigation water. Low flow conditions existing during 1959 did not provide enough dilution water to prevent boron concentrations from reaching class 2 irrigation limits.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	253	144	253	154
Temperature in °F	74	48	70	48
Dissolved oxygen in parts per million	11.9	7.6	11.4	7.6
Percent saturation	110	81	110	81
pH	8.2	7.0	8.2	7.0
Mineral constituents in parts per million				
Calcium (Ca)	14	7.2	16	7.2
Magnesium (Mg)	12	4.4	12	4.4
Sodium (Na)	24	6.5	24	6.5
Potassium (K)	2.8	1.1	2.8	1.2
Carbonate (CO ₃)	12	0.5	12	0.5
Bicarbonate (HCO ₃)	12	4.3	12	4.3
Sulfate (SO ₄)	14	1.6	16	1.6
Chloride (Cl)	20	6.5	20	6.5
Nitrate (NO ₃)	5	0.0	5.0	0.0
Fluoride (F)	2	0.0	0.2	0.0
Boron (B)	1	0.1	0.5	0.1
Silica (SiO ₂)	53	27	53	27
Total dissolved solids in parts per million	146	83	146	83
Percent sodium	38	21	38	21
Hardness as CaCO ₃ in parts per million				
Total	82	36	82	36
Noncarbonate	1	0.0	1	0.0
Turbidity	See 1959	See 1959	10	1
Coliform in most probable number per milliliter (Not Measured)	See 1959	See 1959		
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.50	0.09
Solid alpha			0.14	0.27
Dissolved beta			12.05	0.00
Solid beta			2.61	0.26

WATER QUALITY VARIATIONS



PAYNES CREEK NEAR RED BLUFF STA. 88g)

West Side Stream Unit. The drainage area of the West Side stream unit occupies approximately 4,000 square miles along the west side of Sacramento Valley. Major streams draining the area include Redbank, Elder, Thomas, Stony, Cache, and Putah Creeks. Clear Lake is a large natural lake on Cache Creek and is a prominent feature of the drainage area. Foothills and mountains of the Coast Range cover about 75 percent of the unit. The aggregate natural runoff of the streams of the unit average about 1,900,000 acre-feet per year.

Commercial development in the unit is primarily based on agriculture and livestock raising. The foothills provide excellent grazing lands and the valley and mesa lands are suitable for numerous orchard and field crops. Recreation has been a major attraction in the Clear Lake area for many years and as water developments occur in other portions of this unit, recreation will command a more important place in their economy.

Several small communities, resort areas, and limited mining activities discharge wastes into the streams of this unit. Only minor impairment of water quality in these streams is attributable to this source.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this unit and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Redbank Creek near Red Bluff	192
Elder Creek near Paskenta	194
Elder Creek at Gerber	196
Thomes Creek at Paskenta	198
Thomes Creek near mouth	200
Stony Creek at Black Butte Dam Site	202
Stony Creek near Hamilton City	204
Clear Lake at Lakeport	206
Cache Creek near Lower Lake	208
Cache Creek near Capay	210
Cache Creek, North Fork near Lower Lake	212
Putah Creek near Winters	214

REDBANK CREEK NEAR RED BLUFF (STA. 88d)

Sampling Point Station 88d is situated in Section 22 of Township 26 North, Range 5 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a private bridge approximately 100 feet north of Lowery Road, at the DWR gage, 2 miles southeast of Redbank and 15 miles northwest of Red Bluff.

Period of Record January 1959 through December 1959.

Water Quality Characteristics Based on limited data, water at Station 88d is calcium-magnesium bicarbonate in character, class 1 for irrigation, moderately hard to very hard and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1955	Minimum - 1955
Specific conductance (microhm at 25°C)	See 1955	See 1955	500	200
Temperature in °F			80	60
Dissolved oxygen in parts per million			12	8
Percent saturation			100	80
pH			8.5	7.5
Mineral constituents in parts per million				
Calcium (Ca)			15	10
Magnesium (Mg)			20	15
Sodium (Na)			15	10
Potassium (K)			5	5
Carbonate (CO ₃)			5	5
Bicarbonate (HCO ₃)			200	150
Sulfate (SO ₄)			10	10
Chloride (Cl)			50	40
Nitrate (NO ₃)			5	5
Fluoride (F)			0.2	0.2
Boron (B)			0.1	0.1
Silica (SiO ₂)			21	15
Total dissolved solids in parts per million			200	200
Percent sodium			15	12
Hardness as CaCO ₃ in parts per million				
Total			241	180
Noncarbonate			52	20
Turbidity (Not Measured)				
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS

Specific Conductance
(microhm at 25°C)

600
550
500
450
400
350
300
250
200
150
100
50
0

Discharge
(cubic feet per second)

45
40
35
30
25
20
15
10
5
0

1951 1952 1953 1954 1955 1956 1957 1958 1959

REDBANK CREEK NEAR RED BLUFF (STA. 88d)

ELDER CREEK NEAR PASKENTA (STA. 13e)

Sampling Point The location of Station 13e is within Section 14 of Township 25 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the USGS gage, 2.5 miles downstream from South Fork, 8 miles northeast of Flournoy, and 11 miles north of Paskenta.

Period of Record October 1958 through December 1959.

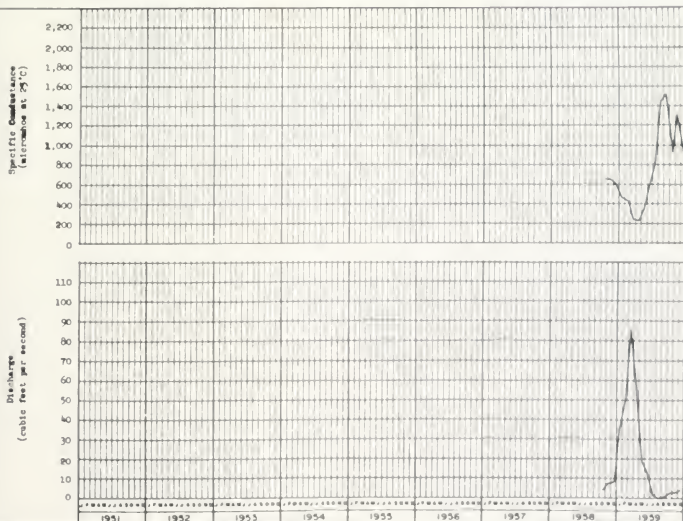
Water Quality Characteristics Analyses of samples show the water at Station 13e to be a bicarbonate type with none of the cations predominant. The water is very hard, but within drinking water standards for mineral content. Occasionally, conductivity causes the water at Station 13e to be class 2 for irrigation. Low flows late in the year offer very little dilution for accretions of poorer quality ground waters, probably accounting for the high conductivity.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	120	214	120	214
Temperature in °F	86	41	86	41
Dissolved oxygen in parts per million	12.4	7.8	11.6	7.8
Percent saturation	115	90	115	90
pH	8.3	7.6	8.3	7.6
Mineral constituents in parts per million				
Calcium (Ca)	69	21	69	21
Magnesium (Mg)	51	12	51	12
Sodium (Na)	156	7.7	146	7.7
Potassium (K)	1.6	0.4	1.6	0.4
Carbonate (CO ₃)	6	0.0	6	0.0
Bicarbonate (HCO ₃)	244	121	244	121
Sulfate (SO ₄)	30	1.9	30	1.9
Chloride (Cl)	494	12	494	12
Nitrate (NO ₃)	5.9	0.0	4.9	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	24	14	24	14
Total dissolved solids in parts per million	770	134	770	134
Percent sodium	52	13	52	13
Hardness as CaCO ₃ in parts per million				
Total	383	103	383	103
Noncarbonate	279	4	279	4
Turbidity	See 1956	See 1959	15	1
Coliforms in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1956	See 1959		
Dissolved alpha			0.68	0.18
Solid alpha			0.27	0.20
Dissolved beta			8.52	0.00
Solid beta			2.41	2.27

WATER QUALITY VARIATIONS



ELDER CREEK NEAR PASKENTA (STA. 13e)

ELDER CREEK AT GERBER (STA. 95a)

Sampling Point Elder Creek station is located within Saucos Grant in Section 2 of Township 25 North, Range 3 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage, 3.5 miles upstream from the mouth, and 1.0 mile west of Gerber.

Period of Record January 1959 through December 1959.

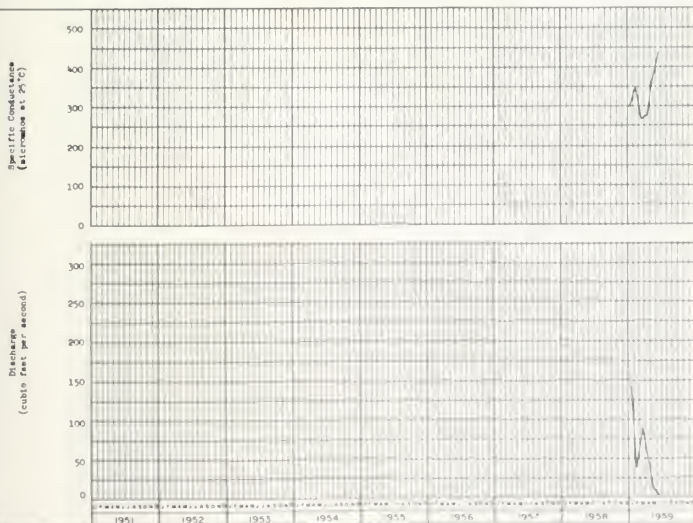
Water Quality Characteristics Based on limited analyses, the water is magnesium-calcium bicarbonate in character, class 1 for irrigation, moderately hard, and within acceptable limits for mineral content in drinking water. Only minor changes in concentrations (about 20 micromhos) occur between the Paskenta station and the Gerber station.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	441	267
Temperature in °F			74	45
Dissolved oxygen in parts per million			11.7	7.1
Percent saturation			108	84
pH			8.3	7.4
Mineral constituents in parts per million				
Calcium (Ca)			36	20
Magnesium (Mg)			27	16
Sodium (Na)			17	9
Potassium (K)			1.4	0.6
Carbonate (CO ₃)			10	0
Bicarbonate (HCO ₃)			224	101
Sulfate (SO ₄)			23	4.8
Chloride (Cl)			29	12
Nitrate (NO ₃)			7.6	1.0
Fluoride (F)			0.2	0.0
Boron (B)			0.2	0.0
Silica (SiO ₂)			24	10
Total dissolved solids in parts per million			248	154
Percent sodium			21	14
Hardness as CaCO ₃ in parts per million				
Total			200	118
Noncarbonate			35	5
Turbidity			0.0	
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.30	
Solid alpha			0.37	
Dissolved beta			7.62	
Solid beta			8.70	

WATER QUALITY VARIATIONS



ELDER CREEK AT GERBER (STA. 95a)

THOMES CREEK AT PASKENTA (STA. 13d)

Sampling Point Station 13d is located in Section 4 of Township 23 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the USGS gage, 0.25 mile upstream from Digger Creek and 0.3 mile upstream from the highway bridge at Paskenta.

Period of Record October 1958 through December 1959.

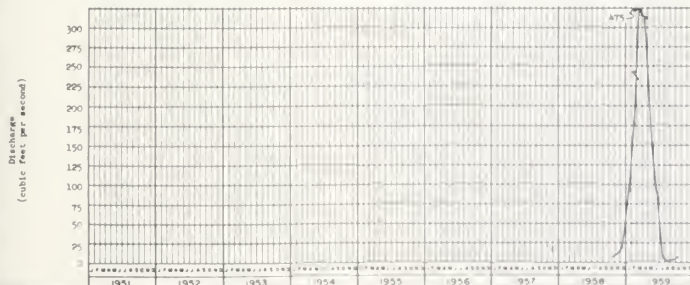
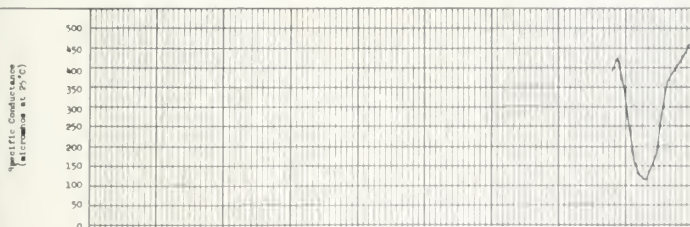
Water Quality Characteristics Analyses show the water at this station to be generally calcium bicarbonate in character, class 1 for irrigation, soft to moderately hard and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	401	116	401	116
Temperature in °F	82	42	82	42
Dissolved oxygen in parts per million	12.9	8.0	12.9	8.0
Percent saturation	127	91	127	91
pH	8.1	7.5	8.1	7.5
Mineral constituents in parts per million				
Calcium (Ca)	52	16	52	16
Magnesium (Mg)	20	1.2	17	1.2
Sodium (Na)	21	2.6	21	2.6
Potassium (K)	2.0	0.1	2.0	0.1
Carbonate (CO ₃)	6	0.0	6	0.0
Bicarbonate (HCO ₃)	188	61	188	61
Sulfate (SO ₄)	40	3.8	40	3.8
Chloride (Cl)	41	2.0	41	2.0
Nitrate (NO ₃)	0.6	0.0	0.6	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	17	8.7	17	8.7
Total dissolved solids in parts per million	267	67	267	67
Percent sodium	21	9	21	9
Hardness as CaCO ₃ in parts per million				
Total	198	51	198	51
Noncarbonate	57	1	57	1
Turbidity	See 1959	See 1959	1	1
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1959	See 1959		
Dissolved alpha			0.58	0.20
Solid alpha			0.60	0.20
Dissolved beta			5.20	1.50
Solid beta			5.20	1.50

WATER QUALITY VARIATIONS



THOMES CREEK AT PASKENTA (STA 13d)

THOMES CREEK NEAR MOUTH (STA. 95b)

Sampling Point Thomes Creek station is located in Section 35 of Township 25 North, Range 3 West, Mt. Diablo Base and Meridian. Monthly water samples were collected from the center of the channel of flow from the Highway 99W bridge at Richfield, 3 miles north of Corning, 14.5 miles south of Red Bluff, and 4.5 miles upstream from the mouth.

Period of Record January 1959 through December 1959.

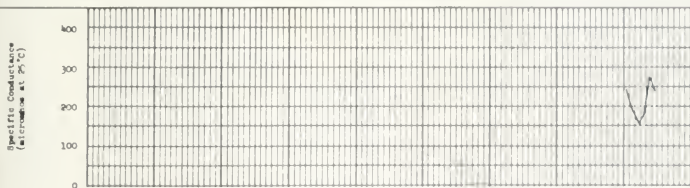
Water Quality Characteristics Water at Station 95b is calcium bicarbonate in character, class 1 for irrigation, slightly to moderately hard, and within the recommended standards for mineral content in drinking water. During the first half of the calendar year the concentration of constituents at this station are slightly higher than at Station 13d about 20 miles upstream. As irrigation commenced and natural runoff decreased in this area, the conductivity differential between the upstream station and Station 95b increased from a few micromhos to about 85 micromhos.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	See 1959	See 1959	277	15
Temperature in °F			7	43
Dissolved oxygen in parts per million			11.9	7.2
Percent saturation			104	81
pH			8.2	7.3
Mineral constituents in parts per million				
Calcium (Ca)			33	20
Magnesium (Mg)			12	4.9
Sodium (Na)			9	1.3
Potassium (K)			1	0.3
Carbonate (CO ₃)			1	0.2
Bicarbonate (HCO ₃)			144	80
Sulfate (SO ₄)			30	15
Chloride (Cl)			12	2.5
Nitrate (NO ₃)			6.5	0.9
Fluoride (F)			2	0.8
Boron (B)			0.2	0.0
Silica (SiO ₂)			16	9.6
Total dissolved solids in parts per million			167	91
Percent sodium			15	9
Hardness as CaCO ₃ in parts per million				
Total			130	71
Noncarbonate			32	5
Turbidity (Not Measured)				
Coliforms in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			30	
Solid alpha			0.00	
Dissolved beta			2.48	
Solid beta			3.69	

WATER QUALITY VARIATIONS



FLOW NOT AVAILABLE

1951 1952 1953 1954 1955 1956 1957 1958 1959

THOMES CREEK NEAR MOUTH (STA. 95b)

STONY CREEK AT BLACK BUTTE DAM SITE (STA. 13c)

Sampling Point Station 13c is situated within Section 29 of Township 23 North, Range 4 West, Mt. Diablo Base and Meridian. Monthly grab samples of water were collected from the right bank in the vicinity of the USGS gage, 120 feet downstream from the diversion dam, and 8.7 miles northwest of Orland.

Period of Record January 1958 through December 1959.

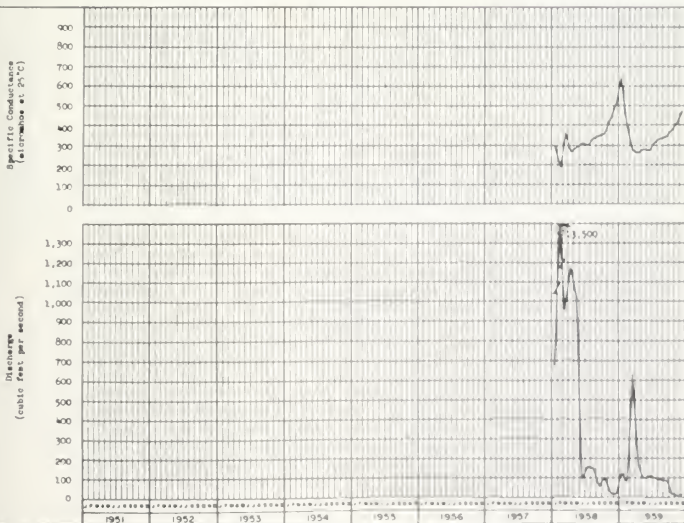
Water Quality Characteristics Stony Creek at Black Butte Dam Site is calcium-magnesium bicarbonate in character, class 1 for irrigation, ranges from slightly hard to very hard and meets drinking water standards for mineral content.

Significant Water Quality Changes A significant decrease in radioactivity was noted during 1959. The total activity decreased from 18.3 $\mu\text{c}/\text{l}$ in May to 5.7 $\mu\text{c}/\text{l}$ in September. The higher value reported is still well within safe limits.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1958	Minimum - 1959
Specific conductance (micromhos at 25°C)	614	94	614	240
Temperature in °F	78	60	78	50
Dissolved oxygen in parts per million	11.3	7.7	10.0	7.0
Percent saturation	104	87	104	90
pH	8.1	7.4	8.1	7.4
Mineral constituents in parts per million				
Calcium (Ca)	55	26	55	26
Magnesium (Mg)	24	0.5	24	0.5
Sodium (Na)	40	7.2	40	10.4
Potassium (K)	2.1	0.4	1.4	0.4
Carbonate (CO ₃)	5	0.0	5	0.0
Bicarbonate (HCO ₃)	220	100	220	121
Sulfate (SO ₄)	60	11	60	11
Chloride (Cl)	84	6.0	84	11
Nitrate (NO ₃)	3.5	0.0	3.5	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.5	0.0	0.5	0.1
Silica (SiO ₂)	25	9.1	25	9.1
Total dissolved solids in parts per million	371	114	371	141
Percent sodium	28	13	28	13
Hardness as CaCO ₃ in parts per million				
Total	214	88	214	100
Noncarbonate	80	0.0	80	0.0
Turbidity	150	3	50	20
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.48	0.00	0.48	0.00
Solid alpha	0.40	0.26	0.40	0.26
Dissolved beta	5.21	1.58	5.21	1.04
Solid beta	12.12	1.54	12.12	1.54

WATER QUALITY VARIATIONS



STONY CREEK AT BLACK BUTTE DAM SITE (STA 13c)

STONY CREEK NEAR HAMILTON CITY (STA. 13a)

Sampling Point Hamilton City station is located in Section 36 of Township 22 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage, 8 miles east of Orland, 2.5 miles southwest of Hamilton City, and 4 miles upstream from the mouth.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Past analyses show the water at the station to be generally calcium bicarbonate to calcium-magnesium bicarbonate in character, slightly to moderately hard and within drinking water standards for mineral content. With one exception in respect to boron (August 1954 - 0.64 ppm), it has been class 1 irrigation water throughout the period of record.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	583	147	531	289
Temperature in °F	86	41	71	48
Dissolved oxygen in parts per million	13.1	5.6	11.8	6.9
Percent saturation	125	66	109	77
pH	8.4	7.0	8.1	7.5
Mineral constituents in parts per million				
Calcium (Ca)	49	29	36	
Magnesium (Mg)	18	6.3	13	
Sodium (Na)	28	6.6	28	13
Potassium (K)	2.3	0.6	0.4	
Carbonate (CO ₃)	8	0.0	0.0	0.0
Bicarbonate (HCO ₃)	197	83	166	139
Sulfate (SO ₄)	22	13	14	
Chloride (Cl)	31	4	64	17
Nitrate (NO ₃)	0.8	0.0	0.0	
Fluoride (F)	0.2	0.0	0.0	
Boron (B)	0.64	0.0	0.2	0.0
Silica (SiO ₂)	18	10	11	
Total dissolved solids in parts per million	301	90	301	166
Percent sodium	24	14	24	17
Hardness as CaCO ₃ in parts per million				
Total	199	65	199	124
Noncarbonate	67	0.0	67	8
Turbidity	450	0.0	30	0
Coliforms in most probable number per milliliter	2,400	0.21	230	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.48	0.00	0.48	
Solid alpha	1.18	0.00	0.40	
Dissolved beta	2.48	0.00	0.00	
Solid beta	9.62	0.00	5.46	

WATER QUALITY VARIATIONS



STONY CREEK NEAR HAMILTON CITY (STA. 13a)

CLEAR LAKE AT LAKEPORT (STA. 41)

Sampling Point Station 41, the only active station on Clear Lake during 1959, is located in Section 24 of Township 14 North, Range 10 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the end of the pier at the foot of Third Street at the north end of the park in Lakeport.

Period of Record April 1951 through December 1959.

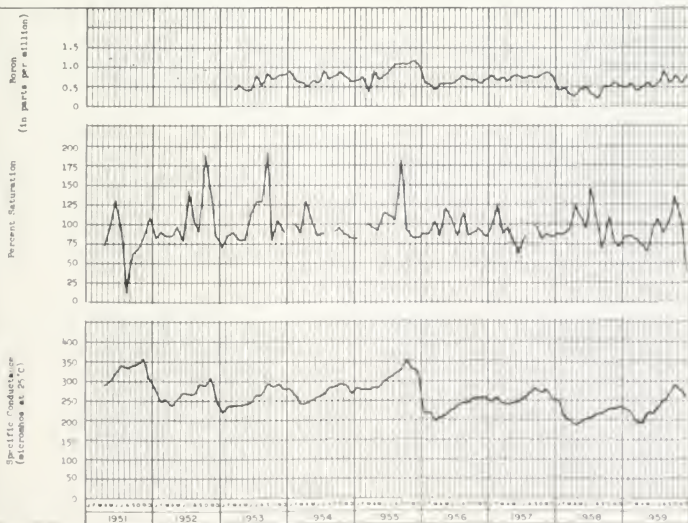
Water Quality Characteristics Antecedent data show the water of Clear Lake to be characteristically calcium-magnesium bicarbonate, slightly hard to moderately hard and within drinking water standards for mineral content. Boron, which has reached 1.23 ppm, frequently causes the water to be class 2 for irrigation use. Geologic formations and runoff from highly mineralized springs are considered to be the source of boron in Clear Lake.

Significant Water Quality Changes Boron continued to frequently exceed the 0.5 ppm maximum for class 1 irrigation use during 1959, causing the water in Clear Lake to be class 2 for eight months of 1959. Total radioactivity increased significantly from 1.2 $\mu\text{c}/\text{l}$ found in May to 15.7 $\mu\text{c}/\text{l}$ in September. However, the September value is still within safe limits.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	158	187	200	191
Temperature in °F	86	41	78	45
Dissolved oxygen in parts per million	16.5	1.8	11.6	4.4
Percent saturation	100	12	114	60
pH	8.7	6.8	8.5	7.1
Mineral constituents in parts per million				
Calcium (Ca)	30	17	24	21
Magnesium (Mg)	20	10	13	11
Sodium (Na)	17	6.4	15	6.7
Potassium (K)	2.8	1.4	1.9	0.8
Carbonate (CO ₃)	11	0.0	7	0
Bicarbonate (HCO ₃)	212	184	165	184
Sulfate (SO ₄)	12	1.8	11	8
Chloride (Cl)	10	1.5	7.5	3.6
Nitrate (NO ₃)	4	0.1	1.9	0.2
Fluoride (F)	0.4	0.0	0.2	0
Boron (B)	1.23	0.2	0.9	0.4
Silica (SiO ₂)	16	0.7	11	3.6
Total dissolved solids in parts per million	199	105	162	107
Percent sodium	19	13	18	14
Hardness as CaCO ₃ in parts per million				
Total	158	82	123	85
Noncarbonate	3	0.0	3	0
Turbidity	140	0.4	140	6
Coliform in most probable number per milliliter	>7,000	<0.045	2,400	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.10	0.00	0.1	0.00
Solid alpha	0.59	0.00	0.26	0.00
Dissolved beta	12.60	0.00	8.01	0.01
Solid beta	10.80	0.00	6.00	0.00

WATER QUALITY VARIATIONS



CLEAR LAKE AT LAKEPORT (STA. 41)

CACHE CREEK NEAR LOWER LAKE (STA. 42)

Sampling Point Station 42, which monitors outflow from Clear Lake to Cache Creek, is situated in Section 6 of Township 12 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at the USGS gage, approximately 500 feet downstream from Cache Creek Dam, 3.5 miles east of State Highway 53, and 5 miles southeast of Lower Lake.

Period of Record April 1951 through December 1959.

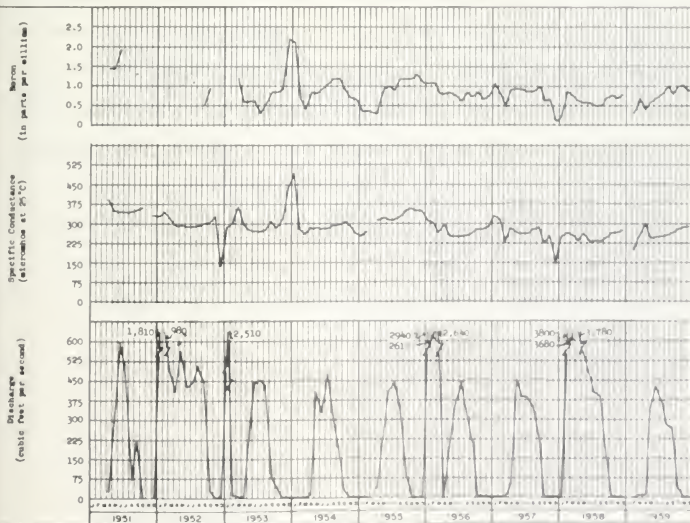
Water Quality Characteristics Analyses show Cache Creek water to be similar to the water found in Clear Lake at Lakeport, calcium-magnesium bicarbonate in character, slightly hard to moderately hard, and to consistently meet drinking water standards for mineral content. Boron frequently causes this water to be class 2 for irrigation use and at times (December 1953 and January 1954) places it in the class 3 irrigation water category. Only minor differences have been noted between the concentration of most constituents found at the Lakeport station on Clear Lake and Station 42.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1969	Minimum - 1969
Specific conductance (micromhos at 25°C)	261	177	201	150
Temperature in °F	84	71	78	65
Dissolved oxygen in parts per million	15.8	4.8	13.7	1.8
Percent saturation	117	40	104	50
pH	8.7	6.5	8.0	6.5
Mineral constituents in parts per million				
Calcium (Ca)	14	8	20	10
Magnesium (Mg)	11	3	11	3
Sodium (Na)	20	7.5	13	3
Potassium (K)	1.9	0.7	1.1	0.5
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	241	4	169	10
Sulfate (SO ₄)	3	1.8	4	1
Chloride (Cl)	36	3	10	4
Nitrate (NO ₃)	1.4	0.0	1.0	0.0
Fluoride (F)	0.8	0.0	0.5	0.0
Boron (B)	2.2	0.0	1.0	0.0
Silica (SiO ₂)	10	0.0	2.0	0.0
Total dissolved solids in parts per million	272	74	167	100
Percent sodium	23	14	20	11
Hardness as CaCO ₃ in parts per million				
Total	200	50	154	54
Noncarbonate	27	0.0	27	0.0
Turbidity	140	9	20	1
Coliform in most probable number per milliliter	>7,000	0.17	2,000	0.03
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.8	0.0	0.6	0.0
Solid alpha	0.51	0.0	0.4	0.0
Dissolved beta	23.8	0.0	2.0	1.0
Solid beta	4.08	0.0	4.0	2.0

WATER QUALITY VARIATIONS



CACHE CREEK NEAR LOWER LAKE (STA. 42)

CACHE CREEK NEAR CAPAY (STA. 80)

Sampling Point The Capay station is located in Section 8 of Township 10 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples of this water were collected from the right bank at the USGS gage, 2 miles upstream from the Clear Lake Water Company diversion dam, and 3 miles northwest of Capay.

Period of Record December 1951 through December 1959.

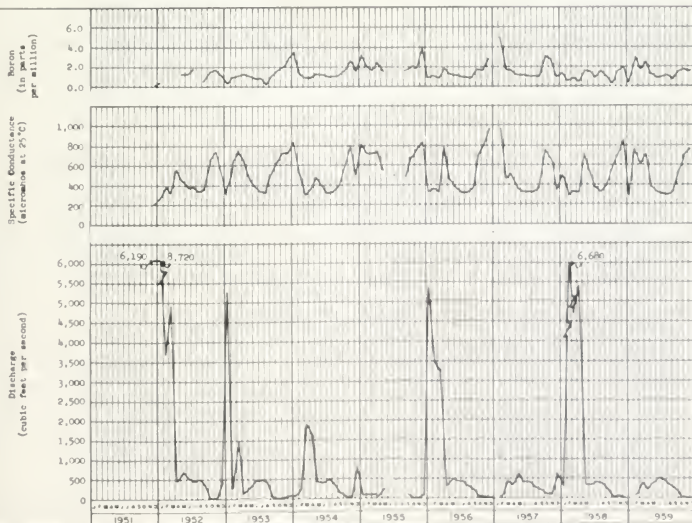
Water Quality Characteristics Cache Creek water at Station 80, as at upstream stations, is magnesium-calcium bicarbonate in character, moderately hard to very hard, within mineral standards for drinking water, and varies from class 1 to 3 for irrigation due to boron. Boron causes the water to be class 2 or 3 for irrigation during the major part of the year, with only flows following heavy precipitation being diluted to class 1 for irrigation. The effects of North Fork tributary flow have perennially been reflected by significant increases in most constituents in Cache Creek between Lower Lake and Capay. Boron concentrations and conductivity have an average increase in this reach of about 0.6 ppm and 240 micromhos, respectively.

Significant Water Quality Changes During 1959, the total radioactivity increased from 6.4 $\mu\text{c}/\text{l}$ in May to 15.16 $\mu\text{c}/\text{l}$ in September. Although these levels are slightly higher than levels detected in previous years, they are within safe limits.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	41	210	41	210
Temperature in °F	81	41	81	41
Dissolved oxygen in parts per million	12.1	7.8	12.1	7.8
Percent saturation	121	87	121	87
pH	8.4	8.8	8.2	7.7
Mineral constituents in parts per million				
Calcium (Ca)	50	11	31	2
Magnesium (Mg)	28	1	31	1
Sodium (Na)	31	12	27	1
Potassium (K)	1.6	2	2.5	1
Carbonate (CO ₃)	11	7	8	1
Bicarbonate (HCO ₃)	343	114	290	114
Sulfate (SO ₄)	56	1	15	1
Chloride (Cl)	140	9	131	1
Nitrate (NO ₃)	2.4	0.3	2.1	0.2
Fluoride (F)	0.3	0.06	0.24	0.06
Boron (B)	5.0	0.06	4.94	0.06
Silica (SiO ₂)	20	3.4	16.6	3.4
Total dissolved solids in parts per million	540	117	423	119
Percent sodium	42	16	40	22
Hardness as CaCO ₃ in parts per million				
Total	348	106	276	107
Noncarbonate	97	10	87	10
Turbidity	1,800	2.3	25	1
Coliform in most probable number per milliliter	2,400	200	620	200
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.81	0.00	0.26	0.21
Solid alpha	0.69	0.00	0.57	0.00
Dissolved beta	7.58	0.30	7.48	2.22
Solid beta	10.52	0.30	10.22	0.30

WATER QUALITY VARIATIONS



CACHE CREEK NEAR CAPAY (STA. 80)

CACHE CREEK, NORTH FORK NEAR LOWER LAKE (STA. 79)

Sampling Point Station 79 is located in Section 31 of Township 14 North, Range 6 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage, 2.7 miles upstream from State Highway 20 bridge, 6 miles east of Clear Lake Oaks, and 10 miles north of Lower Lake.

Period of Record December 1951 through December 1959.

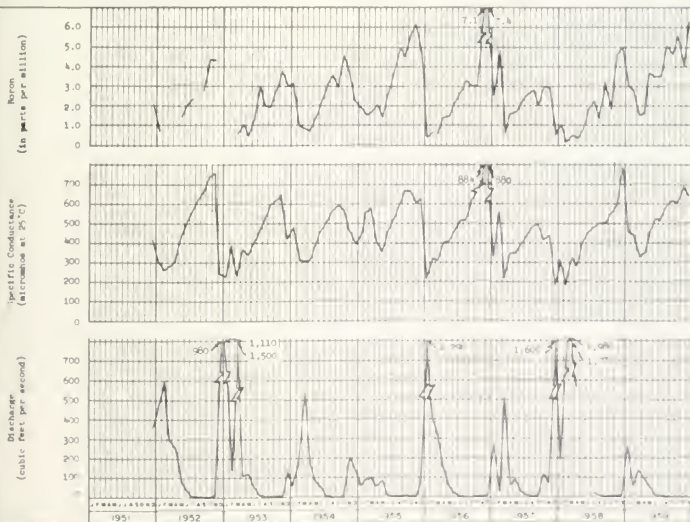
Water Quality Characteristics Samples of North Fork Cache Creek water indicate a characteristically magnesium-calcium bicarbonate water that ranges from slightly hard to very hard but consistently meets drinking water standards for mineral content. Boron concentration usually causes Cache Creek water to be class 2 for irrigation use and frequently class 3. Only during extremely wet seasons is boron found in concentrations less than the 0.5 ppm limit for class 1 irrigation water. North Fork Cache Creek drains an area containing numerous hot springs which have high concentrations of borates and other minerals which, even under pristine conditions, would cause high boron in runoff from the area. The quality of North Fork Cache Creek reflects the effects of drainage from the springs in the area. The concentration of constituents at this station are higher than those found in Clear Lake. During 1959 boron caused the water to be class 3 for irrigation use during ten months of the year and class 2 during the remaining two.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmoh at 75°C)	886	181	714	126
Temperature in °F	86	19	86	66
Dissolved oxygen in parts per million	14.4	1.5	12.9	1.1
Percent saturation	119	57	26	50
pH	8.2	4.4	8.1	4.3
Mineral constituents in parts per million				
Calcium (Ca)	80	16	64	11
Magnesium (Mg)	33	11	22	2
Sodium (Na)	46	12	34	11
Potassium (K)	2.8	1	1.8	0.7
Carbonate (CO ₃)	15	12	3	0.5
Bicarbonate (HCO ₃)	267	26	241	140
Sulfate (SO ₄)	95	9.4	86	17
Chloride (Cl)	145	4.3	141	18
Nitrate (NO ₃)	6.6	3.2	3.4	0.2
Fluoride (F)	7.7	3.2	4.5	0.1
Boron (B)	7.6	3.2	4.4	1.5
Silica (SiO ₂)	24	3.1	21	14
Total dissolved solids in parts per million	806	109	697	260
Percent sodium	57	34	23	70
Hardness as CaCO ₃ in parts per million				
Total	764	76	688	130
Noncarbonate	27	3.1	24	0.1
Turbidity	100	25	75	25
Coliforms in most probable number per milliliter	2,400	200	2,200	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.80	0.00	0.80	0.00
Solid alpha	0.71	0.00	0.71	0.00
Dissolved beta	12.64	0.00	12.64	0.00
Solid beta	0.00	0.00	0.00	0.00

WATER QUALITY VARIATIONS



CACHE CREEK, NORTH FORK NEAR LOWER LAKE STA. 791

PUTAH CREEK NEAR WINTERS (STA. 81)

Sampling Point Station 81 is located in Section 28 of Township 8 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, 1 mile downstream from the USGS gage, 8.2 miles west of Winters.

Period of Record December 1951 through December 1959.

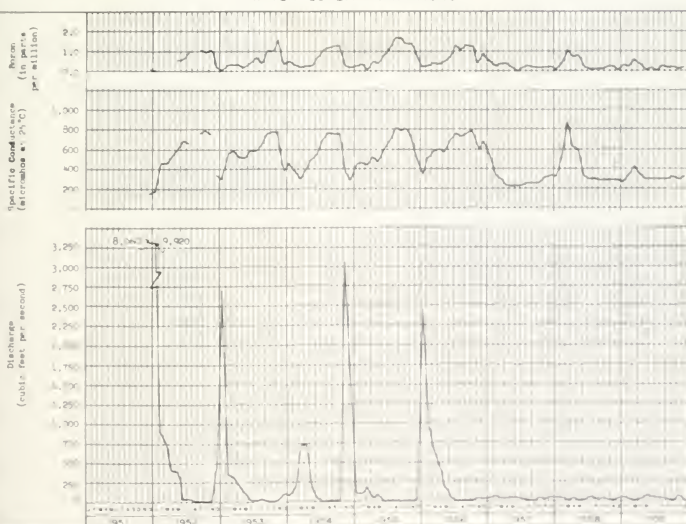
Water Quality Characteristics Past analyses of samples collected at Station 81 indicate a water of calcium-magnesium to magnesium-calcium bicarbonate character, slightly hard to very hard and of acceptable mineral content for drinking water. However, because of boron, it ranges from class 1 to class 2 for irrigation. Runoff from highly mineralized springs and leaching of geologic formations account for the boron in the basin.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1979	Minimum 1979
Specific conductance (microhm at 25°C)	50	46	114	263
Temperature in °F	83	47	84	37
Dissolved oxygen in parts per million	4.8	2.1	12.7	9.1
Percent saturation	142	29	126	58
pH	7.1	6.5	8.2	6.3
Mineral constituents in parts per million				
Calcium (Ca)	113	3	17	8
Magnesium (Mg)	78	13	20	4.3
Sodium (Na)	70	6.6	23	8.7
Potassium (K)	10.7	2	2	0.0
Carbonate (CO ₃)	61	7	7	2
Bicarbonate (HCO ₃)	148	81	203	82
Sulfate (SO ₄)	70	8.8	18	14
Chloride (Cl)	14	2.8	18	3.1
Nitrate (NO ₃)	2.7	0.8	10.2	0.2
Fluoride (F)	1.7	0.2	5.8	0.0
Boron (B)	1.7	0.2	6.6	0.0
Silica (SiO ₂)	29	0.3	18	0.3
Total dissolved solids in parts per million	520	58	240	148
Percent sodium	28	6	28	12
Hardness as CaCO ₃ in parts per million				
Total	368	40	183	98
Noncarbonate	54	0	24	0
Turbidity	2,000	0.8	50	0.8
Coliforms in most probable number per milliliter	>7,000	1001	200	100-15
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.13	0.30	0.10	0.10
Solid alpha	0.02	0.30	0.20	0.06
Dissolved beta	22	0.00	1.81	0.67
Solid beta	3.00	0.00	1.00	0.04

WATER QUALITY VARIATIONS



PUTAH CREEK NEAR WINTERS (STA. 81)

Sacramento Valley Northeast Stream Unit. Several small stream basins which drain the 1,140 square miles east of the northeastern portion of Sacramento Valley are included in this unit. These streams originate in the Sierra Nevada and flow along steep parallel courses to the valley floor. They have only minor tributaries and little development along their route. Principal streams in the unit, from north to south, are Antelope, Mill, Deer, Big Chico, and Butte Creeks. Annual natural mean runoff is about 1,180,000 acre-feet.

The terrain of these basins is almost entirely mountainous with only a few headwater valleys adaptable to irrigated agriculture. Agricultural, livestock raising, mining, recreational, and lumbering activities are carried on in these basins.

There are no significant waste discharges entering streams in this unit.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this unit and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Antelope Creek near Red Bluff	218
Antelope Creek near mouth	220
Mill Creek near Los Molinos	222
Big Chico Creek near Chico	224
Big Chico Creek at Chico	226
Butte Creek near Chico	228

ANTELOPE CREEK NEAR RED BLUFF (STA. 88e)

Sampling Point Red Bluff station is located in Section 8 of Township 27 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, approximately 3 miles east of Highway 99E on Belle Mill Road, and 8.5 miles east of Red Bluff.

Period of Record October 1958 through December 1959.

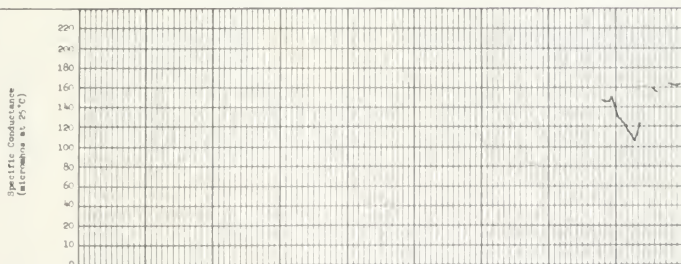
Water Quality Characteristics Antelope Creek water at the Red Bluff station is bicarbonate in type with no predominant cation, soft to slightly hard, class 1 for irrigation, and suitable for industrial and domestic uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (micro-mhos at 75°C)	24	15	24.6	15
Temperature in °F	89	48	88	42
Dissolved oxygen in parts per million	12.4	7.1	12.8	7.1
Percent saturation	96	57	96	57
pH	8.3	7.3	8.3	7.3
Mineral constituents in parts per million				
Calcium (Ca)	14	2.8	14	2.8
Magnesium (Mg)	10.6	4.6	7	4.6
Sodium (Na)	14	6	14	6
Potassium (K)	1	0.1	1	0.1
Carbonate (CO ₃)	4	2	4	2
Bicarbonate (HCO ₃)	10	5	10	5
Sulfate (SO ₄)	11	5	11	5
Chloride (Cl)	11	4.5	11	4.5
Nitrate (NO ₃)	2.1	0.5	2.1	0.5
Fluoride (F)	0.1	0.1	0.1	0.1
Boron (B)	0.2	0.2	0.2	0.2
Silica (SiO ₂)	41	18	41	18
Total dissolved solids in parts per million	145	90	145	90
Percent sodium	11	22	11	22
Hardness as CaCO ₃ in parts per million				
Total	64	41	64	41
Noncarbonate	10	5	10	5
Turbidity	10.8			
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1979			
Dissolved alpha			0.10	
Solid alpha			0.55	
Dissolved beta			9.49	
Solid beta			3.40	

WATER QUALITY VARIATIONS



FLOW NOT AVAILABLE

1951 1952 1953 1954 1955 1956 1957 1958 1959

ANTELOPE CREEK NEAR RED BLUFF (STA. 88e)

ANTELOPE CREEK NEAR MOUTH (STA. 88c)

Sampling Point Station 88c is located in Section 17 of Township 26 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected at State Highway 99E bridge, approximately 1.5 miles above the mouth, and about 9 miles southeast of the town of Red Bluff.

Period of Record October 1958 through December 1959.

Water Quality Characteristics Past analyses show the water to be bicarbonate in type with no predominant cation, slightly hard and class 1 for irrigation. Comparison of analyses of samples of water from Antelope Creek near mouth with those from near Red Bluff show a minor increase (25-100 micromhos) in most constituents. The increase in mineral concentrations is attributable to irrigation return flow and minor waste entering Antelope Creek in the reach between these two stations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	127	74	107	74
Temperature in °F	80	41	80	50
Dissolved oxygen in parts per million	11.4	4.4	11.4	4.4
Percent saturation	116	71	116	71
pH	7.4	7.4	7.4	7.4
Mineral constituents in parts per million				
Calcium (Ca)	25	4.6	25	4.6
Magnesium (Mg)	20	4.6	20	4.6
Sodium (Na)	24	4.4	24	4.4
Potassium (K)	5	1.4	5	1.4
Carbonate (CO ₃)	2	1.4	2	1.4
Bicarbonate (HCO ₃)	160	20	160	20
Sulfate (SO ₄)	31	1.8	31	1.8
Chloride (Cl)	32	6	32	6
Nitrate (NO ₃)	4.7	1.7	4.7	1.7
Fluoride (F)	0.2	0.2	0.2	0.2
Boron (B)	0.7	0.7	0.7	0.7
Silica (SiO ₂)	5.6	2	5.6	2
Total dissolved solids in parts per million	217	81	217	81
Percent sodium	38	19	38	19
Hardness as CaCO ₃ in parts per million				
Total	143	10	143	10
Noncarbonate	15	15	15	15
Turbidity	5	2	5	2
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter	See 1949	See 1949		
Dissolved alpha			0.001	0.001
Solid alpha			0.001	0.001
Dissolved beta			1.00	1.00
Solid beta			0.001	0.001

WATER QUALITY VARIATIONS



ANTELOPE CREEK NEAR MOUTH (STA. 88c)

MILL CREEK NEAR LOS MOLINOS (STA. 88)

Sampling Point Station 88 is located in Section 9 of Township 25 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, below State Highway 99E bridge, about 1.5 miles north of Los Molinos.

Period of Record July 1952 through December 1959.

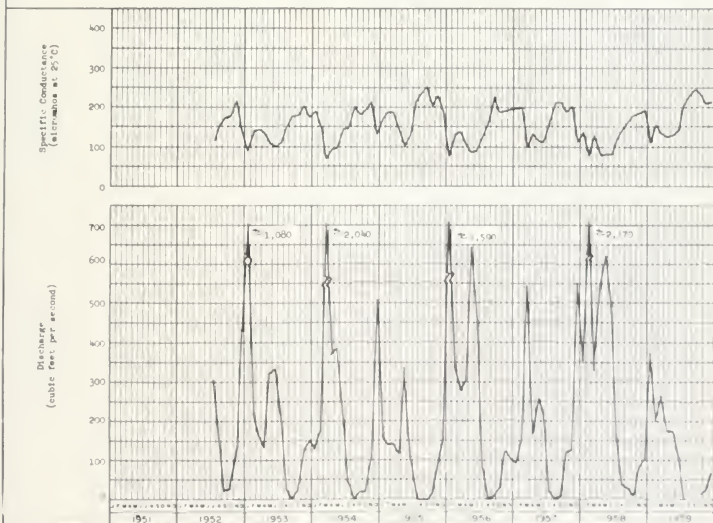
Water Quality Characteristics Generally the water at this station is bicarbonate in type with sodium in excess of the other cations, drinking water requirements for mineral content are met, hardness ranges from soft to slightly hard, and the water is good for most industrial uses. Boron periodically places this water in class 2 for irrigation during periods of low flow.

Significant Water Quality Changes During 1959 mineral concentrations varied somewhat from maximum-minimum ranges established during previous years. Calcium and sulfates reached 20 ppm, chlorides 28 ppm, and boron 0.8 ppm, all representing the highest values reported during the period of record. Silica ranged from 44 to 30 ppm, representing the greatest variation for the period of record. Surface runoff during the year was somewhat lower than average and the smaller amount of dilution water available probably accounted for the increase in mineral concentrations. Also, the considerable use made of the water in Mill Creek for irrigated agriculture and related irrigation returns undoubtedly affected the quality.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	252	7	247	109
Temperature in °F	93	37	56	44
Dissolved oxygen in parts per million	11.4	7.1	18.5	8.6
Percent saturation	140	84	60	94
pH	8.3	6.7	7.8	7.1
Mineral constituents in parts per million				
Calcium (Ca)	20	6.2	20	9.8
Magnesium (Mg)	9.3	1.1	8.3	7.1
Sodium (Na)	21	4.1	20	7.1
Potassium (K)	3.8	1.2	3	1.2
Carbonate (CO ₃)	2.3	0.4	2	0.4
Bicarbonate (HCO ₃)	90	24	91	40
Sulfate (SO ₄)	20	7.7	20	11
Chloride (Cl)	26	2.5	26	9
Nitrate (NO ₃)	7	2	5.5	2.4
Fluoride (F)	1.1	0.1	1	0.1
Boron (B)	0.8	0.1	0.8	0.1
Silica (SiO ₂)	44	26	44	30
Total dissolved solids in parts per million	178	52	178	81
Percent sodium	43	22	41	28
Hardness as CaCO ₃ in parts per million				
Total	88	22	84	38
Noncarbonate	22	0.7	12	5.3
Turbidity	55	0.0	19	0
Coliform in most probable number per milliliter	See 1959	See 1959	2,470	0.02
Radiactivity in micro-micro curies per liter				
Dissolved alpha			0.34	
Solid alpha			0.43	
Dissolved beta			1.44	
Solid beta			2.80	

WATER QUALITY VARIATIONS



MILL CREEK NEAR LOS MOLINOS (STA. 88)

BIG CHICO CREEK NEAR CHICO (STA. 85)

Sampling Point Station 85 is located in Section 9 of Township 22 North, Range 2 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage, approximately 6 miles northeast of Chico and 12.9 miles upstream from the mouth.

Period of Record July 1952 through December 1959.

Water Quality Characteristics Past records show the water to be consistently good to excellent in quality, calcium-magnesium or magnesium-calcium bicarbonate in type, class 1 for irrigation, soft to slightly hard, and very good for domestic and industrial purposes.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohms at 25°C)	221	65	215	105
Temperature in °F	77	36	76	43
Dissolved oxygen in parts per million	13.4	7.6	12.4	7.6
Percent saturation	113	83	100	76
pH	8.5	6.8	8	7.2
Mineral constituents in parts per million				
Calcium (Ca)	20	4.5	20	14
Magnesium (Mg)	9.8	2.8	7.1	6.8
Sodium (Na)	17	2.2	16	3.9
Potassium (K)	2.3	0.4	1.3	0.8
Carbonate (CO ₃)	3	0.0	1	0.0
Bicarbonate (HCO ₃)	115	37	113	49
Sulfate (SO ₄)	6.3	1.8	4.0	1.1
Chloride (Cl)	18	0.4	12	1.5
Nitrate (NO ₃)	1.3	0.0	0.4	0.3
Fluoride (F)	0.2	0.0	0.1	0.0
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	42	30	40	19
Total dissolved solids in parts per million	162	47	146	77
Percent sodium	33	13	31	13
Hardness as CaCO ₃ in parts per million				
Total	92	27	92	40
Noncarbonate	9	0.0	2	0.0
Turbidity	20	0.0	20	0.0
Coliform in most probable number per milliliter	7,000	2.30	620	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



BIG CHICO CREEK NEAR CHICO (STA. 85)

BIG CHICO CREEK AT CHICO (STA. 85a)

Sampling Point Station 85a is located in Section 28 of Township 22 North, Range 1 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the Rose Avenue bridge, at the intersection of Rose and Bidwell Avenues, in the City of Chico.

Period of Record January 1959 through December 1959.

Water Quality Characteristics The water at Station 85a is excellent in quality, a bicarbonate type with calcium dominant over other cations, class 1 for irrigation, soft to slightly hard and has a mineral content which meets drinking water requirements.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (microhms at 25°C)	See (P. 3)	See (P. 3)	100	100
Temperature in °F			73	66
Dissolved oxygen in parts per million			11.2	8.1
Percent saturation			100	71
pH			7.8	7.3
Mineral constituents in parts per million				
Calcium (Ca)			15	8.3
Magnesium (Mg)			1.2	0.4
Sodium (Na)			30	6.1
Potassium (K)			0.5	0.2
Carbonate (CO ₃)			0.5	0.2
Bicarbonate (HCO ₃)			125	28
Sulfate (SO ₄)			18	8.1
Chloride (Cl)			14	2.0
Nitrate (NO ₃)			1.0	0.2
Fluoride (F)			0.2	0.1
Boron (B)			0.8	0.1
Silica (SiO ₂)			47	27
Total dissolved solids in parts per million			168	79
Percent sodium			12	15
Hardness as CaCO ₃ in parts per million				
Total			86	40
Noncarbonate			15.8	0.0
Turbidity	(Not Measured)			
Coliforms in most probable number per milliliter	(Not Measured)			
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.12	0.03
Solid alpha			0.03	0.01
Dissolved beta			2.22	0.01
Solid beta			2.56	0.01

WATER QUALITY VARIATIONS



BIG CHICO CREEK AT CHICO (STA. 85a)

BUTTE CREEK NEAR CHICO (STA. 84)

Sampling Point Station 84 is located in Section 36 of Township 22 North, Range 2 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank at the USGS gage, 0.8 mile downstream from Little Butte Creek, and 7.5 miles east of Chico.

Period of Record July 1952 through December 1959.

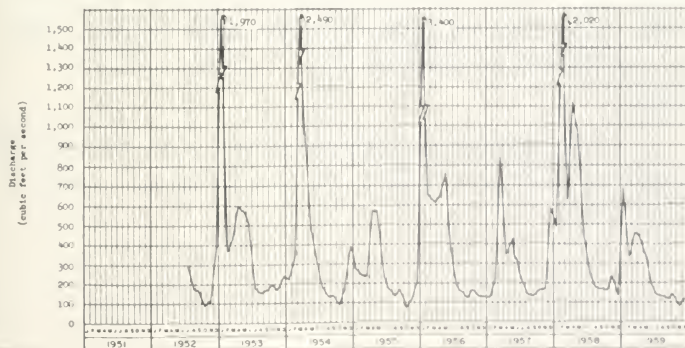
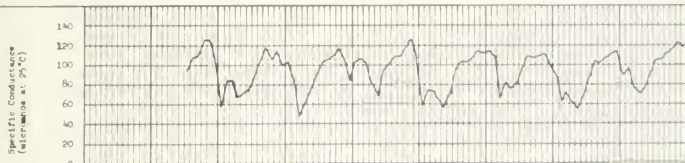
Water Quality Characteristics The character of the water at Station 84 is generally bicarbonate with no predominant cation; however, occasionally it changes to calcium bicarbonate. Chemical analyses show very little variation in quality occurs and that the water is soft, class 1 for irrigation, meets drinking water standards, and is excellent for industrial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhms at 25°C)	107	47	104	71.4
Temperature in °F	84	31	82	40
Dissolved oxygen in parts per million	13.0	7.5	12.8	10.0
Percent saturation	114	70	100	81
pH	8.3	6.3	8.1	6.1
Mineral constituents in parts per million				
Calcium (Ca)	116	4.8	114	6.0
Magnesium (Mg)	6.4	1.2	5.2	1.4
Sodium (Na)	6.4	1.2	5.2	2.4
Potassium (K)	1.4	0.2	1.2	0.2
Carbonate (CO ₃)	76	0.2	75.8	0.2
Bicarbonate (HCO ₃)	76	0.2	75.8	0.2
Sulfate (SO ₄)	4.8	0.2	4.6	0.2
Chloride (Cl)	0.2	0.2	0.0	0.2
Nitrate (NO ₃)	0.2	0.2	0.0	0.2
Fluoride (F)	0.2	0.2	0.0	0.2
Boron (B)	0.2	0.2	0.0	0.2
Silica (SiO ₂)	26	17	24	11
Total dissolved solids in parts per million	94	15	79	11
Percent sodium	23	9.8	18	11
Hardness as CaCO ₃ in parts per million				
Total	58	21	46	11
Noncarbonate	12	0.2	2	0.2
Turbidity	5.0	0.2	20	0.2
Coliforms in most probable number per milliliter	620	7.21	620	7.21
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



BUTTE CREEK NEAR CHICO (STA. 84)

Feather River Basin. Feather River drainage is composed of numerous tributaries which form a dendritic pattern on 3,740 square miles in the northeastern portion of the Central Valley Region. The topography of the area is predominantly mountainous with only 687 square miles classified as valley and mesa lands. The average seasonal runoff of the Feather River Basin is 4,596,000 acre-feet.

Feather River, the major tributary to Sacramento River, rises in large headwater valleys located high in the Sierra Nevada. Flowing out of these valley or meadow areas the river cascades down the steep granitic slopes of the Sierra. In the foothills and along the valley floor the Feather River gradient gradually flattens out and at its mouth the river is considerably stilled.

Lumbering, recreation, and livestock raising are the main economic pursuits in the upper reaches of this basin. In the foothill and valley area agriculture is the predominant enterprise.

Log ponds, small resort areas, and communities located along the waterway all discharge waste into the river system. The only discharges of significant quantity, however, are from the Cities of Oroville (0.8 mgd), Gridley (>0.5 mgd), Yuba City (4 mgd), and Marysville (1.8 mgd). Waste discharges have not created significant impairment problems in this basin.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Feather River at Nicolaus	232
Feather River near Oroville	234
Feather River below Shanghai Bend	236
Indian Creek near Crescent Mills	238

FEATHER RIVER AT NICOLAUS (STA. 20)

Sampling Point Station 20 is situated in Section 12 of Township 12 North, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the USGS gage on Garden Highway bridge at Nicolaus, and 2.9 miles downstream from the confluence with Bear River.

Period of Record April 1951 through December 1959.

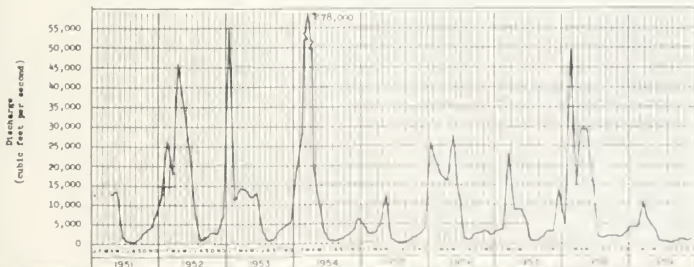
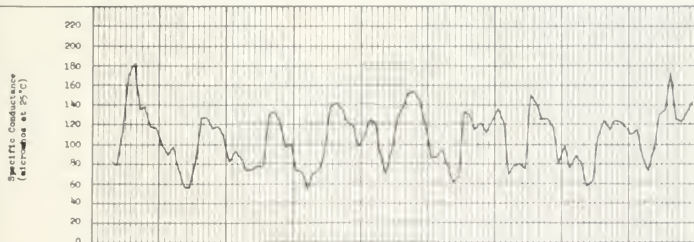
Water Quality Characteristics Chemical classification of past analyses from Station 20 show the water to be calcium-magnesium bicarbonate in character, class 1 for irrigation, soft to slightly hard, and within drinking water standards for mineral content. There is no significant change in water quality between upstream stations and the Nicolaus station, indicating tributary inflow of such streams as the Yuba and Bear Rivers has little effect on quality of Feather River water.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmoh at 25°C)	281	50	172	72.8
Temperature in °F	81	17	71	44
Dissolved oxygen in parts per million	11.5	7.4	12.5	7.4
Percent saturation	114	81	102	80
pH	7.9	6.5	7.7	7.1
Mineral constituents in parts per million				
Calcium (Ca)	7.4	1.1	11	9.4
Magnesium (Mg)	7.4	1.4	7.2	5.4
Sodium (Na)	6.7	1.1	6.8	1.1
Potassium (K)	2.2	0.4	1.8	0.4
Carbonate (CO ₃)	8.0	0.0	8.0	0.0
Bicarbonate (HCO ₃)	99	10	99	10
Sulfate (SO ₄)	7.2	0.0	7.2	0.0
Chloride (Cl)	8.1	0.0	8.1	0.0
Nitrate (NO ₃)	1.6	0.0	1.7	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.18	0.0	0.1	0.0
Silica (SiO ₂)	19	1.0	16	1.0
Total dissolved solids in parts per million	120	16	103	16
Percent sodium	23	1.1	21	1.1
Hardness as CaCO ₃ in parts per million				
Total	75	20	75	14
Noncarbonate	6	0.0	6	0.0
Turbidity	100	0.0	40	0.0
Coliform in most probable number per milliliter	>7,000	0.0	2,400	0.0
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.82	0.00	0.80	0.00
Solid alpha	1.22	0.00	0.81	0.00
Dissolved beta	11.92	0.00	6.44	0.00
Solid beta	0.18	0.00	0.18	0.00

WATER QUALITY VARIATIONS



FEATHER RIVER AT NICOLAUS (STA. 20)

FEATHER RIVER NEAR OROVILLE (STA. 19)

Sampling Point Station 19 is located in Section 2 of Township 19 North, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at the USGS gage 75 feet upstream from State Highway 24 bridge, 2 miles downstream from the confluence of the North and Middle Forks Feather River, and 4 miles northeast of Oroville.

Period of Record April 1951 through December 1959.

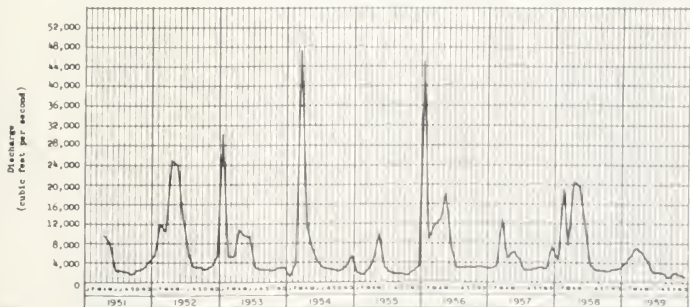
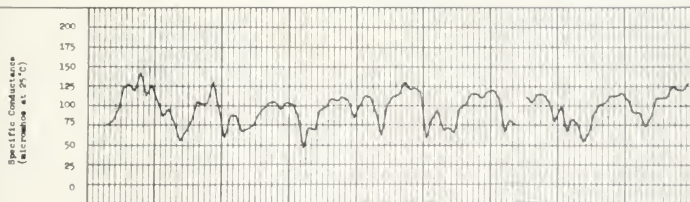
Water Quality Characteristics Past analyses show the water to be generally calcium bicarbonate in character, class 1 for irrigation, soft, and within mineral requirements for drinking water. Only minor changes in the quality of this excellent water have been detected. Most constituents have been found in slightly lower concentrations (averaging about 90 micromhos) at Station 19 as compared to Indian Creek (Station 17d).

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	119	48	87	48
Temperature in °F	78	35	74	40
Dissolved oxygen in parts per million	18.5	4.4	13.1	8.1
Percent saturation	120	48	72	40
pH	8.1	6.8	7.3	7
Mineral constituents in parts per million				
Calcium (Ca)	16	4.8	11	4.8
Magnesium (Mg)	6.2	1.5	4.7	1.2
Sodium (Na)	6.6	0.9	5.6	0.9
Potassium (K)	3.8	0.5	3.3	0.7
Carbonate (CO ₃)	1.0	0.0	1.0	0.0
Bicarbonate (HCO ₃)	77	18	59	18
Sulfate (SO ₄)	5.2	1	4.2	1
Chloride (Cl)	6.0	0.0	6.0	0.0
Nitrate (NO ₃)	1.8	0.0	1.8	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.30	0.0	0.30	0.0
Silica (SiO ₂)	21	0.0	21	0.0
Total dissolved solids in parts per million	98	18	80	18
Percent sodium	25	10	21	10
Hardness as CaCO ₃ in parts per million				
Total	56	22	34	13
Noncarbonate	1	0	1	0
Turbidity	130	0.0	30	0.0
Coliform in most probable number per milliliter	7,000	0.045	230	0.13
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.60	0.00	0.60	0.00
Solid alpha	0.25	0.00	0.25	0.00
Dissolved beta	10.81	0.00	10.81	0.00
Solid beta	10.41	0.00	10.41	0.00

WATER QUALITY VARIATIONS



FEATHER RIVER NEAR OROVILLE (STA. 19)

FEATHER RIVER BELOW SHANGHAI BEND (STA. 20a)

Sampling Point Shanghai Bend station is situated within Section 11 of Township 14 North, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the DWR gage 1.2 miles east of the junction of U. S. Highway 40 Alternate and Barry Road, and 4.5 miles south of Yuba City.

Period of Record July 1958 through December 1959.

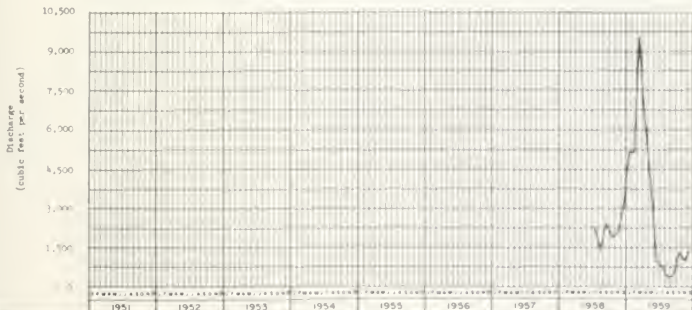
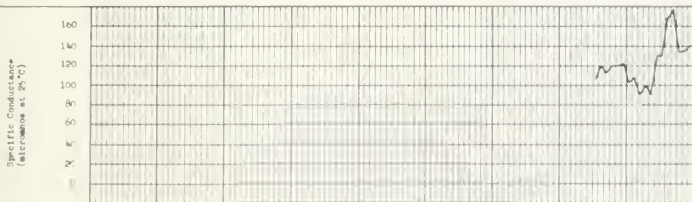
Water Quality Characteristics The water at Station 20a is calcium to calcium-magnesium bicarbonate in character, class 1 for irrigation, soft, and within drinking water standards for mineral content. Comparison of quality between Station 20a and upstream stations indicate no significant changes in mineral concentrations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	176	88.5	176	88.5
Temperature in °F	80	44	80	44
Dissolved oxygen in parts per million	12.2	7.7	12.2	7.9
Percent saturation	108	91	108	94
pH	7.7	7.1	7.7	7.1
Mineral constituents in parts per million				
Calcium (Ca)	17	8.2	17	8.2
Magnesium (Mg)	7.7	3.2	7.7	3.2
Sodium (Na)	7.1	2.9	7.1	2.9
Potassium (K)	2.6	0.5	1.7	0.5
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	00	47	00	47
Sulfate (SO ₄)	11	0.0	11	1.9
Chloride (Cl)	4.8	1.2	4.8	1.2
Nitrate (NO ₃)	0.9	0.0	0.2	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.2	0.0	0.2	0.0
Silica (SiO ₂)	21	11	21	11
Total dissolved solids in parts per million	111	62	111	62
Percent sodium	19	12	19	12
Hardness as CaCO ₃ in parts per million				
Total	74	38	74	38
Noncarbonate	7	0.0	7	0.0
Turbidity	15	5	15	5
Coliform in most probable number per milliliter	2,400	2.3	2,400	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.61	0.17	0.61	0.17
Solid alpha	0.35	0.00	0.35	0.00
Dissolved beta	6.33	4.28	5.53	4.28
Solid beta	6.37	0.64	1.18	0.64

WATER QUALITY VARIATIONS



FEATHER RIVER BELOW SHANGHAI BEND (STA. 20a)

INDIAN CREEK NEAR CRESCENT MILLS (STA. 17d)

Sampling Point Station 17d is located in Section 25 of Township 26 North, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the creek, from the Taylorsville Road bridge (0.7 mile upstream from the USGS gage), 1.5 miles upstream from Dixie Creek and 1 mile south of Crescent Mills.

Period of Record April 1951 through December 1959.

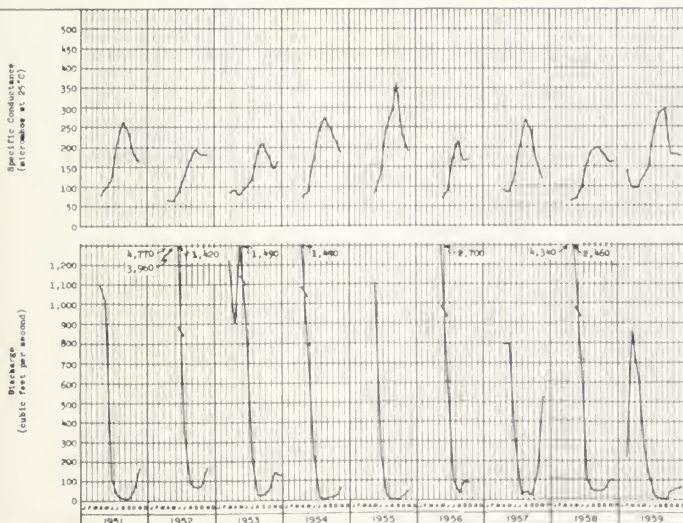
Water Quality Characteristics The water at Station 17d is calcium bicarbonate in character, ranges from soft to moderately hard, consistently meets mineral requirements for drinking water, and is class 1 for irrigation.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (microhmohms at 25°C)	963	63.8	203	27.1
Temperature in °F	76	35	74	35
Dissolved oxygen in parts per million	13.6	6.7	12.1	6.0
Percent saturation	102	68	102	76
pH	7.6	6.7	7.1	6.2
Mineral constituents in parts per million				
Calcium (Ca)	36	6.8	31	12
Magnesium (Mg)	12	2.1	9.8	4.4
Sodium (Na)	21	2.7	18	4.1
Potassium (K)	3.1	0.8	3.1	1.3
Carbonate (CO ₃)	0.7	0.7	0.0	0.0
Bicarbonate (HCO ₃)	201	16	17	50
Sulfate (SO ₄)	1.1	1.0	.8	1.8
Chloride (Cl)	12	0.0	9.5	1.5
Nitrate (NO ₃)	1.1	0.0	0.5	0.1
Fluoride (F)	0.2	0.0	0.1	0.1
Boron (B)	0.1	0.0	0.2	0.1
Silica (SiO ₂)	12	10	27	24
Total dissolved solids in parts per million	291	43	106	66
Percent sodium	34	16	24	17
Hardness as CaCO ₃ in parts per million				
Total	130	26	118	30
Noncarbonate	6	0.0	3	0.0
Turbidity	80	0.8	35	2
Coliform in most probable number per milliliter	>7,000	<0.045	690	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.51	0.00	0.51	0.00
Solid alpha	0.83	0.00	0.17	0.00
Dissolved beta	35.41	0.00	6.40	0.00
Solid beta	28.6	0.00	2.28	0.70

WATER QUALITY VARIATIONS



INDIAN CREEK NEAR CRESCENT MILLS (STA. 17d)

Yuba-Bear Rivers Unit. The Yuba-Bear Rivers Unit is located on the western slope of the Sierra Nevada in the west-central portion of the Central Valley Region. Included within the unit are about 1,490 square miles of land ranging from rugged mountains to rolling foothills with only about 17 square miles classified as valley and mesa. Mean annual runoff of the drainage systems of the Yuba and Bear Rivers are 2,415,000 acre-feet and 356,000 acre-feet, respectively.

Developments in the unit are typical of mountainous areas and include lumbering, mining, resorts, recreational facilities, and livestock raising.

Waste discharges from these activities are relatively minor and, except for the cities of Auburn (0.6 mgd), Nevada City (0.5 mgd), and Grass Valley (1.0 mgd), do not exceed 0.5 million gallons per day.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this unit and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Yuba River near Smartville	242
Yuba River at Marysville	244
Bear River near Wheatland	246
Bear River near mouth	248

YUBA RIVER NEAR SMARTVILLE (STA. 21a)

Sampling Point Station 21a is situated in Section 20 of Township 16 North, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at Highway 20 bridge 4 miles downstream from the confluence of Deer Creek, 5 miles downstream from Narrows Dam, and 2 miles northwest of Smartville.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Antecedent data show the water at Station 21a to be calcium bicarbonate in character, class 1 for irrigation, soft to slightly hard, and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	136	43.1	136	43.1
Temperature in °F	77	19	76	46
Dissolved oxygen in parts per million	14.8	7.5	12.1	8.7
Percent saturation	124	48	100	48
pH	8.4	6.7	7.4	7.2
Mineral constituents in parts per million				
Calcium (Ca)	18	2	15	9.9
Magnesium (Mg)	5.3	1	1.5	0.9
Sodium (Na)	5.7	1.1	4.4	1.9
Potassium (K)	1.8	0.2	1.4	0.2
Carbonate (CO ₃)	2.4	0.2	2.2	0.2
Bicarbonate (HCO ₃)	72	15	57	15
Sulfate (SO ₄)	8.2	1.9	6.3	1.9
Chloride (Cl)	1.8	0.2	1.6	0.2
Nitrate (NO ₃)	1.2	0.2	1.0	0.2
Fluoride (F)	0.2	0.1	0.1	0.1
Boron (B)	0.1	0.1	0.1	0.1
Silica (SiO ₂)	19	12	16	12
Total dissolved solids in parts per million	97	31	93	47
Percent sodium	23	9	17	12
Hardness as CaCO ₃ in parts per million				
Total	63	17	63	28
Noncarbonate	44	10	34	10
Turbidity	150	0.0	55	0.0
Coliform in most probable number per milliliter	>7,000	1.0	230	1.0
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.30	0.00	0.30	0.00
Solid alpha	0.8	0.00	0.77	0.00
Dissolved beta	9.04	0.00	1.62	0.00
Solid beta	10.8	0.00	1.80	0.00

WATER QUALITY VARIATIONS



YUBA RIVER NEAR SMARTVILLE (STA. 21a)

YUBA RIVER AT MARYSVILLE (STA. 21)

Sampling Point Station 21 on Yuba River is located in Section 18 of Township 15 North, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the channel of flow, from Simpson Lane bridge, approximately 1 mile upstream from the confluence with the Feather River.

Period of Record April 1951 through December 1959.

Water Quality Characteristics The water in Yuba River at Station 21 is calcium bicarbonate in character, class 1 for irrigation, soft to slightly hard, and within drinking water standards for mineral content. There is no significant difference noted in the quality of water at Station 21 and the upstream station (21a) near Smartville.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1977	Minimum - 1979
Specific conductance (micro-mhos at 25°C)	284	44	204	49
Temperature in °F	83	38	83	62
Dissolved oxygen in parts per million	14.5	7.4	11.5	8.5
Percent saturation	119	84	108	99
pH	8.0	6.6	7.7	6.1
Mineral constituents in parts per million				
Calcium (Ca)	19	4.2	19	9.6
Magnesium (Mg)	6.4	1.1	6.4	2.9
Sodium (Na)	6.0	1.2	5.9	1.3
Potassium (K)	1.8	0.1	1.8	0.1
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	96	21	96	5
Sulfate (SO ₄)	17	0.0	17	2.9
Chloride (Cl)	5.3	0.0	4.5	1.0
Nitrate (NO ₃)	1.4	0.0	1.4	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.28	0.0	0.1	0.0
Silica (SiO ₂)	21	10	20	14
Total dissolved solids in parts per million	141	10.4	141	48
Percent sodium	19	9	19	11
Hardness as CaCO ₃ in parts per million				
Total	96	18	96	11
Noncarbonate	17	0.0	17	1
Turbidity	220	0.0	17	0.0
Coliform in most probable number per milliliter	>7,000	0.11	230	0.11
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.22	0.00	0.20	0.00
Solid alpha	2.15	0.00	0.72	0.22
Dissolved beta	20.68	0.00	11.25	0.00
Solid beta	5.15	0.00	0.76	0.00

WATER QUALITY VARIATIONS



YUBA RIVER AT MARYSVILLE (STA. 21)

BEAR RIVER NEAR WHEATLAND (STA. 78)

Sampling Point Station 78 is located in Section 3 of Township 13 North, Range 5 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, (at the USGS gage) near U. S. Highway 99E bridge 1 mile southeast of Wheatland.

Period of Record December 1951 through December 1959.

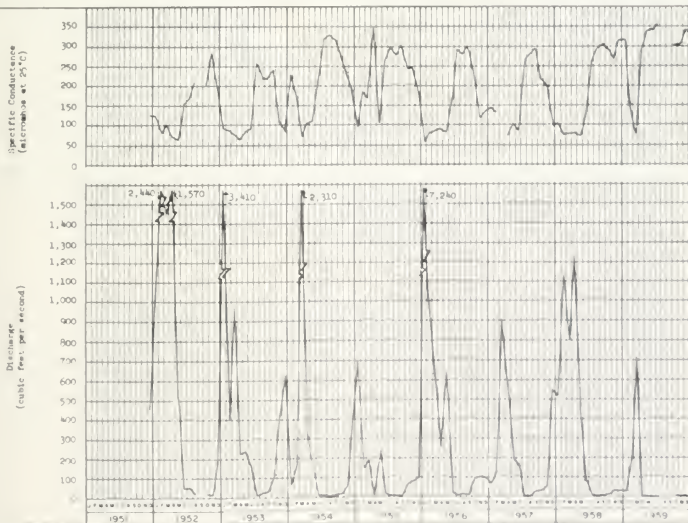
Water Quality Characteristics The water at Station 78 is calcium or calcium-magnesium bicarbonate in character, soft to moderately hard, consistently class 1 for irrigation and within drinking water standards for mineral content.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micro-mhos at 25°C)	156	57.5	156	78.8
Temperature in °F	87	77	78	45
Dissolved oxygen in parts per million	16	6.1	11.6	7.6
Percent saturation	144	75	109	81
pH	8.3	6.8	8.1	7.2
Mineral constituents in parts per million				
Calcium (Ca)	17	7	8	
Magnesium (Mg)	40	7	18	
Sodium (Na)	14	1.6	11	2.3
Potassium (K)	2.6	7.1	3.7	
Carbonate (CO ₃)	4	2	4	3.0
Bicarbonate (HCO ₃)	158	24	156	91
Sulfate (SO ₄)	39	9.8	19	
Chloride (Cl)	14	9	16	4.4
Nitrate (NO ₃)	3.4	2	2	
Fluoride (F)	0.2	0.0	0.0	
Boron (B)	0.69	0.0	0.1	0.2
Silica (SiO ₂)	23	17	19	
Total dissolved solids in parts per million	206	56	226	50
Percent sodium	29	4	14	9
Hardness as CaCO ₃ in parts per million				
Total	173	24	173	32
Noncarbonate	51	7	51	7
Turbidity	3,400	2.2	30	0.9
Coliform in most probable number per milliliter	7,000	0.05	23	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.65	0.00	0.00	
Solid alpha	4.43	0.00	0.00	
Dissolved beta	8.79	0.00	0.00	
Solid beta	8.1	0.00	0.00	

WATER QUALITY VARIATIONS



BEAR RIVER NEAR WHEATLAND (STA. 78)

BEAR RIVER NEAR MOUTH (STA. 20b)

Sampling Point The station near the mouth of the Bear River is located in Section 20 of Township 13 North, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected at Highway 24 (Feather River Boulevard) at Rio Oso, approximately 2 miles upstream from the mouth.

Period of Record November 1958 through December 1959.

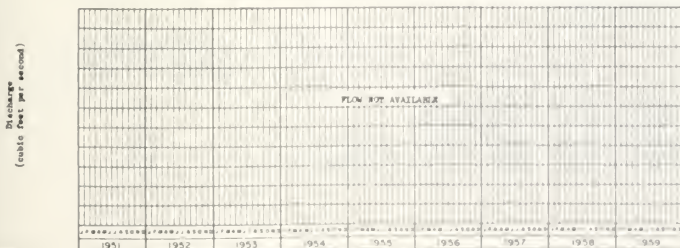
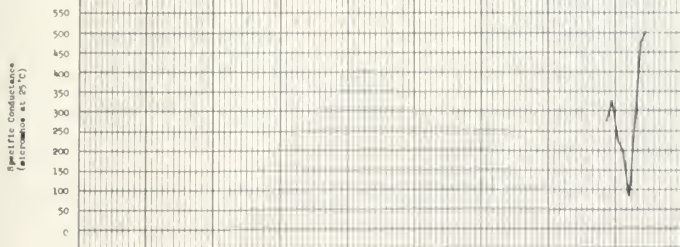
Water Quality Characteristics Water at Station 20b is generally a bicarbonate type with calcium, magnesium, and sodium alternating as predominant cations. The water ranges from soft to moderately hard and meets mineral requirements for drinking and class 1 irrigation water.

Significant Water Quality Changes Radioactivity at Station 20b during 1959 was the highest reported in the Yuba-Bear Rivers Unit, 13.7 $\mu\text{c}/\text{l}$ in May and 17.1 $\mu\text{c}/\text{l}$ in September. These values are well within safe limits.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	211	84	211	84
Temperature in °F	80	47	80	47
Dissolved oxygen in parts per million	12.1	7.1	12.1	7.0
Percent saturation	100	60	100	60
pH	7.9	7.1	7.9	7.1
Mineral constituents in parts per million				
Calcium (Ca)	15	8.8	15	8.8
Magnesium (Mg)	23	2.0	23	2.0
Sodium (Na)	42	1.2	42	1.2
Potassium (K)	8.1	0.4	8.1	0.4
Carbonate (CO ₃)	5.0	0.0	5.0	0.0
Bicarbonate (HCO ₃)	211	84	211	84
Sulfate (SO ₄)	5	11	5	11
Chloride (Cl)	46	4.2	46	4.2
Nitrate (NO ₃)	1.8	0.0	1.8	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	4.8	1.0	4.8	1.0
Total dissolved solids in parts per million	314	56	314	56
Percent sodium	37	12	37	12
Hardness as CaCO ₃ in parts per million				
Total	180	14	180	14
Noncarbonate	51	3	51	3
Turbidity	See 1959	See 1959	1	1
Coliforms in most probable number per milliliter (Not Measured)	See 1959	See 1959		
Radioactivity in micro-micro curies per liter	See 1959	See 1959	0.51	0.00
Dissolved alpha			0.43	0.10
Dissolved beta			15.76	11.05
Solid beta			0.86	0.00

WATER QUALITY VARIATIONS



BEAR RIVER NEAR MOUTH (STA. 20b)

American River Basin. The American River Basin drains the southeast corner of the Sacramento River Valley Basin of the Central Valley and covers approximately 1,940 square miles of the western slopes of the Sierra Nevada. The basin is characterized by foothill and mountainous terrain with elevations varying from 150 to 10,000 feet. Valley and mesa land comprise only 20 square miles in the basin. Mean annual runoff is estimated to be about 2,774,000 acre-feet.

The American River Basin is favored by forest, mineral, and recreational resources, which have all been developed for economic return. Although valley and mesa land comprise only about one percent of the area, considerable orchard development is found in the foothills and on the lower mountain slopes. These developments, along with numerous irrigation and municipal diversions for use in the valley, are the main water users in this watershed.

Impairment problems caused by the discharge of wastes are not of a serious nature under the present regimen. Placerville is the only source of waste discharge in excess of 0.5 mgd along the drainage basin above the valley floor.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
American River at Nimbus Dam	252
American River at Sacramento	254
American River, Middle Fork near Auburn	256
American River, South Fork near Lotus	258

AMERICAN RIVER AT NIMBUS DAM (STA. 22a)

Sampling Point The sampling point for Station 22a is located in Section 16, Township 9 North, Range 7 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank just downstream from the fish screen at Nimbus Fish Hatchery, about 10 miles east of Sacramento.

Period of Record November 1958 through December 1959.

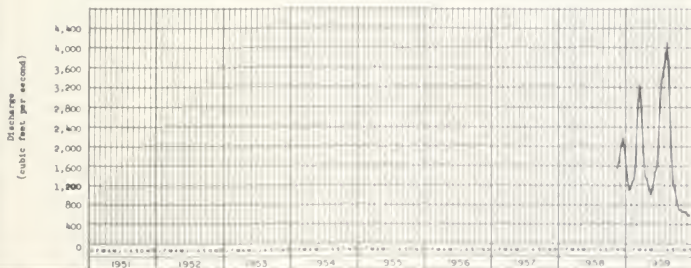
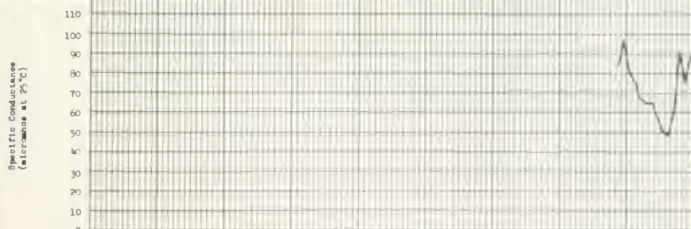
Water Quality Characteristics Past analyses show the water to be a calcium bicarbonate type of excellent mineral quality. From a mineral standpoint the water is suitable for domestic use, class 1 for irrigation, and soft with a maximum recorded hardness of 44 ppm. Water quality at Nimbus Dam station is very similar to that at Stations 22b and 22c located upstream on the Middle Fork and South Fork, respectively.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	110	4.2	100	4.2
Temperature in °F	71	50	71	50
Dissolved oxygen in parts per million	12.5	7.1	12.5	7.1
Percent saturation	120	80	120	80
pH	7.4	6.8	7.4	6.8
Mineral constituents in parts per million				
Calcium (Ca)	13	2.4	13	2.4
Magnesium (Mg)	3.5	2.7	2.8	2.7
Sodium (Na)	4.5	2.3	4.5	2.3
Potassium (K)	1.3	1.1	1.3	1.1
Carbonate (CO ₃)	10.1	1.1	10.1	1.1
Bicarbonate (HCO ₃)	40	10	40	10
Sulfate (SO ₄)	5.8	1.1	5.8	1.1
Chloride (Cl)	12	2.3	12	2.3
Nitrate (NO ₃)	0.5	0.5	0.5	0.5
Fluoride (F)	0.2	0.2	0.2	0.2
Boron (B)	0.1	0.1	0.1	0.1
Silica (SiO ₂)	18	8.5	18	8.5
Total dissolved solids in parts per million	69	40	69	40
Percent sodium	22	14	22	14
Hardness as CaCO ₃ in parts per million				
Total	44	19	44	19
Noncarbonate	11	0.0	11	0.0
Turbidity	See 1950	See 1959	1	1
Coliform in most probable number per milliliter	See 1959	See 1959	2,400	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.90	0.05
Solid alpha			0.77	0.27
Dissolved beta			3.05	0.70
Solid beta			3.66	0.30

WATER QUALITY VARIATIONS



AMERICAN RIVER AT NIMBUS DAM (STA. 22a)

AMERICAN RIVER AT SACRAMENTO (STA. 22)

Sampling Point Station 22 on the American River is located in Section 3, Township 8 North, Range 5 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-stream from the "H" Street bridge in Sacramento.

Period of Record April 1951 through December 1959.

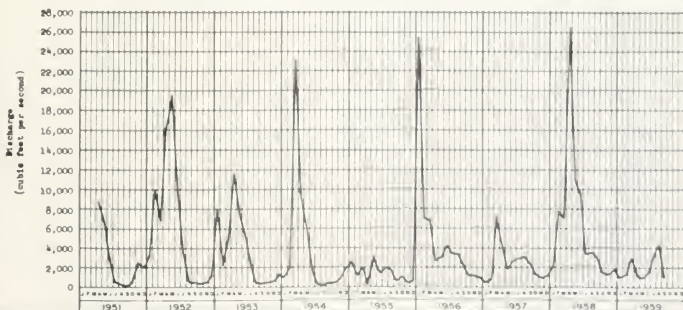
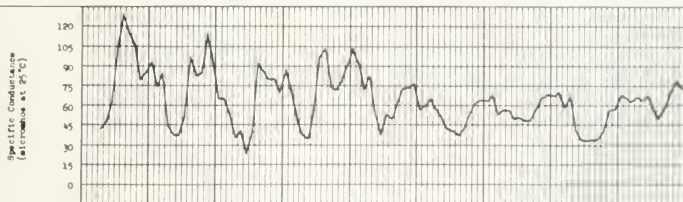
Water Quality Characteristics Antecedent data reveal the water to be generally calcium bicarbonate in character. The mineral quality is excellent, soft, and very similar to that at Station 22a (American River at Nimbus Dam). The criteria for class 1 irrigation water and mineral constituents for domestic use are consistently met by water at this station. Since 1956, when regulation of flow by Folsom Dam was commenced, the quality of water has been noticeably improved and more perennially uniform. The maximum values for concentration of individual constituents during the period of record all occurred prior to 1956.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1969	Minimum - 1969
Specific conductance (micromhos at 25°C,	129	24	77.1	48.1
Temperature in °F	81	41	72	31
Dissolved oxygen in parts per million	14.2	7.1	11.7	7.7
Percent saturation	124	56	106	46
pH	8.1	6.7	7.1	6.3
Mineral constituents in parts per million				
Calcium (Ca)	12	3.4	7.2	6.4
Magnesium (Mg)	5.5	1.7	1.9	1.7
Sodium (Na)	5.1	1.1	1.4	2
Potassium (K)	1.3	0.1	1.1	0.7
Carbonate (CO ₃)	0.3	0.1	0.3	0.1
Bicarbonate (HCO ₃)	54	16	31	20
Sulfate (SO ₄)	4.7	0.3	2.9	1.4
Chloride (Cl)	10	0.7	5.8	1.4
Nitrate (NO ₃)	0.9	0.2	0.6	0.2
Fluoride (F)	0.2	0.1	0.2	0.1
Boron (B)	6.17	0.7	3.1	0.7
Silica (SiO ₂)	15	6.9	11	6.9
Total dissolved solids in parts per million	91	17	55	35
Percent sodium	33	18	24	18
Hardness as CaCO ₃ in parts per million				
Total	50	18	32	21
Noncarbonate	7	0.7	5	0.7
Turbidity	55	0.3	20	0.3
Coliform in most probable number per milliliter	>7,000	0.6	7,000	0.69
Radioactivity in micro-micro curies per liter				
Dissolved alpha	5.5	0.00	7.00	0.00
Solid alpha	0.69	0.00	0.69	0.00
Dissolved beta	15.99	7.00	8.56	0.20
Solid beta	7.89	0.00	3.41	0.23

WATER QUALITY VARIATIONS



AMERICAN RIVER AT SACRAMENTO (STA. 22)

AMERICAN RIVER, MIDDLE FORK NEAR AUBURN (STA. 22b)

Sampling Point Station 22b is located in Section 6, Township 12 North, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, adjacent to the USGS stream gaging station, 1.9 miles upstream from the confluence of the American River with its North Fork, 3.5 miles northeast of Auburn.

Period of Record July 1958 through December 1959.

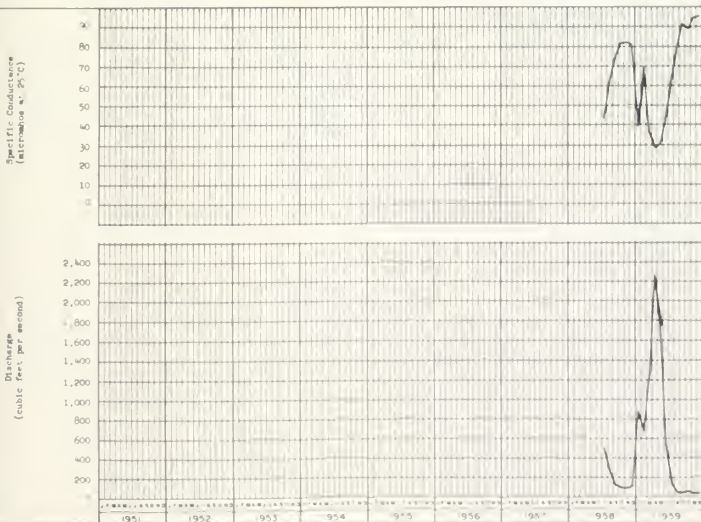
Water Quality Characteristics Past analyses show the water to be excellent in quality, calcium bicarbonate, with extremely low concentrations of dissolved solids. During the period of record total dissolved solids have not exceeded 72 ppm. The water consistently meets the requirements for a class 1 irrigation supply as well as mineral standards for domestic use. The water is soft with a maximum recorded hardness of 37 ppm.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	94.7	48.8	94.7	48.8
Temperature in °F	86	40	77	40
Dissolved oxygen in parts per million	11.8	8	11.2	8
Percent saturation	118	86	110	86
pH	7.5	7	7.5	7.0
Mineral constituents in parts per million				
Calcium (Ca)	2.8	1.8	9.8	1.8
Magnesium (Mg)	1.2	1	2.1	1
Sodium (Na)	1.9	1.5	1.9	1.5
Potassium (K)	0.9	0.7	1.8	0.7
Carbonate (CO ₃)	0.2	0.2	0.2	0.2
Bicarbonate (HCO ₃)	17	14	17	14
Sulfate (SO ₄)	1.5	1.2	5.9	1.5
Chloride (Cl)	1.2	1.2	6.2	1.2
Nitrate (NO ₃)	0.4	0.4	0.1	0.1
Fluoride (F)	0.2	0.2	0.1	0.1
Boron (B)	0.2	0.2	0.1	0.1
Silica (SiO ₂)	1.4	1.4	1.4	1.4
Total dissolved solids in parts per million	79	40	72	25
Percent sodium	25	15	25	16
Hardness as CaCO ₃ in parts per million				
Total	17	10	17	10
Noncarbonate	8	7	7	7
Turbidity	2	1.2	2	1.2
Coliforms in most probable number per milliliter	See 1959	See 1959	230	26
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.61	0.30	0.61	0.30
Solid alpha	0.44	0.17	0.44	0.17
Dissolved beta	2.46	0.00	2.46	1.84
Solid beta	1.24	0	0.20	0.00

WATER QUALITY VARIATIONS



AMERICAN RIVER, SOUTH FORK NEAR LOTUS (STA. 22c)

Sampling Point The station on South Fork American River is located in Section 11, Township 11 North, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, opposite the USGS gaging station located 0.4 mile downstream from the confluence of Greenwood Creek, and 2.4 miles northwest of Lotus.

Period of Record July 1958 through December 1959.

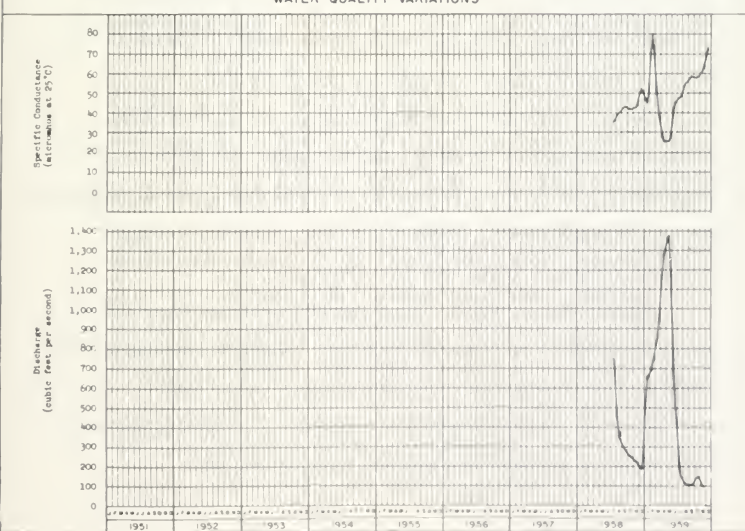
Water Quality Characteristics Analyses of the water show the predominant anion to be bicarbonate and the principal cation calcium. Calcium is not predominant, however, since sodium and magnesium together often comprise over 50 percent of the cations. Water at the station is of excellent mineral quality, class 1 for irrigation, suitable for domestic use, and soft with a maximum recorded hardness of 32 ppm.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1959	Minimum - 1959
Specific conductance (microhm at 75°C)	5	1.6	4	2.8
Temperature in °F	78	4	76	4
Dissolved oxygen in parts per million	11.7	8	12.7	2
Percent saturation	10	87	100	81
pH	7.5	7	7.5	7
Mineral constituents in parts per million				
Calcium (Ca)	6.4	1.2	6.4	1.4
Magnesium (Mg)	1.0	0.2	1.0	0.4
Sodium (Na)	4.1	1.6	4.1	1.6
Potassium (K)	1.0	0.1	0.9	0.1
Carbonate (CO ₃)	0	0.0	0	0
Bicarbonate (HCO ₃)	14	4	14	4
Sulfate (SO ₄)	4.8	0.6	4.8	0.6
Chloride (Cl)	7.2	0.6	7.2	0.6
Nitrate (NO ₃)	1.2	0.3	1.2	0.3
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.5	0.0	0.5	0.0
Silica (SiO ₂)	16	8	16	8
Total dissolved solids in parts per million	58	24	58	24
Percent sodium	29	16	29	16
Hardness as CaCO ₃ in parts per million				
Total	30	10	30	10
Noncarbonate	4	0	4	0
Turbidity	4	1	4	1
Coliform in most probable number per milliliter	See 1959	See 1959	7,000	1,000
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.20	0.00	0.20	0.00
Solid alpha	0.27	0.17	0.27	0.20
Dissolved beta	3.08	1.56	2.42	1.56
Solid beta	5.62	0.00	2.71	0.00

WATER QUALITY VARIATIONS



AMERICAN RIVER, SOUTH FORK NEAR LOTUS (STA. 22c)

San Joaquin River Valley (5b)

The San Joaquin River Valley occupies the south-central portion of the Central Valley Region. Boundaries of the basin are defined by the ridge of the Sierra Nevada to the east, the divide between Tulare Lake Basin and the San Joaquin River to the south, the crest of the Coast Range on the west and the San Joaquin Delta and Mokelumne River Basin to the north. Average east-west width of the basin is 130 miles. About 11,792 square miles, of which 7,993 square miles are mountainous and foothills, are included in the drainage basin.

Between the alluvial fans and foothills of the Coast and Sierra Mountains lies the broad, level San Joaquin Valley. The 95 percent of the land in the San Joaquin Valley is classified as valley and mesa area. The main valley floor contains about 3,670 square miles of fertile agricultural lands with elevations varying from almost sea level in the lower end to about 300 feet at the base of the foothills.

There are no major streams draining the relatively barren foothills and mountains to the west. Major streams, all of which arise in the Sierra Nevada to the east include the Stanislaus, Fresno, Chowchilla, Merced, and Tuolumne Rivers. These rivers drain rugged, mountainous terrain with elevations often exceeding 10,000 feet.

Natural mean seasonal surface runoff in the San Joaquin River Valley is estimated to be 6,385,000 acre-feet. Eighteen sampling stations are being monitored to provide a continuing check on the quality of surface water resources in the San Joaquin River Valley. Monitored stream basins with the number of stations in parentheses are as follows:

San Joaquin River Unit (8)
Fresno River Basin (1)
Chowchilla River and Bear Creek Unit (2)
Merced River Basin (2)
Tuolumne River Basin (3)
Stanislaus River Basin (2)

San Joaquin River Unit. The San Joaquin River Unit includes the following three watersheds: (1) all land below the Sierra Nevada foothill line to the east, (2) drainage basins of minor tributary streams and the valley floor to the west, (3) the headwaters of the San Joaquin River. Valley and mesa lands in the unit include 3,855 square miles, with over 95 percent of these on the floor of the San Joaquin Valley. Mountains and foothills comprise 10,679 square miles, with over 60 percent of these lands located in the Coast Range. Mean seasonal runoff from the unit is 3,264,500 acre-feet.

The topography of the unit is highly variable. The west side terrain is hilly, generally rolling, with elevations varying from about 500 feet to slightly over 3,000 feet above sea level. The valley floor is comparatively level, marred only by stream channels or draws. In contrast, the San Joaquin River Basin in the Sierra Nevada is extremely rugged and drains areas with elevations in excess of 10,000 feet.

Economic activities in the unit are dominated by agriculture, but mining, natural gas, lumber production, livestock raising, and light industry all have a role. Agriculture, however, is by far the largest user of water resources.

Wastes of significant magnitude are discharged from several communities and industries in the unit. Controls have been established to prevent these wastes from becoming major impairment problems. However, irrigation return flows pose a threat to water quality in the San Joaquin River. Major wastes discharging into this river unit are listed in the following tabulation:

City of Fresno	30.0 mgd
City of Turlock	3.3 mgd
City of Oakdale	1.0 mgd
City of Modesto	6.7 mgd
Lee Paper Company	1.5 mgd
Castle Air Force Base	0.6 mgd
City of Chowchilla	0.5 mgd
City of Los Banos	1.5 mgd
City of Atwater	2.5 mgd
City of Merced	10.0 mgd

Seven surface water quality monitoring stations are maintained in this unit. The following tabulation presents the names of the monitoring stations and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
San Joaquin River at Friant	264
San Joaquin River near Mendota	266
San Joaquin River at Fremont Ford Bridge	268
San Joaquin River at Hills Ferry Bridge	270
San Joaquin River near Grayson	272
San Joaquin River at Maze Road Bridge	274
San Joaquin River near Vernalis	276
Salt Slough at San Luis Ranch	278

SAN JOAQUIN RIVER AT FRIANT (STA. 24)

Sampling Point Station 24 is the most upstream monitoring station on the San Joaquin River. It is located in Section 7, Township 11 South, Range 21 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank (100 feet downstream from the USGS gage house), about 2 miles downstream from Friant Dam, 0.5 mile west of Friant.

Period of Record April 1951 through December 1959.

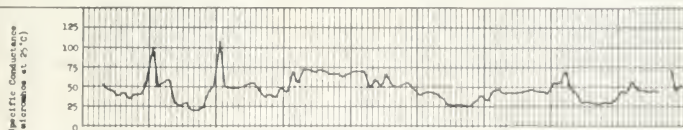
Water Quality Characteristics Water from the San Joaquin River at the Friant station generally exhibits a calcium-sodium bicarbonate characteristic. The mineral quality is excellent with a maximum recorded total dissolved solid concentration of 164 ppm. The water is soft, consistently meets the mineral criteria for drinking water and is class 1 for irrigation.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (microhms at 25°C)	108	81	79.9	43.4
Temperature in °F	64	35	60	46
Dissolved oxygen in parts per million	11.7	9.0	11.4	9.7
Percent saturation	124	83	124	85
pH	8.8	6.2	7.7	6.5
Mineral constituents in parts per million				
Calcium (Ca)	8.4	1.9	4.8	
Magnesium (Mg)	2.3	0.0	1.0	
Sodium (Na)	8.7	0.7	6.6	2.9
Potassium (K)	4.1	0.5	0.6	
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	44	8	32	15
Sulfate (SO ₄)	5.8	0.0	3.8	
Chloride (Cl)	8.0	0.0	7.5	2.5
Nitrate (NO ₃)	2.8	0.0	0.0	
Fluoride (F)	0.3	0.0	0.0	
Boron (B)	0.34	0.0	0.1	0.0
Silica (SiO ₂)	15	7.4	10	
Total dissolved solids in parts per million	87	17	58	35
Percent sodium	52	25	41	20
Hardness as CaCO ₃ in parts per million				
Total	30	6	24	12
Noncarbonate	9	0.0	4	0.0
Turbidity	70	0.0	10	2
Coliform in most probable number per milliliter	>7,000	0.045	620	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.89	0.00	0.89	0.51
Solid alpha	1.67	0.00	0.27	0.22
Dissolved beta	15.01	0.00	17.71	6.85
Solid beta	14.2	0.00	2.77	0.81

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT FRIANT STA. 241

SAN JOAQUIN RIVER NEAR MENDOTA (STA. 25)

Sampling Point Mendota station is located in Section 7, Township 13 South, Range 15 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the foot of the USGS gage house, 2.5 miles downstream from Mendota Dam and 4 miles north of Mendota.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Inasmuch as most of the water impounded by Mendota Dam is imported from the Sacramento-San Joaquin Delta via the Delta-Mendota Canal, water at the station is not necessarily representative of the natural quality of the San Joaquin River. Water from this station exhibits no consistent predominance of any specific cation or anion. Principal cations are sodium and calcium, while bicarbonate and chloride are the major anions. Based on mineral concentrations, the water is good to excellent, is suitable for domestic use, and ranges from soft to very hard.

Significant Water Quality Changes During August 1959, for the first time since sampling was commenced, the water at this station was class 2 for irrigation. Electrical conductivity, percent sodium, and concentrations of sodium and chloride also established new record maximums. Delta-Mendota Canal water near Mendota was a class 2 irrigation supply during August 1959, which probably accounted for the poor quality at this station at this time.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	1,30	1	1,351	291
Temperature in °F	86	42	78	47
Dissolved oxygen in parts per million	16.6	7.2	12.1	1.6
Percent saturation	149	84	114	88
pH	8.7	6.8	7.7	6.8
Mineral constituents in parts per million				
Calcium (Ca)	47	2.9	42	20
Magnesium (Mg)	27	0.9	27	14
Sodium (Na)	131	2.8	131	10
Potassium (K)	4.4	0.6	3.6	1.4
Carbonate (CO ₃)	2.4	0.0	2.4	0.0
Bicarbonate (HCO ₃)	158	14	150	40
Sulfate (SO ₄)	65	1.0	45	24
Chloride (Cl)	215	0.4	205	22
Nitrate (NO ₃)	1.9	0.0	1.6	0.0
Fluoride (F)	3.4	0.0	3.2	0.1
Boron (B)	4.4	0.0	4.4	0.1
Silica (SiO ₂)	31	5.9	17	12
Total dissolved solids in parts per million	753	18	713	116
Percent sodium	65	32	65	42
Hardness as CaCO ₃ in parts per million				
Total	214	8	214	52
Noncarbonate	101	0.0	101	1
Turbidity	170	0.0	170	2
Coliform in most probable number per milliliter	>7,000	0.23	2,400	0.23
Radioactivity in micro-curies per liter				
Dissolved alpha	1.13	0.00	1.13	0.00
Solid alpha	1.67	0.00	0.21	0.00
Dissolved beta	10.41	0.00	5.18	2.32
Solid beta	14.4	0.00	0.93	0.00

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER NEAR MENDOTA (STA. 25)

SAN JOAQUIN RIVER AT FREMONT FORD BRIDGE (STA. 25c)

Sampling Point Station 25c is located in Section 24 of Township 7 South, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the channel of flow, from Fremont Ford highway bridge, 2.1 miles downstream from Salt Slough, 4.5 miles west of Stevinson, and 6.7 miles upstream from the Merced River.

Period of Record July 1955 through December 1959.

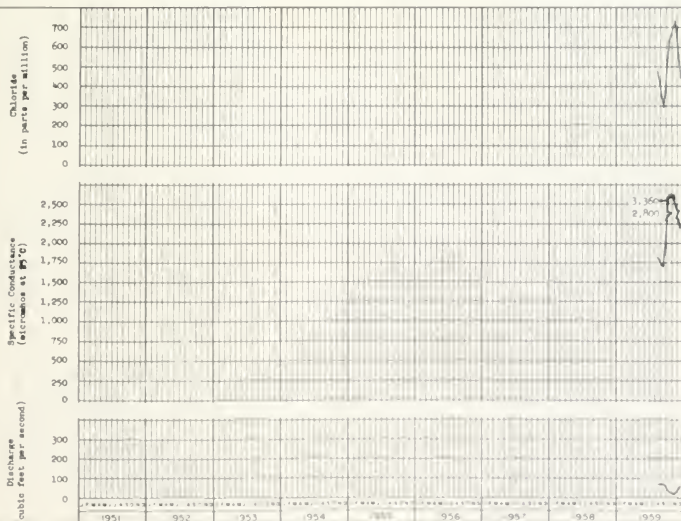
Water Quality Characteristics Water at Station 25c ranges from sodium chloride to sodium-calcium chloride-sulfate in character. Chloride and sulfate concentrations frequently exceed the recommended limit for domestic use. Chlorides, boron and dissolved solids are normally in concentrations sufficient to cause the water to be class 2 or 3 for irrigation. The poor quality water is attributable to ground water accretions and drainage flows tributary to the San Joaquin River between Station 25 near Mendota Pool and Station 25c at Fremont Ford Bridge. These degrading influences cause a significant increase (on the order of 950 micromhos) in the concentration of dissolved minerals between the two stations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	5,410	103	3,360	547
Temperature in °F	84	47	84	47
Dissolved oxygen in parts per million	12.4	8.5	12.4	8.5
Percent saturation	137	99	137	99
pH	8.5	6.6	8.3	7.0
Mineral constituents in parts per million				
Calcium (Ca)	248	8.8	124	30
Magnesium (Mg)	150	1.5	66	13
Sodium (Na)	730	5.6	409	58
Potassium (K)	8.4	1.5	8.4	0.2
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	252	37	250	127
Sulfate (SO ₄)	760	4.8	376	38
Chloride (Cl)	1,330	5.8	1,050	79
Nitrate (NO ₃)	28	0.0	5.4	1.1
Fluoride (F)	0.5	0.0	0.4	0.0
Boron (B)	1.6	0.0	1.6	0.2
Silica (SiO ₂)	37	0.2	37	0.2
Total dissolved solids in parts per million	3,350	67	1,960	303
Percent sodium	62	34	62	48
Hardness as CaCO ₃ in parts per million				
Total	1,240	28	660	136
Noncarbonate	1,080	0.0	455	128
Turbidity	40	20	40	20
Coliform in most probable number per milliliter	>7,000	0.23	>7,000	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.09		0.09	
Solid alpha	0.18		0.18	
Dissolved beta	14.21		14.21	
Solid beta	7.27		7.27	

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT FREMONT FORT BRIDGE (STA. 25c)

SAN JOAQUIN RIVER AT HILLS FERRY BRIDGE (STA. 25b)

Sampling Point Station 25b is located in Section 3 of Township 7 South, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at Hills Ferry Bridge, 300 feet downstream from the Merced River and 3.5 miles northeast of Newman.

Period of Record October 1958 through December 1959.

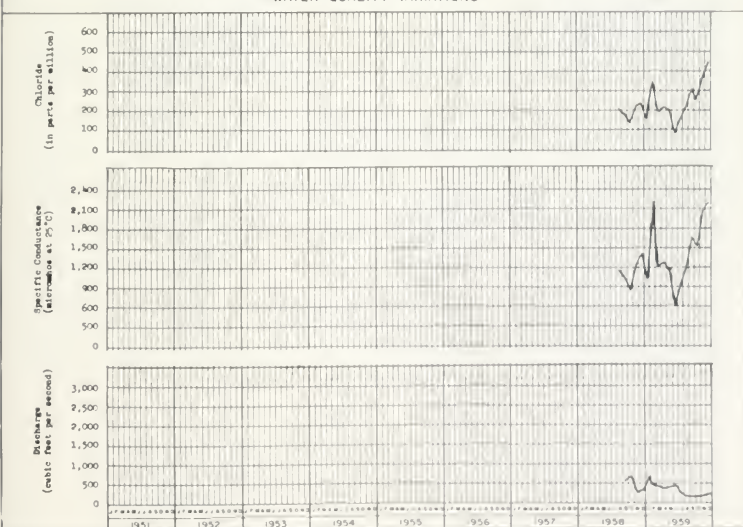
Water Quality Characteristics The water is sodium chloride-sulfate in character and very hard. Concentrations of dissolved solids, chlorides, and boron cause the water to be class 2 for irrigation. Dissolved solids and chlorides frequently exceed the recommended limits for domestic use. The quality of water at this station is not significantly different from that at the next upstream station, Fremont Ford Bridge (Station 25c).

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	2,160	61	2,160	61
Temperature in °F	89	47	78	47
Dissolved oxygen in parts per million	19.4	7.8	12.2	7.8
Percent saturation	131	77	124	77
pH	8.1	7.1	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	99	32	99	32
Magnesium (Mg)	42	14	42	14
Sodium (Na)	144	74	144	74
Potassium (K)	6.0	2.6	6.0	2.8
Carbonate (CO ₃)	4	0.0	4	0
Bicarbonate (HCO ₃)	26.9	140	26.9	142
Sulfate (SO ₄)	75.9	69	75.9	69
Chloride (Cl)	444	89	444	89
Nitrate (NO ₃)	4.9	0.8	4.9	0
Fluoride (F)	0.4	0.1	0.4	0.1
Boron (B)	1.3	0.2	1.3	0.2
Silica (SiO ₂)	33	14	32	14
Total dissolved solids in parts per million	1,320	372	1,320	372
Percent sodium	64	53	64	53
Hardness as CaCO ₃ in parts per million				
Total	421	139	421	139
Noncarbonate	224	23	224	23
Turbidity	35	6	35	6
Coliform in most probable number per milliliter	7,000	0.23	7,000	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT HILLS FERRY BRIDGE (STA. 25b)

SAN JOAQUIN RIVER NEAR GRAYSON (STA. 26)

Sampling Point The location of the Grayson monitoring station is within Section 24, Township 4 South, Range 7 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, adjacent to Laird Slough Bridge, and 2 miles northeast of Westley.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Water at Station 26 varies widely both in quality and character. The water is predominantly a sodium chloride type, however, calcium often constitutes a significant portion of the cations, and bicarbonate at times is the predominant anion. The water ranges from class 1 to class 2 for irrigation; the concentration of chloride at times exceeds the recommended limits for domestic use; total hardness ranges from soft to very hard. Conductivity ranges from 500 to over 1,000 micromhos higher at this station than at the Mendota station, about 65 miles upstream. Irrigation returns, effluent ground water, and waste discharges have caused mineral concentrations to increase along this reach of the river.

Significant Water Quality Changes In 1959 concentrations of one or more minerals exceeded class 1 irrigation limits in samples collected at the Grayson station. The extremely poor quality at the station was attributed to the low flows in San Joaquin River which afforded only minor dilution to poor quality drainage waters.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,660	91	1,660	1,030
Temperature in °F	81	45	80	40
Dissolved oxygen in parts per million	14.5	5.9	13.6	7.2
Percent saturation	171	61	145	63
pH	8.6	6.8	8.5	7.3
Mineral constituents in parts per million				
Calcium (Ca)	72	7.2	72	6.9
Magnesium (Mg)	47	2.1	47	26
Sodium (Na)	290	7.6	290	134
Potassium (K)	5.2	1.1	5.2	4.8
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	248	35	248	178
Sulfate (SO ₄)	164	5.8	164	159
Chloride (Cl)	260	6.0	260	160
Silicate (SiO ₂)	4.4	0.8	1.6	1.0
Fluoride (F)	0.4	0.0	0.2	0.1
Boron (B)	0.8	0.0	0.8	0.4
Silica (SiO ₂)	27	11	27	28
Total dissolved solids in parts per million	286	54	286	612
Percent sodium	61	36	61	58
Hardness as CaCO ₃ in parts per million				
Total	348	24	348	296
Noncarbonate	153	2	153	8
Turbidity	300	0.0	60	0.0
Coliform in most probable number per milliliter	>7,000	2.3	2,400	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	4.07	0.00	0.18	0.00
Solid alpha	1.56	0.00	0.18	0.00
Dissolved beta	15.5	0.00	6.86	4.25
Solid beta	40.0	0.00	5.49	3.58

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT MAZE ROAD BRIDGE (STA. 26a)

Sampling Point Station 26a is located in Section 29, Township 3 South, Range 7 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, 50 feet upstream from El Solyo Ranch irrigation intake and about 300 feet from Maze Road Bridge. Samples from the station were collected by personnel working for the City of San Francisco, an interested and cooperating agency in this program.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Water at Maze Road Bridge is normally sodium chloride in type, ranging from soft to very hard, and generally meeting mineral standards for domestic use. Although water at the station ranges from class 1 to class 2 for irrigation, the mineral quality is considerably better (averaging about 350 micromhos) than at Station 26 near Grayson, located about 11 miles upstream. The improvement in quality reflects the influence of tributary Tuolumne River water, which enters the San Joaquin River between these two stations.

Significant Water Quality Changes During 1959, water samples from the station contained mineral concentrations which exceeded the maximum of record. The water was class 2 for irrigation during May through October. Chloride concentrations in July and September exceeded 250 ppm, the recommended limit for drinking water. Impairment of quality at Station 26a was attributed to low flow conditions not affording enough dilution for poor quality drainage and effluent ground waters entering the river.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,933	97	1,933	607
Temperature in °F	81	46	81	40
Dissolved oxygen in parts per million	14.1	6.3	13.9	6.4
Percent saturation	175	58	171	58
pH	8.4	6.7	8.3	7.5
Mineral constituents in parts per million				
Calcium (Ca)	67	8.3	67	48
Magnesium (Mg)	28	2.7	28	18
Sodium (Na)	157	7.4	157	64
Potassium (K)	7.4	1.2	7.4	6.6
Carbonate (CO ₃)	19	0.0	19	0
Bicarbonate (HCO ₃)	196	31	196	95
Sulfate (SO ₄)	89	5.3	89	45
Chloride (Cl)	295	5	295	134
Nitrate (NO ₃)	5.1	0.0	3.6	1.1
Fluoride (F)	0.6	0.0	0.2	0.1
Boron (B)	0.73	0.0	0.4	0.1
Silica (SiO ₂)	34	9.3	34	20
Total dissolved solids in parts per million	720	54	720	338
Percent sodium	59	38	57	52
Hardness as CaCO ₃ in parts per million				
Total	273	25	273	132
Noncarbonate	118	1	111	44
Turbidity	300	0.0	40	0.0
Coliform in most probable number per milliliter	>7,000	1.3	7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	2.88	0.00	0.11	0.27
Solid alpha	1.56	0.00	0.64	0.27
Dissolved beta	12.7	0.00	5.75	0.08
Solid beta	20.6	0.00	6.56	4.69

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT MAZE ROAD BRIDGE (STA. 26a)

SAN JOAQUIN RIVER NEAR VERNALIS (STA. 27)

Sampling Point Station 27 is located in Section 13 of Township 3 South, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of flow, from Durham Ferry highway bridge, 3 miles downstream from the Stanislaus River, 3.4 miles northeast of Vernalis.

Period of Record April 1951 through December 1959.

Water Quality Characteristics The water at Station 27 is generally sodium chloride to sodium chloride-bicarbonate in character and moderately hard to very hard. Chlorides and dissolved solids occasionally exceed the recommended maximum for domestic use and frequently cause the water to be class 2 for irrigation. A decrease (about 150 micromhos) in the concentration of most mineral constituents is noted between Station 26a at Maze Road Bridge and Station 27 at Vernalis. The improvement in quality between these two stations is attributable to the dilution caused by the excellent quality tributary inflow of the Stanislaus River.

Significant Water Quality Changes Maximum values for the period of record for conductivity, calcium, sodium, potassium, bicarbonate, dissolved solids, percent sodium, and hardness were reported during the latter part of 1959. The maximums were not, however, sufficient to seriously impair the quality of the water at this point.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	1,245	99	1,245	95
Temperature in °F	Rp	45	Rp	45
Dissolved oxygen in parts per million	11.4	4	12.1	7.8
Percent saturation	166	66	124	89
pH	8.5	6.5	8.9	6.5
Mineral constituents in parts per million				
Calcium (Ca)	65	8.4	65	20
Magnesium (Mg)	43	1.9	31	9.2
Sodium (Na)	158	8.0	138	19
Potassium (K)	8.6	0.9	8.6	2.8
Carbonate (CO ₃)	7	0.0	0.0	0.0
Bicarbonate (HCO ₃)	264	32	264	81
Sulfate (SO ₄)	113	2.9	113	30
Chloride (Cl)	274	8	275	46
Nitrate (NO ₃)	5.5	0.4	5.5	1.8
Fluoride (F)	0.4	0.0	0.4	0.1
Boron (B)	0.56	0.0	0.4	0.1
Silica (SiO ₂)	41	10	46	24
Total dissolved solids in parts per million	748	52	748	270
Percent sodium	58	32	58	47
Hardness as CaCO ₃ in parts per million				
Total	246	26	246	88
Noncarbonate	132	0.0	132	20
Turbidity	85	0.0	60	0.0
Coliforms in most probable number per milliliter	24,000	0.62	7,000	0.62
Radioactivity in micro-micro curies per liter				
Dissolved alpha	2.02	0.30	0.51	0.26
Solid alpha	2.10	0.30	0.53	0.17
Dissolved beta	13.02	0.30	8.44	0.30
Solid beta	14.88	0.30	4.45	3.61

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER NEAR VERNALIS (STA. 27)

SALT SLOUGH AT SAN LUIS RANCH (STA. 92a)

Sampling Point Station 92a is located in Section 7 of Township 9 South, Range 11 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the center of the channel of flow, from a bridge about 8 miles north of Los Banos, at San Luis Ranch.

Period of Record November 1958 through December 1959.

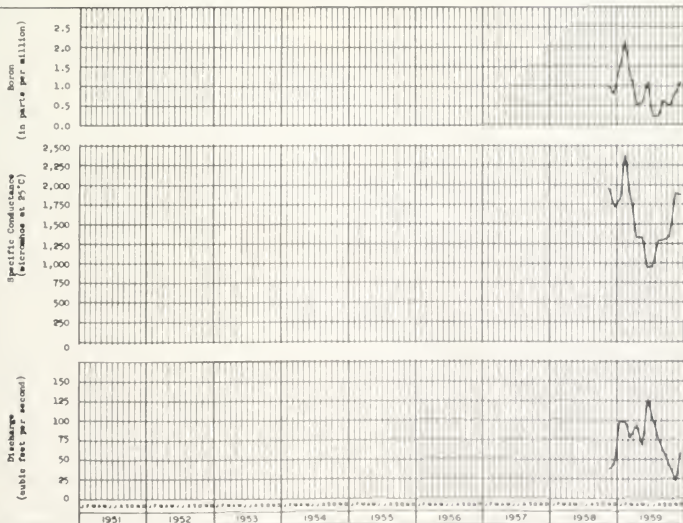
Water Quality Characteristics Water of the slough is generally sodium chloride in character, very hard, and class 2 for irrigation. Dissolved solids, chlorides, and sulfates usually exceed the recommended limits for domestic use. During winter months, storm and surface drainage and ground water accretions supply the preponderance of flow and cause high electrical conductivity on the order of 2,000 micromhos. However, during the irrigation season, return waters (averaging about 1,000 micromhos) contribute significant quantities of flow to the slough and frequently dilute the concentrations of most dissolved minerals to within the limits recommended for domestic use.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	2,370	962	2,370	962
Temperature in °F	78	48	78	48
Dissolved oxygen in parts per million	8.7	1.2	8.7	3.2
Percent saturation	88	30	88	30
pH	7.4	6.4	7.4	6.4
Mineral constituents in parts per million				
Calcium (Ca)	95	40	95	40
Magnesium (Mg)	55	23	55	23
Sodium (Na)	338	117	338	117
Potassium (K)	7.8	3.2	7.8	3.2
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	244	160	244	160
Sulfate (SO ₄)	461	93	461	93
Chloride (Cl)	425	162	425	162
Nitrate (NO ₃)	8	1.5	8.0	1.5
Fluoride (F)	0.4	0.1	0.4	0.1
Boron (B)	2.2	0.3	2.2	0.3
Silica (SiO ₂)	27	10	27	10
Total dissolved solids in parts per million	1,460	560	1,460	560
Percent sodium	62	53	62	53
Hardness as CaCO ₃ in parts per million				
Total	462	218	462	218
Noncarbonate	248	84	248	84
Turbidity	50	8	50	8
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.40	0.09	0.40	0.09
Solid alpha	0.27	0.27	0.27	0.27
Dissolved beta	4.90	3.37	4.90	3.37
Solid beta	2.14	0.78	2.14	0.78

WATER QUALITY VARIATIONS



SALT SLOUGH AT SAN LUIS RANCH (STA. 92a)

Fresno River Basin. The Fresno River Basin drains a portion of the lower western slope of the Sierra Nevada in Madera County. The Fresno River rises at an elevation of about 7,000 feet and flows westerly into the San Joaquin Valley floor. Natural runoff varies from little or no flow in later summer, to flash floods during the rainy season, averaging about 103,000 acre-feet annually.

In the river basin, above the valley floor, only 4 square miles out of 270 are classified as valley or mesa lands. Development is limited to livestock raising and recreation.

There are no significant waste discharges entering the basin. A surface water sampling station is maintained on the Fresno River near Daulton to monitor quality of runoff.

FRESNO RIVER NEAR DAULTON (STA. 113)

Sampling Point The Daulton station is located in Section 3, Township 10 South, Range 19 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-stream from Hensley Bridge, about 1.6 miles upstream from USGS gaging station, and about 5.3 miles southeast of Daulton.

Period of Record January 1958 through December 1959.

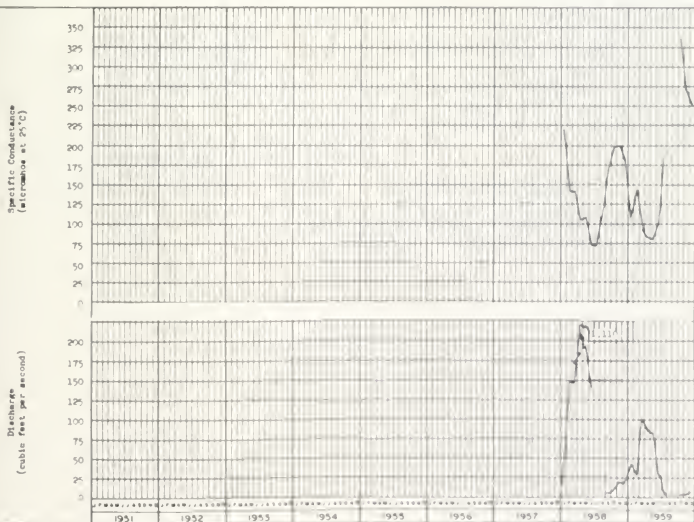
Water Quality Characteristics Past analyses show Fresno River water to be calcium-sodium bicarbonate-chloride in character, soft to slightly hard, class 1 for irrigation, and that the water meets the criteria for domestic use.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	16	72	116	78.9
Temperature in °F	78	66	78	66
Dissolved oxygen in parts per million	13	7.1	13.2	8
Percent saturation	100	71	99	91
pH	7.5	6.6	7.5	6.6
Mineral constituents in parts per million				
Calcium (Ca)	15	9.8	11	6.4
Magnesium (Mg)	1.5	0.9	2.1	1.1
Sodium (Na)	81	1.1	10	1.1
Potassium (K)	2.4	0	1.2	0.9
Carbonate (CO ₃)	1.1	0	1.2	0
Bicarbonate (HCO ₃)	6.4	26	14	16
Sulfate (SO ₄)	1.2	4.4	70	1.1
Chloride (Cl)	1.2	0	0.4	0
Nitrate (NO ₃)	0.3	0	0.2	0
Fluoride (F)	0.3	0	0.2	0
Boron (B)	0.1	0	0.3	0
Silica (SiO ₂)	0.1	1.7	26	0.1
Total dissolved solids in parts per million	231	51	231	59
Percent sodium	60	29	57	29
Hardness as CaCO ₃ in parts per million				
Total	73	1.8	73	23
Noncarbonate	21	0	21	0
Turbidity	26	0.8	7	0.7
Coliforme in most probable number per milliliter	>7,000	0.23	2,400	0.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.70	0.05	0.70	0.05
Solid alpha	0.55	0.05	0.05	0.05
Dissolved beta	0.40	0.20	0.40	0.20
Solid beta	1.88	1.16	1.88	1.16

WATER QUALITY VARIATIONS



FRESNO RIVER NEAR DAULTON STA. 113)

Chowchilla River and Bear Creek Unit. The Chowchilla River drains about 238 square miles of the lower slopes of the Sierra Nevada in Madera and Mariposa Counties. Bear Creek drains the foothills in western Merced and eastern Mariposa County. The Chowchilla River Basin has a mean annual runoff of about 91,300 acre-feet. Information is not available on the mean annual runoff from Bear Creek Basin.

Topography in these two basins, above the valley floor, is mostly rolling foothills. The Chowchilla River headwaters are at about 6,000 feet in a fairly well forested terrain. Ground covering changes with elevation from forest to range grass and a scattering of scrub trees and brush in the foothills. Livestock raising is the only significant use made of the foothill areas in the unit. Only minor quantities of waste enter these streams and there is no discernible water quality impairment problems in the unit.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this unit and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Chowchilla River at Buchanan Dam Site	286
Bear Creek at Merced	288

CHOWCHILLA RIVER AT BUCHANAN DAM SITE (STA. 114)

Sampling Point Station 114 is located in Section 22, Township 8 South, Range 18 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at a USGS gage located about 4.3 miles west of Raymond.

Period of Record January 1958 through December 1959.

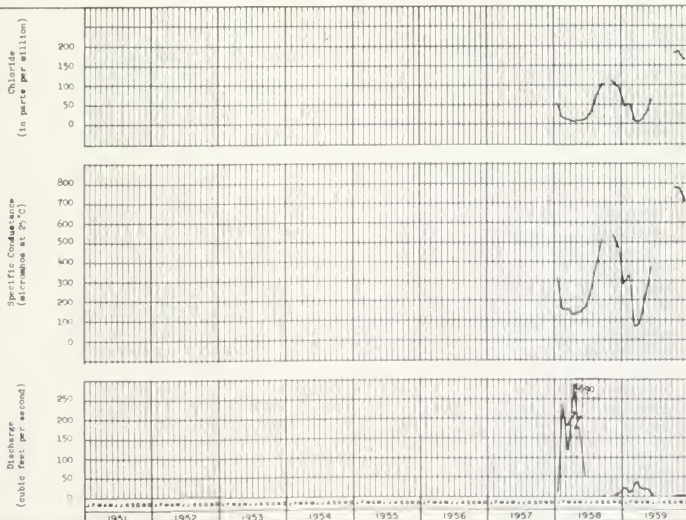
Water Quality Characteristics Water in the stream is a mixed calcium-sodium bicarbonate-chloride type from January through June. The stream is usually dry through the summer. When flow starts again in October, the water is a mixed sodium-calcium chloride type. The water is normally class 1 for irrigation and occasionally class 2 due to high chloride concentrations. Water in Chowchilla River ranges from soft to very hard, limiting it for domestic and industrial uses.

Significant Water Quality Changes During November 1959, chlorides reached 190 ppm, the maximum value of record. Runoff from the basin was extremely small during the late months of 1959. The lack of dilution water afforded by the resultant low flow probably accounted for the high concentration of chlorides.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhos at 25°C)	781	68.2	781	68.2
Temperature in °F	82	49	76	51
Dissolved oxygen in parts per million	11.6	7.7	11.2	8.9
Percent saturation	118	86	105	94
pH	8.4	7.1	7.9	7.1
Mineral constituents in parts per million				
Calcium (Ca)	26	6.4	26	6.4
Magnesium (Mg)	11	1.5	5.5	1.5
Sodium (Na)	81	5.2	81	5.2
Potassium (K)	3.9	0.8	2.4	0.8
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	118	27	118	27
Sulfate (SO ₄)	12	1.0	5.8	1.0
Chloride (Cl)	190	6.0	190	6.5
Nitrate (NO ₃)	0.9	0.0	0.4	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Borax (B)	0.21	0.0	0.2	0.0
Silica (SiO ₂)	41	12	26	12
Total dissolved solids in parts per million	481	50	481	50
Percent sodium	47	30	47	32
Hardness as CaCO ₃ in parts per million				
Total	202	22	202	22
Noncarbonate	104	0.0	104	0.0
Turbidity	90	0.0	3	0.0
Coliform in most probable number per milliliter	>7,000	0.23	620	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.40	0.17	0.40	
Solid alpha	0.64	0.17	0.64	
Dissolved beta	5.19	3.05	2.90	
Solid beta	11.55	1.00	11.55	

WATER QUALITY VARIATIONS



CHOWCHILLA RIVER AT BUCHANAN DAM SITE (STA. 114)

BEAR CREEK AT MERCED (STA. 111a)

Sampling Point The station is located in Section 24, Township 7 South, Range 13 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from State Highway 99 bridge at the north end of Merced.

Period of Record October 1958 through December 1959.

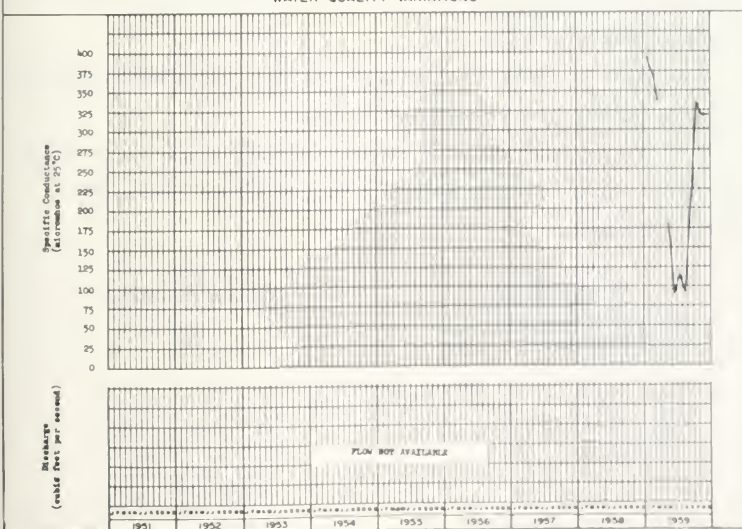
Water Quality Characteristics The water is a mixed calcium-magnesium bicarbonate type, low in concentrations of dissolved solids, and of excellent mineral quality for most beneficial uses. However, hardness ranges from soft to moderately hard, limiting it for domestic and industrial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	See 1959	See 1959	20	10
Temperature in °F			50	41
Dissolved oxygen in parts per million			12	1
Percent saturation			99	80
pH			8.3	7.2
Mineral constituents in parts per million				
Calcium (Ca)			14	8.8
Magnesium (Mg)			20	2.7
Sodium (Na)			18	4.2
Potassium (K)			2.6	1.2
Carbonate (CO ₃)			1	1
Bicarbonate (HCO ₃)			95	40
Sulfate (SO ₄)			20	4.8
Chloride (Cl)			18	2.6
Nitrate (NO ₃)			2.3	1.0
Fluoride (F)			0.2	0.0
Boron (B)			1	0.2
Silica (SiO ₂)			1	16
Total dissolved solids in parts per million			217	62
Percent sodium			23	18
Hardness as CaCO ₃ in parts per million				
Total			165	47
Noncarbonate			11	1.5
Turbidity			7	3
Coliform in most probable number per milliliter (Not Measured)				
Radioactivity in micro-micro curies per liter				
Dissolved alpha			0.24	0.09
Solid alpha			0.37	0.38
Dissolved beta			12.36	3.73
Solid beta			6.49	0.06

WATER QUALITY VARIATIONS



BEAR CREEK AT MERCED (STA. 111a)

Merced River Basin. Merced River drains a watershed area of about 1,035 square miles in Mariposa and Merced Counties. The river originates at an elevation of 11,000 feet in the Sierra Nevada and drops to about 400 feet as it flows out of the foothills into San Joaquin Valley. From its headwater the river flows almost due westward 135 miles to its mouth on the San Joaquin River. Mean annual runoff from the basin is estimated to be 1,027,000 acre-feet. Snowmelt sustains flow in the Merced River throughout most of the summer.

Terrain in the basin is very rugged at the headwaters, and steep canyon walls comprise a large portion of the watershed along the river. Yosemite Valley is the major attraction in this basin and development is primarily based on the tourist trade attracted by the scenic and geologic wonders of the valley. Lumbering, mining, livestock raising, and recreational services comprise the balance of significant economic pursuits in the basin.

Waste discharges entering the Merced River system are insignificant in volume and do not create impairment problems.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Merced River below Exchequer Dam	292
Merced River near Stevinson	294

MERCED RIVER BELOW EXCHEQUER DAM (STA. 32a)

Sampling Point Exchequer Dam station is located in Section 14, Township 4 South, Range 15 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS gage house, 0.5 mile downstream from Exchequer Dam and 5 miles northeast of Merced Falls.

Period of Record April 1951 through December 1959.

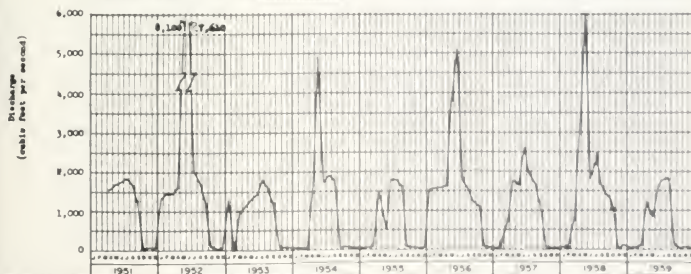
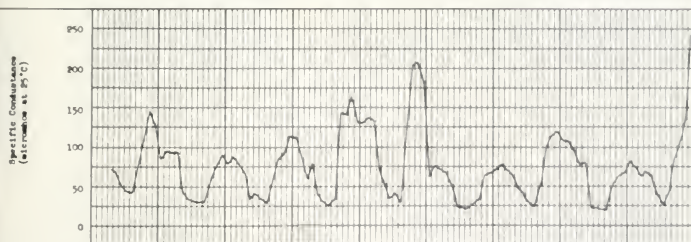
Water Quality Characteristics The water is calcium bicarbonate in character, soft to slightly hard, and of excellent mineral quality for nearly all beneficial uses. The quality of water at Station 32a reflects the amount of runoff in the stream; in a low water year, conductivity ranges from 25 to 250 micromhos, and in a wet water year, conductivity ranges from 20 to 110 micromhos.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1947	Minimum - 1949
Specific conductance (micromhos at 25°C)	242	50	180	26.0
Temperature in °F	77	44	—	44
Dissolved oxygen in parts per million	11.1	1.8	10.0	7.0
Percent saturation	129	47.5	104	77
pH	8.1	7	8.1	6.8
Mineral constituents in parts per million				
Calcium (Ca)	29	2.4	11	7.4
Magnesium (Mg)	5.5	1	4.5	1.5
Sodium (Na)	21	0.9	21	0.4
Potassium (K)	1.9	0.2	1.7	0.4
Carbonate (CO ₃)	3	0	3	0
Bicarbonate (HCO ₃)	127	9	127	41
Sulfate (SO ₄)	8.1	0.4	4.3	2.9
Chloride (Cl)	7.0	0.2	6.2	0.2
Nitrate (NO ₃)	2.9	0	2.9	0
Fluoride (F)	0.2	0	0.2	0
Boron (B)	0.73	0	0.73	0
Silica (SiO ₂)	16	3.7	14	0.1
Total dissolved solids in parts per million	158	13	158	17
Percent sodium	37	10	37	12
Hardness as CaCO ₃ in parts per million				
Total	91	8	79	9
Noncarbonate	12	0	12	0
Turbidity	400	0.0	35	0.0
Coliforms in most probable number per milliliter	>7,000	0.0045	7,000	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.54	0.00	0.54	0.00
Solid alpha	1.61	0.00	1.61	0.01
Dissolved beta	13.3	0.00	13.3	0.00
Solid beta	8.1	0.00	8.1	0.00

WATER QUALITY VARIATIONS



MERCED RIVER BELOW EXCHEQUER DAM (STA. 32a)

MERCED RIVER NEAR STEVINSON (STA. 32)

Sampling Point Station 32 is located in Section 36, Township 6 South, Range 9 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank (100 feet upstream from a USGS gage), about 6 miles northwest of Stevinson.

Period of Record April 1951 through December 1959.

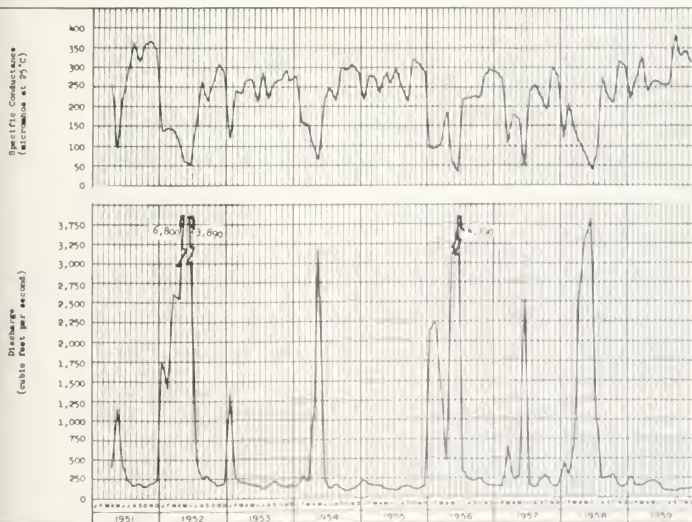
Water Quality Characteristics Water in the river near Stevinson is a mixed calcium-magnesium bicarbonate type, slightly hard and of excellent mineral quality for nearly all beneficial uses. Comparison of mineral quality of water at Station 32 with that at Merced River below Exchequer Dam (Station 32a) reveal a proportionately large increase (about 200 micromhos) in dissolved solids. However, quality of water is still excellent at both stations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	985	33.6	985	222
Temperature in °F	83	41	81	53
Dissolved oxygen in parts per million	12.0	7.0	11.1	8.0
Percent saturation	126	78	104	84
pH	8.3	6.6	7.7	6.6
Mineral constituents in parts per million				
Calcium (Ca)	28	4.8	27	19
Magnesium (Mg)	13	0.3	8.4	8.1
Sodium (Na)	45	2.0	43	19
Potassium (K)	3.1	0.5	2.8	2.3
Carbonate (CO ₃)	14	0	14	0
Bicarbonate (HCO ₃)	162	18	162	40
Sulfate (SO ₄)	22	9.3	17	11
Chloride (Cl)	52	0.0	38	9.0
Nitrate (NO ₃)	2.8	0.0	2.8	2.7
Fluoride (F)	0.4	0.0	0.1	0.0
Boron (B)	0.31	0.0	0.2	0.0
Silica (SiO ₂)	49	7.5	35	28
Total dissolved solids in parts per million	248	21.6	248	143
Percent sodium	55	14	53	37
Hardness as CaCO ₃ in parts per million				
Total	115	13	108	66
Noncarbonate	18	0.0	0	0
Turbidity	60	0.0	35	0
Coliforms in most probable number per milliliter	7,000	0.46	7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.55	0.00	0.41	0.27
Solid alpha	0.76	0.00	0.53	0.26
Dissolved beta	13.50	0.00	3.70	2.13
Solid beta	19.74	0.00	3.30	0.00

WATER QUALITY VARIATIONS



Tuolumne River Basin. Tuolumne River Basin contains approximately 1,540 square miles on the western slopes of the Sierra Nevada in the east-central portion of the Central Valley Region. Headwaters derive from glacial lakes high in the mountains, from where the stream flows south-westerly for 150 miles to its junction with the San Joaquin River. Average annual discharge of Tuolumne River Basin is estimated to be 1,900,000 acre-feet.

Above the San Joaquin Valley floor the terrain of the basin is classified as mountainous-foothill area. The upper portion drains a few meadows and plateaus, but the river soon drops into a steep canyon and flows through a gorge for a distance of about 80 miles. Elevation varies from 300 feet at the foothill line to over 13,000 feet at the crest of the Sierra Nevada.

Economic developments in the basin are typical of mountainous areas and include resort areas, lumbering, mining, livestock raising, and recreational facilities.

Numerous wastes from individual domestic, lumbermill, and resort developments, discharge into the Tuolumne watershed. These waste discharges are minor in quantity and have not caused significant impairment problems.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Tuolumne River below Don Pedro Dam	298
Tuolumne River at Hickman-Waterford Bridge	300
Tuolumne River at Tuolumne City	302

TUOLUMNE RIVER BELOW DON PEDRO DAM (STA. 31a)

Sampling Point Don Pedro Dam station is located in Section 3, Township 3 South, Range 14 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, about one-quarter mile downstream from the dam and approximately 5 miles upstream from La Grange.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Analyses show the water to be calcium bicarbonate in character, excellent in mineral quality, soft, and suitable for all beneficial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	154	13.4	66.1	28.3
Temperature in °F	71	44	67	48
Dissolved oxygen in parts per million	11.6	7.7	9.9	7.8
Percent saturation	133	60	86	75
pH	8.8	6.2	7.1	6.7
Mineral constituents in parts per million				
Calcium (Ca)	7.1	3.8	4.3	3.6
Magnesium (Mg)	2.7	1.1	1.5	1.4
Sodium (Na)	5.1	2.7	2.7	1.7
Potassium (K)	1.0	0.3	0.7	0.7
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	4.0	2.0	2.0	1.9
Sulfate (SO ₄)	2.9	0.0	2.9	0.0
Chloride (Cl)	10	0.0	2.5	0.5
Nitrate (NO ₃)	0.8	0.0	0.2	0.0
Nitrite (NO ₂)	0.2	0.0	0.1	0.0
Boron (B)	0.17	0.0	0.1	0.0
Silica (SiO ₂)	22	4.0	7.2	3.8
Total dissolved solids in parts per million	117	13	46	19
Percent sodium	44	13	43	17
Hardness as CaCO ₃ in parts per million				
Total	38	4	28	8
Noncarbonate	7	0.0	7	0.0
Turbidity	55	0.0	32	0.0
Coliforms in most probable number per milliliter	>7,000	0.045	230	0.21
Radioactivity in micro-micro curies per liter				
Dissolved alpha	2.25	0.00	0.51	0.00
Solid alpha	1.07	0.00	1.07	0.35
Dissolved beta	9.57	0.92	9.57	6.05
Solid beta	7.01	0.00	4.22	0.00

WATER QUALITY VARIATIONS



TUOLUMNE RIVER BELOW DON PEDRO DAM (STA. 31a)

TUOLUMNE RIVER AT HICKMAN-WATERFORD BRIDGE (STA. 30)

Sampling Point The station is located within Section 34, Township 3 South, Range 11 East, Mt. Diablo Base and Meridian. Monthly grab samples were taken from Hickman-Waterford Bridge about one-half mile south of Waterford.

Period of Record April 1951 through December 1959.

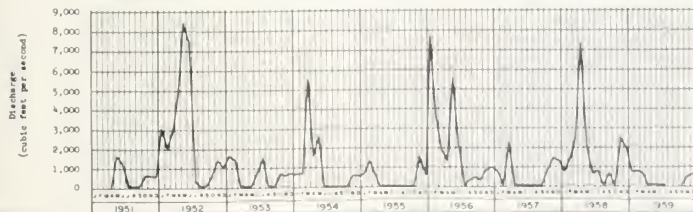
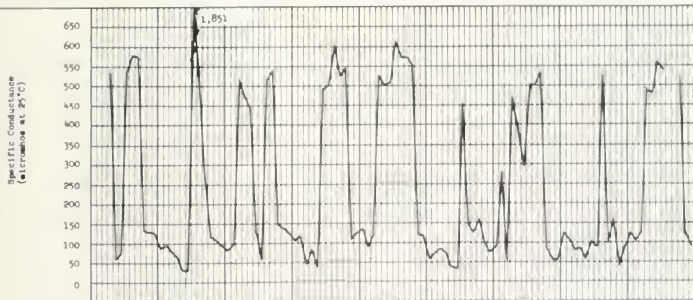
Water Quality Characteristics The water is a mixed calcium-sodium-magnesium bicarbonate type during most of the year. During the summer months when the flow in the river is low due to upstream diversion for irrigation, the river becomes sodium chloride in character. Tuolumne River is class 1 for irrigation, soft to moderately hard, and suitable for nearly all beneficial uses. A source of water quality degradation is abandoned flowing gas wells which discharge saline water into the river along the reach in the valley floor east of the station. Degradation from the saline well water is apparent during low flow periods when less dilution water is available. Station 30 is located approximately 25 miles downstream from Don Pedro Dam (Station 31a). Comparison of quality at these two stations show a considerable increase (from about 50 to 450 micromhos) in mineral concentrations at the downstream station.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhms at 25°C)	513	18	555	91.3
Temperature in °F	80	45	78	50
Dissolved oxygen in parts per million	13.5	5.7	10.6	6.8
Percent saturation	136	70	113	77
pH	8.4	6.6	8.3	7.1
Mineral constituents in parts per million				
Calcium (Ca)	55	2.8	55	8.4
Magnesium (Mg)	17	1.2	17	3.9
Sodium (Na)	117	1.6	117	7.0
Potassium (K)	10	0.5	10	5.1
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	164	12	164	26
Sulfate (SO ₄)	26	0.0	26	8.6
Chloride (Cl)	206	0.0	206	12
Nitrate (NO ₃)	8.6	0.0	8.6	0.8
Fluoride (F)	0.3	0.0	0.1	0.1
Boron (B)	0.18	0.0	0.1	0.0
Silica (SiO ₂)	55	6.8	43	38
Total dissolved solids in parts per million	355	19	384	53
Percent sodium	54	16	54	32
Hardness as CaCO ₃ in parts per million				
Total	206	11	206	25
Noncarbonate	72	0.0	72	3
Turbidity	45	0.0	30	0.0
Coliform in most probable number per milliliter	7,000	<0.045	620	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.08	0.00	0.41	0.26
Solid alpha	0.83	0.00	0.31	0.00
Dissolved beta	12.43	0.00	4.05	1.16
Solid beta	8.37	0.00	5.10	1.22

WATER QUALITY VARIATIONS



TUOLUMNE RIVER AT HICKMAN-WATERFORD BRIDGE (STA. 30)

TUOLUMNE RIVER AT TUOLUMNE CITY (STA. 31)

Sampling Point Tuolumne City station is located in Section 7, Township 4 South, Range 8 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from Shiloh Road bridge, about 8 miles west of Modesto.

Period of Record April 1951 through December 1959.

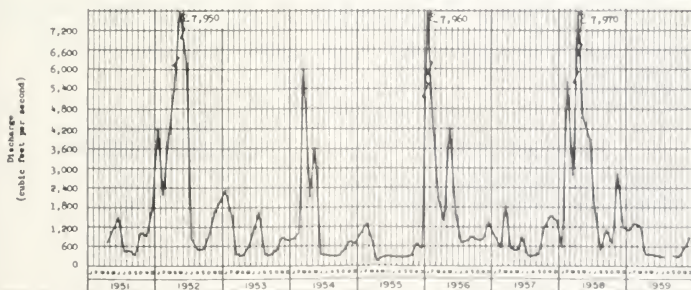
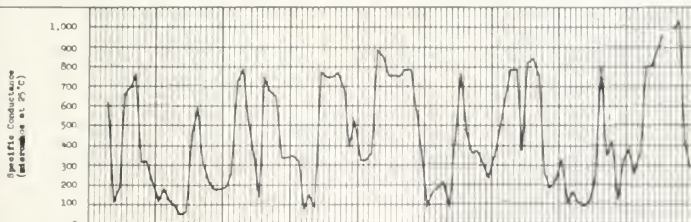
Water Quality Characteristics Antecedent data show the water is predominantly a sodium-calcium chloride type changing to sodium chloride type during the summer months. Tuolumne River water occasionally becomes class 2 for irrigation due to an excess of chloride concentration. Hardness ranges from slightly hard to very hard, limiting its beneficial uses. The City of Modesto discharges treated sewage into the river approximately 8 miles upstream from Station 31. No detectable change in quality has been noted at this station that can be attributed to the Modesto waste discharges.

Significant Water Quality Changes During 1959, the mineral concentrations observed at the station were the highest of record. During seven months, April through October 1959, conductivity exceeded 790 micromhos and reached a maximum of 1,030 micromhos in October. The poor quality was probably a result of the lack of dilution waters caused by the low runoff conditions experienced in 1959.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	1,300	74.5	1,300	242
Temperature in °F	84	45	77	50
Dissolved oxygen in parts per million	18.4	2.5	10	5.4
Percent saturation	224	29	113	61
pH	8.9	6.3	8.4	7.2
Mineral constituents in parts per million				
Calcium (Ca)	58	1.6	58	25
Magnesium (Mg)	18	1.4	18	8.6
Sodium (Na)	127	1.8	127	29
Potassium (K)	10	0.9	10	6.8
Carbonate (CO ₃)	3.0	0.0	3.0	0.0
Bicarbonate (HCO ₃)	184	20	184	48
Sulfate (SO ₄)	26	29.0	26	0.0
Chloride (Cl)	238	6.9	238	14
Nitrate (NO ₃)	4.6	0.0	4.6	1.4
Fluoride (F)	0.4	0.0	0.1	0.1
Bromine (Br)	0.15	0.0	0.2	0.0
Silica (SiO ₂)	57	6.5	46	18
Total dissolved solids in parts per million	588	14.2	588	140
Percent sodium	56	25	56	47
Hardness as CaCO ₃ in parts per million				
Total	220	14	220	58
Noncarbonate	72	0.0	72	19
Turbidity	45	0.0	17	1
Coliform in most probable number per milliliter	>7,000	0.18	7,000	62
Radioactivity in micro-curie per liter				
Dissolved alpha	1.23	0.00	1.23	0.06
Solid alpha	1.10	0.00	0.42	0.09
Dissolved beta	24.3	0.00	6.14	2.47
Solid beta	6.1	0.00	1.97	0.00

WATER QUALITY VARIATIONS



TUOLUMNE RIVER AT TUOLUMNE CITY (STA. 31)

Stanislaus River Basin. The Stanislaus River drains a narrow basin on the western slope of the Sierra Nevada in northeastern San Joaquin River Basin. Enclosed within the boundaries of the watershed are 983 square miles of mountains and foothills. The drainage basin slopes westward, from an elevation of over 10,000 feet at the crest of the Sierra Nevada, to about 20 feet at its confluence with the San Joaquin River. Mean annual runoff of the Stanislaus River is about 1,210,000 acre-feet.

Extremely rugged topography, which includes bare granite peaks and steep canyons, limit development along the upper reaches of the basin. At lower elevations, the ridges and valleys are covered with timber which have promoted lumbering operations, while the foothills provide grazing land suitable for livestock raising. Other commercial pursuits are generally associated with recreation, mining activities, or catering to the tourist trade attracted by the scenery and colorful history of this area.

Waste discharges enter the drainage basin in small volumes. There has been no serious impairment of water quality caused by these small waste discharges.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Stanislaus River below Tulloch Dam	306
Stanislaus River near mouth	308

STANISLAUS RIVER BELOW TULLOCH DAM (STA. 29a)

Sampling Point The Tulloch Dam station is located within Section 1, Township 1 South, Range 12 East, Mt. Diablo Base and Meridian. The monthly water samples were collected downstream from Tulloch Dam and approximately 6 miles northeast of Knights Ferry.

Period of Record July 1956 through December 1959.

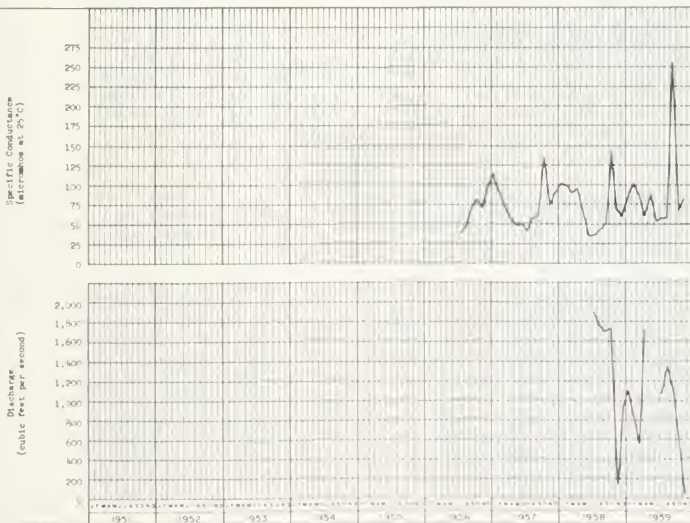
Water Quality Characteristics Past analyses show the water to be calcium-magnesium bicarbonate in character, soft, excellent in mineral quality and suitable for all beneficial uses. Quality of water at the station does not vary significantly, even with wide fluctuations in flow, indicating that regulation by upstream water resources developments apparently stabilizes mineral concentrations.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohm at 25°C)	255	15.1	255	15.1
Temperature in °F	72	68	79	68
Dissolved oxygen in parts per million	13.1	6.6	10.1	6.6
Percent saturation	104	78	94	71
pH	7.6	6.8	7.4	6.9
Mineral constituents in parts per million				
Calcium (Ca)	25	5.3	25	9.2
Magnesium (Mg)	9.4	5	9.4	3.9
Sodium (Na)	15	1.4	15	2.3
Potassium (K)	2.4	3.5	2.4	0.5
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	143	17	143	25
Sulfate (SO ₄)	7.0	0.0	7.0	6.7
Chloride (Cl)	7.0	0.0	7.0	0.5
Nitrate (NO ₃)	1.3	0.0	1.3	0.0
Fluoride (F)	0.2	0.0	0.1	0.1
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	18	10	18	16
Total dissolved solids in parts per million	175	25	175	37
Percent sodium	24	4	24	11
Hardness as CaCO ₃ in parts per million				
Total	101	14	101	29
Noncarbonate	19	0.0	10	0.0
Turbidity	80	0.0	80	0.0
Coliforms in most probable number per milliliter	>7,000	0.045	2,400	0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.41	0.00	0.41	0.00
Solid alpha	0.04	0.01	0.26	0.01
Dissolved beta	5.33	0.00	5.33	0.00
Solid beta	1.08	1.24	1.08	1.24

WATER QUALITY VARIATIONS



STANISLAUS RIVER BELOW TULLOCH DAM (STA. 29a)

STANISLAUS RIVER NEAR MOUTH (STA. 29)

Sampling Point Station 29 is located in Section 17, Township 3 South, Range 7 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the foot of a DWR gage house, about 1 mile above the junction of the Stanislaus River with the San Joaquin River and about 9 miles south of Manteca.

Period of Record April 1951 through December 1959.

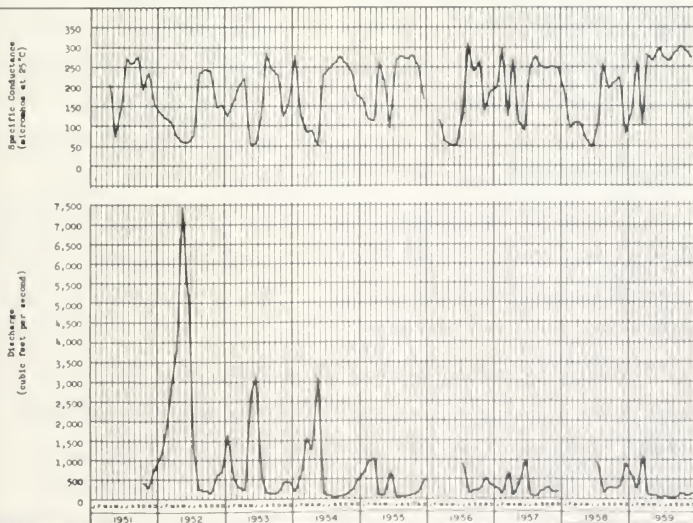
Water Quality Characteristics Antecedent data show the water to be a mixed calcium-magnesium bicarbonate type, soft to moderately hard, and excellent in mineral quality for nearly all beneficial uses. Mineral concentrations in water at the Stanislaus River mouth station are about 50 percent higher (40 to 250 micromhos) than those found at the upstream station at Tulloch Dam. This mineral pickup is attributed to tributary drainage and waste discharges.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmoh at 25°C)	310	47.3	267	69
Temperature in °F	84	41	84	51
Dissolved oxygen in parts per million	13.0	6.4	13.3	6.6
Percent saturation	134	61	134	74
pH	8.7	6.8	7.5	7.1
Mineral constituents in parts per million				
Calcium (Ca)	30	5.5	26	25
Magnesium (Mg)	12	1.2	10	9.8
Sodium (Na)	19	1.5	19	3.5
Potassium (K)	3.4	0.7	2.9	2.1
Carbonate (CO ₃)	5	0.0	5	0.0
Bicarbonate (HCO ₃)	165	24	163	49
Sulfate (SO ₄)	11	0.0	9.0	7.7
Chloride (Cl)	15	0.0	14	1.8
Nitrate (NO ₃)	2.6	0.0	1.7	1.6
Fluoride (F)	0.2	0.0	0.1	0.1
Boron (B)	0.37	0.0	0.1	0.0
Silica (SiO ₂)	37	9.3	33	32
Total dissolved solids in parts per million	210	32	207	69
Percent sodium	28	13	26	15
Hardness as CaCO ₃ in parts per million				
Total	123	21	116	42
Noncarbonate	8	0.0	3	0.0
Turbidity	100	0.0	85	0.0
Coliform in most probable number per milliliter	>7,000	0.23	7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.67	0.00	0.30	0.09
Solid alpha	1.22	0.00	0.53	0.44
Dissolved beta	10.01	0.00	2.59	2.52
Solid beta	11.32	0.00	2.07	2.66

WATER QUALITY VARIATIONS



STANISLAUS RIVER NEAR MOUTH (STA. 29)

Sacramento-San Joaquin Delta Drainage (5c)

The Sacramento-San Joaquin Delta comprises the central portion of the great Central Valley Basin. The drainage area extends north to the City of Sacramento, south to the vicinity of Vernalis, east to the crest of the Sierra Nevada Range and west to Carquinez Straits. The watersheds of the Calaveras, Cosumnes and Mokelumne Rivers are included in the area. Major streams entering the delta area include the Sacramento River from the north, the San Joaquin River from the south and the Calaveras, Cosumnes and Mokelumne Rivers from the east. The Sacramento-San Joaquin Delta Drainage comprises approximately 4,154 square miles, approximately 2,390 square miles of which are classified as mountain and foothill terrain.

A broad gentle-sloping plain, cut into islands by numerous waterways, lies between the foothills on the east and Carquinez Straits on the west. This fertile agricultural land comprises an area of 1,764 square miles and is referred to as the Sacramento-San Joaquin Delta.

Prominent uses of water in the delta include irrigation, power development, salinity control, export under operation of the Central Valley Project and East Bay Municipal Utility District, and water associated recreation. Many varied industries are located in the western end of the delta and depend upon the river for a source of process and cooling waters.

Twenty-four sampling stations are being monitored to obtain information and to provide a continuing check on quality of surface water resources in the delta. Monitored basins with the number of sampling stations in parentheses are as follows:

Sacramento-San Joaquin Delta (18)
Cosumnes River Basin (2)
Mokelumne River Basin (2)
Calaveras River Basin (2)

Sacramento-San Joaquin Delta. The central delta area comprises over 50 islands and tracts reclaimed, since 1852, from former tule swamps and overflow lands. Included in the area are about 469,000 acres lying generally below an elevation of five feet above sea level. A survey in 1955 determined that approximately 386,000 acres were agricultural and 83,000 acres nonagricultural. Acreage classified nonagricultural included approximately 42,000 acres of water surface made up of three major flooded areas and a maze of interconnected waterways.

The interest of many public agencies in the water quality of the delta prompted the planning and organization of a water quality surveillance program and the establishment of a large number of monitoring stations on the maze of rivers, canals, and sloughs making up the water channels in the delta. The escaping of poor quality water trapped in the more or less deadend portions of the delta channels when heavy drafts at the Central Valley Project pumping plants induce movement, and the effect of irrigation and drainage practices, have caused considerable concern in the past.

The quality of water in the delta area is influenced primarily by five factors: (a) the tidal motivated incursion of saline water from Suisun Bay and the Pacific Ocean into the delta, (b) flow conditions in streams tributary to the delta, (c) Central Valley Project diversions to delta upland areas, (d) irrigation diversions to and return flows from the many irrigated islands in the delta area, and (e) accretions from ground water aquifers in the delta.

The following tabulation presents the names of stations maintained to monitor quality of surface water in the delta and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Lindsey Slough near Rio Vista	314
Sacramento River at Rio Vista	316
Delta Cross Channel near Walnut Grove	318
Little Potato Slough at Terminous	320
San Joaquin River at Mossdale Bridge	322
San Joaquin River at Garwood Bridge	324
San Joaquin River at Antioch	326
Stockton Ship Channel on Rindge Island	328
Old River near Tracy	330
Old River at Clifton Court Ferry	332
Old River at Orwood Bridge	334
Old River at Mandeville Island	336
Grant Line Canal at Tracy Road Bridge	338
Delta-Mendota Canal near Tracy	340
Delta-Mendota Canal near Mendota	342
Italian Slough near Mouth	344
Indian Slough near Brentwood	346
Rock Slough near Knightsen	348

LINDSEY SLOUGH NEAR RIO VISTA (STA. 110)

Sampling Point Lindsey Slough station is located in Section 25, Township 5 North, Range 2 West, Mt. Diablo Base and Meridian. Monthly grab samples are taken from the boat landing on the right bank at California Packing Corporation's Montezuma Ranch headquarters, and about 6 miles north of Rio Vista.

Period of Record October 1952 through December 1959.

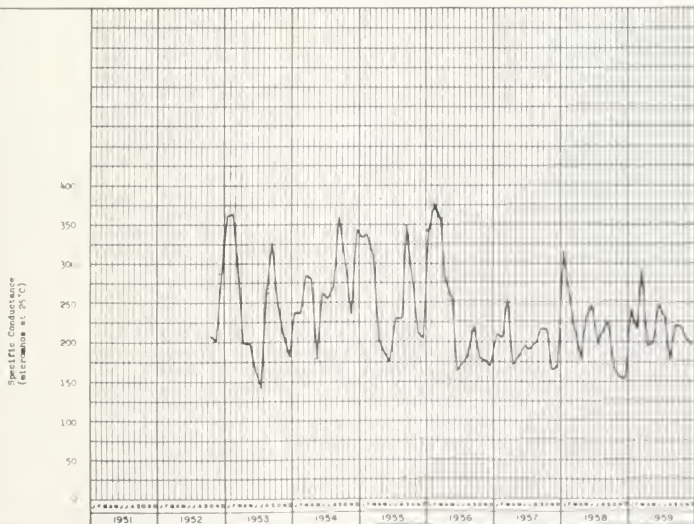
Water Quality Characteristics The water is a complex calcium-magnesium-sodium bicarbonate type, low in mineral content and well suited for domestic, industrial and class 1 irrigation uses. Because tidal oscillations and pumping maintain a nearly constant flow through the slough, the water quality at this station has remained fairly consistent throughout the period of record and generally reflects the quality of the Sacramento River at Rio Vista (Station 16).

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	177	181	292	178
Temperature in °F	80	44	78	49
Dissolved oxygen in parts per million	11.4	8.5	10.8	7.5
Percent saturation	104	69	98	84
pH	8.0	7.1	7.5	7.3
Mineral constituents in parts per million				
Calcium (Ca)	24	11	16	14
Magnesium (Mg)	19	6.5	9.2	8.3
Sodium (Na)	32	9.0	24	14
Potassium (K)	3.3	1.2	2.3	2.2
Carbonate (CO ₃)	8	8	8	8
Bicarbonate (HCO ₃)	100	16	112	78
Sulfate (SO ₄)	24	13	19	17
Chloride (Cl)	29	6	20	9.8
Nitrate (NO ₃)	1.4	0.4	1.1	0.5
Fluoride (F)	1.4	0.8	0.1	0.1
Boron (B)	3.7	0.3	0.7	0.8
Silica (SiO ₂)	24	15	20	18
Total dissolved solids in parts per million	230	87	177	108
Percent sodium	91	22	5	29
Hardness as CaCO ₃ in parts per million				
Total	139	42	100	60
Noncarbonate	12	0	8	0.7
Turbidity	700	7	100	12
Coliform in most probable number per milliliter	>7,000	2.3	2,400	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



LINDSEY SLOUGH NEAR RIO VISTA (STA. 110)

SACRAMENTO RIVER AT RIO VISTA (STA. 16)

Sampling Point Station 16 is located in Section 31, Township 4 North, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a pier on the right bank at the U. S. Department of Army installation located about 1.0 mile south of Rio Vista.

Period of Record April 1951 through December 1959.

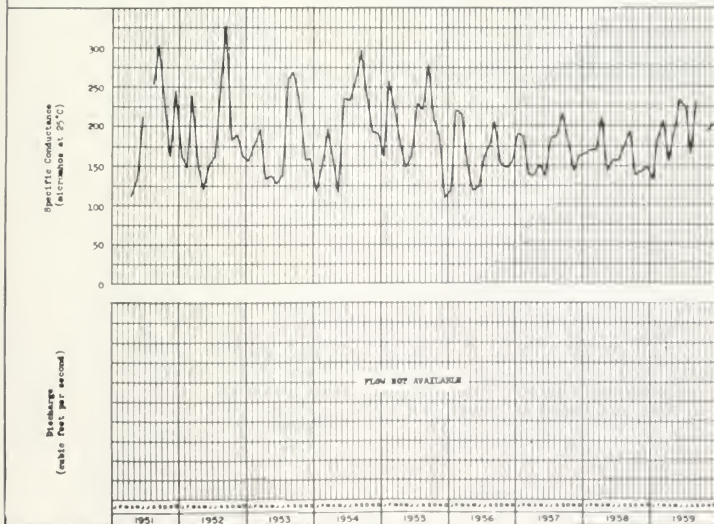
Water Quality Characteristics Antecedent data reveal the water to be a mixed calcium-magnesium-sodium bicarbonate type, soft to slightly hard, and generally within mineral requirements for domestic and class 1 irrigation use.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micro-mhos at 25°C)	126	100	933	139
Temperature in °F	78	43	73	48
Dissolved oxygen in parts per million	18.8	5.0	16.8	7.7
Percent saturation	178	68	188	86
pH	8.2	6.8	7.7	7.0
Mineral constituents in parts per million				
Calcium (Ca)	26	8.8	15	15
Magnesium (Mg)	12	4.1	11	7.2
Sodium (Na)	26	4.4	19	6.7
Potassium (K)	2.9	0.8	2.0	1.7
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	164	43	106	38
Sulfate (SO ₄)	20	3.1	15	10
Chloride (Cl)	26	3.1	16	6.3
Nitrate (NO ₃)	1.4	0.1	0.9	0.1
Fluoride (F)	0.4	0.0	0.1	0.1
Boron (B)	0.39	0.05	0.1	0.0
Silica (SiO ₂)	26	3.8	3.2	1.8
Total dissolved solids in parts per million	204	68	146	83
Percent sodium	77	10	78	23
Hardness as CaCO ₃ in parts per million				
Total	122	40	81	48
Noncarbonate	10	0.0	9	0.0
Turbidity	600	1	70	12
Coliforms in most probable number per milliliter	70,000	0.69	>7,000	0.69
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.56	0.00	0.09	0.09
Solid alpha	0.79	0.00	0.27	0.20
Dissolved beta	20.20	0.00	2.28	1.56
Solid beta	11.41	2.00	11.41	1.03

WATER QUALITY VARIATIONS



SACRAMENTO RIVER AT RIO VISTA (STA. 16)

DELTA CROSS CHANNEL NEAR WALNUT GROVE (STA. 98)

Sampling Point Station 98 is located in Section 35, Township 5 North, Range 4 East, Mt. Diablo Base and Meridian. The monthly water samples were collected on the left bank about 0.2 mile downstream from the control gates when the gates are open, or from the Walnut Grove bridge over the Sacramento River when the gates are closed.

Period of Record September 1952 through December 1959.

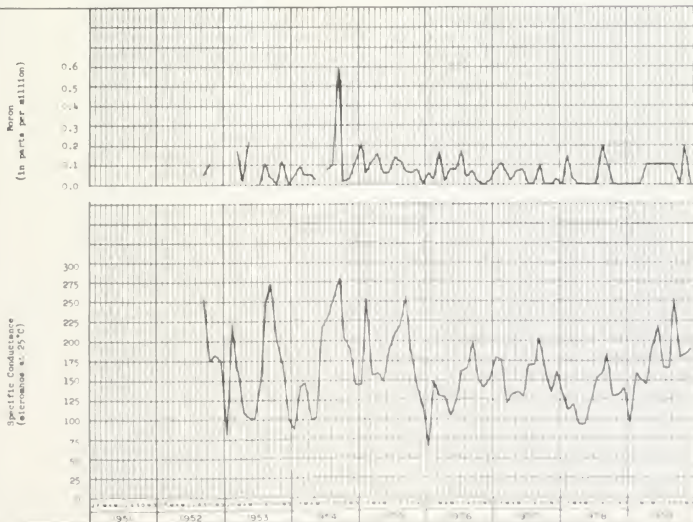
Water Quality Characteristics The Delta Cross Channel is a unit of the Central Valley Project and is comprised of artificial and natural channels used to divert water from the Sacramento River near Walnut Grove. The diverted water flows through various channels of the delta to the intake of the Tracy pumping plant. The water at this station is calcium-magnesium bicarbonate in character, soft to slightly hard, and generally within accepted mineral limits for domestic and class 1 irrigation use.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	260	84	243	97
Temperature in °F	77	44	74	40
Dissolved oxygen in parts per million	12.2	7.7	11.5	7
Percent saturation	121	63	120	70
pH	8.2	6.8	7.5	7
Mineral constituents in parts per million				
Calcium (Ca)	241	5.9	188	11
Magnesium (Mg)	11	2.7	9.8	9
Sodium (Na)	24	1.6	21	4.3
Potassium (K)	2.0	0.5	1.7	1.1
Carbonate (CO ₃)	0.41	0.0	0.41	0.0
Bicarbonate (HCO ₃)	123	33	118	35
Sulfate (SO ₄)	19	4.0	13	13
Chloride (Cl)	20	1.3	17	5.8
Nitrate (NO ₃)	2.1	0.0	1.2	0.0
Fluoride (F)	8.3	0.0	8.3	0.0
Boron (B)	0.40	0.0	0.2	0.0
Silica (SiO ₂)	26	1.5	24	17
Total dissolved solids in parts per million	175	43	148	64
Percent sodium	37	15	35	20
Hardness as CaCO ₃ in parts per million				
Total	40	28	84	37
Noncarbonate	13	3.0	8	0
Turbidity	140	0.9	50	1
Coliform in most probable number per milliliter	70,000	6.62	1,000	4.2
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.43	0.0	0.43	0.0
Solid alpha	0.32	0.0	0.17	0.0
Dissolved beta	7.38	0.0	3.79	0.0
Solid beta	7.0	0.0	5.49	0.0

WATER QUALITY VARIATIONS



DELTA CROSS CHANNEL NEAR WALNUT GROVE (STA. 98)

LITTLE POTATO SLOUGH AT TERMINOUS (STA. 99)

Sampling Point Station 99 is located in Section 13, Township 3 North, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were taken from a boat dock on the east bank, about 250 feet north of State Highway 12 bridge.

Period of Record September 1952 through December 1959.

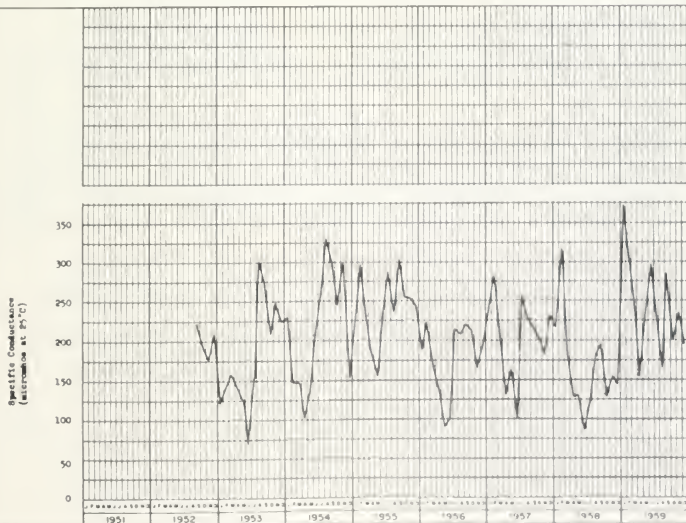
Water Quality Characteristics Antecedent data reveal the water to be a complex calcium-magnesium-sodium bicarbonate type of excellent mineral quality, slightly hard to moderately hard, class 1 for irrigation and suitable for domestic uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1949	Minimum - 1949
Specific conductance (microhm at 25°C)	275	7	275	7
Temperature in °F	77	47	77	48
Dissolved oxygen in parts per million	10.5	7.2	10.5	7.2
Percent saturation	100	80	100	80
pH	7.7	6.8	7.7	6.8
Mineral constituents in parts per million				
Calcium (Ca)	24	2	24	2
Magnesium (Mg)	18	1.1	18	1.1
Sodium (Na)	12	2	27	8
Potassium (K)	2.8	2	1.5	1.7
Carbonate (CO ₃)	127	28	11	8
Bicarbonate (HCO ₃)	28	2.4	13	2.6
Sulfate (SO ₄)	28	2	22	2
Chloride (Cl)	7.8	1	1.3	1
Nitrate (NO ₃)	2.5	0	2.5	0
Fluoride (F)	1	0	1	0
Boron (B)	24	21	21	21
Silica (SiO ₂)	24	21	21	21
Total dissolved solids in parts per million	223	42	223	94
Percent sodium	42	22	38	24
Hardness as CaCO ₃ in parts per million				
Total	116	26	116	55
Noncarbonate	46	19	46	19
Turbidity	150	2	41	2
Coliform in most probable number per milliliter	>7,000	0.23	7,000	13
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



LITTLE POTATO SLOUGH AT TERMINOUS (STA. 99)

SAN JOAQUIN RIVER AT MOSSDALE BRIDGE (STA. 102)

Sampling Point Station 102 is located in Section 4, Township 2 South, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a boat landing on the left bank, just downstream from Mossdale Bridge on U. S. Highway 50, about 12 miles south of Stockton and 7 miles northeast of Tracy.

Period of Record September 1952 through December 1959.

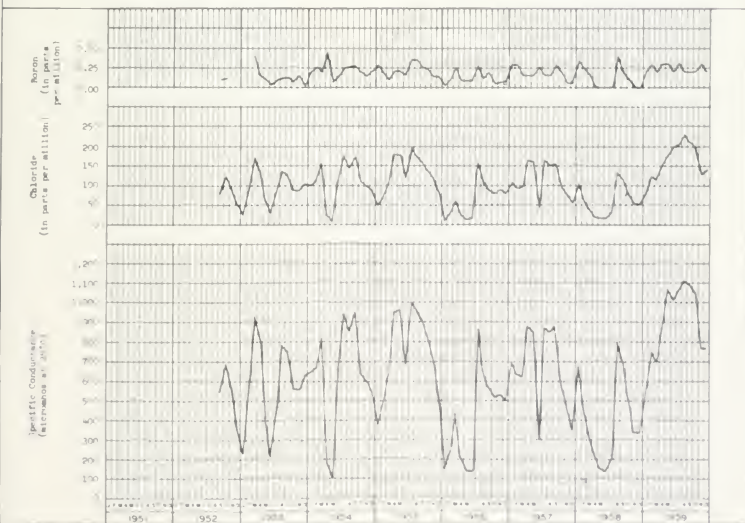
Water Quality Characteristics Water at the station is predominantly a sodium chloride type, moderately hard to very hard, and not recommended for domestic and some industrial uses. During the summer months the concentration of dissolved solids is often sufficiently high to place the water in class 2 for irrigation. The quality of water at this station is influenced by tidal action, fresh water inflow to the delta, irrigation diversions and return flows, and ground water accretions. Wide seasonal variations in quality are characteristics of the water at Station 102.

Significant Water Quality Changes The 1959 water year was considerably below normal and the quality of water reflected the effects of low flow conditions. During 1959, the maximum values for conductivity and chlorides occurred in August, 1,110 micromhos and 232 ppm, respectively. The conductivity value of 1,110 micromhos is the maximum of record at this station and represents a significant increase over the 1958 maximum of 793 micromhos. Because of conductivity and chlorides the water was class 2 for irrigation from May to October 1959.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhms at 25°C)	110	43	7-148	538
Temperature in °F	88	48	89	48
Dissolved oxygen in parts per million	12.7	4.4	11.7	4.4
Percent saturation	176	51	151	51
pH	8.5	6.5	8.2	7.2
Mineral constituents in parts per million				
Calcium (Ca)	18	7.2	58	14
Magnesium (Mg)	48	2.4	25	2.4
Sodium (Na)	182	2.6	112	60
Potassium (K)	0.4	0.7	6.4	5.9
Carbonate (CO ₃)	0.4	0.7	0.4	0.7
Bicarbonate (HCO ₃)	304	12	263	103
Sulfate (SO ₄)	115	115	115	75
Chloride (Cl)	182	2.3	212	89
Nitrate (NO ₃)	1.7	0.3	3.4	0.8
Fluoride (F)	0.4	0.2	0.2	0.2
Boron (B)	0.4	0.3	0.3	0.2
Silica (SiO ₂)	14	0.3	27	2
Total dissolved solids in parts per million	633	58	633	304
Percent sodium	77	12	54	1
Hardness as CaCO ₃ in parts per million				
Total	246	26	246	116
Noncarbonate	99	2.3	95	12
Turbidity	2	0.3	85	0.8
Coliform in most probable number per milliliter	>7,000	<2,000	>7,000	0-14
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT MOSSDALE BRIDGE (STA. 102)

SAN JOAQUIN RIVER AT GARWOOD BRIDGE (STA. 101)

Sampling Point The station is located in Section 16, Township 1 North, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a boat landing on the left bank, upstream from State Highway 4 bridge and approximately 4 miles west of Stockton.

Period of Record September 1952 through December 1959.

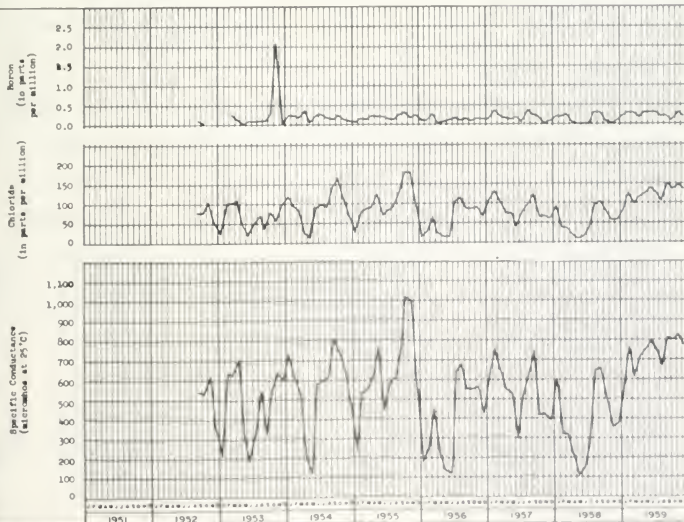
Water Quality Characteristics Water at Station 101 is predominantly sodium chloride in character, moderately hard, class 1 for irrigation and suitable for domestic use. Analyses of samples collected at this station indicate Sacramento River water, which traverses the delta through the many interconnected channels, and water from other streams tributary to the delta, significantly affects the quality of San Joaquin River at Garwood Bridge. Quality of water generally improves at the Garwood Bridge station (averaging about 250 micromhos) when mineral concentrations are compared to the next upstream station at Mossdale Bridge.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (micromhos at 25°C)	1,200	100	1,100	100
Temperature 10 °F	60	40	60	40
Dissolved oxygen in parts per million	11.6	7.0	11.6	7.0
Percent saturation	120	70	120	70
pH	8.4	7.2	8.4	7.2
Mineral constituents in parts per million				
Calcium (Ca)	54	7.8	46	10
Magnesium (Mg)	20	2.5	17	15
Sodium (Na)	110	0.2	109	4.4
Potassium (K)	8.5	1.2	7.3	6.7
Carbonate (CO ₃)	10	0.5	9.5	0.5
Bicarbonate (HCO ₃)	104	17	87	100
Sulfate (SO ₄)	60	5.8	54	26
Chloride (Cl)	180	11	169	79
Nitrate (NO ₃)	4.7	0.4	4.3	0.6
Fluoride (F)	0.5	0.1	0.4	0.0
Boron (B)	2.1	0.3	1.8	0.1
Silica (SiO ₂)	21	2.0	19	4.8
Total dissolved solids in parts per million	575	69	506	200
Percent sodium	57	17	57	51
Hardness as CaCO ₃ in parts per million				
Total	227	31	196	113
Noncarbonate	64	0.1	63	5
Turbidity	310	0.0	310	0.0
Coliform in most probable number per milliliter	>7,000	2.3	>7,000	62
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



SAN JOAQUIN RIVER AT GARWOOD BRIDGE (STA. 101)

SAN JOAQUIN RIVER AT ANTIOCH (STA. 28)

Sampling Point The Antioch station is located in Section 18, Township 2 North, Range 2 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank at old Antioch Water Works pier, Fulton Shipyard Road, near the northeast city limits of Antioch.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Quality of water at Station 28 is affected by sea-water incursion from San Francisco Bay. During the summer and fall months when outflow from the delta is insufficient to repel sea-water incursion, the water is sodium chloride in character, poor in quality, class 3 for irrigation and unsuitable for nearly all domestic and industrial uses. In the winter and spring, when river outflow from the delta increases, the water is sodium bicarbonate in character, excellent in quality, class 1 for irrigation, and within mineral quality requirements for domestic use.

Significant Water Quality Changes During 1959, analyses showed quality of water at the Antioch station was generally poorer from May through December than in previous years of record. In the first four months of 1959, conductivity did not exceed 500 micromhos; however, in May and June it increased to about 1,000 micromhos, and during the remaining six months conductivity was in excess of 1,500 micromhos with a maximum of 6,010 micromhos occurring in July. The high specific conductance of water at Antioch, in 1959, is attributable to the low outflow from the delta. Low outflow resulted from the considerably below normal precipitation over much of the drainage area tributary to the delta.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1959
Specific conductance (microhm at 25°C)	6,134	100	4,733	100
Temperature in °F	77	42	77	42
Dissolved oxygen in parts per million	18.8	7.0	18.7	7.0
Percent saturation	177	70	177	70
pH	8.0	6.8	8.0	6.8
Mineral constituents in parts per million				
Calcium (Ca)	61	8	74	8
Magnesium (Mg)	148	3	66	3
Sodium (Na)	1,387	7	1,387	7
Potassium (K)	72	1	72	1
Carbonate (CO ₃)	8,000	1	8,000	1
Bicarbonate (HCO ₃)	13	1	13	1
Sulfate (SO ₄)	204	7	204	7
Chloride (Cl)	1,993	1	1,993	1
Nitrate (NO ₃)	6.1	1	6.1	1
Fluoride (F)	1	1	1	1
Boron (B)	1.1	1	1.1	1
Silica (SiO ₂)	23	1	18	16
Total dissolved solids in parts per million	1,600	10	1,600	10
Percent sodium	77	11	77	11
Hardness as CaCO ₃ in parts per million				
Total	790	2	790	2
Noncarbonate	641	2	641	2
Turbidity	150	1	150	1
Coliform in most probable number per milliliter	24,000	1	24,000	1
Radioactivity in micro-micro curies per liter				
Dissolved alpha	8.64	1	62	1
Solid alpha	18.15	1	11	1
Dissolved beta	16.35	1	11	1
Solid beta	16.35	1	11	1

WATER QUALITY VARIATIONS



STOCKTON SHIP CHANNEL ON RINDGE ISLAND (STA. 100)

Sampling Point Stockton Ship Channel station is located in Section 27, Township 2 North, Range 5 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from a boat landing on the right bank of the ship channel, at the southeast corner of Rindge Tract, and near the junction of Fourteen Mile Slough.

Period of Record September 1952 through December 1959.

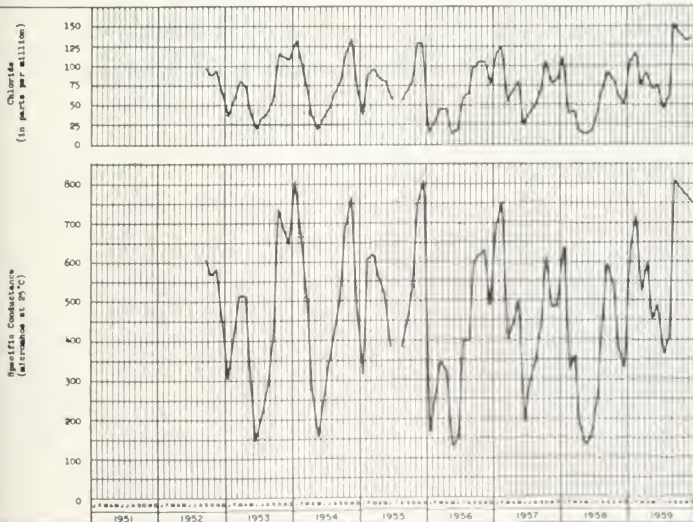
Water Quality Characteristics Antecedent data show the water to be predominantly sodium chloride in character during the winter months, changing to sodium bicarbonate during the summer months. The water is normally well within the limits for class 1 irrigation and domestic uses, and is in the moderately hard range.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)		1	810	56
Temperature in °F	84	41	8	50
Dissolved oxygen in parts per million	12.4	8.8	12.5	8.8
Percent saturation	119	11	132	73
pH	8.4	6.9	8.1	7.2
Mineral constituents in parts per million				
Calcium (Ca)	92	2	80	26
Magnesium (Mg)	22	7	16	13
Sodium (Na)	187	12	188	35
Potassium (K)	8.6	2.2	8.6	3.4
Carbonate (CO ₃)	2	0	0.5	0
Bicarbonate (HCO ₃)	198	42	198	133
Sulfate (SO ₄)	98	7	90	26
Chloride (Cl)	152	14	159	48
Nitrate (NO ₃)	14	0.4	13.6	1.1
Fluoride (F)	0.6	0	0.2	0
Boron (B)	1.6	0	0.3	0
Silica (SiO ₂)	23	1.4	18	1.4
Total dissolved solids in parts per million	473	83	461	203
Percent sodium	57	13	57	40
Hardness as CaCO ₃ in parts per million				
Total	210	16	182	109
Noncarbonate	124	0.0	64	4
Turbidity	90	1	86	1
Coliform in most probable number per milliliter	>7,000	0.62	>7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



STOCKTON SHIP CHANNEL ON RINDGE ISLAND (STA. 100)

OLD RIVER NEAR TRACY (STA. 103)

Sampling Point Station 103 is located in Section 6, Township 2 South, Range 5 East, Mt. Diablo Base and Meridian. Samples were collected from the trash rack of a pump intake on the left bank, 500 feet from Lammers Road about 5.0 miles northwest of Tracy.

Period of Record October 1952 through December 1959.

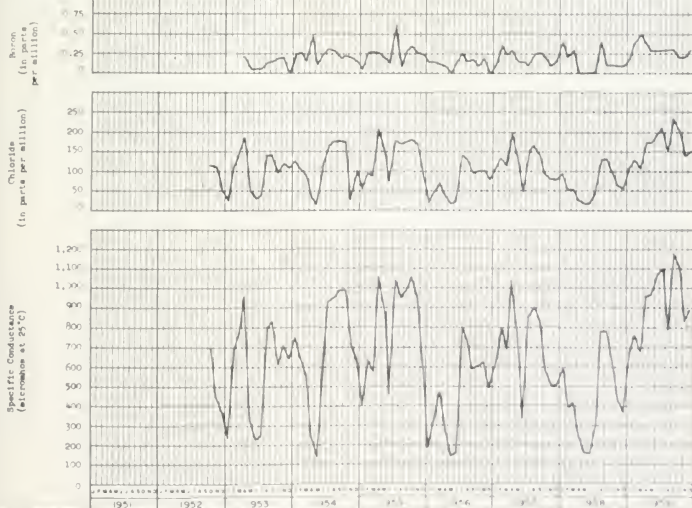
Water Quality Characteristics Water at Old River near Tracy is predominantly a sodium chloride type, frequently class 2 for irrigation (usually during the summer months), moderately hard to very hard, and not recommended for domestic and some industrial uses.

Significant Water Quality Changes The 1959 maximum values for conductivity and chlorides (1,180 micromhos and 234 ppm, respectively) established new maximums for the period of record and represent a significant increase of values found at this station in prior years of record. These high values are attributed to the low flow conditions existing in streams tributary to this portion of the delta.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1949	Minimum - 1959
Specific conductance (microhms at 25°C)	1,175	115	1,150	160
Temperature in °F	81	45	80	40
Dissolved oxygen in parts per million	16.8	5.5	16.0	6.5
Percent saturation	188	48	170	71
pH	8.5	7.2	8.1	7.2
Mineral constituents in parts per million				
Calcium (Ca)	67	9.2	67	53
Magnesium (Mg)	32	3.8	32	23
Sodium (Na)	1,114	19	114	71
Potassium (K)	7.2	1.2	7.2	5.4
Carbonate (CO ₃)	6.5	6.5	6.5	0.0
Bicarbonate (HCO ₃)	215	38	215	118
Sulfate (SO ₄)	81	9.1	81	74
Chloride (Cl)	234	17	234	177
Nitrate (NO ₃)	5.1	5.1	5.1	0.3
Fluoride (F)	2.4	2.4	2	0.0
Boron (B)	0.63	0.0	0.5	0.2
Silica (SiO ₂)	27	11	19	13
Total dissolved solids in parts per million	673	81	673	376
Percent sodium	54	37	54	49
Hardness as CaCO ₃ in parts per million				
Total	248	36	248	147
Noncarbonate	129	3	129	40
Turbidity	11.5	0.0	11	0.3
Coliform in most probable number per milliliter	7,000	1,400	7,000	2.1
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



OLD RIVER NEAR TRACY (STA. 103)

OLD RIVER AT CLIFTON COURT FERRY (STA. 104)

Sampling Point Clifton Court Ferry station is located in Section 20, Township 1 South, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the ferry on the left bank, about 0.3 mile downstream from a tide stage recorder, 6.0 miles southeast of Byron, 10 miles northwest of Tracy.

Period of Record September 1952 through December 1959.

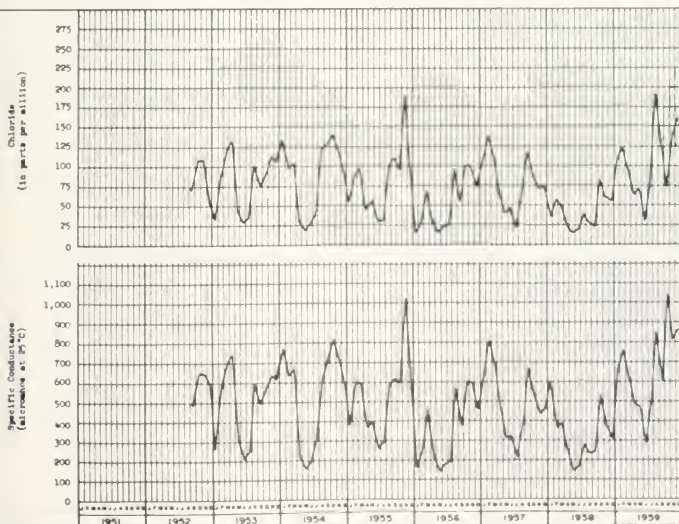
Water Quality Characteristics Antecedent data reveal the water to be predominantly sodium chloride in character, slightly hard to very hard, and occasionally class 2 for irrigation due to high conductivity and chloride concentrations associated with sea-water incursion and poor quality river inflows.

Significant Water Quality Changes During 1959, the maximum values for conductivity and chloride were 1,040 micromhos and 198 ppm, respectively. As at other delta stations maximums of record occurred, representing a significant increase over previously reported values. Low flow conditions in streams tributary to the delta probably caused these maximum values.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohms at 25°C)	500	100	500	100
Temperature in °F	80	40	80	40
Dissolved oxygen in parts per million	12.5	6.5	12.5	6.5
Percent saturation	100	10	100	10
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	120	20	120	20
Magnesium (Mg)	20	5	20	5
Sodium (Na)	270	10	270	10
Potassium (K)	5	1	5	1
Carbonate (CO ₃)	10	2	10	2
Bicarbonate (HCO ₃)	190	40	190	40
Sulfate (SO ₄)	70	10	70	10
Chloride (Cl)	190	10	190	10
Nitrate (NO ₃)	5	1	5	1
Fluoride (F)	0.7	0.1	0.7	0.1
Iron (Fe)	0.5	0.1	0.5	0.1
Silica (SiO ₂)	31	1.0	31	1.0
Total dissolved solids in parts per million	500	80	500	80
Percent sodium	67	15	67	15
Hardness as CaCO ₃ in parts per million				
Total	230	30	230	30
Noncarbonate	87	1	87	1
Turbidity	120	5.0	85	5.0
Coliforms in most probable number per milliliter	>7,000	12	>7,000	12
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



OLD RIVER AT CLIFTON COURT FERRY (STA. 104)

OLD RIVER AT ORWOOD BRIDGE (STA. 108)

Sampling Point Station 108 is located in Section 17, Township 1 North, Range 4 East, Mt. Diablo Base and Meridian.. Monthly grab samples were collected from a boat dock on the right bank, at Atchison, Topeka and Santa Fe Railroad bridge and about 6.0 miles northeast of Byron.

Period of Record September 1952 through December 1959.

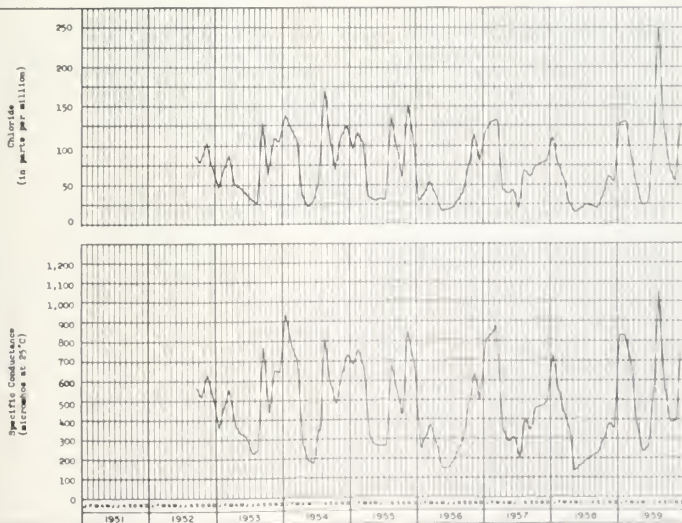
Water Quality Characteristics Past analyses show the water to be sodium chloride in character during the winter and summer months and a complex sodium-calcium-magnesium bicarbonate character during the spring and fall months. Samples of water from Station 108, with one exception, usually met class 1 irrigation water requirements. Old River water also met mineral quality standards for domestic use and ranged from slightly hard to moderately hard.

Significant Water Quality Changes During August 1959 conductivity and chlorides were 1,050 micromhos and 250 ppm, respectively. These values, which are new maximums for the period of record, changed the classification of the water for irrigation use from class 1 to class 2. High quality Sacramento River water, drawn across the delta by the Tracy Pumping Plant in conjunction with increased releases of stored water from Shasta and Folsom reservoirs, partially alleviated the sea-water incursion problem believed responsible for these high values.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,700	100	1,000	200
Temperature in °F	79	44	78	49
Dissolved oxygen in parts per million	10.4	6.2	10.4	8
Percent saturation	91	69	91	70
pH	8.1	7.2	7.7	8.1
Mineral constituents in parts per million				
Calcium (Ca)	55	9.3	18	16
Magnesium (Mg)	27	3	16	7
Sodium (Na)	153	12	153	20
Potassium (K)	4.3	1.3	4.2	2.1
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	164	42	124	24
Sulfate (SO ₄)	139	7.7	134	24
Chloride (Cl)	250	10	240	24
Nitrate (NO ₃)	13	0.0	13	0.0
Fluoride (F)	0.5	0.0	0.5	0.0
Boron (B)	0.4	0.0	0.4	0.0
Silica (SiO ₂)	26	1.4	19	1.0
Total dissolved solids in parts per million	500	96	404	144
Percent sodium	48	10	38	10
Hardness as CaCO ₃ in parts per million				
Total	244	16	228	24
Noncarbonate	164	2	162	2
Turbidity	110	5	95	5
Coliform in most probable number per milliliter	>7,000	62	>7,000	62
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



OLD RIVER AT ORWOOD BRIDGE (STA 108)

OLD RIVER AT MANDEVILLE ISLAND (STA. 112)

Sampling Point Station 112 is located in Section 6, Township 2 North, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the northwest side of Mandeville Island, approximately 1.0 mile from the mouth of Old River, and about 5.0 miles northwest of Mandeville School, along the levee road.

Period of Record December 1954 through December 1959.

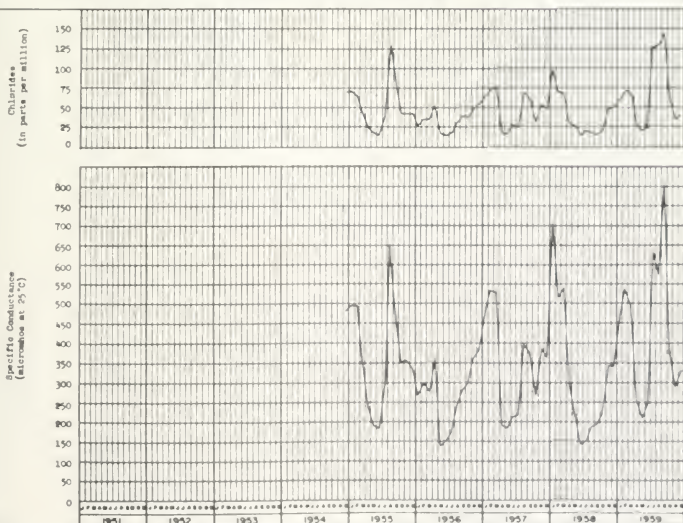
Water Quality Characteristics Water at Station 112 is a complex sodium-calcium-magnesium bicarbonate-chloride type, class 1 for irrigation, slightly to moderately hard and suitable for domestic and some industrial uses. The Old River channel is the main carrier of high quality Sacramento River water while it traverses the delta en route to the Tracy Pumping Plant.

Significant Water Quality Changes During September 1959, conductivity and chloride values of 801 micromhos and 145 ppm, respectively, established new maximums of record.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm/cm at 25°C)	81	3	81	3
Temperature in °F	79	43	79	43
Dissolved oxygen in parts per million	12.8	8	12.8	8
Percent saturation	97	67	94	69
pH	8.8	7.8	7.8	7.8
Mineral constituents in parts per million				
Calcium (Ca)	10	6	38	15
Magnesium (Mg)	19	3.9	28	6.9
Sodium (Na)	15	1	100	18
Potassium (K)	9.4	1.2	9.4	2
Carbonate (CO ₃)	3.5	0.5	2.5	0.5
Bicarbonate (HCO ₃)	104	42	194	72
Sulfate (SO ₄)	68	1	13	18
Chloride (Cl)	145	4	145	20
Nitrate (NO ₃)	8.1	0.3	2.6	0.6
Fluoride (F)	2.1	0.2	0.1	0.1
Boron (B)	0.50	0.2	0.4	0.2
Silica (SiO ₂)	22	1.7	15	14
Total dissolved solids in parts per million	457	84	457	138
Percent sodium	62	31	62	36
Hardness as CaCO ₃ in parts per million				
Total	217	4	172	66
Noncarbonate	129	0.8	71	8
Turbidity	50	3	50	3
Coliform in most probable number per milliliter	>7,000	1.3	7,000	2.3
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



OLD RIVER AT MANDEVILLE ISLAND (STA. 112)

GRANT LINE CANAL AT TRACY ROAD BRIDGE (STA. 103a)

Sampling Point The Grant Line Canal station is located in Section 30, Township 1 South, Range 5 East, Mt. Diablo Base and Meridian. The monthly water samples were collected from a boat dock at Tracy Road Bridge approximately 5 miles north of Tracy.

Period of Record July 1958 through December 1959.

Water Quality Characteristics Past analyses show the water to be sodium chloride in character, moderately hard to very hard, and class 1 to 2 for irrigation.

Significant Water Quality Changes The maximum 1959 values for conductivity and chlorides (September sample) were 1,130 micromhos and 230 ppm, respectively, representing a significant increase over the 1958 values for these characteristics. Even though these values are new maximums of record, because of the short period of record, it is difficult to ascertain if significant changes occurred.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	1,130	112	1,130	536
Temperature in °F	86	47	86	47
Dissolved oxygen in parts per million	11.3	4.9	11.3	7.5
Percent saturation	140	50	140	86
pH	8.9	7.8	8.9	7.8
Mineral constituents in parts per million				
Calcium (Ca)	64	18	64	27
Magnesium (Mg)	30	7.5	27	13
Sodium (Na)	139	19	139	59
Potassium (K)	8.6	1.4	8.6	2.5
Carbonate (CO ₃)	12	0.0	12	0.0
Bicarbonate (HCO ₃)	205	67	205	104
Sulfate (SO ₄)	77	33	77	45
Chloride (Cl)	243	54	230	87
Nitrate (NO ₃)	2.7	0.2	2.7	0.2
Fluoride (F)	0.4	0.0	0.3	0.0
Iron (Fe)	0.4	0.0	0.4	0.1
Silica (SiO ₂)	24	0.2	24	0.2
Total dissolved solids in parts per million	658	108	658	336
Percent sodium	54	49	54	51
Hardness as CaCO ₃ in parts per million				
Total	272	76	272	121
Noncarbonate	104	21	104	36
Turbidity	75	1.2	40	1.2
Coliform in most probable number per milliliter	See 1959	See 1959	7,300	1,200
Radioactivity in micro-curries per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



GRANT LINE CANAL AT TRACY ROAD BRIDGE (STA. 103a)

DELTA-MENDOTA CANAL NEAR TRACY (STA. 93)

Sampling Point Station 93 is located in Section 30, Township 1 South, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from left bank downstream from Byron-Bethany Road crossing, about 1 mile from Tracy Pumping Plant, about 10 miles northwest of Tracy.

Period of Record July 1952 through December 1959.

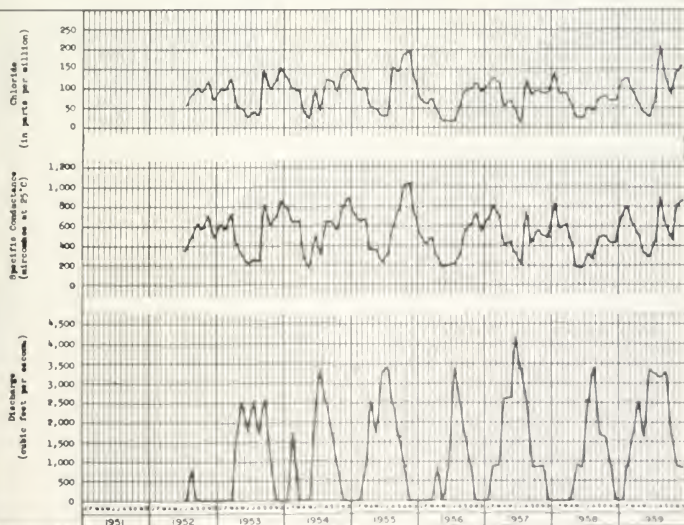
Water Quality Characteristics The water at Station 93 is predominantly sodium chloride in character, changing to sodium bicarbonate in the late spring due to increased runoff, and then reverting back again to sodium chloride. The canal water contains moderate amounts of dissolved solids and is usually class 1 for irrigation. In August 1959, a high chloride concentration and percent sodium placed the water in class 2. The hardness is slightly hard to moderately hard, limiting some domestic and industrial water uses.

Significant Water Quality Changes During 1959, the maximum values for conductivity and chloride were 886 micromhos and 208 ppm, respectively. The 208 ppm chloride concentration is a maximum of record and placed the water in class 2 for irrigation use.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	1,190	42	550	200
Temperature in °F	80	42	79	1
Dissolved oxygen in parts per million	11.1	6.2	11.1	6.8
Percent saturation	121	65	121	78
pH	8.7	6.7	8.1	7.3
Mineral constituents in parts per million				
Calcium (Ca)	54	8.8	24	21
Magnesium (Mg)	26	2.9	15	0.6
Sodium (Na)	123	13	120	27
Potassium (K)	5.2	1.0	4.6	1.0
Carbonate (CO ₃)	8.0	0.0	8.0	0.0
Bicarbonate (HCO ₃)	166	38	134	81
Sulfate (SO ₄)	109	4.8	34	34
Chloride (Cl)	208	17	208	31
Nitrate (NO ₃)	5.7	0.0	5.8	0.8
Fluoride (F)	0.5	0.0	0.2	0.1
Boron (B)	0.98	0.0	0.5	0.1
Silica (SiO ₂)	28	1.0	18	1.2
Total dissolved solids in parts per million	571	93	499	161
Percent sodium	66	38	66	42
Hardness as CaCO ₃ in parts per million				
Total	234	41	193	89
Noncarbonate	122	2	73	11
Turbidity	140	1	140	1
Coliform in most probable number per milliliter	>7,000	0.23	7,000	0.23
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.27		0.27	
Solid alpha	2.88		0.99	
Dissolved beta	12.39		7.58	
Solid beta	7.24		1.61	

WATER QUALITY VARIATIONS



DELTA-MENDOTA CANAL NEAR TRACY (STA. 93)

DELTA-MENDOTA CANAL NEAR MENDOTA (STA. 92)

Sampling Point The Mendota station is located in Section 19, Township 13 South, Range 15 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, about 1 mile upstream from the gates to Mendota Pool and about 2 miles north of Mendota.

Period of Record July 1952 through December 1959.

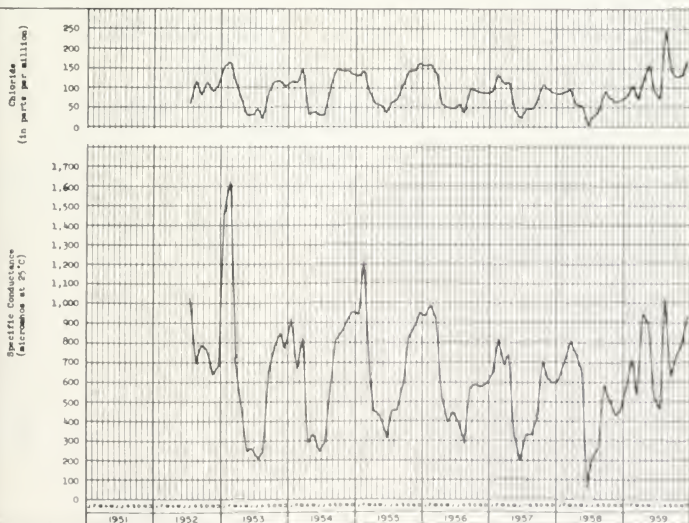
Water Quality Characteristics Past analyses show the water to be predominantly sodium chloride in character with moderate concentrations of dissolved solids, moderately hard to very hard, and normally class 1 for irrigation during the pumping season. The water, during August, is frequently class 2 for irrigation because of conductivity, chlorides, and percent sodium, but is within class 1 requirements throughout the remainder of the year. A comparison of the quality of water at Tracy station with that at Station 92 cannot effectively be made since sampling prior to September 1959 did not give consideration to the time of travel of the water in the canal.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1979
Specific conductance (microhm at 25°C)	1,630	61.5	1,090	457
Temperature in °F	81	43	76	48
Dissolved oxygen in parts per million	11.8	6.4	10.5	7.1
Percent saturation	158	77	88	77
pH	8.5	7.0	7.8	7.1
Mineral constituents in parts per million				
Calcium (Ca)	67	13	41	23
Magnesium (Mg)	35	4.1	26	18
Sodium (Na)	212	4.2	142	55
Potassium (K)	5.2	1.5	4.6	3.6
Carbonate (CO ₃)	4	0	0.03	0.79
Bicarbonate (HCO ₃)	249	26	159	88
Sulfate (SO ₄)	154	29	67	40
Chloride (Cl)	245	1.8	245	70
Nitrate (NO ₃)	2.0	0.6	1.4	0.8
Fluoride (F)	0.4	0.0	0.2	0.1
Boron (B)	0.89	0.0	0.4	0.1
Silica (SiO ₂)	40	12	18	13
Total dissolved solids in parts per million	900	35	573	294
Percent sodium	67	30	67	49
Hardness as CaCO ₃ in parts per million				
Total	311	21	210	146
Noncarbonate	146	0.8	88	26
Turbidity	180	0.0	85	3.8
Coliform in most probable number per milliliter	>7,000	0.045	620	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.50	0.00	0.50	0.00
Solid alpha	1.6	0.00	0.27	0.00
Dissolved beta	29.41	0.00	12.05	0.00
Solid beta	8.1	0.00	0.28	0.00

WATER QUALITY VARIATIONS



DELTA-MENDOTA CANAL NEAR MENDOTA (STA. 92)

ITALIAN SLOUGH NEAR MOUTH (STA. 106)

Sampling Point Station 106 is located in Section 7, Township 1 South, Range 4 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at a pump house on the northwestern side of Clifton Court Tract, about 3.0 miles southeast of Byron.

Period of Record September 1952 through December 1959.

Water Quality Characteristics Italian Slough water is predominantly sodium chloride in character and slightly to very hard. The water is occasionally class 2 for irrigation due to high conductivity, chloride and boron. Italian Slough, one of several dead-end sloughs in the southwestern delta, is used as an intake channel by the Byron-Bethany Irrigation District to divert water during the irrigation season from Old River to a portion of the delta uplands area. Due to the proximity of this station to Old River, the quality of water in the slough is largely dependent upon the quality of water in Old River.

Significant Water Quality Changes In Italian Slough the August 1959 values for conductivity and chloride were 1,000 micromhos and 232 ppm, respectively. These values establish new maximums of record for conductivity and chloride concentrations in the slough. These maximums occurred at the same time that the 1959 maximums of 1,050 micromhos conductivity and 250 ppm chlorides occurred at Station 108 on Old River. They reflect the influence of Old River on the quality at Station 106.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1 Yr	Minimum - 1 Yr
Specific conductance (microhms at 25°C)	1,100	40	1,100	200
Temperature in °F	84	4	84	50
Dissolved oxygen in parts per million	10.1	0.4	9.1	0.1
Percent saturation	100	60	91	10
pH	8.5	7.8	7.5	7.9
Mineral constituents in parts per million				
Calcium (Ca)	51	9.1	40	18
Magnesium (Mg)	22	1.3	18	0.8
Sodium (Na)	145	1.1	144	2.1
Potassium (K)	4.6	1.2	4.6	1.1
Carbonate (CO ₃)	28	0.1	28	0.1
Bicarbonate (HCO ₃)	141	19	124	14
Sulfate (SO ₄)	41	1	40	24
Chloride (Cl)	232	10	232	10
Nitrate (NO ₃)	1.8	0.2	1.1	0.4
Fluoride (F)	0.4	0.2	0.2	0.2
Boron (B)	1.1	0.3	1.1	0.3
Silica (SiO ₂)	25	1.2	17	14
Total dissolved solids in parts per million	571	88	471	147
Percent sodium	68	36	68	38
Hardness as CaCO ₃ in parts per million				
Total	208	38	208	77
Noncarbonate	135	3	135	11
Turbidity	85	2	60	2
Coliform in most probable number per milliliter	>7,000	0.62	>7,000	62
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



ITALIAN SLOUGH NEAR MOUTH (STA. 106)

INDIAN SLOUGH NEAR BRENTWOOD (STA. 107)

Sampling Point Station 107 is located in Section 22, Township 1 North, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the East Contra Costa Irrigation District canal at Pump No. 1 on Bixler Road, at the head of Indian Slough, approximately 3.0 miles north of Byron.

Period of Record September 1952 through December 1959.

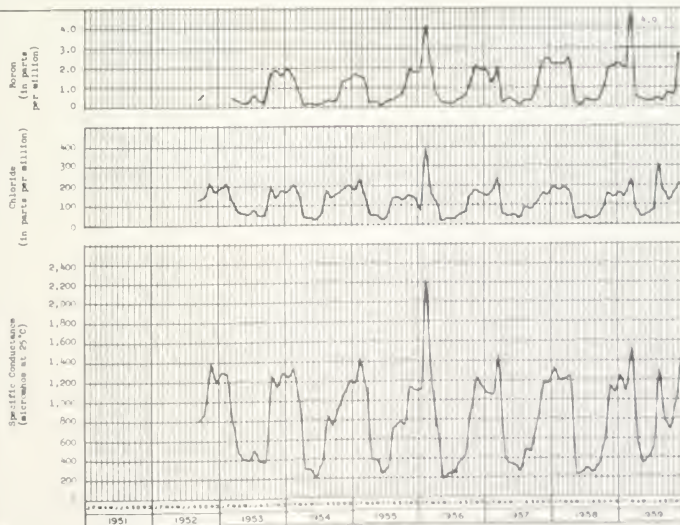
Water Quality Characteristics Past analyses show the water to be predominantly sodium bicarbonate during the winter months and sodium chloride in the summer. The water ranges from excellent to poor in quality, moderately hard to very hard, and is frequently class 2 or 3 for irrigation during the winter months due to high conductivity, chlorides and boron.

Significant Water Quality Changes Early in 1959, when the irrigation pumps were not operating, conductivity increased to 1,530 micromhos, chlorides to 235 ppm and boron to 4.9 ppm. These values reflect the mineral build-up caused by poor quality accretions from ground water into the deadend slough. Operation of irrigation pumps on the slough resulted in water from Old River flowing through the slough with a subsequent improvement in the quality of water as reflected by a decrease to 344 micromhos, 38 ppm chloride and 0.3 ppm boron. However, in August, due to the low flow conditions in streams tributary to the delta, water quality was again impaired and conductivity increased to 1,300 micromhos, chlorides to 312 ppm and boron 0.4 ppm. In the latter part of the summer, water releases from Shasta and Folsom reservoirs were increased to dilute the poor quality water in the delta. As a result, by October conductivity dropped to 705 micromhos and chlorides to 122 ppm. In December, when all irrigation pumping had ceased, conductivity again increased to 1,420 micromhos, chlorides to 208 ppm, and boron to 2.7 ppm.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1957	Minimum - 1957
Specific conductance (micromhos at 25°C)	8,890	120	1,500	150
Temperature in °F	81	60	60	60
Dissolved oxygen in parts per million	16	5.4	12.5	6.1
Percent saturation	140	40	100	70
pH	8.4	6.8	8.0	7.0
Mineral constituents in parts per million				
Calcium (Ca)	1,200	6.5	250	10
Magnesium (Mg)	77	4.7	150	5
Sodium (Na)	264	18	150	10
Potassium (K)	6.2	1.4	6.0	1.0
Carbonate (CO ₃)	7.8	0.5	7.0	0.5
Bicarbonate (HCO ₃)	17	4.0	13.0	3.0
Sulfate (SO ₄)	78	11	60	10
Chloride (Cl)	190	20	170	20
Nitrate (NO ₃)	1.8	0.5	1.3	0.5
Fluoride (F)	0.1	0.05	0.05	0.05
Boron (B)	4.0	0.2	4.0	0.2
Silica (SiO ₂)	201	7.8	200	10
Total dissolved solids in parts per million	1,940	112	850	210
Percent sodium	69	30	60	40
Hardness as CaCO ₃ in parts per million				
Total	570	40	400	30
Noncarbonate	275	5	130	5
Turbidity	160	1	150	1
Coliforms in most probable number per milliliter	7,000	20	150	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



ROCK SLOUGH NEAR KNIGHTSEN (STA. 109)

Sampling Point Station 109 is located in Section 34, Township 2 North, Range 3 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the Tule Lane bridge 300 feet south of Contra Costa Canal intake gates, and near the head of Rock Slough about 2 miles north-east of Knightsen.

Period of Record September 1952 through December 1959.

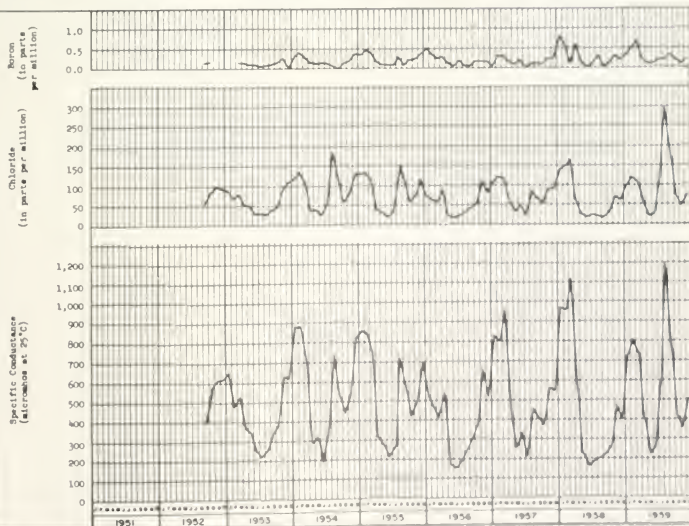
Water Quality Characteristics Rock Slough water is generally a chloride type with no predominant cation except during the irrigation season when it becomes sodium chloride. The water is generally within limits of class 1 irrigation water, suitable for domestic use from a mineral standpoint, and slightly hard to moderately hard. Significant seasonal variations in quality are noted at this station. These variations reflect the changing quality of Old River water and are probably attributable to the effects of accretions from ground water, surface drainage, and sea-water intrusion.

Significant Water Quality Changes In August 1959 maximum values for conductivity and chlorides, 1,190 micromhos and 295 ppm, respectively, established new maximums of record.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1949	Minimum 1949
Specific conductance (micromhos at 25°C)	1,100	100	1,000	200
Temperature in °F	80	35	80	35
Dissolved oxygen in parts per million	10.5	1.5	10.5	1.5
Percent saturation	110	10	110	10
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	100	10	100	10
Magnesium (Mg)	50	5	50	5
Sodium (Na)	175	15	175	15
Potassium (K)	10	1	10	1
Carbonate (CO ₃)	10	1	10	1
Bicarbonate (HCO ₃)	50	5	50	5
Sulfate (SO ₄)	50	5	50	5
Chloride (Cl)	250	25	250	25
Nitrate (NO ₃)	20	2	20	2
Fluoride (F)	10	1	10	1
Boron (B)	20	2	20	2
Silica (SiO ₂)	20	2	20	2
Total dissolved solids in parts per million	600	60	600	60
Percent sodium	60	10	60	10
Hardness as CaCO ₃ in parts per million				
Total	265	10	265	10
Noncarbonate	15	1	15	1
Turbidity	100	1	100	1
Coliform in most probable number per milliliter	>7,000	<60	>7,000	<60
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.23	0.00	1.23	0.00
Solid alpha	1.57	0.00	1.57	0.00
Dissolved beta	12.5	0.00	12.5	0.00
Solid beta	100	0.00	100	0.00

WATER QUALITY VARIATIONS



ROCK SLOUGH NEAR KNIGHTSEN (STA. 109)

Cosumnes River Basin. The Cosumnes River watershed lies in the central portion of the Central Valley Region. The basin contains approximately 537 square miles, all of which are classified as mountains and foothills. It is bounded by the drainage divide of the Sierra Nevada Range on the east, by the American River drainage on the north, and by the Mokelumne River drainage on the south. The Cosumnes River flows into the Mokelumne River near Thornton. The Cosumnes River at Michigan Bar has a total annual flow of approximately 374,000 acre-feet.

Prominent uses of surface water in the basin include developments devoted to recreation, irrigation, and fish and wildlife propagation and preservation. The basin's natural resources are used primarily for recreational activities such as hunting, fishing, boating, swimming, and picnicking.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Cosumnes River at Michigan Bar	352
Cosumnes River at McConnell	354



COSUMNES RIVER AT MICHIGAN BAR (STA. 94)

Sampling Point Michigan Bar station is located in Section 36, Township 8 North, Range 8 East, Mt. Diablo Base and Meridian. The monthly water samples were collected at mid-channel from the county road bridge, at the USGS stream gaging station, 5.5 miles southwest of Latrobe and about 12 miles downstream from the confluence of North and Middle Forks.

Period of Record July 1952 through December 1959.

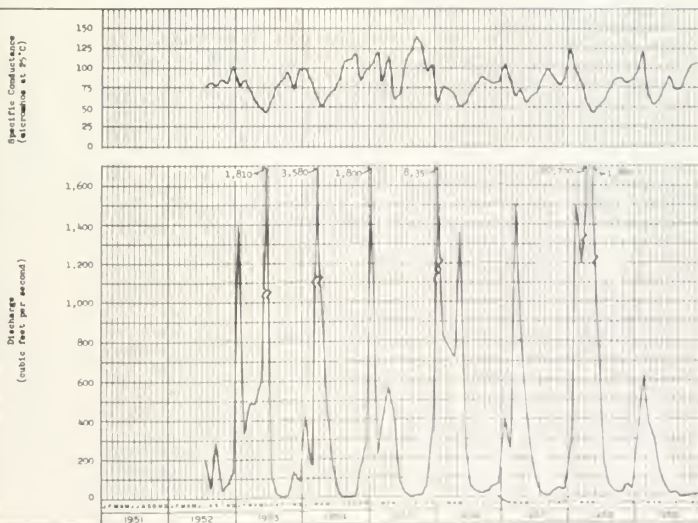
Water Quality Characteristics Past analyses of samples of this water show it to be calcium bicarbonate in character, soft, of excellent mineral quality, and suitable for all beneficial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	139	40.2	103	55.4
Temperature in °F	86	40	86	44
Dissolved oxygen in parts per million	19.9	6.1	19.9	7.7
Percent saturation	130	79	129	80
pH	8.8	6.9	8.1	7.1
Mineral constituents in parts per million				
Calcium (Ca)	9.4	4.2	8.9	7.9
Magnesium (Mg)	6.1	1.5	9.9	1
Sodium (Na)	5.6	1.8	5.1	9.4
Potassium (K)	1.8	0.5	1.4	0.4
Carbonate (CO ₃)	3	0.8	2	0.8
Bicarbonate (HCO ₃)	77	36	55	37
Sulfate (SO ₄)	6.7	0.6	3.7	1
Chloride (Cl)	5.2	0.2	5.2	1.1
Silicate (SiO ₂)	1.2	0.2	0.4	0.1
Fluoride (F)	0.1	0.02	0.08	0.0
Boron (B)	0.24	0.02	0.2	0.0
Silica (SiO ₂)	21	16	17	10
Total dissolved solids in parts per million	98	31	62	4
Percent sodium	27	14	21	10
Hardness as CaCO ₃ in parts per million				
Total	58	18	51	20
Noncarbonate	10	0.8	10	0.8
Turbidity	300	0.2	14	0.6
Coliform in most probable number per milliliter	See 1959	See 1959	2,300	20
Radioactivity in micro-micro curies per liter				
Dissolved alpha			1.20	0.00
Solid alpha			1.00	0.00
Dissolved beta			1.00	0.00
Solid beta			1.00	0.00

WATER QUALITY VARIATIONS



COSUMNES RIVER AT MICHIGAN BAR (STA. 94)

COSUMNES RIVER AT McCONNELL (STA. 94a)

Sampling Point The station is located in Section 20, Township 6 North, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected in mid-channel from U. S. Highway 99 bridge, at the USGS gage, approximately 7.7 miles north of Galt.

Period of Record July 1958 through December 1959.

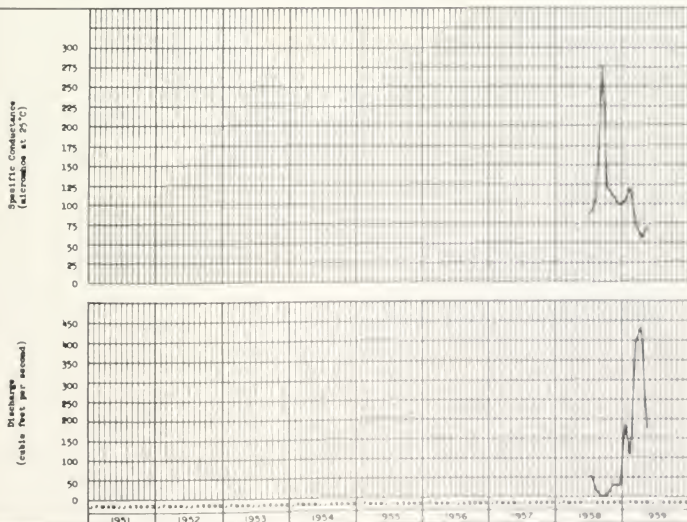
Water Quality Characteristics Past analyses show the water to be similar in quality to the upstream station at Michigan Bar, calcium bicarbonate in character, soft, and of excellent mineral quality for all beneficial uses. Only very minor increases in conductivity were noted between Station 94 at Michigan Bar and Station 94a, indicating no significant sources of degradation.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 17%	Minimum - 17%
Specific conductance (micromhos at 25°C)	271	100	230	83
Temperature in °F	78	47	68	37
Dissolved oxygen in parts per million	11.7	3.7	10.1	3.1
Percent saturation	117	56	101	51
pH	8.1	7.2	7.7	6.9
Mineral constituents in parts per million				
Calcium (Ca)	20	5.5	17	4.7
Magnesium (Mg)	10	3.4	8.7	2.9
Sodium (Na)	10	2.6	8.7	2.2
Potassium (K)	1	0.3	0.8	0.2
Carbonate (CO ₃)	1	0.3	0.8	0.2
Bicarbonate (HCO ₃)	145	27	125	23
Sulfate (SO ₄)	7.1	2.0	6.1	1.7
Chloride (Cl)	6	1.8	5.2	1.5
Nitrate (NO ₃)	1.8	0.5	1.6	0.4
Fluoride (F)	0.1	0.03	0.08	0.02
Boron (B)	0.1	0.03	0.08	0.02
Silica (SiO ₂)	55	16	47	14
Total dissolved solids in parts per million	198	46	171	40
Percent sodium	20	14	18	13
Hardness as CaCO ₃ in parts per million				
Total	21	20	18	19
Noncarbonate	6	5.8	5	5.1
Turbidity %	12	0.5	10	0.4
Coliforms in most probable number per milliliter	230	21	200	18
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.50	0.06	0.43	0.05
Solid alpha	1.26	0.09	1.08	0.07
Dissolved beta	3.07	0.30	2.67	0.26
Solid beta	7.75	2.13	6.68	1.83

WATER QUALITY VARIATIONS



COSUMNES RIVER AT McCONNELL (STA. 94a)

Mokelumne River Basin. The Mokelumne River watershed lies in the central portion of the Central Valley Region. It contains about 630 square miles, approximately 626 of which are mountains and foothills. The remainder are valley and mesa lands. The river drains a portion of the western slope of the Sierra Nevada. It is bordered by the Cosumnes River drainage on the north, and Calaveras River drainage on the south. The Mokelumne River enters the delta near Thornton. At Clements the Mokelumne River has a total annual flow of approximately 780,000 acre-feet.

Approximately four square miles of the Mokelumne River drainage basin are potential agricultural lands. The most prominent uses of surface water in this basin are for recreation, power development, fish and wildlife propagation and preservation, and export by Mokelumne Aqueduct for municipal use by the East Bay Municipal Utility District. Natural resources of the basin are utilized for recreational activities such as hunting, fishing, boating, swimming and picnicking.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Mokelumne River near Lancha Plana	358
Mokelumne River at Woodbridge	360



MOKELUMNE RIVER NEAR LANCHA PLANA (STA. 23a)

Sampling Point Station 23a is located in Section 4, Township 4 North, Range 10 East, Mt. Diablo Base and Meridian. The monthly water samples were collected from the left bank, about 1.0 mile east of Lancha Plana, 3.0 miles downstream from Pardee Dam, and 5.0 miles upstream from Camanche Creek.

Period of Record April 1951 through December 1959.

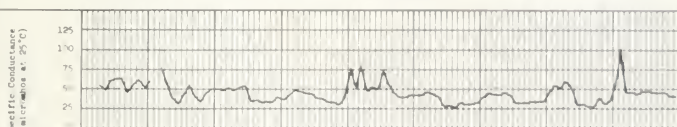
Water Quality Characteristics Antecedent data reveal the water to be predominantly calcium bicarbonate in character, soft, of excellent mineral quality, and suitable for all beneficial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	108	26.6	108	19.5
Temperature in °F	67	41	68	48
Dissolved oxygen in parts per million	13.6	8.8	11.1	9.4
Percent saturation	136	89	128	88
pH	7.5	6.2	7.2	6.1
Mineral constituents in parts per million				
Calcium (Ca)	7.2	2.8	5.6	4.8
Magnesium (Mg)	2.9	0.2	2.1	0.3
Sodium (Na)	5.2	0.7	1.6	1.7
Potassium (K)	1.7	0.2	1.1	0.2
Carbonate (CO ₃)	8.5	5.0	8.0	4.0
Bicarbonate (HCO ₃)	11	10	10	10
Sulfate (SO ₄)	9.6	1.1	9.6	1.1
Chloride (Cl)	6	1	4	1
Nitrate (NO ₃)	0.7	0.2	0.1	0.2
Fluoride (F)	0.2	0.2	0.1	0.2
Boron (B)	0.35	0.2	0.1	0.2
Silica (SiO ₂)	16	6.5	11	8
Total dissolved solids in parts per million	81	21	81	11
Percent sodium	44	14	27	17
Hardness as CaCO ₃ in parts per million				
Total	38	9	38	14
Noncarbonate	26	1	26	0.5
Turbidity	90	0.1	30	0.1
Coliform in most probable number per milliliter	7,000	100	500	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.20	0.20	0.20	0.20
Solid alpha	1.17	0.00	0.44	0.00
Dissolved beta	0.49	0.30	0.40	0.30
Solid beta	1.52	0.30	0.36	0.25

WATER QUALITY VARIATIONS



MOKELUMNE RIVER NEAR LANCHA PLANA (STA. 23a)

MOKELUMNE RIVER AT WOODBRIDGE (STA. 23)

Sampling Point Station 23 is located in Section 34, Township 4 North, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank at a USGS gaging station about 0.4 mile downstream from the Woodbridge Irrigation District dam.

Period of Record April 1951 through December 1959.

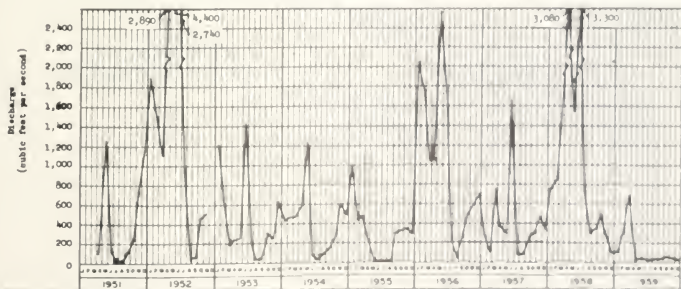
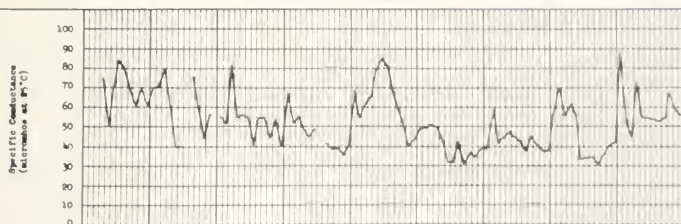
Water Quality Characteristics Past analyses show the water to be predominantly calcium bicarbonate in character, soft, and of excellent mineral quality suitable for all beneficial uses. Only a slight increase in the concentration of mineral constituents has been noted between the upstream Station 23a at Lancha Plana and Station 23.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	87.1	30.2	87.1	45.8
Temperature in °F	74	44	74	53
Dissolved oxygen in parts per million	11.0	7.9	11.6	8.2
Percent saturation	14	75	100	80
pH	7.8	6.1	7.1	6.4
Mineral constituents in parts per million				
Calcium (Ca)	8.4	2.7	7.8	6.4
Magnesium (Mg)	4.4	0.5	2.1	2.2
Sodium (Na)	4.9	1.4	4.5	2.2
Potassium (K)	0.6	0.4	0.0	0.8
Carbonate (CO ₃)	0.2	0.2	0.0	0.0
Bicarbonate (HCO ₃)	37	1.8	30	17
Sulfate (SO ₄)	11	1.0	11	1.0
Chloride (Cl)	6.0	0.0	4.5	2.0
Nitrate (NO ₃)	2.4	0.0	2.4	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.22	0.0	0.2	0.0
Silica (SiO ₂)	1.0	0.8	1.4	1.2
Total dissolved solids in parts per million	71	20	55	31
Percent sodium	36	16	28	16
Hardness as CaCO ₃ in parts per million				
Total	32	9	12	16
Noncarbonate	14	0.0	14	0.0
Turbidity	70	0.0	20	0.8
Coliforms in most probable number per milliliter	>7,000	0.62	2,400	0.62
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.41	0.00	0.41	0.00
Solid alpha	1.10	0.00	1.10	0.41
Dissolved beta	25.5	0.00	6.41	4.75
Solid beta	15.76	1.00	2.46	0.30

WATER QUALITY VARIATIONS



MOKELUMNE RIVER AT WOODBRIDGE (STA. 23)

Calaveras River Basin. The Calaveras River watershed contains approximately 395 square miles in the central portion of the Central Valley Region. The basin drains the mountainous and foothill terrain along the western slopes of the Sierra Nevada.

The Calaveras River parallels the course of the Mokelumne and Stanislaus Rivers, whose basins border it on the north and south, respectively, and flows westward into the San Joaquin River below Stockton. Total mean annual runoff, measured at Jenny Lind, has been approximately 199,000 acre-feet.

Very unproductive top soil, coupled with a relatively rugged topography, have limited development in the basin. Mining, livestock raising, and lumbering operations are carried on to a minor degree. Recreational activities have increased in recent years and are playing an increasingly important role in the economy of the basin. The most prominent uses of surface water are for recreation and irrigation diversion.

Waste discharges entering the waterways of this basin are small in volume and have not caused any impairment problems.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Calaveras River at Jenny Lind	364
Calaveras River near Stockton	366



CALAVERAS RIVER AT JENNY LIND (STA. 16a)

Sampling Point Station 16a is located in Section 27, Township 3 North, Range 10 East, Mt. Diablo Base and Meridian. Samples were collected from the right bank, about 225 feet downstream from Milton Road bridge, and about 0.2 mile south of Jenny Lind.

Period of Record April 1951 through December 1959.

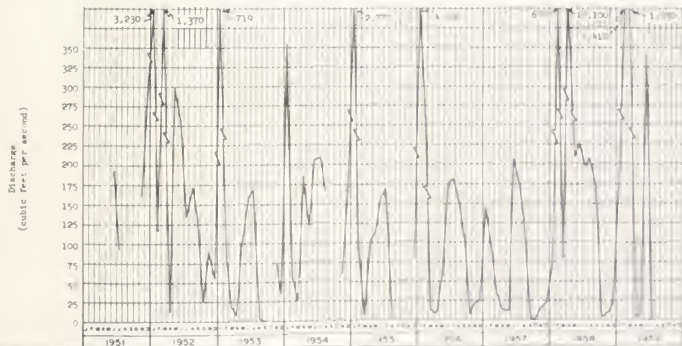
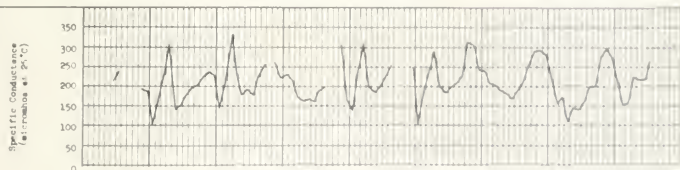
Water Quality Characteristics Past analyses show the water to be predominantly a calcium bicarbonate type, slightly to moderately hard, and class 1 for irrigation. Calaveras River water at Station 16a meets drinking water standards for mineral content and is suitable for most industrial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1951	Minimum - 1951
Specific conductance (micromhos at 25°C)	132	80	88	10
Temperature in °F	89	48	82	55
Dissolved oxygen in parts per million	11	1.6	10.9	1.4
Percent saturation	124	84	99	80
pH	8.1	7.2	7.7	7.2
Mineral constituents in parts per million				
Calcium (Ca)	10	1.8	28	1.8
Magnesium (Mg)	2	2.4	2.4	2.4
Sodium (Na)	12	2.4	8.9	4.5
Potassium (K)	8	1.2	7.4	1.2
Carbonate (CO ₃)	7	1.2	7.4	1.2
Bicarbonate (HCO ₃)	18	1.2	16.8	1.2
Sulfate (SO ₄)	15	1.2	13.8	1.2
Chloride (Cl)	14	1.2	12.8	1.2
Nitrate (NO ₃)	2.2	1.2	1.2	1.2
Fluoride (F)	1.5	1.2	1.2	1.2
Boron (B)	20	1.2	18.8	1.2
Silica (SiO ₂)	21	1.2	19.8	1.2
Total dissolved solids in parts per million	204	42	162	38
Percent sodium	18	1.2	16	1.2
Hardness as CaCO ₃ in parts per million				
Total	152	42	110	48
Noncarbonate	67	1.2	65.8	1.2
Turbidity	160	0.2	158	0.2
Coliform in most probable number per milliliter	7,300	1.2	7,288	1.2
Radioactivity in micro-micro curies per liter				
Dissolved alpha	4.8	1.2	3.6	1.2
Solid alpha	1.08	1.2	0.88	1.2
Dissolved beta	8.28	1.2	7.08	1.2
Solid beta	11.5	1.2	10.3	1.2

WATER QUALITY VARIATIONS



CALAVERAS RIVER AT JENNY LIND (STA 16a)

CALAVERAS RIVER NEAR STOCKTON (STA. 16b)

Sampling Point The Stockton station is located in Section 26, Township 2 North, Range 6 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected in mid-channel from West Lane bridge.

Period of Record July 1958 through December 1959.

Water Quality Characteristics Samples of Calaveras River water near Stockton are predominantly calcium bicarbonate, slightly hard to moderately hard and of excellent mineral quality for irrigation. No significant difference is noted in the water quality at Station 16a at Jenny Lind and Station 16b.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1951	Minimum - 1951
Specific conductance (micromhos at 25°C)	254	124	214	
Temperature in °F	88	47	71	
Dissolved oxygen in parts per million	12.4	4.0	5.1	
Percent saturation	124	40	51	
pH	8.1	7.4	7.4	
Mineral constituents in parts per million				
Calcium (Ca)	26	20	26	
Magnesium (Mg)	8.8	0.0	8.8	
Sodium (Na)	7.0	1.1	7.2	
Potassium (K)	1.0	0.0	1.0	
Carbonate (CO ₃)	2.0	0.0	2.0	
Bicarbonate (HCO ₃)	18	0	18	
Sulfate (SO ₄)	18	0	18	
Chloride (Cl)	8.2	0.0	8.2	
Nitrate (NO ₃)	0.7	0.2	0.7	
Fluoride (F)	0.1	0.0	0.1	
Boron (B)	0.0	0.0	0.0	
Silica (SiO ₂)	20	10	11	
Total dissolved solids in parts per million	177	114	177	
Percent sodium	14	11	14	
Hardness as CaCO ₃ in parts per million				
Total	183	78	183	
Noncarbonate	1	0	1	
Turbidity	1			
Coliform in most probable number per milliliter	See 1959	See 1956	60	11
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.24			
Solid alpha	0.08			
Dissolved beta	2.51			
Solid beta	3.50			

WATER QUALITY VARIATIONS

INSUFFICIENT FLOW TO OBTAIN DATA
FOR GRAPHICAL ILLUSTRATION

1951 1952 1953 1954 1955 1956 1957 1958 1959

CALAVERAS RIVER NEAR STOCKTON (STA. 16b)

Tulare Lake Drainage (5d)

Tulare Lake Drainage includes the southern third of the Great Central Valley and the mountainous drainage areas coterminous to the valley on three sides. The drainage area extends approximately 130 miles southerly from the San Joaquin River to the drainage divide of the Tehachapi Mountains. Average width of the basin is 120 miles with the eastern boundary defined by the crest line of the southern portion of the Sierra Nevada, and the western boundary by the drainage divide along the coastal ranges. The Tulare Lake Drainage encompasses an area of 16,518 square miles, of which 7,773 square miles is valley and mesa and 8,745 square miles are mountains and foothills.

The valley portion of Tulare Lake Drainage consists of relatively flat-bottomed terrain bordered on three sides by gently sloping alluvial fans. Lowlands of the valley floor range in elevation from 200 feet above sea level at Tulare Lake to 500 feet along the southern end. The valley floor is broken by several ridges, such as Kettlemen Hills and Elk Hills, which have crest elevations of over 1,000 feet. Stream systems in this basin are tributary to evaporation sumps in the trough of the valley, chiefly Tulare and Buena Vista Lake beds. In the past, however, during years of heavy floods the low divide between Buena Vista and Tulare Lakes and between Tulare Lake and the San Joaquin River drainage were overtopped. During such periods, surface runoff flowed out of the Tulare Lake Drainage into the San Joaquin River.

Mountainous terrain bounding the three sides of the valley area rise from the valley floor as gently rolling foothills grading upwards to a rugged mountainous terrain. The Sierra Nevada Range on the

east dominated by Mt. Whitney rises to altitudes greater than 14,000 feet. The Coast Ranges to the west rise to 6,000 feet, to the south the valley is enclosed by the coastal and Tehachapi Mountains, which rise to altitudes of about 8,000 feet.

Natural mean seasonal surface runoff for the basin is estimated to be 3,310,000 acre-feet. The principal hydrographic units are the Kings, Kern, Kaweah, and Tule Rivers, all originating in the Sierra Nevada Range. Flows are sustained by the Sierra Nevada seasonal snowpack. No streams of importance enter the valley area from the Coast Ranges or the Tehachapi Mountains. Monitored streams with the number of stations in parentheses are as follows:

- Kings River Basin (3)
- Kaweah River Basin (1)
- Tule River Basin (1)
- Kern River Basin (3)

Kings River Basin. The Kings River Basin is located in the Sierra Nevada in Fresno and Tulare Counties. The basin contains an area of 7,163 square miles, classified as mountainous foothill terrain, with 162 miles classified as valley and mesa land. It is bounded on the north by the San Joaquin River drainage divide, on the east by the Sierra crest line, on the south by the Kaweah River drainage basin, and by Tulare Lake bed on the west. During high flows a portion of the overflow from Kings River is tributary to the San Joaquin River via Fresno Slough.

The Kings River originates near the Sierra crest line at an altitude in excess of 10,000 feet. From their headwaters these streams flow eastward through Kings Canyon National Park where terrain is extremely rugged and mountainous with deeply entrenched, steep walled canyons. Mountainous area slowly gives way to a moderately rugged foothill terrain at Piedra. Kings River flows into the San Joaquin Valley at an elevation of 500 feet above sea level and terminates in Tulare Lake at an elevation of 200 feet above sea level. Total average annual runoff in the Kings River is 1,715,000 acre-feet.

In the upper reaches of the Kings River (Kings Canyon National Park) development is primarily limited to recreation. Lumbering, ranching, recreation, and hydroelectric power developments are the chief industries between the park and the base of the foothills.

Waste discharges entering the waterways of Kings River Basin above the foothill line are negligible. Impairment of quality of runoff by these waste discharges has not been serious and has not caused a discernible problem.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Kings River below North Fork	372
Kings River below Pine Flat Dam	374
Kings River below Peoples Weir	376

KINGS RIVER BELOW NORTH FORK (STA. 33c)

Sampling Point Station 33c on Kings River is located in Section 21, Township 12 South, Range 26 East, Mt. Diablo Base and Meridian.

Monthly grab samples were collected at mid-stream, from the highway bridge located 0.8 mile downstream from the North Fork confluence.

Period of Record September 1955 through December 1959.

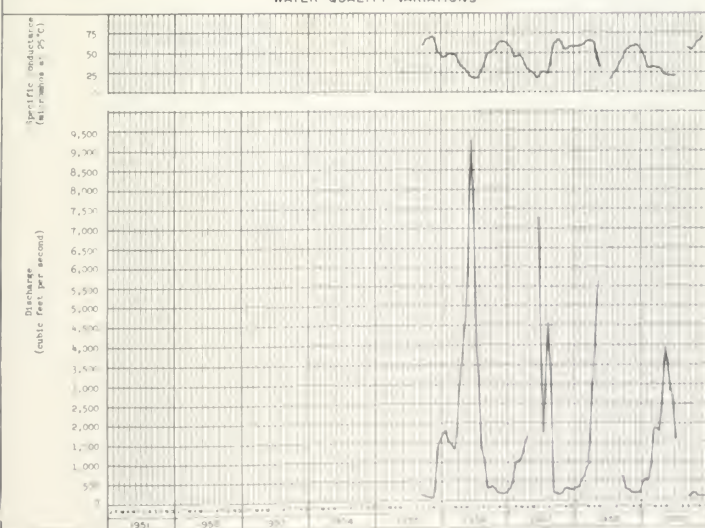
Water Quality Characteristics Antecedent data show Kings River at Station 33c to be characterized by nearly equivalent calcium, magnesium and sodium cations. Bicarbonate is the predominant anion. The mineral quality of the water is excellent, class 1 for irrigation, suitable for drinking water, and soft with a maximum recorded hardness of 39 ppm. Quality of Kings River at this station is representative of the major portion of inflow to Pine Flat Reservoir.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1973	Minimum - 1973
Specific conductance (micromhos at 25°C)	71.7	25.3	71.6	25.2
Temperature in °F	73	40	71	40
Dissolved oxygen in parts per million	12.0	1.0	12.0	1.0
Percent saturation	100	20	100	20
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	80	10	80	10
Magnesium (Mg)	10	1	10	1
Sodium (Na)	20	1	20	1
Potassium (K)	10	1	10	1
Carbonate (CO ₃)	10	1	10	1
Bicarbonate (HCO ₃)	10	1	10	1
Sulfate (SO ₄)	10	1	10	1
Chloride (Cl)	10	1	10	1
Nitrate (NO ₃)	10	1	10	1
Fluoride (F)	10	1	10	1
Boron (B)	10	1	10	1
Silica (SiO ₂)	10	1	10	1
Total dissolved solids in parts per million	100	20	100	20
Percent sodium	10	1	10	1
Hardness as CaCO ₃ in parts per million				
Total	20	1	20	1
Noncarbonate	10	1	10	1
Turbidity	1.0	0.1	1.0	0.1
Coliform in most probable number per milliliter	7,000	100	7,000	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.1	0.01	0.1	0.01
Solid alpha	0.07	0.01	0.07	0.01
Dissolved beta	0.07	0.01	0.07	0.01
Solid beta	0.07	0.01	0.07	0.01

WATER QUALITY VARIATIONS



KINGS RIVER BELOW PINE FLAT DAM (STA. 33b)

Sampling Point Pine Flat Dam station is located in Section 2, Township 13 South, Range 24 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the bridge located about 3,000 feet downstream from Pine Flat Dam.

Period of Record September 1955 through December 1959.

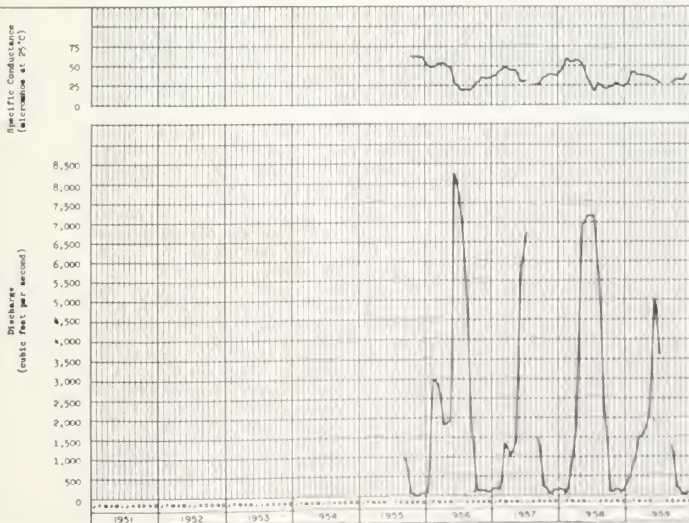
Water Quality Characteristics Water from Kings River below Pine Flat Dam is calcium bicarbonate or occasionally a calcium-sodium bicarbonate type. The water is class 1 for irrigation, meets the criteria for domestic use, and is soft (maximum recorded hardness of 24 ppm). Mineral quality at this station is qualitatively similar to that at Station 33c (Kings River below North Fork) located about 25 miles upstream.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1957	Minimum 1958
Specific conductance (micromhos at 25°C)	18.5	8.8	15.7	8.0
Temperature in °F	70	40	58	30
Dissolved oxygen in parts per million	18.0	7.8	13.0	5.0
Percent saturation	100	80	100	75
pH	9.6	6.7	8.1	6.1
Mineral constituents in parts per million				
Calcium (Ca)	8.8	3.0	4.5	3.0
Magnesium (Mg)	4.8	1.0	2.5	1.0
Sodium (Na)	4.8	1.0	2.5	1.0
Potassium (K)	2.1	0.5	1.0	0.5
Carbonate (CO ₃)	1.0	0.5	0.5	0.5
Bicarbonate (HCO ₃)	1.0	0.5	0.5	0.5
Sulfate (SO ₄)	1.0	0.5	0.5	0.5
Chloride (Cl)	1.0	0.5	0.5	0.5
Nitrate (NO ₃)	1.0	0.5	0.5	0.5
Fluoride (F)	0.2	0.1	0.1	0.1
Bromine (Br)	0.1	0.0	0.0	0.0
Silica (SiO ₂)	1.4	0.5	0.5	0.5
Total dissolved solids in parts per million	40	14	12	20
Percent sodium	10	10	27	20
Hardness as CaCO ₃ in parts per million				
Total	24	6	10	30
Noncarbonate	1.1	0.5	0.5	0.5
Turbidity	2.8	0.2	1.0	1
Coliform in most probable number per milliliter	7,000	100	7,000	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.53	0.00	0.41	0.00
Solid alpha	1.00	0.00	0.00	0.00
Dissolved beta	7.52	0.00	7.52	4.77
Solid beta	6.68	3.29	6.68	6.11

WATER QUALITY VARIATIONS



KINGS RIVER BELOW PINE FLAT DAM (STA. 33b)

KINGS RIVER BELOW PEOPLES WEIR (STA. 34)

Sampling Point Station 34 is situated in Section 1, Township 17 South, Range 22 East, Mt. Diablo Base and Meridian. The point of monthly grab sample collection is from the left bank, at the stream gage located about 1/4 mile downstream from the diversion weir, approximately 2 miles south of Kingsburg, 12 miles northeast of Hanford.

Period of Record April 1951 through December 1959.

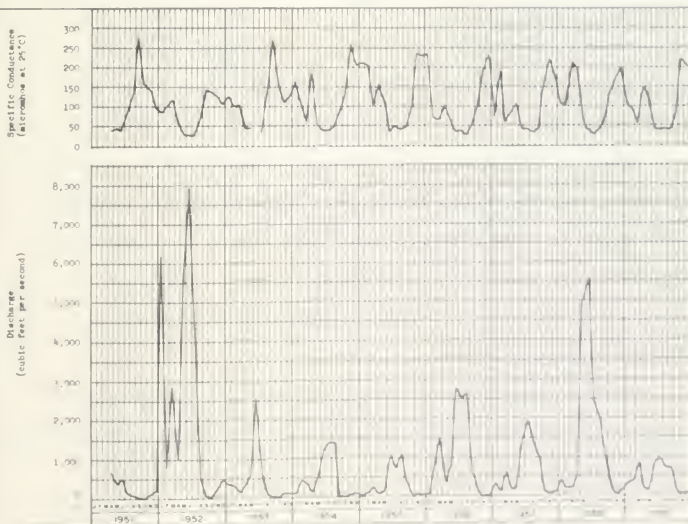
Water Quality Characteristics Water from Station 34 has bicarbonate as the predominant anion with no specific cation predominating. The water is excellent, ranges from soft to moderately hard, and meets class 1 irrigation criteria and mineral standards for drinking water. Concentrations of mineral constituents in Kings River at Station 34 are considerably higher (over 100 percent or averaging approximately 60 micromhos) than at the upstream Station 33b below Pine Flat Dam.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1951	Minimum 1952
Specific conductance (micromhos at 75°C)	76	27	65	25
Temperature in °F	86	41	75	35
Dissolved oxygen in parts per million	12.8	1.6	12.4	2.4
Percent saturation	121	21	118	24
pH	8.6	7.8	7.7	7.7
Mineral constituents in parts per million				
Calcium (Ca)	76	11	74	11
Magnesium (Mg)	1	1	1	1
Sodium (Na)	24	1	1	1
Potassium (K)	8	1	1	1
Carbonate (CO ₃)	76	1	74	1
Bicarbonate (HCO ₃)	110	1	110	1
Sulfate (SO ₄)	14	1	14	1
Chloride (Cl)	46	1	46	1
Nitrate (NO ₃)	6.2	1	6.2	1
Fluoride (F)	1	1	1	1
Boron (B)	12	1	12	1
Silica (SiO ₂)	26	1	26	1
Total dissolved solids in parts per million	193	14	192	24
Percent sodium	41	19	41	19
Hardness as CaCO ₃ in parts per million				
Total	97	1	96	1
Noncarbonate	6	1	6	1
Turbidity	11	1	11	1
Coliform in most probable number per milliliter	>7	1	7	1
Radioactivity in micro-micro curies per liter				
Dissolved alpha	168	1	168	1
Solid alpha	168	1	168	1
Dissolved beta	168	1	168	1
Solid beta	168	1	168	1

WATER QUALITY VARIATIONS



KINGS RIVER BELOW PEOPLES WEIR STA. 34

Kaweah River Basin. The Kaweah River Basin is located in the Sierra Nevada east of Visalia and extends from Sequoia National Park to Three Rivers in Tulare County. The basin has an area of 520 square miles of mountainous and hilly terrain. It is bounded on the north and north-east by the Kings River watershed, on the east and southeast by the Kern River drainage divide, and on the south by the Tule River drainage area. Kaweah River flows into the San Joaquin Valley at Lemon Cove where the channel splits into several distributaries which eventually drain into Tulare Lake evaporation sump.

Forks of the Kaweah River head in an extremely rugged, mountainous area with alpine peaks rising above 10,000 feet. Steep walled canyons and ravines are characteristic of the waterways in the upper reaches. Progressing downslope the topography undergoes a gradual transition to rolling foothills and broader river valleys. The Kaweah River flows out of the hydrographic unit at Three Rivers at an elevation of 800 feet above sea level. Total average annual runoff of the Kaweah River Basin is 416,000 acre-feet.

Economic activities in the Kaweah River Basin consist primarily of recreation, ranching, hydroelectric power development, and lumbering. Approximately seven miles downstream from Three Rivers, near Lemon Cove, Terminus Dam is presently under construction by the U. S. Corps of Engineers. The structure will provide flood control, irrigation, and other benefits to nearby areas.

Numerous domestic wastes discharge into the waterways of this basin, however, these are comparatively minor and have created no noticeable impairment problems.

A surface water sampling station is maintained on Kaweah River near Three Rivers to monitor quality of runoff from the basin.



KAWEAH RIVER NEAR THREE RIVERS (STA. 35)

Sampling Point Kaweah River water is sampled in Section 33, Township 17 South, Range 28 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, at the USGS gaging station about 2.5 miles downstream from the South Fork confluence, 3 miles southeast of Three Rivers, approximately 1/2 mile east of Cobbles Lodge.

Period of Record April 1951 through December 1959.

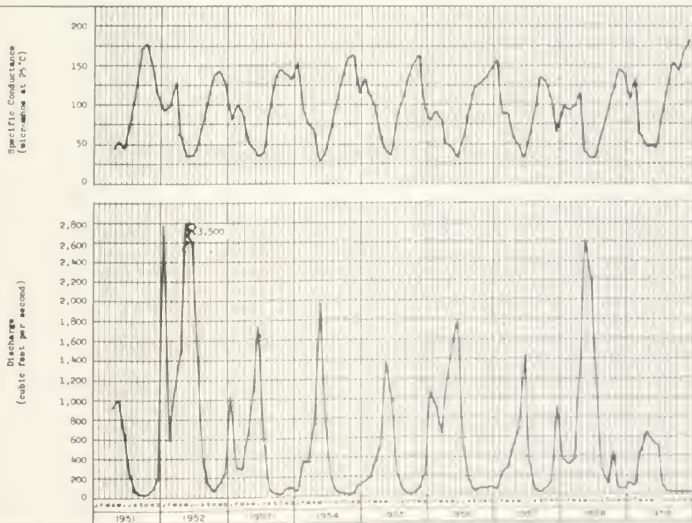
Water Quality Characteristics Water in Kaweah River at Station 35 is a calcium bicarbonate type; however, at times no cation is predominant. The water has been of excellent mineral quality, soft to slightly hard, class 1 for irrigation (with one exception) and meets drinking water criteria. In December 1953, a boron concentration of 0.56 ppm was recorded, which exceeds class 1 irrigation standards.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Range	Minimum of Range	Maximum of Range	Minimum of Range
Specific conductance (micromhos at 25°C)	300	30	300	30
Temperature in °F	80	30	80	30
Dissolved oxygen in parts per million	11	0.5	11	0.5
Percent saturation	125	50	125	50
pH	8.5	6.5	8.5	6.5
Mineral constituents in parts per million				
Calcium (Ca)	40	1.5	40	1.5
Magnesium (Mg)	10	0.5	10	0.5
Sodium (Na)	10	0.5	10	0.5
Potassium (K)	10	0.5	10	0.5
Carbonate (CO ₃)	10	0.5	10	0.5
Bicarbonate (HCO ₃)	10	0.5	10	0.5
Sulfate (SO ₄)	10	0.5	10	0.5
Chloride (Cl)	10	0.5	10	0.5
Nitrate (NO ₃)	10	0.5	10	0.5
Fluoride (F)	10	0.5	10	0.5
Boron (B)	10	0.5	10	0.5
Silica (SiO ₂)	27	1.5	27	1.5
Total dissolved solids in parts per million	122	20	122	20
Percent sodium	43	10	43	10
Hardness as CaCO ₃ in parts per million				
Total	68	10	68	10
Noncarbonate	9	0.5	9	0.5
Turbidity	100	0.5	100	0.5
Coliform in most probable number per milliliter	7,000	100	7,000	100
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.04	0.005	0.04	0.005
Solid alpha	0.2	0.005	0.2	0.005
Dissolved beta	20.5	0.005	20.5	0.005
Solid beta	21.7	0.005	21.7	0.005

WATER QUALITY VARIATIONS



Tule River Basin. The Tule River Basin is located on the western slopes of the Sierra Nevada in the southern part of the Central Valley Region. The basin extends from the southern boundary of Sequoia National Park to Porterville in Tulare County and contains an area of 390 square miles. It is bounded on the north by the Kaweah River drainage divide, on the east by the Kern River watershed crest line, on the south by a drainage divide of Deer Creek, and on the west by the San Joaquin Valley. Tule River flows westward into the San Joaquin Valley and terminates at Tulare Lake sump.

The terrain along the upper reaches of the Tule River Basin is extremely rugged cut by steep-walled canyons and ravines, with mountainous ridges rising to altitudes greater than 7,000 feet. Progressing toward the San Joaquin Valley, the topography undergoes a gradual change to foothills interspersed by relatively broad river valleys. The Tule River flows out of the Sierra foothills, at an elevation of about 500 feet, into the San Joaquin Valley at Porterville. The average annual runoff of the Tule River is 140,000 acre-feet.

Activities in Tule River Basin include recreation, ranching, hydroelectric power development and limited lumbering and orcharding. The newly constructed Success Dam on the Tule River, four miles east of Porterville, provides flood control and other benefits to nearby areas.

Waste discharges are relatively minor in quantity and have not created any deleterious effects on quality of water in the basin.

A surface water sampling station is maintained on Tule River near Porterville to monitor quality of runoff from this basin.



TULE RIVER NEAR PORTERVILLE (STA. 91)

Sampling Point Prior to September 1959, Station 91 was located in Section 25, Township 21 South, Range 28 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected at mid-stream from the county road bridge, 0.1 mile downstream from the South Fork, 8.8 miles east of Porterville. In September 1959, it was necessary to move the station due to construction of Success Dam. The new location is in Section 3, Township 22 South, Range 28 East, Mt. Diablo Base and Meridian. Monthly water samples were collected at mid-stream, from Worth Bridge, about 3 miles downstream from the location described for the former sampling station.

Period of Record July 1952 through December 1959.

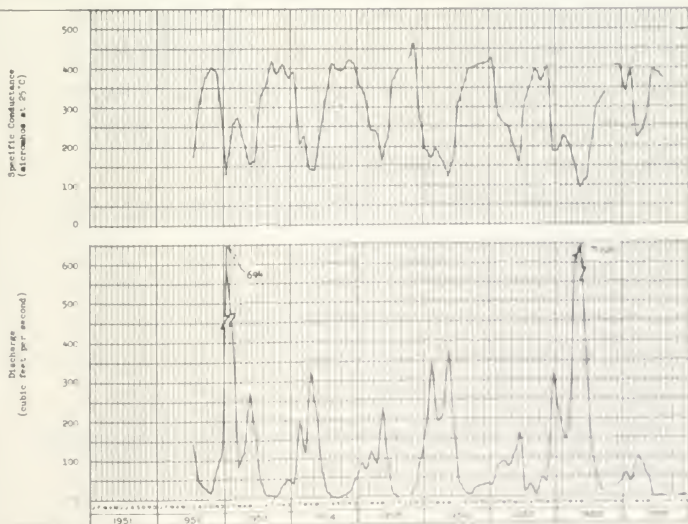
Water Quality Characteristics Tule River at Station 91 generally exhibits a calcium bicarbonate characteristic, although a calcium-sodium bicarbonate type water has been recorded on several occasions. Mineral quality of the water is excellent and meets class 1 irrigation requirements and drinking water standards. Tule River water ranges from soft to very hard with the hardness attributable to the natural leaching of soils and rocks of the watershed.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	524	21	124	224
Temperature in °F	80	41	70	41
Dissolved oxygen in parts per million	11.4	8.1	10.3	8.1
Percent saturation	127	70	127	70
pH	8.4	7.0	8	7.0
Mineral constituents in parts per million				
Calcium (Ca)	118	12	11	
Magnesium (Mg)	20	1.2	11	
Sodium (Na)	81	4.4	11	18
Potassium (K)	5	1	2	
Carbonate (CO ₃)	1		2	10.4
Bicarbonate (HCO ₃)	284	41	284	184
Sulfate (SO ₄)	7.7	1.0	1.8	
Chloride (Cl)	20	2.0	10	8.0
Nitrate (NO ₃)	4.6	0.0	4.6	
Fluoride (F)	0.2	0	0.2	
Boron (B)	0.22	0.03	0.2	0.03
Silica (SiO ₂)	15	1.6	27	
Total dissolved solids in parts per million	324	66	324	147
Percent sodium	27	18	20	18
Hardness as CaCO ₃ in parts per million				
Total	324	15	284	18
Noncarbonate	118	1.2	11	1.2
Turbidity	40	1.0	35	1.0
Coliform in most probable number per milliliter	See 1959	See 1959	120	10.18
Radioactivity in micro-micro curies per liter				
Dissolved alpha				
Solid alpha				
Dissolved beta				
Solid beta				

WATER QUALITY VARIATIONS



Kern River Basin. The Kern River Basin is located on the southwestern slope of the Sierra Nevada, and extends southwesterly from Sequoia National Park near Mt. Whitney to Bakersfield in Tulare and Kern Counties. The basin includes an area of 2,420 square miles, almost all in mountainous and hilly topography. It is bounded on the northwest by watersheds of the Kaweah and Tule Rivers and other minor streams draining into San Joaquin Valley, on the east and southeast by the Sierra Nevada crest line dominated by Mt. Whitney, and on the south by the drainage divide of minor intermittent streams draining into the San Joaquin Valley. Emerging from the foothills and into the valley area at Bakersfield, Kern River flows down a gently sloping alluvial fan to Buena Vista Lake.

Above the confluence of North Fork and South Fork Kern River, at Isabella Reservoir, the watershed is extremely rugged, rising to altitudes of about 13,000 feet. Deep, steep-walled canyons have been carved into the mountainous terrain by the Kern River. Below Isabella Reservoir the topography is moderately rugged, grading to rolling foothills toward the edge of the San Joaquin Valley. Total average annual runoff in the Kern River is about 736,000 acre-feet.

The headwater area of the Kern River in Sequoia National Park is generally inaccessible, and hence, development consists of limited recreation. Below the park, in the foothills, development consists chiefly of lumbering, ranching, hydroelectric power development, and recreation. Farming and crude oil production is the chief industry in the valley area. Isabella Dam, located about 22 miles northeast of Bakersfield on the Kern River, provides flood control and other benefits to the basin.

Waste discharges in the watershed are extremely small above the valley floor and have caused no impairment problems.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Kern River near Kernville	388
Kern River below Isabella Dam	390
Kern River near Bakersfield	392

KERN RIVER NEAR KERNVILLE (STA. 36b)

Sampling Point Kernville station is located in Section 14, Township 23 South, Range 32 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, at the USGS stream gaging station, about 3 miles upstream from the confluence with Salmon Creek, 15 miles north of Kernville.

Period of Record September 1955 through December 1959.

Water Quality Characteristics Kern River at Station 36b is characterized by a calcium-sodium bicarbonate type water, which is soft and relatively low in dissolved solids. The mineral quality of this water consistently meets the criteria for a class 1 irrigation supply and for drinking water.

Significant Water Quality Changes During 1959 the only significant quality change occurred in July when a boron concentration of 0.8 ppm rendered water at this station class 2 for irrigation. The reason for this relatively high boron concentration has not yet been ascertained.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum 1977	Minimum 1977
Specific conductance (microhos at 25°C)	140	30	130	30
Temperature in °F	88	36	84	43
Dissolved oxygen in parts per million	12.8	2.0	12.1	1.0
Percent saturation	100	20	100	10
pH	10.2	7.0	9.0	7.0
Mineral constituents in parts per million				
Calcium (Ca)	100	10	110	10
Magnesium (Mg)	10	1	10	1
Sodium (Na)	100	10	110	10
Potassium (K)	10	1	10	1
Carbonate (CO ₃)	100	10	110	10
Bicarbonate (HCO ₃)	100	10	110	10
Sulfate (SO ₄)	100	10	110	10
Chloride (Cl)	100	10	110	10
Nitrate (NO ₃)	100	10	110	10
Fluoride (F)	100	10	110	10
Boron (B)	100	10	110	10
Silica (SiO ₂)	100	10	110	10
Total dissolved solids in parts per million	27	20	127	20
Percent sodium	75	20	80	20
Hardness as CaCO ₃ in parts per million				
Total	100	10	110	10
Noncarbonate	100	10	110	10
Turbidity	10	1	10	1
Coliforms in most probable number per milliliter	7,000	4	7,000	4
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.0	0.1	1.0	0.1
Solid alpha	1.0	0.1	1.0	0.1
Dissolved beta	2.7	0.1	2.7	0.1
Solid beta	9.5	0.1	9.5	0.1

WATER QUALITY VARIATIONS



KERN RIVER BELOW ISABELLA DAM (STA. 36a)

Sampling Point Station 36a is located in Section 30, Township 26 South, Range 33 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the right bank, 500 feet downstream from the outfall tunnel of Isabella Dam.

Period of Record September 1955 through December 1959.

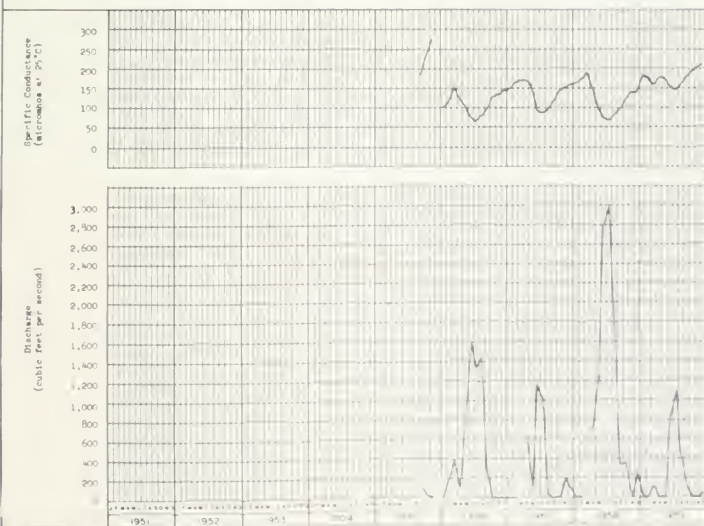
Water Quality Characteristics Water at Isabella Dam station is consistently a bicarbonate type with either calcium or calcium-sodium cations being predominant. Mineral quality of the water is excellent, soft to slightly hard, meets the criteria for class 1 irrigation use and drinking water standards. Comparison of analyses of samples from Kern River at Station 36b with those from Station 36c, located about 10 miles upstream and above Isabella Reservoir, show that in this reach there is an increase in specific conductance of from 10 to 92 micromhos. The reason for this increase has not as yet been ascertained; but it is probably attributable to the concentration of minerals caused by evaporation from Isabella Reservoir.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - L.P.	Minimum - L.P.
Specific conductance (microhm/cm at 25°C)	271	80.4	61.1	148
Temperature in °F	78	61	78	61
Dissolved oxygen in parts per million	10.8	7.7	10.8	7.7
Percent saturation	100	97	100	97
pH	7.8	6.8	7.8	6.8
Mineral constituents in parts per million				
Calcium (Ca)	98	13	18	13
Magnesium (Mg)	18	1.2	18	1.2
Sodium (Na)	18	1.2	18	1.2
Potassium (K)	4	0.2	4	0.2
Carbonate (CO ₃)	18	1.2	18	1.2
Bicarbonate (HCO ₃)	18	1.2	18	1.2
Sulfate (SO ₄)	18	1.2	18	1.2
Chloride (Cl)	18	1.2	18	1.2
Nitrate (NO ₃)	18	1.2	18	1.2
Fluoride (F)	18	1.2	18	1.2
Boron (B)	18	1.2	18	1.2
Silica (SiO ₂)	18	1.2	18	1.2
Total dissolved solids in parts per million	171	82	115	94
Percent sodium	82	18	18	18
Hardness as CaCO ₃ in parts per million				
Total	91	13	91	13
Noncarbonate				
Turbidity	40	20	20	20
Coliform in most probable number per milliliter	17,000	1,000	1,000	1,000
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.0	1.00	1.00	1.00
Solid alpha	1.15	1.00	1.00	1.00
Dissolved beta	1.15	1.00	1.00	1.00
Solid beta	1.15	1.00	1.00	1.00

WATER QUALITY VARIATIONS



Station
Number

11
11a
12
12b
12c
12d
13
13a
13b
13c
14
14a
15
15a
16
16a
16b
17
17a
17d
17e
18
18a
19
20
20a
20b
21
21a
22
22a
22b
22c

Station
Number

80
81
84
85
85a
87
87a
88
88a
88b
88c
88d
88e
88g
91
92
92a
93
94
94a
95a
95b
97
98
99
100
101
101a
102
103
103a
104
106
107
108
108a
108b
109



Lahontan Region (No. 6)

The Lahontan Region extends from the Oregon border on the north to the southern boundary of the Mojave River Basin on the south, and comprises that area situated between the California-Nevada border to the east and the Sierra Nevada to the west. The region contains about 33,000 square miles and varies in width from less than 20 miles in the north to over 170 miles, across the Mojave Desert and Antelope Valley, in the south.

The terrain of the region is characterized by basins of interior drainage or sinks surrounded by mountain peaks. Areas classified as valley and mesa lands cover about 10,000 square miles, most of which are considered irrigable. The eastern slopes of the Sierra Nevada dominate the mountainous portions of the Lahontan Region.

The region has an estimated mean seasonal runoff of 3,177,000 acre-feet. Principal streams in the Lahontan area include the Susan, Truckee, Carson, Walker, Owens and Mojave Rivers. To provide a continuing check on the quality of surface runoff in this region, 12 sampling stations are maintained on the following surface water sources as indicated in the following tabulation. The number of sampling stations on each source is shown in parentheses.

Susan River (1)
Lake Tahoe (3)
Truckee River (2)

Carson River (2)
Walker River (2)
Mojave River (2)*

* The Mojave River is in Southern California and will be discussed in Part II of this bulletin.



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Truckee River (2)

Carson River (2)
Walker River (2)
Mojave River (2)*

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Susan River Basin

Susan River Basin is a subbasin in the closed Honey Lake Basin located in the northeastern portion of California. The river originates on the slopes of the Sierra Nevada and flows eastward into Honey Lake Valley. Susan River watershed contains about 238 square miles and has an estimated mean seasonal runoff of 50,900 acre-feet.

Timber covered mountains and foothills comprise 157 square miles in the basin. Valley and mesa lands, some of which are also heavily forested, cover the remaining 81 square miles of watershed tributary to Susan River. Lumbering is the primary industry, followed in importance by the beef industry and agriculture devoted to the support of livestock.

Waste discharges entering the Susan River are primarily those associated with lumbermills and domestic or urban developments. Waste effluent from the City of Susanville is the only discharge in excess of 0.5 mgd entering the river. Serious impairment of water quality by waste discharges has not occurred or been reported in the Susan River.

A water quality monitoring station is maintained on Susan River at Susanville (17b) to monitor quality of runoff from the basin.

SUSAN RIVER AT SUSANVILLE (STA. 17b)

Sampling Point Station 17b is located in Section 31 of Township 30 North, Range 12 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the left bank, USGS gaging station, 0.5 mile west of Susanville, 1.1 miles upstream from Piute Creek, 24 miles above the mouth at Honey Lake.

Period of Record April 1951 through December 1959.

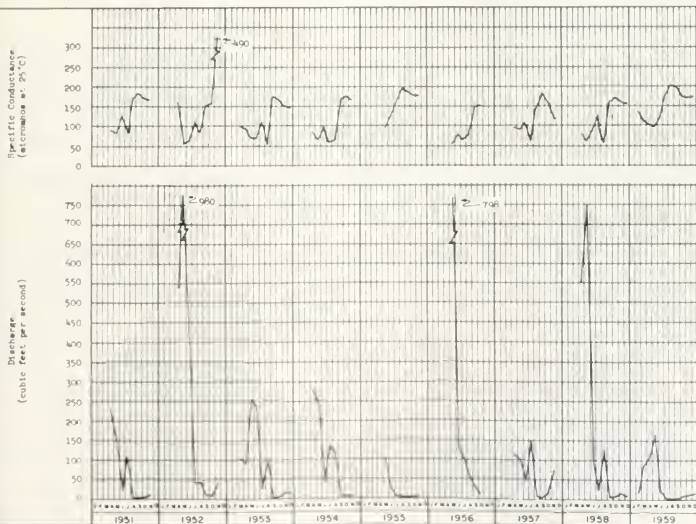
Water Quality Characteristics Past samples show water from Station 17b to be calcium-magnesium bicarbonate in character, class 1 for irrigation, varying from soft to moderately hard. Susan River water consistently meets drinking water standards. Only minor changes in water quality occur.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhm at 25°C)	480	55.5	201	48.2
Temperature in °F	75	33	64	33
Dissolved oxygen in parts per million	12.1	6.6	11.9	7.3
Percent saturation	98	71	95	76
pH	8.1	6.8	7.6	7.3
Mineral constituents in parts per million				
Calcium (Ca)	20	5.7	19	8.8
Magnesium (Mg)	16	2.8	10	4.6
Sodium (Na)	8.4	1.5	8.4	3.0
Potassium (K)	4.5	0.3	4.5	0.7
Carbonate (CO ₃)	8	8.0	8.8	0.0
Bicarbonate (HCO ₃)	238	32	134	56
Sulfate (SO ₄)	2.8	0.0	1.9	0.4
Chloride (Cl)	12	0.8	2.3	0.5
Nitrate (NO ₃)	1	0.8	0.5	0.4
Fluoride (F)	0.2	0.2	0.1	0.0
Boron (B)	0.24	0.0	0.0	0.0
Silica (SiO ₂)	42	16	42	23
Total dissolved solids in parts per million	370	42	152	72
Percent sodium	20	11	19	12
Hardness as CaCO ₃ in parts per million				
Total	120	23	93	41
Noncarbonate	2	0.8	2	0.4
Turbidity	5	0.5	15	1
Coliform in most probable number per milliliter	7,000	2,045	2,400	506
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.32	0.00	0.30	0.26
Solid alpha	1.65	0.00	0.55	0.00
Dissolved beta	13.7	7.00	7.64	4.16
Solid beta	25.1	0.00	4.03	0.00

WATER QUALITY VARIATIONS



SUSAN RIVER AT SUSANVILLE (STA. 17b)

Truckee River Basin

The Truckee River drains an area near the central portion of the Lahontan Region at the "elbow-bend" in the California-Nevada border. The California portion of the river basin is predominantly alpine with 621 of the 805 square miles classed as mountainous. Mean annual runoff from the California portion of the basin exceeds 580,000 acre-feet.

Lake Tahoe, formed by the down-dropping of a fault block along the Sierran-Nevadan fault, is one of the prominent physical features of the Truckee River Basin. With a mean water surface elevation of 6,228 feet and an approximate 120 miles of shore line, it has become an internationally known recreation and vacation attraction. Valley and mountain meadow land contain 184 square miles in the California portion of the watershed. Development is primarily associated with recreation. Lumbering is carried on to a minor degree in the basin.

Waste discharged into the waterways of the area have been small in quantity; however, those entering Lake Tahoe have caused some concern as to their effect on this important body of water. A review of available data reveals that no significant impairment to the lake has been detected under present conditions.

The following tabulation presents the names of stations maintained to monitor quality of surface water in this basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Lake Tahoe at Bijou	402
Lake Tahoe at Tahoe Vista	404
Lake Tahoe at Tahoe City	406
Truckee River near Truckee	408
Truckee River near Farad	410

LAKE TAHOE AT BIJOU (STA. 39)

Sampling Point Bijou station is located on the south end of the lake in Section 33 of Township 13 North, Range 18 East, Mt. Diablo Base and Meridian. The monthly grab samples were collected from a pier in Bijou.

Period of Record April 1951 through December 1959.

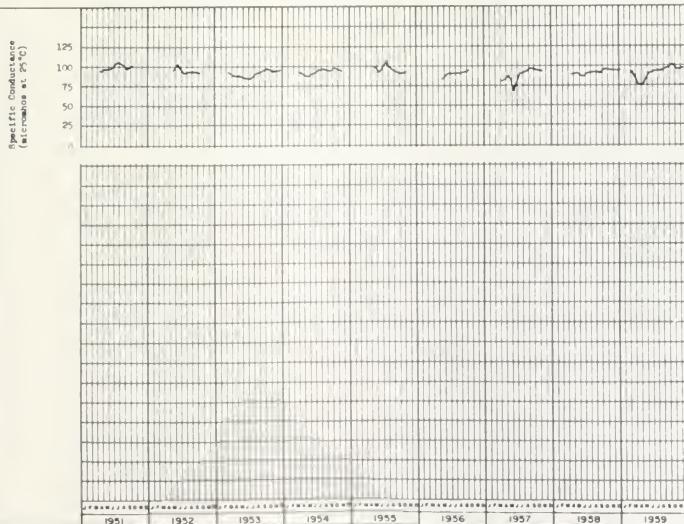
Water Quality Characteristics Lake Tahoe water is calcium bicarbonate in character, soft, and excellent for all uses. The quality is very similar to that reported at Tahoe Vista (Station 37) and Tahoe City (Station 38), with only minor variations noted. Variation in quality has been nearly imperceptible during the period of record.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	100	80.3	10	75.4
Temperature in °F	74	4	14	44
Dissolved oxygen in parts per million	14.2	7.2	9.9	7.6
Percent saturation	100	100	100	70
pH	8.1	6.8	7.7	7.3
Mineral constituents in parts per million				
Calcium (Ca)	100	7.6	9.4	8.4
Magnesium (Mg)	1.8	1.6	2.1	1.7
Sodium (Na)	4.2	4.1	1.2	5.2
Potassium (K)	21	8.1	1.9	1.7
Carbonate (CO ₃)	1	0.0	0.3	0.0
Bicarbonate (HCO ₃)	11	16	52	42
Sulfate (SO ₄)	4.8	1.0	2.9	2.0
Chloride (Cl)	6	3.0	4.1	1.8
Nitrate (NO ₃)	1.5	0.0	0.5	0.0
Fluoride (F)	1.1	0.0	0.0	0.0
Boron (B)	0.25	0.0	0.2	0.0
Silica (SiO ₂)	18	8.8	13	11
Total dissolved solids in parts per million	86	44	67	51
Percent sodium	37	23	32	25
Hardness as CaCO ₃ in parts per million				
Total	40	21	37	27
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	0.5	0.0	15	0.3
Coliform in most probable number per milliliter	7,300	0.004	23	0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.17	0.00	0.10	0.00
Solid alpha	1.20	0.00	0.51	0.37
Dissolved beta	14.77	0.00	7.23	2.64
Solid beta	9.83	0.00	3.77	0.00

WATER QUALITY VARIATIONS



LAKE TAHOE AT BIJOU (STA. 39)

LAKE TAHOE AT TAHOE VISTA (STA. 37)

Sampling Point Station 37 is located in Section 14 of Township 16 North, Range 17 East, Mt. Diablo Base and Meridian. Monthly grab samples were taken from the end of a pier located on the north end of the lake 0.1 mile west of Tahoe Vista, 8 miles northeast of Tahoe City.

Period of Record April 1951 through December 1959.

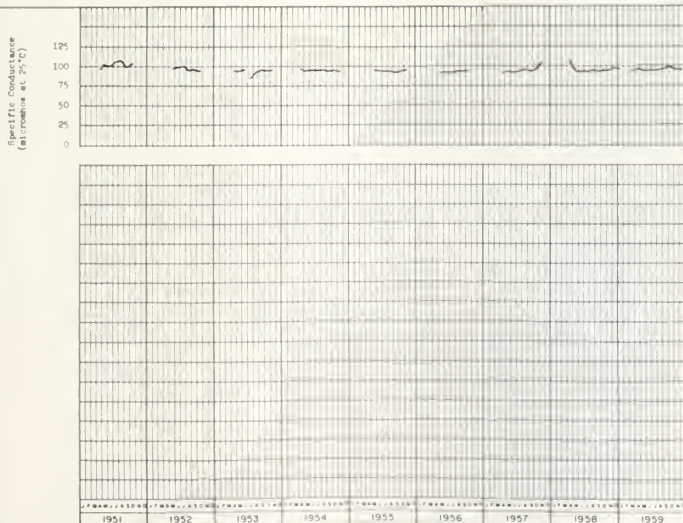
Water Quality Characteristics Antecedent data show the mineral character of the lake, at Station 37, to be calcium bicarbonate, low in concentration of all constituents, soft, of excellent mineral quality, and suitable for nearly all beneficial uses. The water varies very little in mineral quality.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohms at 25°C)	60	49.8	60	41.6
Temperature in °F	72	43	68	44
Dissolved oxygen in parts per million	10.4	6.4	10	7.9
Percent saturation	98	63	100	81
pH	8.4	6.8	8.5	7
Mineral constituents in parts per million				
Calcium (Ca)	11	8.4	11	9.4
Magnesium (Mg)	4.4	2.1	4.4	1
Sodium (Na)	8.2	5.7	8	5.8
Potassium (K)	2.4	1.5	7	1.6
Carbonate (CO ₃)	2.1	2.1	0	1.7
Bicarbonate (HCO ₃)	4	7.4	6	39
Sulfate (SO ₄)	6	4.5	2.9	2.7
Chloride (Cl)	6.1	3.5	4.8	1.5
Nitrate (NO ₃)	.2	7.8	8	3.1
Fluoride (F)	.1	0.03	0.05	0.05
Boron (B)	1.9	0.0	1.1	0.0
Silica (SiO ₂)	18	11	14	11
Total dissolved solids in parts per million	73	59	67	62
Percent sodium	34	22	34	24
Hardness as CaCO ₃ in parts per million				
Total	41	31	39	32
Noncarbonate	2	1	2	0.5
Turbidity	25	0.25	1	0.3
Coliform in most probable number per milliliter	240	10.85	23	0.04
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.43	0.35	1.7	0.00
Solid alpha	7.00	0.00	7.3	0.18
Dissolved beta	13.58	0.00	13.64	1.07
Solid beta	1.1	0.00	7.55	2.81

WATER QUALITY VARIATIONS



LAKE TAHOE AT TAHOE VISTA (STA. 37)

LAKE TAHOE AT TAHOE CITY (STA. 38)

Sampling Point Station 38 is located on the west side of Lake Tahoe in Section 7 of Township 15 North, Range 17 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected on the upstream side of the control gates at the outlet of the lake (Truckee River).

Period of Record April 1951 through December 1959.

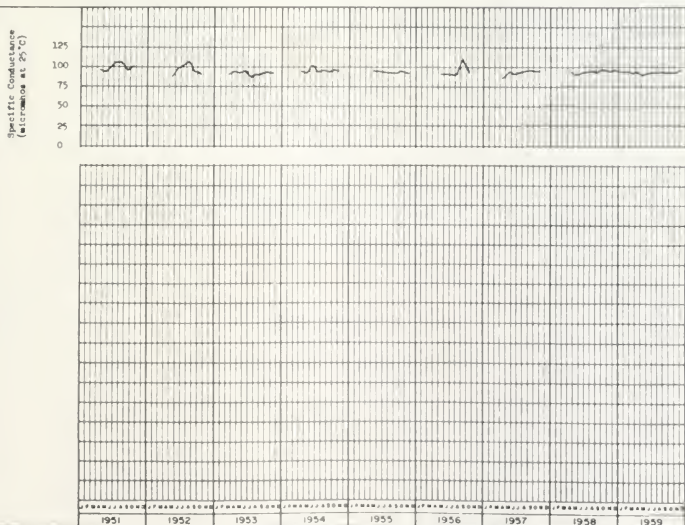
Water Quality Characteristics Water at Station 38 is calcium bicarbonate in character, soft, and in all respects very similar to the water at Tahoe Vista (Station 37). These waters are consistently excellent in quality and within mineral requirements for nearly all beneficial uses.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	100	50	75	40
Temperature in °F	72	50	60	40
Dissolved oxygen in parts per million	11.0	7.1	8.7	7.1
Percent saturation	99	94	88	74
pH	8.5	7.2	7.5	7.0
Mineral constituents in parts per million				
Calcium (Ca)	115	85	100	85
Magnesium (Mg)	35	15	25	15
Sodium (Na)	20	10	15	10
Potassium (K)	5.0	0.00	5.0	0.0
Carbonate (CO ₃)	7	0.01	6.99	0.0
Carbonate (HCO ₃)	75	44	49	44
Sulfate (SO ₄)	6.8	0.00	6.8	0.0
Chloride (Cl)	6	0.0	6	0.0
Nitrate (NO ₃)	0.5	0.0	0.5	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.28	0.0	0.28	0.0
Silica (SiO ₂)	35	11	24	11
Total dissolved solids in parts per million	95	58	64	60
Percent sodium	36	22	31	27
Hardness as CaCO ₃ in parts per million				
Total	43	30	38	30
Noncarbonate	2	0.0	2.0	0.0
Turbidity	25	0.10	10	0.2
Coliform in most probable number per milliliter	2,400.	0.02	230.	0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.44	0.00	0.10	0.00
Solid alpha	1.77	0.00	0.37	0.00
Dissolved beta	14.4	0.00	5.56	5.13
Solid beta	8.28	0.00	3.55	2.67

WATER QUALITY VARIATIONS



LAKE TAHOE AT TAHOE CITY (STA. 38)

TRUCKEE RIVER NEAR TRUCKEE (STA. 52)

Sampling Point Station 52 is located 1.4 miles upstream from Donner Creek, 2.5 miles southwest of Truckee, in Section 28 of Township 17 North, Range 16 East, Mt. Diablo Base and Meridian. Monthly water samples were collected on the left bank, at the USGS gage, approximately 11.5 miles downstream from the outlet of Lake Tahoe.

Period of Record April 1951 through December 1959.

Water Quality Characteristics Chemical classification of water from Truckee River, at Station 52, shows the water usually to be calcium bicarbonate in character, class 1 for irrigation, and excellent for domestic and industrial use. Tributary inflow in the reach between Lake Tahoe and Truckee has no apparent effects on water quality in the Truckee River.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (microhmohms at 25°C)	125	47.2	106	14.9
Temperature in °F	69	19	65	47
Dissolved oxygen in parts per million	12.4	7.2	10.1	7.5
Percent saturation	88	73	100	76
pH	8.4	6.8	7.9	7.1
Mineral constituents in parts per million				
Calcium (Ca)	82	5.9	44	6.8
Magnesium (Mg)	1.9	0.9	1.3	1.2
Sodium (Na)	3.6	2.2	7.7	2.6
Potassium (K)	3.4	2.4	1.1	1.8
Carbonate (CO ₃)	2.2	0.3	2.0	0.0
Bicarbonate (HCO ₃)	1.9	1.4	57	26
Sulfate (SO ₄)	4.9	1.1	4.8	2.1
Chloride (Cl)	6.2	0.3	4.4	0.8
Nitrate (NO ₃)	1.2	0.0	0.4	0.1
Fluoride (F)	0.2	0.0	0.4	0.0
Boron (B)	14	0.0	0.1	0.0
Silica (SiO ₂)	20	12	13	13
Total dissolved solids in parts per million	96	34	73	44
Percent sodium	30	18	32	19
Hardness as CaCO ₃ in parts per million				
Total	48	12	42	22
Noncarbonate	1	0.4	2	0.1
Turbidity	18	0.3	120	0.3
Coliform in most probable number per milliliter	>7,000	0.045	231	0.04
Radioactivity in micro-micro curies per liter				
Dissolved alpha	5	0.30	0.30	0.26
Solid alpha	104	0.30	0.55	0.17
Dissolved beta	16.1	0.00	0.49	1.03
Solid beta	0.26	0.00	8.54	1.26

WATER QUALITY VARIATIONS



TRUCKEE RIVER NEAR TRUCKEE (STA. 52)

TRUCKEE RIVER NEAR FARAD (STA. 53)

Sampling Point Station 53 is located approximately 2 miles upstream from the California-Nevada state line in Section 12 of Township 18 North, Range 17 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank at the USGS gage.

Period of Record April 1951 through December 1959.

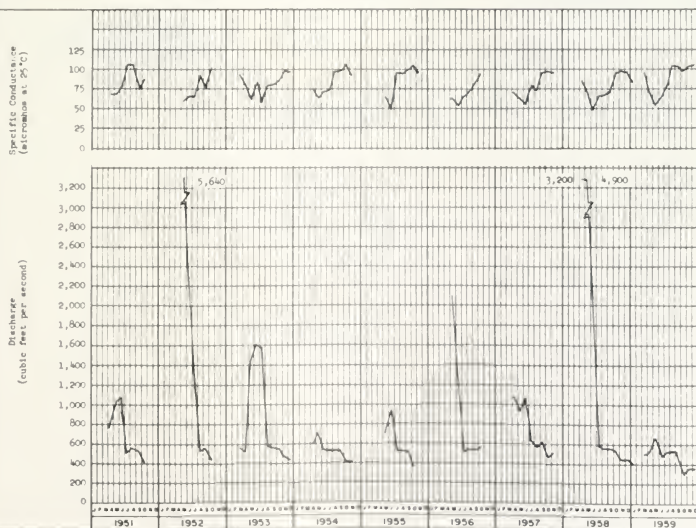
Water Quality Characteristics Analyses show water at Station 53 to be bicarbonate in type, generally with no predominant cation, although during periods of high runoff the water becomes calcium bicarbonate in character. The water is class 1 for irrigation, soft, and meets drinking water requirements for mineral content. Very little change is noted between Station 52 (near Truckee) and Station 53. In some instances the quality improves slightly in this reach of the river due to the excellent quality of tributary inflow.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1979	Minimum - 1959
Specific conductance (micromhos at 25°C)	187	49	116	54.7
Temperature in °F	66	14	60	38
Dissolved oxygen in parts per million	7.2	7.4	10.4	7.8
Percent saturation	138	70	190	75
pH	8.2	6.3	7.2	7.3
Mineral constituents in parts per million				
Calcium (Ca)	12	5.3	11	5.8
Magnesium (Mg)	1.7	1	2.1	2.1
Sodium (Na)	3.4	2.4	9.4	2.8
Potassium (K)	2.8	0.6	1.2	0.7
Carbonate (CO ₃)		0.0	0.0	0.0
Bicarbonate (HCO ₃)	24	24	62	28
Sulfate (SO ₄)	4.8	0.0	4.8	1.0
Chloride (Cl)	1.2	0.0	5.2	1.8
Nitrate (NO ₃)	0.65	0.0	0.4	0.1
Fluoride (F)	0.2	0.0	0.0	0.0
Boron (B)	0.17	0.0	0.0	0.0
Silica (SiO ₂)	22	8.3	16	13
Total dissolved solids in parts per million	81	37	81	46
Percent sodium	14	17	34	18
Hardness as CaCO ₃ in parts per million				
Total	44	18	39	23
Noncarbonate	1	0.0	1	0.0
Turbidity	50	0.3	50	0.3
Coliform in most probable number per milliliter	>7,000	0.06	2,400	0.06
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.32	0.00	0.30	0.00
Solid alpha	0.76	0.00	0.44	0.27
Dissolved beta	8.85	0.00	6.37	2.76
Solid beta	7.78	0.00	7.78	3.09

WATER QUALITY VARIATIONS



TRUCKEE RIVER NEAR FARAD (STA. 53)

Carson River Basin

The Carson River Basin drains an area in California of 449 square miles in the central portion of the Lahontan Region. The Carson River system originates in the Sierra Nevada and flows eastward into the State of Nevada. The California watershed of the Carson River is classified as mountainous and foothill terrain. Annual natural mean runoff from the California portion of the basin is about 389,300 acre-feet.

The economy in the basin is based on livestock raising, supplemented by recreational activities. There are no significant waste discharges entering the river system.

The following tabulation presents the names of stations maintained to monitor quality of surface waters in the basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Carson River, West Fork at Woodfords	414
Carson River, East Fork near Markleeville	416

CARSON RIVER, WEST FORK AT WOODFORDS (STA. 115a)

Sampling Point The Woodfords station is located in Section 3⁴ of Township 11 North, Range 19 East, Mt. Diablo Base and Meridian. Samples were collected monthly at the USGS gage on the left bank 3-1/2 miles downstream from Willow Creek, 0.8 mile west of Woodfords.

Period of Record August 1958 through December 1959.

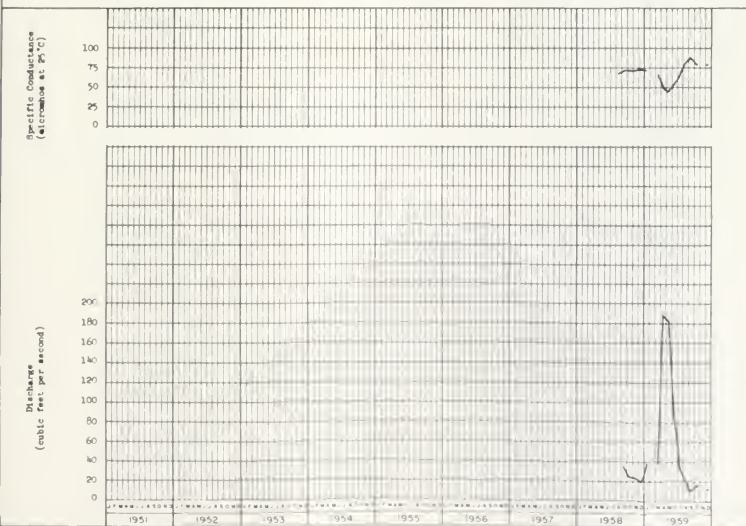
Water Quality Characteristics Past analyses show the water at Station 115a to be characteristically calcium bicarbonate, soft, class 1 for irrigation, and within drinking water requirements for mineral content. Total radioactivity reached 15.16 $\mu\text{c}/\text{l}$ in September 1959, which is slightly higher than is usually encountered in surface streams, but below the recommended safe limit.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	87.2	45.5	87.2	45.5
Temperature in °F	64	34	60	34
Dissolved oxygen in parts per million	11.6	7.4	11.6	8.2
Percent saturation	100	78	100	78
pH	8.0	7.1	7.8	7.1
Mineral constituents in parts per million				
Calcium (Ca)	18	6.0	10	6.0
Magnesium (Mg)	2.9	0.9	2.7	0.9
Sodium (Na)	5.4	1.9	5.4	1.9
Potassium (K)	1.8	0.6	1.8	0.6
Carbonate (CO ₃)	0.0	0.0	0.0	0.0
Bicarbonate (HCO ₃)	51	25	51	25
Sulfate (SO ₄)	6.0	0.0	6.0	0.0
Chloride (Cl)	2.5	0.5	2.5	0.5
Nitrate (NO ₃)	1.5	0.0	1.5	0.0
Fluoride (F)	0.1	0.0	0.1	0.0
Boron (B)	0.1	0.0	0.1	0.0
Silica (SiO ₂)	24	10	24	10
Total dissolved solids in parts per million	85	36	85	36
Percent sodium	30	13	30	13
Hardness as CaCO ₃ in parts per million				
Total	34	18	34	18
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	10	0.5	10	0.5
Coliform in most probable number per milliliter	230	0.13	230	0.13
Radioactivity in micro-micro curies per liter				
Dissolved alpha	1.36	0.00	0.33	0.00
Solid alpha	0.54	0.00	0.54	0.00
Dissolved beta	8.79	4.16	8.79	4.16
Solid beta	6.67	0.96	6.04	0.96

WATER QUALITY VARIATIONS



CARSON RIVER, WEST FORK AT WOODFORDS (STA. 115a)

CARSON RIVER, EAST FORK NEAR MARKLEEVILLE (STA. 115)

Sampling Point Station 115 is located in Section 27 of Township 10 North, Range 20 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected in mid-stream, from State Highway 4 bridge, about 2.5 miles south of Markleeville.

Period of Record September 1958 through December 1959.

Water Quality Characteristics The water is calcium bicarbonate in character, soft to slightly hard, class 1 for irrigation use, within drinking water requirements for mineral content and suitable for all but the most exacting industrial uses.

Significant Water Quality Changes None.

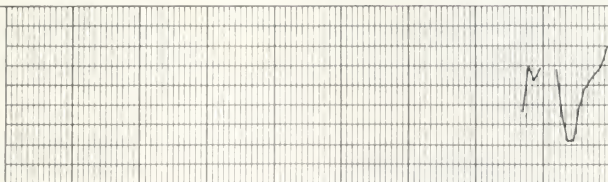
WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	75	54.5	175	54.5
Temperature in °F	64	33	64	33
Dissolved oxygen in parts per million	11.7	8.2	11.7	8.2
Percent saturation	92	78	92	78
pH	7.2	6.8	7.2	6.8
Mineral constituents in parts per million				
Calcium (Ca)	16	5.2	16	5.2
Magnesium (Mg)	8.1	7	8.1	7
Sodium (Na)	12	3.4	12	3.4
Potassium (K)	2.5	0.9	2.5	0.9
Carbonate (CO ₃)	0	0.0	0	0.0
Bicarbonate (HCO ₃)	27	27	27	27
Sulfate (SO ₄)	17	2.9	17	2.9
Chloride (Cl)	7.2	0.8	7.2	0.8
Nitrate (NO ₃)	0.6	0.0	0.6	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.2	0.0	0.2	0.0
Silica (SiO ₂)	28	12	28	12
Total dissolved solids in parts per million	132	42	132	42
Percent sodium	30	20	30	21
Hardness as CaCO ₃ in parts per million				
Total	62	19	62	19
Noncarbonate	3	0.0	3	0.0
Turbidity	10	3	4	3
Coliform in most probable number per milliliter	230	0.13	230	0.13
Radioactivity in micro-curies per liter				
Dissolved alpha	0.17	0.00		0.00
Solid alpha	0.43	0.42	0.43	
Dissolved beta	3.36	0.84		0.84
Solid beta	8.11	2.81		2.81

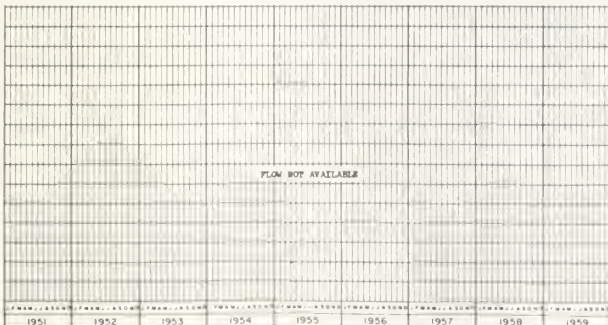
WATER QUALITY VARIATIONS

Specific Conductance
(micromhos at 25°C)

175
150
125
100
75
50
25
0



FLOW NOT AVAILABLE



CARSON RIVER, EAST FORK NEAR MARKLEEVILLE (STA. 115)

Walker River Basin

Walker River Basin encompasses the northern section of Mono County in the central part of the Lahontan Region. The California portion of this basin includes about 910 square miles along the eastern slopes of the Sierra Nevada. Estimated mean annual runoff in this portion of the basin is 484,000 acre-feet.

The terrain of the Walker River watershed is predominantly mountainous with only 61 square miles in California classified as valley and mesa lands. Livestock raising, the production of winter feed, and recreation sustain the existing economy of the basin.

There are no significant waste discharges entering the waterways of the basin. Quality impairment of runoff from the watershed area by waste discharges has been negligible.

The following tabulation presents the names of stations maintained to monitor quality of surface water in the basin and the page on which each is discussed:

<u>Monitoring Station</u>	<u>Page Number of Station Discussion</u>
Walker River, West near Coleville	420
Walker River, East near Bridgeport	422

WALKER RIVER, WEST NEAR COLEVILLE (STA. 116)

Sampling Point Station 116 is located in Section 9 of Township 6 North, Range 23 East, Mt. Diablo Base and Meridian. Monthly water samples were collected from the left bank 300 feet downstream from U. S. Highway 395 bridge, 700 feet downstream from East Fork, 500 feet downstream from USGS gage, 13 miles southeast of Coleville.

Period of Record August 1958 through December 1959.

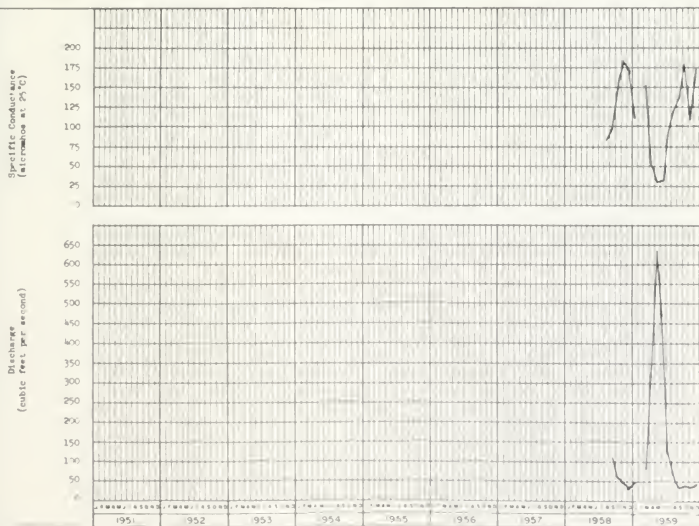
Water Quality Characteristics Analyses of West Walker River indicate a water generally bicarbonate in type with no predominant cation, class 1 for irrigation, soft and within drinking water requirements for mineral content. However, during months of higher flows this water was calcium bicarbonate in character, with calcium receding in prominence in months when surface runoff diminished.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	185	31.3	180	31.3
Temperature in °F	67	33	67	33
Dissolved oxygen in parts per million	11.6	7.0	11.6	7.0
Percent saturation	104	70	104	76
pH	8.2	7.3	8.2	7.3
Mineral constituents in parts per million				
Calcium (Ca)	17	4.2	17	4.2
Magnesium (Mg)	4.1	0.2	4.3	0.2
Sodium (Na)	22	1.7	22	1.7
Potassium (K)	2.1	0.3	2.1	0.3
Carbonate (CO ₃)	4.8	0.0	4.8	0.0
Bicarbonate (HCO ₃)	88	17	88	17
Sulfate (SO ₄)	13	0.0	11	0.0
Chloride (Cl)	7.5	0.2	7.0	0.2
Nitrate (NO ₃)	0.9	0.0	0.9	0.0
Fluoride (F)	0.2	0.0	0.2	0.0
Boron (B)	0.3	0.0	0.3	0.0
Silica (SiO ₂)	19	5.1	19	5.1
Total dissolved solids in parts per million	119	24	118	24
Percent sodium	52	14	52	14
Hardness as CaCO ₃ in parts per million				
Total	56	12	56	12
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	15	1	15	1
Coliform in most probable number per milliliter	620	<0.045	620	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.20	0.00	0.20	0.17
Solid alpha	1.36	0.08	0.26	0.08
Dissolved beta	4.16	0.64	4.16	0.64
Solid beta	6.54	0.00	6.54	0.00

WATER QUALITY VARIATIONS



WALKER RIVER, WEST NEAR COLEVILLE (STA. 116)

WALKER RIVER, EAST NEAR BRIDGEPORT (STA. 116a)

Sampling Point Station 116a is located in Section 34 of Township 6 North, Range 25 East, Mt. Diablo Base and Meridian. Monthly grab samples were collected from the right bank, 500 feet downstream from Bridgeport Reservoir and 5 miles north of Bridgeport.

Period of Record August 1958 through December 1959.

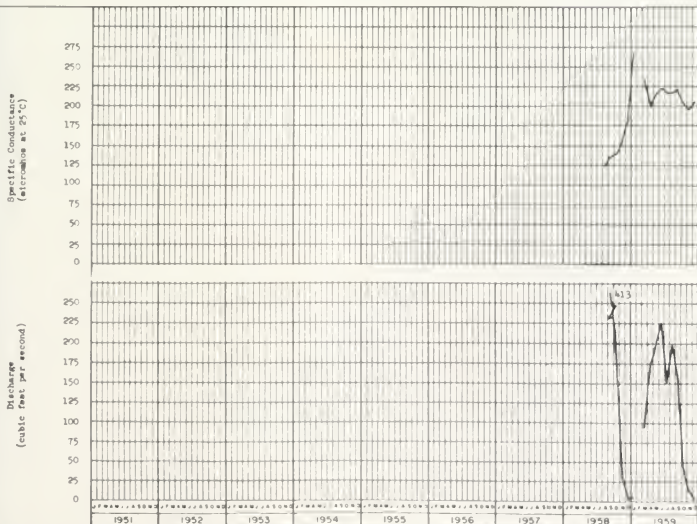
Water Quality Characteristics During periods of high flow, the water is calcium bicarbonate in character, class 1 for irrigation, and meets drinking water standards for mineral content. During low flow conditions the water becomes bicarbonate in type with no predominant cation.

Significant Water Quality Changes None.

WATER QUALITY RANGES

Item	Maximum of Record	Minimum of Record	Maximum - 1959	Minimum - 1959
Specific conductance (micromhos at 25°C)	268	124	268	124
Temperature in °F	73	40	70	40
Dissolved oxygen in parts per million	10.9	6.6	10.9	6.6
Percent saturation	100	69	100	69
pH	8.5	7.2	8.1	7.3
Mineral constituents in parts per million				
Calcium (Ca)	26	7.2	26	22
Magnesium (Mg)	8.7	4.4	4.9	4.4
Sodium (Na)	30	7.3	30	14
Potassium (K)	5.4	1.9	4.4	3.2
Carbonate (CO ₃)	6	2.0	6	4.0
Bicarbonate (HCO ₃)	124	67	124	90
Sulfate (SO ₄)	36	0.0	36	0.0
Chloride (Cl)	7.5	0.8	7.5	1.5
Nitrate (NO ₃)	2.0	0.0	1.8	0.0
Fluoride (F)	0.6	0.1	0.6	0.1
Boron (B)	0.2	0.0	0.2	0.0
Silica (SiO ₂)	26	12	26	16
Total dissolved solids in parts per million	187	80	187	124
Percent sodium	43	24	43	27
Hardness as CaCO ₃ in parts per million				
Total	82	45	82	59
Noncarbonate	0.0	0.0	0.0	0.0
Turbidity	60	2	60	2
Coliform in most probable number per milliliter	2,400	<0.045	2,400	<0.045
Radioactivity in micro-micro curies per liter				
Dissolved alpha	0.42	0.33		0.33
Solid alpha	0.33	0.00	0.33	
Dissolved beta	0.54	0.36		0.36
Solid beta	3.60	0.00		0.00

WATER QUALITY VARIATIONS



WALKER RIVER, EAST NEAR BRIDGEPORT (STA. 116a)







SURFACE WATER QUALITY, OTHER AGENCIES' MONITORING PROGRAM

One objective of this bulletin is to present all available data of a continuous and reliable nature on quality of surface waters in California. Several agencies, not under the administration of the State of California, Department of Water Resources, have surface water quality monitoring programs comparable to the department's. Mineral analyses of samples collected from surface waters in California under other agencies' programs are presented in Appendix B of this bulletin.

To supplement analyses of other agencies a brief description of their sampling stations and, when known, period of record, are included in the following alphabetical listing:

AMERICAN RIVER AT FAIR OAKS (STA. 22d)

Sampling Point Station 22d is located in Section 13 of Township 9 North, Range 6 East, Mt. Diablo Base and Meridian. Samples were collected 1,000 feet below Old Fair Oaks Bridge, 2.4 miles east of Fair Oaks, 0.4 mile downstream from Nimbus Dam. Samples were collected quarterly and analyzed by the U. S. Bureau of Reclamation.

Period of Record January 1938 through October 1959.

CACHE SLOUGH BELOW LINDSEY SLOUGH (STA. 110a)

Sampling Point Station 110a is located in Section 31, Township 5 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected at the surface, at Liberty Island Ferry, about 0.5 mile downstream from Lindsey Slough, 6 miles north of Rio Vista. Samples were collected quarterly, usually in January, April, July, and October, and analyzed by the U. S. Bureau of Reclamation.

Period of Record April 1952 through October 1959.



SURFACE WATER QUALITY, OTHER AGENCIES' MONITORING PROGRAM

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Sampling Point Station 22d is located in Section 13 of Township 9 North, Range 6 East, Mt. Diablo Base and Meridian. Samples were collected 1,000 feet below Old Fair Oaks Bridge, 2.4 miles east of Fair Oaks, 0.4 mile downstream from Nimbus Dam. Samples were collected quarterly and analyzed by the U. S. Bureau of Reclamation.

Period of Record January 1938 through October 1959.

CACHE SLOUGH BELOW LINDSEY SLOUGH (STA. 110a)

Sampling Point Station 110a is located in Section 31, Township 5 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected at the surface, at Liberty Island Ferry, about 0.5 mile downstream from Lindsey Slough, 6 miles north of Rio Vista. Samples were collected quarterly, usually in January, April, July, and October, and analyzed by the U. S. Bureau of Reclamation.

Period of Record April 1952 through October 1959.

CARQUINEZ STRAIT AT MARTINEZ (STA. 28a)

Sampling Point The Martinez station is located in Section 13, Township 2 North, Range 3 West, Mt. Diablo Base and Meridian. Samples were collected from the left bank at Benicia-Martinez ferry slip. A U. S. Bureau of Reclamation continuous salinity recorder is located at the sampling point. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation when the salinity recorder was operating.

Period of Record March 1955 through December 1959.

CONTRA COSTA CANAL AT FIRST PUMP LIFT (STA. 109a)

Sampling Point The station is located in Section 25, Township 2 North, Range 2 East, Mt. Diablo Base and Meridian. The samples were collected at the discharge of the first pump lift of Contra Costa Canal approximately 0.5 mile southeast of Oakley. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation.

Period of Record January 1955 through December 1959.

DUTCH SLOUGH AT FARRAR PARK BRIDGE (STA. 108b)

Sampling Point Farrar Park Bridge station is located in Section 22, Township 2 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected at Farrar Park Bridge (Bethel Island Bridge) about 4 miles east of Oakley. A U. S. Bureau of Reclamation continuous salinity recorder is located at the sampling point. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation when the salinity recorder was in operation.

Period of Record May 1955 through December 1959.

FALSE RIVER AT WEBB PUMP (STA. 112a)

Sampling Point The station is located in Section 36, Township 3 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected on the south side of Webb Tract, approximately 10 miles northeast of Antioch. A U. S. Bureau of Reclamation continuous salinity recorder is located at this site. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation.

Period of Record May 1955 through December 1959.

MOKELUMNE RIVER BELOW COSUMNES RIVER (STA. 23b)

Sampling Point The Cosumnes River station is located in Section 29, Township 5 North, Range 5 East, Mt. Diablo Base and Meridian. Samples were collected below the confluence of the Cosumnes and Mokelumne Rivers about 2 miles north of Thornton. The U. S. Bureau of Reclamation collected and analyzed monthly samples from this station during periods when the Delta Cross Channel gates are open.

Period of Record June 1952 through December 1959, except for 1956.

MOKELUMNE RIVER BELOW GEORGIANA SLOUGH (STA. 23c)

Sampling Point The station is located in Section 7, Township 3 North, Range 4 East, Mt. Diablo Base and Meridian. Samples were collected and analyzed monthly by the U. S. Bureau of Reclamation during periods when the Delta Cross Channel gates are open below the confluence of Georgiana Slough, near State Highway 12 bridge crossing approximately 3 miles east of Isleton.

Period of Record May 1952 through December 1959, except for 1956.

OLD RIVER AT HOLLAND TRACT (STA. 108a)

Sampling Point The station is located in Section 19, Township 2 North, Range 4 East, Mt. Diablo Base and Meridian. Samples were collected from the left bank on the east side of Holland Tract about 5 miles northeast of Knightsen. A U. S. Bureau of Reclamation continuous salinity recorder is located at the sampling point. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation when the salinity recorder is operating.

Period of Record March 1952 through December 1959.

SACRAMENTO RIVER NEAR MALLARD SLOUGH (STA. 15c)

Sampling Point Station 15c is located in Section 5, Township 2 North, Range 1 East, Mt. Diablo Base and Meridian. Samples were taken from the left bank and at Pacific Gas and Electric Company dock at Pittsburg. A U. S. Bureau of Reclamation continuous salinity recorder is located at this sampling point. Samples were collected and analyzed by the U. S. Bureau of Reclamation when the salinity recorder was operating.

Period of Record March 1955 through December 1959.

SACRAMENTO RIVER AT SNODGRASS SLOUGH (STA. 97)

Sampling Point The station is located in Section 22, Township 6 North, Range 4 East, Mt. Diablo Base and Meridian. Samples were taken from the left bank at the structure housing a U. S. Bureau of Reclamation continuous salinity recorder, at Greens Landing approximately 2 miles northeast of Courtland. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation.

Period of Record June 1938 through December 1959.

SACRAMENTO RIVER AT TOLAND LANDING (STA. 15a)

Sampling Point Station 15a is located in Section 21, Township 3 North, Range 2 East, Mt. Diablo Base and Meridian. The samples were collected from the right bank at the structure housing a U. S. Bureau of Reclamation continuous salinity recorder, approximately 6 miles downstream from Rio Vista. Samples were collected and analyzed monthly by the U. S. Bureau of Reclamation when the salinity recorder was operating.

Period of Record July 1952 through December 1959.

SAN JOAQUIN RIVER AT BRANDT BRIDGE (STA. 101a)

Sampling Point Station 101a is located in Section 9, Township 1 North, Range 6 East, Mt. Diablo Base and Meridian. Samples were collected at the tide stage recorder on Brandt Bridge approximately 7.0 miles south of Stockton. Samples were collected every three months and analyzed by the U. S. Bureau of Reclamation.

Period of Record August 1940 through June 1945; March 1948 through December 1955; and March 1957 through December 1959.

SAN JOAQUIN RIVER AT CROWS LANDING BRIDGE (STA. 26b)

Sampling Point Station 26b is located in Section 7 of Township 6 South, Range 9 East, Mt. Diablo Base and Meridian. Samples were collected at Crows Landing Bridge 4.5 miles northeast of Crows Landing. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation.

Period of Record June 1952 through December 1959.

SAN JOAQUIN RIVER NEAR DOS PALOS (STA. 25a)

Sampling Point The station is located in Section 12 of Township 11 South, Range 13 East, Mt. Diablo Base and Meridian. Samples were collected about

0.7 mile downstream from the head of Temple Slough and 7 miles east of Dos Palos. Samples were collected and analyzed monthly by the U. S. Bureau of Reclamation.

Period of Record September 1938 through December 1959.

SAN JOAQUIN RIVER AT JERSEY POINT (STA. 28b)

Sampling Point The station is located in Section 6, Township 2 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected from the left bank on the northern portion of Jersey Island approximately 9 miles northeast of Antioch. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation when the salinity recorder located at this site is operating.

Period of Record July 1952 through December 1959.

SAN JOAQUIN RIVER ABOVE MERCED RIVER (STA. 30a)

Sampling Point Station 30a is located in Section 3 of Township 7 South, Range 9 East, Mt. Diablo Base and Meridian. Samples were collected 0.2 mile upstream from Hills Ferry Bridge, 0.1 mile upstream from the mouth of the Merced River 2.3 miles northeast of Newman. Samples were collected and analyzed monthly by the U. S. Bureau of Reclamation.

Period of Record January 1938 through December 1959.

SAN JOAQUIN RIVER AT PATTERSON WATER COMPANY (STA. 27a)

Sampling Point The station is located in Section 15 of Township 5 South, Range 8 East, Mt. Diablo Base and Meridian. Samples were collected at Patterson Water Company Intake at Patterson Bridge 3.6 miles northeast of Patterson. Monthly samples were collected and analyzed by the U. S. Bureau of Reclamation.

Period of Record June 1938 through December 1959.

SAN JOAQUIN RIVER ABOVE SALT SLOUGH (STA. 111b)

Sampling Point Station 111b is located in Section 26 of Township 7 South, Range 10 East, Mt. Diablo Base and Meridian. Samples were collected from the surface approximately 4 miles upstream from Salt Slough 3 miles south of Stevinson. Samples were collected and analyzed monthly by the U. S. Bureau of Reclamation.

Period of Record September 1955 through December 1959.

SAN JOAQUIN RIVER AT SAN ANDREAS LANDING (STA. 112b)

Sampling Point The station is located in Section 13, Township 3 North, Range 3 East, Mt. Diablo Base and Meridian. Samples were collected from the left bank at Andrus Island approximately 6 miles south of Isleton. This station is maintained and operated by the U. S. Bureau of Reclamation.

Period of Record March 1952 through December 1959.

SAN JOAQUIN RIVER AT WEST STANISLAUS IRRIGATION DISTRICT INTAKE (STA. 27b)

Sampling Point The station is located in Section 10 of Township 4 South, Range 7 East, Mt. Diablo Base and Meridian. Samples were collected at the intake canal to West Stanislaus Irrigation District 4 miles north of Westley 0.2 mile upstream from Tuolumne River. Samples were collected monthly and analyzed by the U. S. Bureau of Reclamation.

Period of Record June 1938 through December 1959.

SAN JOAQUIN RIVER AT WHITEHOUSE (STA. 24b)

Sampling Point Station 24b is located in Section 25 of Township 13 South, Range 15 East, Mt. Diablo Base and Meridian. Samples were collected approximately 12 miles upstream from Mendota Dam at the head of Willow Slough. Conductivity is determined semimonthly and mineral analyses

made bimonthly during the irrigation season by the U. S. Bureau of Reclamation.

Period of Record November 1953 through December 1959.

A P P E N D I X A
P R O C E D U R E S A N D C R I T E R I A

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Field Methods and Procedures

Agencies which participated in the field sampling program during 1959 are listed below, together with the number of stations sampled by each agency:

<u>Agency</u>	<u>Number of stations sampled</u>
Department of Water Resources	146
Department of Fish and Game	1
United States Corps of Engineers	4
United States Bureau of Reclamation	22
City and County of San Francisco	<u>5</u>
Total	178

Water samples are collected in May and September for mineral, radiological and bacterial examination. Water samples are collected the other ten months for partial mineral analysis and bacterial examination. The samples collected monthly for bacterial examination (see Table A-1 for explanation) are kept in portable ice boxes until mailed to the laboratory in special containers. Every effort is made to get the samples to the laboratory as quickly as possible.

At the time the samples are collected for laboratory examination, field determinations are made for dissolved oxygen (DO), by the modified Winkler method; water temperature; and pH. Visual inspection is made of the stream or lake and the physical conditions are noted.

Where possible, the sampling stations have been selected so as to be at or near stream gaging stations so that gage heights can also

be recorded at the time the water samples are collected. Instantaneous stream discharges at the time of sample collection are then obtained.

Laboratory Methods and Procedures

Methods of mineral and bacterial analysis, in general, are those described in the American Public Health Association publication "Standard Methods for the Examination of Water and Sewage", 10th Edition, 1955. In some cases, the methods described in the following publications also have been employed:

U. S. Geological Survey, "Methods of Water Analysis", 1950.

California Department of Public Works, Division of Water Resources, "Methods of Analysis", October 1955.

Table A-1 indicates the constituents analyzed for in the various types of analysis performed in connection with this program.

TABLE A-1
Types of Analysis

Constituent	:Standard: : mineral:	Partial : mineral :	Bacterial	: Radiological
Specific Conductance	X	X		
pH ^a	X	X		
Total dissolved solids	X			
Percent sodium	X	X		
Hardness	X	X		
Turbidity	X	X		
Coliform			X	
Temperature ^b	X	X		
Dissolved oxygen ^b	X	X		
Calcium	X	X		
Magnesium	X	X		
Sodium	X	X		
Potassium	X			
Carbonate	X	X		
Bicarbonate	X	X		
Sulfate	X			
Chloride	X	X		
Nitrate	X			
Fluoride	X			
Boron	X	X		
Silica	X			
Phosphate	X			
Zinc ^c	X			
Iron ^c	X			
Copper ^c	X			
Aluminum ^c	X			
Manganese ^c	X			
Arsenic ^c	X			
Hexavalent chromium ^c	X			
Dissolved alpha				X
Solid alpha				X
Dissolved beta				X
Solid beta				X

- a pH is determined both in the field and in the laboratory.
b Field determination.
c These constituents are normally designated as heavy metals.

The methods and procedures of sample preparation and determination of radioactivity in surface waters is as follows:

I. Sample Preparation

- A. On receipt in the laboratory, each sample is well mixed, and two 250-ml portions taken. Each is acidified with a few drops of glacial acetic acid, and two drops of colloidal graphite suspension (Aquadag) added.
- B. Each portion is filtered under suction through a membrane ("Millipore") filter, which retains suspended particles of approximately 0.2 microns diameter and larger. Filters are treated with an antistatic preparation (Merix Anti-Static No. 79-OL) to eliminate any extraneous electrostatic charge.
- C. The filtrate is placed in a 250-ml volumetric flask, inverted, and the mouth placed in a 1-3/4" x 1/4" aluminum culture dish in a "chicken-feeder" type arrangement. The flask is supported by a ring stand; the dish rests on a hotplate adjusted so that the sample is taken to dryness at a temperature well below boiling.
- D. At this point, there are duplicate samples of both suspended solids and dissolved material from each original water sample ready for determination of radioactive content.

II. Determination of Radioactivity

- A. Two determinations are made on each sample, one for gross beta, one for gross alpha radioactivity. This represents a total of eight determinations for each original sample.
- B. Beta activity is determined with an internal gas flow counter operating in the proportional region, using argon-methane mixture as a flow gas. Background determinations are made before the first sample count each day, and then after each two sample counts throughout the day. Determinations of counter efficiency are made with a reference standard (thallium - 204) at least twice daily. Each determination of sample and background count rate is made for a total of 1000 counts.
- C. Alpha activity is determined with a scintillation counter utilizing an activated zinc sulfide phosphor. Sample, background and efficiency measurements are made in the same manner as are the beta measurements. Uranium 238 is used as an alpha reference standard. Each determination of sample and background count rate is made for a pre-set time of 32 minutes.

III. Calculations

- A. Results are expressed as micro-micro curies per liter ($\mu\mu\text{c/l}$). One micro-micro curie is equivalent to 2.22 disintegrations per minute. Four values are reported for each sample:
 - (a) beta activity in the solids retained on the filter,
 - (b) beta activity in the filtrate (dissolved material),
 - (c) alpha activity in the solids, and (d) alpha activity in the filtrate.
- B. Sample counts are corrected for background and geometric efficiency.
- C. Standard statistical procedures are utilized to compute the 0.9 error. The final result is expressed (symbolically) as $x \pm y \mu\mu\text{c/l}$. This means that in a series of determinations on the same sample, the value of x should fall between $x - y$ and $x + y$, 90% of the time.

Water Quality Criteria

Criteria used by the Department of Water Resources in the evaluation of the acceptability of water for the most common beneficial uses are described hereinafter. In general, the values presented herein should be considered only as guides to judgment, and not as absolute limiting standards.

Criteria for Drinking Water

Chapter 7 of the California Health and Safety Code contains laws and standards relating to domestic water supply. Section 4010.5 of this code refers to the drinking water standards promulgated by the United States Public Health Service for water used on interstate carriers. These criteria have been adopted by the State of California. They are set forth in detail in United States Public Health Report, Volume 61, No. 11, March 15, 1946, re-issued in March 1956.

According to Section 4.2 of the above-named report, chemical substances in drinking water, either natural or treated, should not exceed the concentrations shown in Table A-2.

TABLE A-2

LIMITING CONCENTRATIONS OF MINERAL
CONSTITUENTS FOR DRINKING WATER

United States Public Health Service
Drinking Water Standards, 1946

Constituent	:	Parts per million
<u>Mandatory</u>		
Fluoride (F)		1.5
Lead (Pb)		0.1
Selenium (Se)		0.05
Hexavalent chromium (Cr ⁺⁶)		0.05
Arsenic (As)		0.05
<u>Nonmandatory but Recommended Values</u>		
Iron (Fe) and manganese (Mn) together		0.3
Magnesium (Mg)		125
Chloride (Cl)		250
Sulfate (SO ₄)		250
Copper (Cu)		3.0
Zinc (Zn)		15
Phenolic compounds in terms of phenol		0.001
Total solids - desirable		500
- permitted		1,000

Interim standards for certain mineral constituents have recently been adopted by the California State Board of Public Health. Based on these standards, temporary permits may be issued for drinking water failing to meet the United States Public Health Service Drinking Water Standards, provided the mineral constituents in the following tabulation are not exceeded.

UPPER LIMITS OF TOTAL SOLIDS AND SELECTED MINERALS IN
DRINKING WATER AS DELIVERED TO THE CONSUMER

	<u>Permit</u>	<u>Temporary Permit</u>
Total solids	500 (1000)*	1500 ppm
Sulfates (SO ₄)	250 (500)*	600 ppm
Chlorides (Cl)	250 (500)*	600 ppm
Magnesium (Mg)	125 (125)*	150 ppm

* Numbers in parentheses are maximum permissible, to be used only where no other more suitable waters are available in sufficient quantity for use in the system.

The California State Board of Health recently has defined the maximum safe amounts of fluoride ion in drinking water in relation to mean annual temperature.

<u>Mean annual temperature in °F</u>	<u>Mean monthly maximum fluoride ion concentration in ppm</u>
50	1.5
60	1.0
70 - above	0.7

The relationship of infant methomoglobinemia (a reduction of oxygen content in the blood, constituting a form of asphyxia) to nitrates in the water supply has led to limitation of nitrates in drinking water. The California State Department of Public Health has recommended a tentative limit of 10 ppm nitrogen (44 ppm nitrates) for domestic waters. Water containing higher concentrations of nitrates may be considered to be of questionable quality for domestic and municipal use.

Limits may be established for other organic mineral substances if their presence in water renders it hazardous, in the judgment of state or local health authorities.

An additional factor with which water users are concerned is hardness. Hardness is due principally to calcium and ~~magnesium~~ salts and is generally evidenced by inability to develop suds when using soap. The United States Geological Survey has suggested the following four degrees of hardness:

TABLE A-3
Hardness Classification of Waters
U. S. Geological Survey

Range of hardness in parts per million	:	Relative classification
0 - 55		Soft
56 - 100		Slightly hard
101 - 200		Moderately hard
Greater than 200		Very hard

According to the International Commission on Radiological Protection¹, tentatively concurred in by the National Committee on Radiation Protection², if the Radium - 226 and Radium - 228 activity in water is substantially less than 10 $\mu\text{c}/\text{l}$, the maximum permissible concentration of otherwise unidentified radionuclides in water for individuals in the population at large may be considered to be 100 $\mu\text{c}/\text{l}$.

For the purposes of the environmental survey of surface water made for this report, it has been assumed that the total dissolved and solid alpha activity is derived from Ra^{226} and Ra^{228} .

1 "Report on Decisions of the 1959 Meeting of the Interantional Committee on Radiological Protection (ICRP)". Radiology, Vol. 74, No. 1, January 1960, pp. 116-119.

2 Somatic Radiation Dose for the General Population, Ad Hoc Committee of the National Committee on Radiation Protection and Measurements. Science, Vol. 131, No. 3399, February 19, 1960, pp. 482-486.

During the 1959 reporting year, the highest alpha activity observed in monitored surface waters was 1.86 $\mu\text{c}/\text{l}$. Consequently, it is believed that the maximum permissible concentration of 100 $\mu\text{c}/\text{l}$, as recommended by the I.C.R.P., was met by all stations sampled in the Surface Water Monitoring Program during 1959.

Criteria for Irrigation Water

Because of the diverse climatological conditions, crops, soils, and irrigation practices in California, criteria which may be set up to evaluate the suitability of water for irrigation use must necessarily be of a general nature, and judgment must be used in their application to individual cases. Suggested limiting values for total dissolved solids, chloride concentration, percent sodium and boron concentration for three general classes of irrigation water are shown in Table A-4.

TABLE A-4

QUALITATIVE CLASSIFICATION OF IRRIGATION WATERS

	Class 1	Class 2	Class 3
	Excellent to good	Good to injurious	Injurious to
			unsatisfactory
Chemical properties	(Suitable for most	(Possibly harmful	(Harmful to
	plants under any	for some crops	most crops and
	conditions of soil	under certain	unsatisfactory
	and climate)	soil conditions)	for all but the
			most tolerant)
Total dissolved solids			
In ppm	Less than 700	700 - 2,000	More than 2,000
In conductance, $EC \times 10^6$	Less than 1,000	1,000 - 3,000	More than 3,000
Chloride ion concentration			
In milliequivalents			
per liter	Less than 5	5 - 10	More than 10
In ppm	Less than 175	175 - 350	More than 350
Sodium in percent of			
base constituents	Less than 60	60 - 75	More than 75
Boron in ppm			
	Less than 0.5	0.5 - 2.0	More than 2.0

Criteria for Industrial Water

The water quality criteria for the diversified uses of water in industry range from the exacting requirements for make-up water for high pressure boilers to the minimum requirements for water washdown and metallurgical processing.

Because of the large number of industrial uses of water and widely varied quality requirements, it is practicable to suggest only very broad criteria of quality. These variable conditions make it desirable to consider water quality requirements in broad and general terms only, and, where possible, for groups of related industries rather than individually. The general quality requirements of several individual and major groups of water uses are listed in Table A-5.

The values shown in this table are those suggested in the Progress Report of the Committee on Quality of Tolerance of Water for Industrial Uses in the Journal of the New England Water Works Association, Volume 54, 1940.

Criteria for Fish and Aquatic Life

Water of suitable quality and quantity is a fundamental requirement for the existence of an abundant supply of fish and aquatic life. It is very important that water quality conditions be such as to maintain an abundant supply of food required by fish and other desirable forms of aquatic life. Streams utilized for the propagation of fish and aquatic life should be free of toxic or harmful concentrations of mineral and organic substances and excessive turbidity. Extensive field and laboratory studies conducted by the United States Fish and Wildlife Service show that, among other things, the water in streams supporting a mixed fauna of fresh water fish such as bluegill, bass, crappie and catfish should have the following properties:

- (a) Dissolved oxygen not less than 5 ppm (at least 6 ppm for Salmonids),
- (b) pH range between 6.5 and 8.5,
- (c) Ionizable salts, as indicated by conductivity, between 150 and 500 micromhos at 25° Centigrade, and in general not exceeding 1,000 micromhos,
- (d) Ammonia not exceeding 1.5 ppm.

Mineral salts of high toxicity to fish are those of silver, mercury, copper, zinc, lead, cadmium, nickel, trivalent and hexavalent chromium, and others. Some pairs of toxicants, such as copper and zinc (also copper and cadmium, nickel and zinc) are far more toxic when combined than when they occur individually. Other toxic substances, when combined,

TABLE A-5

WATER QUALITY TOLERANCE FOR INDUSTRIAL USES^a

Allowable limits in parts per million

Use	Turbidity	Color	Hardness as CaCO ₃	Iron ^b as Fe	Manganese as Mn	Total solids	Alkalinity as CaCO ₃	Odor, taste	Hydrogen sulfide	Miscellaneous Requirements	
										Health	Other
Air conditioning	---	---	---	0.5	0.5	---	---	Low	1	---	No corrosiveness, slime formation
Baking	10	10	---	0.2	0.2	---	---	Low	0.2	Potable ^b	---
Brewing	---	---	---	---	---	---	---	---	---	---	---
Light Beer	---	---	---	0.1	0.1	500	75	Low	0.2	Potable ^b	NaCl less than 275 ppm (pH 6.5-7.0).
Dark Beer	---	---	---	0.1	0.1	1,000	150	Low	0.2	Potable ^b	NaCl less than 275 ppm (pH 7.0 or more)
Canning	---	---	---	---	---	---	---	---	---	---	---
Legumes	10	---	25-75	0.2	0.2	---	---	Low	1	Potable ^b	---
General	10	---	250	0.2	0.2	---	---	Low	1	Potable ^b	---
Carbonated beverages	2	10	---	0.2	0.2	850	50-100	Low	0.2	Potable ^b	---
Confectionery	---	---	---	0.2	0.2	100	---	Low	0.2	Potable ^b	Organic color plus oxygen consumed less than 10 ppm.
Cooling	50	---	50	0.5	0.5	---	---	---	---	---	pH above 7.0 for hard candy.
Food: General	10	---	---	0.2	0.2	---	---	Low	---	Potable ^b	No corrosiveness, slime formation.
Ice	5	5	---	0.2	0.2	---	---	Low	---	Potable ^b	---
Laundering	---	---	50	0.2	0.2	---	---	---	---	Potable ^b	SiO ₂ less than 10 ppm.
Plastics, clear,	---	---	---	---	---	---	---	---	---	---	---
Uncolored	2	2	---	0.02	0.02	200	---	---	---	---	---
Paper and pulp:	---	---	---	---	---	---	---	---	---	---	---
Groundwood	50	20	180	1.0	0.5	---	---	---	---	---	No grit, corrosiveness.
Draft pulp	25	15	100	0.2	0.1	300	---	---	---	---	---
Soda and sulfide	15	10	100	0.1	0.05	200	---	---	---	---	---
High-grade	---	---	---	---	---	---	---	---	---	---	---
Light papers	5	5	50	0.1	0.05	200	---	---	---	---	---
Rayon (viscose):	---	---	---	---	---	---	---	---	---	---	---
Pulp production	5	5	8	0.05	0.03	100	total 50; hydroxide 8	---	---	---	Al ₂ O ₃ less than 8 ppm, SiO ₂ less than 25 ppm, Cu less than 5 ppm.
Manufacture	0.3	---	55	0.0	0.0	---	total 135; hydroxide 8	---	---	---	pH 7.8 to 8.3
Tanning	20	10-100	50-135	0.2	0.2	---	---	---	---	---	---
Textiles: General	5	20	---	0.25	0.25	---	---	---	---	---	Constant composition. Residual alumina less than 0.5 ppm.
Dyeing	5	5-20	---	0.25	0.25	200	---	---	---	---	---
Wool scouring	---	---	---	1.0	1.0	---	---	---	---	---	---
Cotton bandage	5	5	---	0.2	0.2	---	---	Low	---	---	---

a-Moore, E. W., Progress Report of the Committee on Quality Tolerances of Water for Industrial Uses: Journal New England Water Works Association, Volume 51, Page 271, 1940.

b-Potable water, conforming to U. S. P.H.S. standards, is necessary.

c-Limit given applies to both iron alone and the sum of iron and manganese.

neutralize each other through antagonism or chemical reaction (e.g., free cyanide combines with toxic heavy metal cations, such as nickel and copper ions, to form relatively harmless metallocyanide complexes).

The increasing use of household and industrial detergents, as well as the expansion in the manufacture and use of agriculture insecticides, poses serious hazards to fish and aquatic life. Preliminary studies, for example, indicate that one of the most common household detergents is lethal to relatively hardy fish at very low concentrations. This detergent was lethal to fish in fresh water at concentrations below 0.1 ppm and below 0.005 ppm in salt water. The increase in toxicity in salt water can probably be attributed to the fact that marine fishes must ingest water to maintain their osmotic balance.

Development and use of water resources, including the construction of dams for storage of water, frequently affects water temperatures which in turn affect fish and other aquatic life. Optimum water temperatures for cold water fish, such as trout and salmon, normally lie between 32° and 65° Fahrenheit. The cold water species are generally intolerant of temperatures above 75° Fahrenheit and will seek the lower temperature where possible. Warm water fish such as minnows, carp, catfish, perch, sunfish, and bass normally live in water having temperatures ranging from near 32° to 86° Fahrenheit. Acclimatization enables certain warm water species to live in waters having temperatures as high as 90° Fahrenheit, although they will migrate, where possible, to waters below 86° Fahrenheit.

A P P E N D I X B

BASIC DATA

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TABLE B-1
ANALYSES OF SURFACE WATER

THE FIFTH COASTAL REGION (NO. 1)

ANTYLOPE CREEK NEAR TENNANT (STA. 10)

[illegible]

F. 0.13 gH

b. Laboratory pH

Sum of calcium and magnesium in ppm.

Iron (Fe) aluminum (Al) organic (As) copper (Cu) lead (Pb) and hexavalent chromium (Cr⁶⁺), reported here as 0 except as shown.

^a Derived from conductivity vs TDS curves

(f) Determined by addition of analyzed constituents

Geometric determination

b. Annual median and range.

Murrel analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Health District, City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Imperial Irrigation District, California Department of Water Resources (CDWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

BIG RIVER NEAR MOUTH (STA. 6c)

Date and time sampled P.S.T.	Discharge Time in site in days	Specific Conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent sodium in ppm	Hardness as CaCO ₃ Total N.C. ppm	Total hardness by - Calcium by - MPN/ml	Analyzed by	
			Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Other constituents					
1959																		
1/6	Not Available	143	14	4.1	9.8	2.0	0.0	57	15	8.5	1.0	0.2		99	28	52	5	USGS
1/6			14	7.34	0.43	0.05	0.05	0.23	0.31	0.24	0.02	0.01						
2/5	200 (est.)	157	14	6.0	8.6	1.0	0.0	78	4.8	7.5	0.2	0.1	1.8	99	23	62	0	
2/30			14	0.75	0.37	0.03	0.05	1.28	0.13	0.21	0.00	0.01						
3/3	Not Available	146	14	6.1	9.2	0.7	0.0	74	9.6	7.2	0.0	0.0	0.0	103	25	60	0	
3/3			14	0.70	0.30	0.02	0.00	1.21	0.26	0.20	0.00	0.00						
4/1	Not Available	120	12	3.8	7.5	0.8	0.0	54	5.8	6.5	0.9	0.0	0.0	89	27	44	0	
4/1			12	0.70	0.33	0.02	0.00	0.79	0.12	0.12	0.00	0.00						
5/4	100 (est.)	166	17	4.5	8.7	1.2	0.0	70	7.2	8.0	0.1	0.0	1.8	103	23	61	0	
5/4			17	0.75	0.39	0.03	0.00	1.29	0.12	0.23	0.00	0.00						
6/1	50 (est.)	210	21	7.4	12	1.3	0.0	114	9.0	7.0	0.0	0.1	0.37	138	24	83	0	
6/1			21	1.05	0.52	0.03	0.00	1.27	0.19	0.20	0.00	0.01						
7/13	25 (est.)	216	22	7.3	12	1.6	0.0	118	9.0	7.0	0.2	0.0	0.4	130	23	85	0	
7/13			22	1.10	0.52	0.04	0.00	1.23	0.08	0.20	0.00	0.00						
8/3	20 (est.)	213	22	6.8	13	1.9	0.0	117	9.0	7.8	0.0	0.2	0.5	133	25	83	0	
8/3			22	0.56	0.57	0.05	0.00	1.22	0.12	0.22	0.00	0.01						
9/7	10 (est.)	211	21	6.9	14	1.6	0.0	112	9.0	10	0.0	0.1	0.5	132	27	81	0	
9/7			21	1.05	0.61	0.04	0.00	1.24	0.12	0.28	0.00	0.01						
10/5	8 (est.)	206	30	3.6	14	2.0	0.0	122	1.0	16	0.3	0.1	1.8	145	25	90	0	
10/5			30	1.50	0.61	0.05	0.00	2.00	0.08	0.45	0.00	0.01						
11/2	5 (est.)	233	24	8.8	16	2.4	0.0	126	1.4	8.5	0.0	0.0	0.5	154	26	96	0	
11/2			24	0.72	0.70	0.06	0.00	2.07	0.29	0.24	0.00	0.00						
12/8	39 (est.)	235	25	8.1	14	2.8	0.0	127	8.0	14	0.0	0.0	0.4	152	23	96	0	
12/8			25	0.67	0.61	0.07	0.00	2.08	0.17	0.39	0.00	0.00						

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in gpm

d Iron (Fe), aluminum (Al), uranic (As), copper (Cu), lead (Pb), manganese (Mn), and hazardous chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBOPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

SOUTHERN CALIFORNIA REGION (NO. 1)

BUTTE CREEK NEAR MACDOUGL, (STA. 1d)

Date and time of day P.S.T.	Discharge in cfs	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Hardness as CaCO ₃ ppm	Turbid- ity ppm	Analyzed by																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																										
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)					Other constituents																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																									
1/5/59		15	10.7	88	69.8	7.3	6.8	2.4	4.4	2.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

• Daily Mean

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and barium (Ba) reported here on 0.0 except as shown

e Derived from conductivity vs TDS curves

f Determined by addition of standard

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, at United States Public Health Service

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County, California, Central District (SBCFCD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Public Health (LADPH), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated

TABLE B-1

NORTH COASTAL REGION (NO. 1)

THE NEAR DOB BYOG (STA. 54)

Date and time of day and P.S.T.	Oversize Temp in cts	Dissolved oxygen in % Sat	Specific Gravity at 25°C	Mineral constituents in parts per million													Total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Turbid- ity in ppm	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)	Other constituents				
1959																				
1/7 1440	370	58 10.6	103	7.4 ^h	4.1 0.70	3.1 0.31	4.4 0.19	1.0 0.03	0.0 0.00	54 0.89	12 0.25	4.5 0.13	1.3 0.02	0.1 0.01	0.2 0.01	10	79 ^f 15 58	8	U305	
2/17 1450	45	11.690		7.4 ^h	3.9 0.62	3.9 0.32	4.1 0.19	1.6 0.04	0.0 0.00	52 0.85	11 0.23	2.2 0.06	0.1 0.01	0.1 0.01	0.1 0.01	10	71 ^f 16 46	3		
3/5 1230	52	1.034		7.7 ^h	4.6 0.70	3.8 0.38	4.4 0.19	0.8 0.02	0.0 0.00	48 0.75	11 0.23	2.8 0.08	0.0 0.00	0.1 0.00	0.1 0.00	11	78 ^f 15 54	6		
4/9 1330	64	176		8.2 ^h	5.8 1.00	4.8 0.48	5.4 0.23	0.7 0.02	0.0 0.00	87 1.00	8.6 0.78	3.0 0.00	0.0 0.00	0.1 0.01	0.2 0.01	11	98 ^f 13 74	3		
5/14 1550	68	8.1	68	8.2 ^h	7.2 1.35	6.9 0.59	6.0 0.36	1.1 0.03	0.0 0.00	120 1.97	7.7 0.15	5.0 0.11	0.0 0.00	0.0 0.00	0.3 0.00	9.0	123 ^f 13 97	0 1		
6/10 0830	85	26.6	64	8.2 ^h	8.6 1.35	8.6 0.71	8.4 0.37	1.6 0.04	0.0 0.00	128 2.10	17 0.35	6.0 0.17	0.0 0.00	0.1 0.01	0.4 0.01	11	145 ^f 14 108	3		
7/14 1140	90	8.1	80	8.2 ^h	8.4 1.35	8.4 0.59	10.1 0.69	1.4 0.04	0.0 0.00	122 2.00	17 0.35	5.8 0.16	0.2 0.01	0.1 0.01	0.3 0.01	4.2	134 ^f 17 102	2		
8/5 1315	116	2.5	80	9.4	11.6	23.0 ^c	11.4 ^c	0.0 ^c	0.0 ^c	118 1.84	17 0.35	6.0 0.14	0.0 0.00	0.1 0.01	0.4 0.01	13	138 ^f 19 100	8 10		
9/15 0800	89	41.2	63	8.6	89	9.1 1.25	10.4 ^c	1.4 ^c	0.0 ^c	116 1.90	17 0.35	10 0.28	0.0 0.00	0.0 0.00	0.4 0.00	13	145 ^f 20 100	5 1		
10/7 0820	87	17.8	56	9.1	87	8.1 ^h	11.4 ^c	1.8 ^c	0.0 ^c	145 2.35	17 0.35	6.0 0.17	0.0 0.00	0.8 0.01	0.8	158 ^f 17 119	0 3			
11/5 0960	91	12.4	49	10.4	91	7.7 ^h	10.4 ^c	1.8 ^c	0.0 ^c	144 2.31	17 0.35	7.5 0.21	0.0 0.00	0.8 0.01	0.8	158 ^f 16 118	2 2			
12/9 1528	104	9.6	40	11.2	104	8.2 ^h	12 0.52	1.8 ^c	0.0 ^c	146 2.39	17 0.35	9.0 0.28	0.0 0.00	1.1 0.01	1.1	168 ^f 16 136	16 10			

g. Field pH.

In Laboratory pH.

Sum of calcium and magnesium in eqm.

Copper (Cu), calcium (Ca), magnesium (Mg), iron (Fe), zinc (Zn), manganese (Mn), cobalt (Co), nickel (Ni), chromium (Cr⁺⁶), reported here as 0.0 except as shown.

d Iron (Fe), aluminum (Al), organic (AS), copper (Cu), and zinc (Zn) concentrations.

e Derived from conductivity vs IUS curves.

f Determined by addition of analyzed constituents.

g. Gravimetric determination.

Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

UNITED STATES GEOLOGICAL SURVEY, UNITED STATES DEPARTMENT OF THE INTERIOR, BUREAU OF RECLAMATION (USBR); UNITED STATES PUBLIC HEALTH SERVICE (USPHS); SON BERNARDINO COUNTY FLOOD

[illegible]

Control District (S&B Co.), metropolitan water district serving Los Angeles County, California.
D.L.H. = Lohmeyer Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE 3-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

ELZ RIVER AT SCOTIA (STA. 6)

Date and time sampled P.S.T.	Dissolved oxygen in % in air	Dissolved oxygen in ppm	Specific conductance at 25°C %S ₂₅	Major constituents in parts per million										Total dissolved solids in ppm	Percent dissolved solids on CO ₂ in ppm	Total hardness in ppm	Total calcium in ppm	Total magnesium in ppm	Assigned by			
				equivalents per million																Other constituents		
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)								Residue (R)	
1959																						
1/21	5.140	47	11.5	98	165	7.4	1.38 ^c	0.0	78	5.8	0.15			0.1	95 ^e	13	75	11	30	Median 2.3	USGS	
1/30	6.350	49	11.2	97	152	7.5	1.38 ^c	0.0	73	4.0	0.11			0.0	87 ^e	13	66	6	9	Maximum 600.		
2/4	8.090	60	10.5	104	245	7.3	1.28 ^e	0.0	112	12	0.34			0.1	141 ^e	20	99	7	60	Maximum 40.005		
3/3	6.080	60	10.4	104	143	7.4	1.38 ^c	0.0	72	3.0	0.08			0.0	82 ^e	12	68	9	50			
4/7	3.830	61	10.2	102	186	7.4	1.75 ^c	0.0	95	4.0	0.1	0.0	0.0	0.1	111 ^f	10	87	8	5			
1/30	960	62	10.0	102	231	7.5	2.10 ^e	1.28	210	6.0	0.17			0.2	133 ^g	12	122	15	2			
7/14	174	78	12.3	148	263	8.2	2.38 ^e	0.17	213	8.5	0.24			0.1	151 ^g	15	119	4	1			
11/5	109	69	11.1	122	215	8.0	2.18 ^e	0.13	179	7.5	0.21			0.1	124 ^g	16	109	23	6			
1/30	76	75	11.9	139	243	8.3	2.4	1.6	120	8.2	0.23			0.2	136 ^f	18	106	8	15			
11/5	168	63	10.8	111	312	8.1	2.88 ^e	0.0	161	8.4	0.24			0.2	179 ^g	14	144	12	2			
11/4	174	54	10.6	98	341	7.7	3.13 ^e	0.0	180	10	0.28			0.1	196 ^g	13	159	11	5			
12/9	126	48	11.1	98	318	7.6	3.06 ^e	0.0	168	8.0	0.23			0.2	183 ^g	14	150	12	10			
13/5																						

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

0.00

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Contaminant analysis by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

KEL RIVER, MIDDLE FORK AT DOS RIOS (STA. 5C)

Date and time of day P.S.T.	Discharge rate in cfs	Temp. in air °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Temp. of water °F	Assigned by						
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					Other constituents					
1/6/59	1.400	56	10.5	100	7.4 ^b	4.4	0.70	4.2	0.18	1.1	0.0	0.70	55	0.70	12	0.25	4.2	0.13	0.0	0.1	0.2	10		70 ^f	1.4	
1/7/59																										
2/17	8.670	60			7.6 ^f	4.1	0.70	3.1	0.03	0.6	0.0	0.70	54	0.70	15	0.31	2.2	0.0	0.0	0.0	0.0	2.9		71 ^f	1.1	
3/30																										
4/9	1.559	54			7.5 ^b	6.4	0.75	2.9	0.0	0.1	0.0	0.70	78	1.9	2.0	0.70	2.0	0.0	0.0	0.0	0.0	8.8		7 ^f	1.0	
5/14	0.08	64	7.6	79	7.7	1.8	0.70	1.6	0.13	0.8	0.0	0.70	70	9.6	2.2	0.70	2.2	0.0	0.0	0.0	0.0	7.7		70 ^f	1.1	
6/10	1.96	76	8.1	120	8.1 ^b	2.0	0.70	0.7	0.03	0.0	0.0	0.70	113	1.3	0.8	0.70	0.70	0.13	0.70	0.0	1.1	16		24 ^f	1.1	
7/14	24.1	74	7.6	87	8.1 ^b	1.1	0.70	0.7	0.03	0.1	0.0	0.70	95	17	4.1	0.70	0.13	0.70	0.0	0.0	1.1	16		24 ^f	1.1	
8/5	11.1	76	9.1	136	8.2 ^b	1.2	0.70	0.5	0.03	0.0	0.0	0.70	116	1.2	1.6	0.70	1.6	0.0	0.0	0.0	0.1	0.1		140 ^f	1.0	
9/15	1.4	68	9.2	100	7.6 ^b	1.0	0.70	0.7	0.03	1.8	0.0	0.70	115	1.3	2.0	0.70	0.20	0.0	0.0	0.0	0.1	0.1		20 ^f	1.0	
10/15	3.0	50	8.8	76	7.7 ^b	1.1	0.70	0.7	0.03	0.0	0.0	0.70	124	1.2	1.6	0.70	0.0	0.0	0.0	0.0	0.1	0.1		140 ^f	1.0	
11/7	2.1	60	1.0	91	8.1 ^b	1.1	0.70	0.7	0.03	0.0	0.0	0.70	115	1.3	2.0	0.70	0.20	0.0	0.0	0.0	0.1	0.1		20 ^f	1.0	
12/9	1.5	60			8.1 ^b	1.1	0.70	0.7	0.03	0.0	0.0	0.70	115	1.3	2.0	0.70	0.20	0.0	0.0	0.0	0.1	0.1		20 ^f	1.0	

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4. *Learn from the community & women* As

* Derived from product ID's as ID's were

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TABLE B-1
ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

ELC RIVER, SOUTH FORK NEAR KIRKLAND (STA. 7)

Date and time of day PST	Oscillator Temp in °F	Dissolved oxygen ppm	Specific Conductance at 25°C μ mho/cm	Mineral constituents in μ g/L												Other constituents	Total dissolved solids in ppm	Percent lead in ppm	Total hardness as CaCO ₃ ppm	Total Chloride as Cl ⁻ ppm	Analyzed by		
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Barium (Ba)	Silica (SiO ₂)								
1959																							
1/21	1,610	46	11.6	98	135	7.3												86 ^a	19	58	1	85	Median 2-3
1/230																							
2/6	1,650	48	11.3	97	121	7.5												72 ^a	17	52	4	8	Maximum 52
1800																							
3/3	1,510	52	11.1	100	133	7.2												70 ^a	18	54	0	45	Maximum <0.045
1840																							
4/6	1,277	59	9.3	92	132	7.3												76 ^a	15	58	4	50	
1600																							
5/5	402	61	9.0	91	170	7.3												101 ^f	16	71	0	2	
1130																							
6/1	180	74	8.1	94	189	7.3												112 ^a	16	89	5	2	
1845																							
7/14	68	69	7.7	85	225	8.0												133	17	101	0	1	
0905																							
8/4	37	70	7.9	88	223	7.9												132 ^a	17	98	0	5	
1130																							
9/8	32	70	8.1	90	246	7.9												136 ^f	17	107	0	2	
0930																							
10/6	52	60	8.1	81	264	7.5												156 ^a	16	117	1	1	
0940																							
11/3	48	53	11.2	102	266	7.9												156 ^a	17	114	0	3	
1505																							
12/9	44	41	11.9	93	262	7.9												155 ^a	14	124	11	10	
1100																							

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{+6}), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Temescal Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER

Date and time of analysis, P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm	pH (measured at 25°C)	Mineral constituents in equivalents per million										Total dissolved in ppm	Hardness as CaCO ₃ ppm	Temp. stability, ppm	Assigned by				
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					Boron (B)	Silica (SiO ₂)	Other constituents	
1959																				URGB	
1/19	2,800	41	191	7.5			18	0.0	72			5.5				136 ^a	41	56	0	15	Median 96.
1/30																					
2/11	3,130	39	170	7.2			12	0.0	74			4.8			0.0	122 ^a	31	59	0	10	Maximum 7,000.
6/30																					
3/28	2,250		197	7.0			14	0.0	77			2.8			0.0	140 ^a	38	66	3	4	Minimum 0.20
4/9	1,445	53	181	7.3			12	0.0	72			3.0			0.1	129 ^a	30	62	3	0	
5/7	645	54	241	7.7			20	0.0	97			6.0			0.1	153 ^a	36	74	0	2	
1/30																Fe 0.03 Al 0.14 d PO ₄ 0.50					
6/16	1,500	63	204	7.4			16	0.0	98			3.5			0.1	146 ^a	32	74	0	1	
9/30																					
7/14	1,560	70	152	7.5			13	0.0	78			4.8			0.1	108 ^a	36	51	0	1	
8/6																					
8/4	1,140	6.9	170	7.8			12	0.0	91			6.0			0.1	121 ^a	27	74	0	2	
10/6																					
2/9/80	2,980	63	173	7.2			15	2.6	81			8.0			0.1	134 ^a	36	55	0	6	
10/1/80																PO ₄ 0.75 Al 0.07 d Zn 0.04					
10/1/80	2,550	55	181	7.2			10	7.5	86			5.2			0.1	134 ^a	38	56	0		
11/4																					
11/30	2,550	51	153	7.0			14	0.0	74			4.8			0.1	109 ^a	26	49	0	6	
12/3																					
11/30	2,350	43	145	7.0			14	0.0	72			5.0			0.1	104 ^a	37	52	0	20	

Field pH.

Laboratory pH.

Sum of calcium and magnesium in eqm.

Sum of calcium and magnesium in epm.

Iron (Fe), aluminum (Al), organic (As), copper

Derived from conductivity vs. TDS curves

Determined by addition of c

g. Gravimetric determination.

Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

Mineral analyses made by United States Geological Survey (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood

Contract: City of Long Beach, Department of Public Health (LADPH); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); Metropolitan Water District of Southern California (MWD); San Bernardino County Flood Control District (SBFCFD).

Control Chart (CCL) and Control Chart for Defects (CCD) were used. The CCL was used for the Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE 3-1

ANALYSES OF SURFACE WATER

NORTE COSTAL REGION (NO. 1)

Klamath River Near Klamath (STA. 3)

Date and time of sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in %	Specific conductance at 25 °C (microhm/cm)	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Turbidity in ppm	Coliforms in MPN/ml	Analyzed by	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)							Other constituents
1/29	20,600	44	12.4	101	136	7.3 ^d				0.0	70	1.15	3.5			0.1		88 ^d	16	60	3	25	Median 6.2
1/30	23,900	44	12.4	101	131	7.5 ^d				0.0	67	1.10	2.5			0.0		85 ^d	14	56	1	4	Maximum >7,000
2/3	30,800	47	11.5	97	129	7.5 ^d				0.0	69	1.13	2.5			0.0		84 ^d	12	60	3	60	Minimum 0.05
1000	30,400	56	10.1	96	103	7.5 ^d				0.0	57	0.93	2.0			0.0		67 ^d	9	54	7	85	
1600	13,200	54	9.1	84	125	7.9 ^d				0.0	66	1.08	3.5			0.0		81 ^f	13	56	2	10	
1650	9,880	62	9.8	101	125	7.5 ^d				0.0	70	1.15	2.5			0.1		81 ^d	13	65	8	5	
1800	4,040	72	8.0	91	164	7.8 ^d				0.0	92	1.51	4.8			0.1		106 ^d	19	72	0	1	
7/15	2,930	73	7.9	91	175	7.5 ^d				0.0	96	1.57	3.2			0.1		113 ^d	18	76	0	5	
8/5	2,350	72	9.6	108	184	7.8 ^d				0.0	100	1.64	6.0			0.1		129 ^f	22	80	0	15	
10/1	3,900	58	10.0	97	197	7.9 ^d				0.0	101	1.56	5.4			0.1		136 ^d	25	78	0	2	
11/20	3,000	50	10.3	91	202	7.5 ^d				0.0	105	1.72	6.5			0.1		131 ^d	23	80	0	5	
0805	2,960	40	10.6	86	188	7.5 ^d				0.0	98	1.51	6.5			0.1		126 ^d	23	85	5	20	
12/10																							
0900																							

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituent.

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR), as indicated.

TABLE 3-1

ANALYSES OF SURFACE WATER

BAYVIEW CANTAL REGION (NO. 1)

ELAMITE RIVER NEAR DELTA VALLEY (STA. 26)

Date and time sampled P.S.T.	Discharge in cfs in ft ³ /sec	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million										Total solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ in ppm	Turb- id- ity in ppm	Asphyx- ia in ppm
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					
1/9/59	4,792	44	240	16	8.9	10	1.6	0.0	1.26	11	5.3	2.3	0.2	134	29	77	1	
2/5 1/50	5,150	46	120	17	9.7	14	1.6	0.0	1.00	29	4.5	1.0	0.2	145	29	6	0	
3/5 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
4/7 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
5/13 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
6/14 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
7/13 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
8/14 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
9/8 1/11	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
10/13 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
11/13 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	
12/13 1/50	4,730	50	107	14	6.1	6	1.2	0.0	84	9.6	4	0.3	0.0	100	19	88	1	

a. F is pH

b. Laboratory pH

c. Sum of calcium and magnesium in ppm

d. Iron (Fe), aluminum (Al), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺) reported here as 0 except as shown

e. Determined by conductivity as TDS curves

f. Determined by addition of unreacted constituents

g. Gravimetric determination

h. Asphyxiation and oxygen acceptor test. Collected from analyses of Bayview monthly samples made by California Department of Public Health, Division of Laboratories, United States Public Health Service

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch, USGS, United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service, JSDPS, San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health, LADPH, City of Long Beach, Department of Public Health (LADPH), Technical Testing Laboratories, Inc. (TTL), California Department of Water Resources (CDWR) as indicated

TABLE B-1
ANALYSES OF SURFACE WATER
NORTH COASTAL REGION (NO. 1)
KIAMATI RIVER AT SOMESBAR (STA. 2)

Date and time sampled (P.S.T.)	Discharge in cfs	Temp. in °F	Dissolved oxygen in ppm	% Sat	pH (at 25°C)	Mineral constituents in parts per million										Total dissolved solids in ppm	Percent barium on CaCO ₃ Total in ppm	Hardness as CaCO ₃ Total in ppm	Turbidity in NTU	Analyzed by																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																													
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)	Silica (SiO ₂)	Other constituents																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																										
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a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from survey of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR); as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER

NEW COASTAL REGION (NO. 1)

NAVARRO RIVER NEAR NAVARRO (STA. 86)

Date and time of day P.S.T.	Discharge Temp in °F in °C	Dissolved oxygen ppm	Specific conductance at 25 °C µmhos/cm	Calcium (Ca) ppm	Magnesium (Mg) ppm	Sodium (Na) ppm	Potassium (K) ppm	Carbon- dioxide (CO ₂) ppm	Bicarbonate (HCO ₃) ppm	Sulfate (SO ₄) ppm	Chloride (Cl) ppm	Nitrate (NO ₃) ppm	Fluoride (F) ppm	Boron (B) ppm	Silica (SiO ₂) ppm	Other constituents	Total dissolved solids in ppm	Total Hardness as CaCO ₃ ppm	Temp. in °C in °F	Coliforms per ml	Assigned by
1/20 1:40	24.0	46.11.3	211	22	7.8	10	1.1	0.0	105	12	9.0	0.7	0.1	0.1	17		132	87	1		Med. Lab.
2/5 1:00	26.0	48.11.3	205	21	7.5	9.2	1.0	0.0	102	11	8.5	0.1	0.1	0.1	19		128	83	0		Med. Lab.
1/2 1:00	27.0	48.7	198	18	7.8	8.5	1.1	0.0	94	7.7	8.0	0.6	0.1	0.1	21		121	77	0		Med. Lab.
1/15 1:55	27.0	48.7	207	21	7.2	9.1	1.2	0.0	101	10.6	7.5	0.8	0.0	0.0	17		125	80			Med. Lab.
1/15 1:40	27.0	48.7	211	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		126	80			Med. Lab.
1/15 1:30	27.0	48.7	201	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		151	108			Med. Lab.
1/15 1:00	27.0	48.7	205	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		163	114			Med. Lab.
1/15 1:15	27.0	48.7	210	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		157	108			Med. Lab.
1/15 1:00	27.0	48.7	208	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		160	113	0		Med. Lab.
1/15 1:00	27.0	48.7	210	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		159	108	3		Med. Lab.
1/15 1:00	27.0	48.7	210	21	7.8	9.5	1.2	0.0	102	10.6	7.5	0.8	0.0	0.0	17		166	118	3		Med. Lab.

F. O. J.

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1. The first step is to identify the problem or question that needs to be answered. This involves understanding the context and the specific requirements of the task.

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TABLE B-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

NOYO RIVER NEAR PORT BRAGO (STA. 10c)

Data collection sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent sulfate	Hardness on CaCO ₃ Total N.C. in ppm	Tur- bidity - Coliform ^h MPN/ml	Analyzed by I		
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)	Silica (SiO ₂)
1959																						USGS
1/22 0800	74	46	11.0	92	7.1	12	4.9	7.8	0.9	0.0	64	5.2	8.5	0.1	0.1	1.7		89 ^f	25	50	0	Median 0.56
2/5	124	46	11.3	95	7.2	12	3.8	8.5	0.6	0.0	59	4.8	8.8	0.0	0.1	2.0		88 ^f	28	45	0	Maximum 2.400
3/3 1000	142	48	10.9	94	7.1	7.6	5.6	7.0	0.6	0.0	53	5.8	6.5	0.6	0.1	2.1		81 ^f	26	42	0	Minimum 0.045
4/2 0830	244	54	10.4	97	7.2	12	2.9	7.1	0.8	0.0	56	3.8	5.5	0.4	0.0	1.9		80 ^f	26	42	0	
5/4 1645	32	59	9.9	97	7.2	14	4.4	9.0	0.7	0.0	71	5.8	7.8	0.1	0.0	1.9	Fe 0.06 Al 0.01 d PO ₄ 0.05	96 ^f	27	53	0	1
6/1 1600	18	60	9.9	99	7.2	15	4.7	9.8	1.2	0.0	77	5.4	9.0	0.0	0.0	1.7	Fe 0.03 Al 0.02 d PO ₄ 0.00	100 ^f	27	57	0	1
7/13 ^a 1700	6.6	70	9.0	100	7.3	18	9.0	8.5	3.1	0.0	84	13	82	1.0	0.1	1.8		240 ^f	58	82	13	
8/14 ^a 0730	3.8	63	9.6	99	7.2	18	10	63	3.9	0.0	82	15	29	0.0	0.1	1.8		267 ^f	60	88	21	
9/7 ^a 1850	2.9	67	9.3	100	7.3	56	118	912	52	0.0	84	17	1,760	1.3	0.1	1.9	Al 0.20 PO ₄ 0.00 d	2,980 ^f	74	624	555	4
10/5 ^a 1645	4.8	60	9.5	95	7.3	14	11	31	2.9	0.0	86	14	48	0.2	0.1	2.2		185 ^f	44	82	11	
11/3 1100	4.5	52	9.6	86	7.2	16	5.4	13	0.9	0.0	85	5.0	11	0.3	0.2	1.8		112 ^f	31	62	0	
12/9 0800	6.3	41	10.9	85	7.3	11	7.3 ^c	11	0.0	0.0	86	0.0	8.0	0.0	0.3	0.3		113 ^e	26	67	0	10

^a Sample Influenced by sea water, not included in evaluation.

* Sample influenced by sea water, not included in evaluation.

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by titration of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE A-1

ANALYSES OF SURFACE WATER

RIVER CANTAL MEDIAN (NO. 1)

OUTLET CREEK NEAR LORRAINE (TFA, 53)

Date and time sampled PST	Discharge in cfs at PST	Desired system pH	Specific conductance at 25°C µmhos/cm	parts per million										Total dissolved solids in ppm	Per cent solids in ppm	Hardness as CaCO ₃ in ppm	Tur- bidity in NTU	Asphyxiated by 1
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents			
1/29/59	1,730	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
1/7/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
2/17/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
3/1/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
3/15/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
3/29/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
4/12/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
4/26/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
5/10/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
5/24/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
6/7/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
6/21/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
7/5/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
7/19/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
8/2/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
8/16/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
8/30/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
9/13/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
9/27/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
10/11/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
10/25/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
11/8/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
11/22/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
12/6/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
12/20/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1
12/30/60	1,510	55	125	13	4.3	4.6	1.6	0.0	6.0	6.7	4.2	1.1	0.1	0.2	10	76 ^f	16	1

a. For light

b. Laboratory pH

c. Sum of calcium and magnesium in ppm

d. Iron (Fe), manganese (Mn), copper (Cu), and lead (Pb) concentrations less than 1.0 ppm are not reported here as they are negligible

e. Derived from conductivity in TDS units

f. Determined by titration in 100 ml of water

g. Gross water treatment

h. Annual and 10-day average

i. Maximum 10-day average

j. Maximum 10-day average

k. Maximum 10-day average

l. Maximum 10-day average

m. Maximum 10-day average

n. Maximum 10-day average

o. Maximum 10-day average

TABLE B-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

REDWOOD CREEK AT ORICK (STA. 3b)

[illegible]

a Field pH.

h) Laboratory of

6 Laboratory pH.

^c Sum of calcium and magnesium in epm.

d Iron (Fe), aluminum (Al), arsenic (As), cop

a Derived from conductivity vs TDS curves.

† Determined by addition of Ca^{2+} Determined by addition of
C₁₂-methyl- β -cyclodextrin.

^g Gravimetric determination.

TABLE B-1

ANALYSES OF SURFACE WATER

RIVER COASTAL REGION (NO. 1)

RUSSIAN RIVER AT QUINCYVILLE (STA. 10)

Date and time of day P.S.T.	Discharge in cfs	Temp. in °C	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Major constituents in equivalents per million										Total solids in ppm	pH	Hardness in ppm CaCO ₃	Total dissolved solids in ppm	Total solids in ppm	pH	Temperature in °C	pH	pH
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Barium (Ba)	Other constituents							
1959																							
1/29	1,130	52	11.1	41	200	7.3				136	273	8.8	0.0	0.0	0.0			136	136	136	136	136	136
1/30																							
2/1	1,300	52	11.3	40	200	7.1				129	271	8.1	0.0	0.0	0.0			129	129	129	129	129	129
2/2																							
2/3	2,030	52	9.5	39	201	7.1				132	276	7.6	0.0	0.0	0.0			132	132	132	132	132	132
2/4																							
2/5	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/6																							
2/7	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/8																							
2/9	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/10																							
2/11	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/12																							
2/13	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/14																							
2/15	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/16																							
2/17	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/18																							
2/19	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/20																							
2/21	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/22																							
2/23	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/24																							
2/25	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/26																							
2/27	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/28																							
2/29	1,000	50	11.3	40	194	7.1				131	275	6.9	0.0	0.0	0.0			131	131	131	131	131	131
2/30																							

a. P.S.T.

b. Laboratory pH

c. Sum of calcium and magnesium in ppm

d. Total P₂O₅ (phosphorus) in ppm

e. Determined from conductivity at 25°C

f. Determined by addition of analyzed constituents

g. Chloride concentration

h. Amount in water and range in ppm

i. Amount in water and range in ppm

j. Amount in water and range in ppm

k. Amount in water and range in ppm

l. Amount in water and range in ppm

m. Amount in water and range in ppm

n. Amount in water and range in ppm

o. Amount in water and range in ppm

p. Amount in water and range in ppm

q. Amount in water and range in ppm

r. Amount in water and range in ppm

TABLE B-1

ANALYSES OF SURFACE WATER

NORTE COASTAL REGION (NO. 1)

RUSSIAN RIVER NEAR REDLANDS (SFA. 9)

Date and time of day sampled P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm %Sat	Specific Conductance (microhm/cm at 25°C)	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total N.C. in ppm	Turbidity in ppm	Coliform MPN/ml	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						
1959																			USGS	
1/7 0930	50	10.1	90	7.3	1.56 ^c	0.30	6.9	0.00	0.00	7.5	0.21	0.2			112 ^d	78	16	16	Maximum 2,100.	
2/6 0830	64.1	10.1	90	7.5	2.28 ^c	0.36	8.2	0.00	0.00	5.5	0.16	0.06			146 ^d	114	7	5	Minimum 0.13	
3/2 1445	1,280	9.5	96	7.5	2.28 ^c	0.31	7.2	0.00	0.00	5.8	0.16	0.3			146 ^d	114	7	20		
4/1 1330	947	9.1	96	7.5	1.98 ^c	0.31	7.1	0.00	0.00	5.5	0.16	0.2			128 ^d	14	96	6	16	
5/11 1610	1085	9.0	107	7.9	2.25	1.35	8.4	1.4	0.00	8.6	0.18	0.1	0.5	1.3	160 ^d	12	130	0	30	
6/11 1145	145	8.5	97	7.7	2.65 ^c	0.38	8.8	0.00	0.00	7.0	0.20	0.5			165 ^d	13	130	0	10	
7/1 1445	170	8.5	90	7.7	2.34 ^c	0.38	8.8	0.00	0.00	4.8	0.14	0.4			146 ^d	14	117	0	10	
8/12 1545	163	8.5	102	7.9	2.28 ^c	0.38	8.7	0.00	0.00	4.5	0.13	0.6			142 ^d	15	112	0	3	
9/4 0630	200	7.6	83	7.7	1.15	0.39	8.5	1.5	0.00	7.2	0.15	0.0	0.4	1.1	135 ^d	15	107	0	3	
10/15 0900	313	8.0	83	7.5	2.06 ^c	0.31	7.2	0.00	0.00	6.0	0.17	0.4			132 ^d	13	103	0	8	
11/4 1615	313	8.3	84	7.5	2.38 ^c	0.43	1.9	0.00	0.00	1.4	0.39	0.4			204 ^d	23	142	0	5	
12/3 1530	337	10.6	96	7.9	2.13 ^c	0.33	7.7	0.00	0.00	5.2	0.13	0.5			134 ^d	14	105	0	3	

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity via TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range by United States Geological Survey, California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

RUSSIAN RIVER, EAST FORK AT TOPPER VALLEY POWERHOUSE (STA. 10A)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	%Sat	Specific conductance at 25°C	pH	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Per- cent sulfate Total N.C. ppm	Tur- bid- ity in ppm	Coliform ^b MPN/ml	Analyzed by 1
							Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silic- ic acid (SiO ₂)					
1959																			Median 6-2	USGS			
1/6 1120	312	42	11.3	89	148	7.5 ^b		5.6 0.24	0.00	0.00	72 1.18		5.0 0.11			0.3		62	3	95	Maximum 620		
2/6 1100	307	44	11.1	91	138	7.4 ^a		5.0 0.22	0.00	0.00	67 1.10		3.0 0.08			0.3		63	8	97	Minimum 0.045		
3/2 1115	163	47	11.1	94	114	7.3 ^a		3.3 0.14	0.00	0.00	56 0.92		3.0 0.08			0.1		66 ^a	11	54	8	70	
4/2 1100	212	54	9.9	92	135	7.3 ^a		4.3 0.19	0.00	0.00	70 1.15		3.0 0.08			0.3		81 ^e	14	48	1	40	
5/13 1245	75	62	8.6	88	155	7.8 ^b	4.7 0.19	5.4 0.23	0.3 0.01	0.00	83 1.36	6.3 0.13	2.8 0.08	0.0 0.00	0.1 0.01	0.3 0.01	Fe, 0.01; Al, 0.04; d PO ₄ , 0.05	91 ^f	15	67	0	2	
6/11 1100	302	50	9.3	82	126	7.4 ^a		4.2 0.18	0.00	0.00	71 1.16		3.5 0.10			0.2		76 ^e	13	62	4	20	
7/1 1220	307	59	8.9	86	137	7.4 ^a		4.4 0.19	0.00	0.00	76 1.25		2.5 0.07			0.2		82 ^e	14	60	0	0.8	
8/13 0840	380	67	7.7	83	148	7.5 ^a		4.7 0.20	0.00	0.00	89 1.34		2.5 0.07			0.3		89 ^e	13	66	0	3	
9/3 1325	260	70	8.0	89	160	7.4 ^a	6.0 0.19	5.3 0.23	1.1 0.03	0.00	87 1.43	5.8 0.12	3.1 0.09	0.0 0.00	0.0 0.00	7.2	Al, 0.01; Cu, 0.02; d PO ₄ , 0.00	91 ^f	14	72	1	2	
10/4 Dry																							
11/4 Dry																							
12/3 1305	16	46	10.5	90	208	7.3 ^a		7.4 0.32	0.00	0.00	116 1.99		5.2 0.15			0.7		125 ^e	15	94	0	3	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs. TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, San Bernardino County, California (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); California Water Resources Institute (CWRI); California Water Resources Institute (CWRI); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DMR); as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER

SOUTHEASTERN REGION (No. 1)

BAJAWA RIVER AT BANGKAR (Sta. 26)

Date and time of day of sample PST	Discharge in cfs	Temp in deg F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in —equivalents per million—										Total dis- solved solids in ppm	Per- cent total solids in ppm	Hardness as CaCO ₃ in ppm	Total Hard- ness in ppm	Accepted by PST	
						parts per million															
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)
1959 1/29	1,140	59	12.9	90	89.6	7.3	12 0.75	2.9 0.75	1.9 0.75	0.1 0.05	0.0 0.05	45 0.75	3.8 0.75	2.5 0.75	0.0 0.05	0.0 0.05		58	9	42	5
1/29	2,080	61	13.1	100	89.4	7.5	14 0.75	2.8 0.75	1.7 0.75	0.1 0.05	0.0 0.05	44 0.75	2.9 0.75	2.7 0.75	0.0 0.05	0.0 0.05		66	7	67	3
2/1	2,480	64	12.5	102	98	7.4	13 0.75	3.5 0.75	2.1 0.75	0.3 0.05	0.0 0.05	53 0.75	6.7 0.75	2.0 0.75	0.5 0.05	0.0 0.05		69	9	67	4
3/5	2,480	60	11.4	120	81.4	7.4	12 0.75	6.2 0.75	0.1 0.05	0.0 0.05	0.0 0.05	46 0.75	2.9 0.75	2.7 0.75	0.0 0.05	0.0 0.05		75	8	55	1
6/8	2,480	51	11.2	130	76.8	7.4	11 0.75	4.0 0.75	1.6 0.75	0.1 0.05	0.0 0.05	42 0.75	2.9 0.75	1.8 0.75	0.1 0.05	0.0 0.05		62	7	64	1
7/1	1,480	56	13.6	111	60.2	7.7	13 0.75	4.7 0.75	0.3 0.05	0.0 0.05	0.0 0.05	33 0.75	1.1 0.75	2.1 0.75	0.2 0.05	0.0 0.05		61	11	50	1
8/2	1,480	57	11.1	97	109	7.9	12 0.75	3.4 0.75	3.2 0.75	0.8 0.05	0.0 0.05	50 0.75	7.0 0.75	3.2 0.75	1.6 0.75	0.0 0.05		77	12	62	4
8/2	270	71	8.4	96	175	7.7	15 0.75	4.8 0.75	1.9 0.75	1.8 0.75	0.0 0.05	28 0.75	4.0 0.75	6.1 0.75	0.3 0.75	0.0 0.05		109	24	56	1
8/2	1,110	60	11.1	97	142	7.7	19 0.75	4.0 0.75	1.9 0.75	1.2 0.75	0.0 0.05	72 0.75	11 0.75	1.3 0.75	0.4 0.75	0.1 0.05		94	11	64	5
11/8	290	60	11.1	97	142	7.6	18 0.75	3.6 0.75	1.8 0.75	0.8 0.75	0.0 0.05	74 0.75	13 0.75	1.3 0.75	0.2 0.75	0.1 0.05		120	14	60	7
11/8	174	67	11.8	140	138	7.7	14 0.75	3.3 0.75	1.4 0.75	0.8 0.05	0.0 0.05	74 0.75	11 0.75	1.3 0.75	0.2 0.75	0.1 0.05		94	11	65	1
12/1	Not Sampled																				

PST and pH

1. Laboratory pH

2. Same 10 minutes after precipitation

3. Same 10 minutes after precipitation

4. Same 10 minutes after precipitation

5. Same 10 minutes after precipitation

6. Same 10 minutes after precipitation

7. Same 10 minutes after precipitation

8. Same 10 minutes after precipitation

9. Same 10 minutes after precipitation

10. Same 10 minutes after precipitation

11. Same 10 minutes after precipitation

12. Same 10 minutes after precipitation

13. Same 10 minutes after precipitation

14. Same 10 minutes after precipitation

15. Same 10 minutes after precipitation

1. Same 10 minutes after precipitation

TABLE 3-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

SCOTT RIVER NEAR FORT JONES (STA. 15)

Date sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm %Sat	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Temp. in °F	Coliform ^b MPN/ml	Analyzed by
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Brom- (Br)	Silico- nate (SiO ₂)				
1959																				USGS
1/	Ret Sampled																			
2/4	645	44	12.2	99	15	10	2.4	0.1	0.0	95	2.9	3.9	1.0	0.0	0.0	0.0	100	6	80	2
3/5	790	48	10.8	99	17	11	2.0	0.5	0.0	156	2.0	4.8	1.1	0.1	0.0	20	111	7	87	0
3/5	1,260	52	10.0	90	116	8.3	1.8	0.2	0.0	70	1.9	1.5	0.4	0.0	0.0	16	74	6	59	2
4/7	1600				0.75	0.28	0.08	0.01	0.00	1.15	0.04	0.06	0.01	0.00	0.00					
5/6	582	58	9.7	94	159	7.5	2.0	0.3	0.0	94	4.8	3.0	1.0	0.1	0.0	17	100	5	82	5
6/4	512	67	9.1	98	168	7.4	3.0	1.0	0.0	108	9.6	2.5	1.1	0.0	0.0	16	111	7	88	4
7/15	76	62	8.2	83	270	7.6	4.6	0.8	0.0	168	7.0	3.8	2.2	0.1	0.0	19	164	7	136	0
8/11	45	68	10.7	115	267	8.0	4.8	1.0	0.0	166	9.0	4.0	0.3	0.1	0.0	19	165	7	134	0
9/8	28	69	10.5	115	237	8.1	5.4	1.5	0.0	146	7.0	5.8	0.7	0.0	0.1	20	151 ^f	9	122	6
10/13	53	61	9.6	97	273	8.0	5.4	0.7	0.0	170	10	6.0	1.0	0.1	0.1	17	172	7	147	8
11/10	58	46	12.3	103	273	7.9	5.2	0.6	0.0	170	5.0	3.5	1.2	0.1	0.0	20	167	7	143	4
12/30	58	37	12.5	101	272	7.5	5.0	0.2	0.0	171	6.0	9.0	1.6	0.0	0.0	21	176	8	142	2

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 2-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

SMITH RIVER NEAR CRESCENT CITY (STA. 3A)

[illegible]

9 Field pH.

b Laboratory pH.

Sum of calcium and magnesium in eqm.

4. Iron (Fe), aluminum (Al), arsenic (As),

and Iron (Fe), aluminum (Al), arsenic (As), cop-

e Derived from conductivity vs. IUS curves.

Determined by addition of a

g Gravimetric determination.

h. Annual median and range, respectively. Calculated from analyses of duplicate monthly samples used by California Department of Public Health, Division of Laboratories, at United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDOH); Tempe, Arizona, Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

NORTH CENTRAL REGION (NO. 1)

TRINITY RIVER NEAR BOOPA (STA. 14)

Date and time sampled P.S.T.	Duration in site	Observed stream stage, ft	Specific conductance at 25°C, $\mu\text{mhos/cm}$	Metal constituents in parts per million										Total solids in ppm	Percent acid in ppm	Hardness as CaCO_3 , ppm	Turbidity, NTU	Coliforms, MPN/100 ml	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO_3)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Nitrate (NO_3)	Fluoride (F)	Barytes (Ba)	Silica (SiO_2)				
1959																			
1/30	6,750	43	12.1	97	139	7.4													USGS
7/30																			
8/1	7,300	44	12.6	103	137	7.6													
8/15																			
3/5	13,600	47	11.5	97	125	7.4													
8/60																			
4/8	10,700	50	10.8	96	108	7.5													
10/00																			
5/6	4,790	54	10.6	98	117	7.5													
12/00																			
6/3	4,030	62	9.3	95	105	7.4													
6/5																			
7/5	994	76	8.9	105	165	8.1													
1/00																			
8/6	528	73	8.6	99	202	7.8													
8/30																			
9/10	424	74	7.6	88	222	7.6													
9/10																			
10/8	562	59	9.5	93	222	7.9													
10/30																			
11/5	545	54	11.2	104	229	7.9													
13/00																			
12/10	517	46	11.8	99	243	7.9													
13/00																			

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in gpm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{6+}), reported here as 0.0 except as shown.

0.00

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Nitrogen analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 2-1

ANALYSES OF SURFACE WATER

NORTH COASTAL REGION (NO. 1)

VAN DYCKE NEAR BRIDGETTLE (STA. 26)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	pH at 25°C	Mineral constituents in equivalents per million										Total solids in ppm	Per- cent solid- in ppm	Hazardous as CECs Total N.C. ppm	Tur- bidity NTU	Coliform ^b MPN/ml	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- ate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents				
1/7/59	3,930	62	10.6	10.4	7.6															UD08
1/7/59	3,930	62	10.6	10.4	7.6	12	5.1	3.5	0.0	51	9.6	3.5	1.8	0.1	0.1		71 ^f	13	51	9
9/18/59	9,700	43		71.5	7.4	8.8	3.5	3.5	0.0	37	1.8	0.0	0.0	0.1	0.2		47 ^f	21	28	0
9/18/59	9,700	43		71.5	7.4	8.8	3.5	3.5	0.0	37	1.8	0.0	0.0	0.1	0.2		47 ^f	21	28	0
3/6/59	876	48		59.0	7.8	13	3.3	2.8	0.0	54	5.8	1.8	0.0	0.0	0.0		64 ^f	11	46	2
5/12/59	512	53		125	7.9	17	3.8	3.3	0.4	66	5.8	2.5	1.5	0.0	0.0		77 ^f	11	58	4
5/7/59	224	60	9.2	92	150	7.8	19	5.5	0.6	81	5.8	3.0	0.0	0.0	0.1	Al 0.03 PO ₄ 0.00 ^d	89 ^f	8	70	4
6/9/59	86	74	9.7	112	181	8.4	24	4.7	3.3	94	11	3.5	0.7	0.0	0.0		107 ^f	10	85	5
7/15/59	19	65	8.9	94	223	8.1	2.08 ^c	7.0	0.0	122	0.0	4.2	0.1	0.0	0.0		138 ^f	13	104	4
8/14/59	3.0	70	8.3	92	296	7.4	39	10	1.8	6	150	17	0.1	0.1	0.1		172 ^f	13	140	7
9/8/59	7	68	8.4	92	257	7.9	35	7.4	0.0	135	23	5.8	0.7	0.1	0.2		159 ^f	14	118	7
10/7/59	14	59	10.2	100	242	7.9	34	10	7.7	0.0	126	35	4.0	0.1	0.1		164 ^f	11	198	25
11/4/59	20	53	10.9	100	239	7.9	27	9.7	0.0	126	0.0	0.0	0.0	0.0	0.0		146 ^f	17	101	0
12/9/59	6.5	46	11.2	94	319	7.9	3.04 ^c	11	0.0	166	8.0	0.0	0.0	0.0	0.0		198 ^f	14	159	16
14/5									0.00	2.72	0.00	0.23	0.23							

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in gpm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Determined by conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-2
ANALYSES OF SURFACE WATER

SAN FRANCISCO BAY REGION (NO. 2)

ARMY DEL VALLE AT VETERANS ADMINISTRATION HOSPITAL (STA. 71)

Date and time sample collected P.S.T.	Discharge Temp in °F in op- eration	Dissolved oxygen ppm	Specific conductance micro-mhos at 25°C	pH	Mineral constituents in parts per million													Total dis- solved solid in ppm	Per- cent total solid in ppm	Hardness as CaCO ₃ ppm	Turbid- ity ppm	Coliform MPN/ml	Analyzed by																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																										
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					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)	Other constituents																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																
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a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-2

ANALYSES OF SURFACE WATER

SAN FRANCISCO BAY REGION (NO. 2)

CARQUIN? STRAITS AT MARTIN? (STA. 24a)

[illegible]

Table 1

15

1

2

10

TABLE B-2
ANALYSES OF SURFACE WATER
SAN FRANCISCO BAY ESTUARY (NO. 2)

COCOTE CREEK NEAR MARINER (STA. B2)

Date and time sampled P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm	Specific conductance in μ mhos/cm at 25°C	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent suspended solids in ppm	Hardness as CaCO ₃ in ppm	Turbidity in NTU	Analyzed by	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonates (CO ₃)	Bicarbonates (HCO ₃)	Sulfates (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)
1999																			USGS	
1/8 1350	22	54	11.4	105	7.9	2.78 ^c		14	0.61	0.0	154		11		0.0		139	13	1	Median 23.
2/10 0930	14	51	9.9	88	7.7	2.76 ^c		12	0.52	0.0	154		10		0.2		138	12	10	Maximum 660.
3/5 1705	9.1	56	10.8	102	7.9	3.02 ^c		13	0.57	0.0	147		11		0.1		151	30	35	Minimum 0.23
4/1 1600	36	58	10.9	106	7.7	2.76 ^c		13	0.57	0.0	146		9.0		0.1		138	18	15	
5/12 0755	80	53	10.6	97	7.7	3.2	1.5	14	0.61	1.2	153	32	10	0.8	0.0	5.8	140	15	3	
6/9 1045	107	56	10.6	100	7.7	2.76 ^c		14	0.61	0.0	155		11		0.1		140	13	20	
7/2 1545	121	56	10.5	100	7.7	2.74 ^c		15	0.65	0.0	155		9.2		0.1		137	10	5	
8/4 0725	93	53	9.8	90	7.5	2.76 ^c		16	0.70	0.0	152		8.2		0.1		141	11	2	
9/9 1430	93	58	9.7	94	7.6	34	1.4	15	0.65	2.6	162	27	9.8	1.0	0.1	11	143	10	1	
10/15 1500	79	56	10.0	96	7.7	3.12 ^c		15	0.65	0.0	166		15		0.2		156	20	5	
11/5 1200	112	58	9.9	96	7.7	3.02 ^c		17	0.73	0.0	172		12		0.1		159	11	10	
12/9 1400	104	57	10.3	99	8.1	3.22 ^c		17	0.74	0.0	184		25		0.1		161	10	20	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Annual maximum. United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Tasting Laboratories, Inc. (TTL); or California Department of Water Resources (DWR) as indicated.

TABLE B-2
ANALYSES OF SURFACE WATER
SAN FRANCISCO BAY REGION (NO. 2)

L. OATON CREEK AT LOT GATOR (HTA. 74)

[illegible]

100

11

100

TABLE B-2
ANALYSES OF SURFACE WATER

SAN FRANCISCO BAY REGION (NO. 2)

NAPA RIVER NEAR ST. HELENA (STA. 79)

Date and time sampled P.S.T.	Discharge in cfs	Temp in deg F	Dissolved oxygen in ppm	Specific conductance at 25° C in μ mhos/cm	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent total solids in ppm	Total Hardness as CaCO_3 in ppm	Tur- bidity in NTU	Analyzed by ^h	
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO_3)	Bicar- bonate (HCO_3)	Sul- fate (SO_4)	Chlo- ride (Cl)	Ni- trate (NO_3)	Fluo- ride (F)						Brom- ine (Br)
1959																		USGS		
1/7 1315	31	50	10.0	8.8	264	7.2	24	0.0	80	0.0	25	0.6			176 ^e	38	86	20	60	Median 230
2/6 1640	21	56	11.9	11.3	218	7.8	20	0.0	92	0.0	18	0.4			145 ^e	38	72	0	0	Maximum 7,000
3/9 1680	44	64	9.3	9.7	218	7.3	16	0.0	87	0.0	12	0.4			145 ^e	33	70	0	2	Minimum 0.62
4/1 1130	24	68	8.8	9.6	256	7.3	23	2	102	0.07	17	0.5			171 ^e	37	86	0	10	
5/11 1440	5.7	75	8.1	9.5	319	7.3	26	3.4	0.0	142	0.0	20	0.5	0.6	212 ^f	30	110	0	50	
6/11 1000	1.1	72	7.8	8.8	352	7.1	20	0.0	170	0.0	16	0.5			234 ^g	24	138	0	15	
7/1 1600	0.4	80	13.2	16.2	346	7.9	18	0.0	181	0.0	13	0.4			239 ^g	21	143	0	20	
8/12 (ext.)	0.3	70	6.2	6.9	331	7.4	26	0.0	183	0.0	11	0.4			226 ^g	33	116	0	5	
9/4 Dry																				
10/15 1035	0.8	64	9.6	10.0	379	7.5	21	0.0	187	0.0	28	0.5			252 ^g	23	153	0	2	
11/4 1700	0.46	60	9.9	9.9	381	7.4	21	0.0	196	0.0	16	0.4			254 ^g	23	153	0	2	
12/3 1645	1.0	51	10.8	9.6	395	7.3	22	0.0	177	0.0	34	0.6			265 ^g	23	158	13	4	

^a Field pH.

^b Laboratory pH.

^c Sum of calcium and magnesium in ppm.

^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

^e Derived from conductivity vs TDS curves.

^f Determined by addition of analyzed constituents.

^g Gravimetric determination.

^h Annual median and range, respectively.

ⁱ Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

^j Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-3

ANALYSES OF SURFACE WATER

CENTRAL COASTAL REGION (NO. 3)

PALAZO RIVER NEAR CHITTENDEN (STA. 77)

Date collected sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Hard- ness as CaCO ₃ ppm	Tur- bid- ity ppm	Analyzed by
						Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Bor- on (B)	Silice- ous (SiO ₂)				
1559	32	50	9.9	87	1.360	7.7	12.50 ^b	75	3.26	0.0	3.48	79	2.23	0.4		858 ^e	21	695	340	3	Median 36.5
1/8 1030	21	50	9.5	84	1.380	7.7	11.70 ^b	100	4.35	0.0	3.4	93	2.68	0.5		871 ^e	27	590	316	1	Maximum 9.400
2/10 0740	200	60	8.4	84	634	7.5	5.00 ^c	36	1.57	0.0	2.07	37	1.04	0.3		400 ^e	24	251	61	25	Minimum 0.02
3/5 1100	37	69	8.1	89	1.160	7.7	7.9 ^c	77	3.35	0.0	3.33	69	1.95	0.4		733 ^e	30	397	124	35	
4/1 1505	21	64	7.7	80	1.340	8.1	95	73	3.19	0.0	4.20	288	99	0.7		921 ^f	30	536	192	1	
5/12 1045	18	76	8.7	102	1.420	7.9	10.78 ^b	131	5.70	0.0	4.66	118	3.33	0.6		897 ^e	35	536	156	20	
6/9 0745	5.9	72	6.8	77	1.460	7.9	10.20 ^c	153	6.57	0.0	4.69	125	3.52	0.7		935 ^e	38	530	124	20	
7/2 1420	5.2	62	8.4	85	1.320	8.1	9.90 ^c	139	6.78	0.0	5.04	112	3.72	0.7		834 ^e	40	494	241	10	
8/4 0815	6.2	69	8.0	88	1.430	8.1	65	177	7.70	12	6.90	67	4.29	0.7		870 ^f	48	400	0	15	
9/9 1145	1.8	67	8.1	88	1.290	8.0	10.11 ^c	94	4.09	0.0	4.42	92	2.73	0.5		769 ^e	29	507	145	35	
10/7 1500	12	58	8.3	81	1.350	7.7	7.60 ^c	126	5.28	0.0	4.18	125	3.52	0.7		893 ^e	36	480	137	3	
11/4 1240	13	50	9.7	86	1.270	7.8	100	85	3.70	5.5	3.71	208	2.82	0.5		860 ^f	25	558	254		
12/9 1650																					

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in gpm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR), as indicated.

TABLE B-3
ANALYSES OF SURFACE WATER

CENTRAL COASTAL REGION (NO. 3)

SAN BENITO RIVER NEAR VALLEY FIRE STATION (STA. 77a)

Date analysis sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance (at 25°C)	pH	Mineral constituents in parts per million												Total dis- solved solids in ppm	Re- sidual alkali in ppm	Hardness as CaCO ₃ ppm	Total alkali in ppm	Analyzed by
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					
1959																			USGS			
1/8	40	52	10.4	94	1.330	8.3	49	99	122	3.3	0.0	257	68	0.9	0.3	1.0	11	530	99	Median 4.2		
1/8	(est.)						27.45	8.15	5.31	0.08	0.00	8.02	5.35	1.92	0.01							
2/9	15	50	11.0	97	1.610	8.1	43	101	196	2.6	0.0	479	334	0.0	0.3	1.5	11	594	99	Maximum 230.		
10/6	(est.)						21.5	8.33	8.53	0.07	0.63	7.85	6.95	1.08	0.00							
7/5	30	67	9.0	97	1.220	8.1	24	114	114	2.7	0.0	503	192	1.0	0.5	1.0	15	476	64	Minimum 0.23		
10/5	(est.)						11.2	9.40	4.50	0.07	0.00	8.24	4.00	1.0	0.03							
4/1	12	79	8.3	101	1.370	8.1	30	98	149	3.2	27	417	202	0.0	0.2	3.3	4.0	476	89			
14.0	(est.)						1.50	8.02	6.40	0.08	0.50	6.83	5.87	0.00	0.01							
5/12	Dry																					
6/8	2.5	81	8.0	99	1.730	8.1	38	103	216	7.2	19	464	372	0.0	0.2	1.6	11	518	114	9		
1630	(est.)						1.50	8.46	9.50	0.18	0.63	7.44	7.89	3.52	0.00							
7/2	Dry																					
8/3	1.5																					
2340	(est.)						28	168	292	6.0	20	463	395	0.7	0.4	2.3	16	572	145			
9/9	Dry						2.59	8.05	12.70	0.15	0.97	7.59	9.66	0.01								
10/7	Dry																					
11/4	2.5	63	9.8	101	1.680	8.1			216		15	512	148			1.9		590	146	1		
1350	(est.)								9.40		0.50	8.39	4.17									
12/9	1.3	56	11.2	106	1.790	8.1			209		21	501	142			1.6		546	147	6		
1200	(est.)								9.09		0.70	8.28	4.00									

Field pH₁

Laboratory pH

Sum of calcium and magnesium in ppm.

Sum of calcium and magnesium in epiL

Derived from conductivity vs TDS curves

^f Determined by addition of analyzed constituents.

Gravimetric determination.

h Annual median and range, respectively.

^a Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPDH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-3
ANALYSES OF SURFACE WATER

CENTRAL COASTAL REGION (NO. 3)

SOQUEL CREEK AT SOQUEL (STA. 76)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	%Sat	Specific conductance at 25°C	pH	Mineral constituents in parts per million										Total dis- solved solids in ppm	Per- cent solids in ppm	Hard- ness as CaCO ₃ ppm	Tur- bid- ity N.C. ppm	Analyzed by	
							Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- ate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)
1959																		USGS				
1/7 1640	25	52	10.4	94	703	7.5	41	1.78		0.0	183	3.00	32	1.47		0.0	444 ^e	25	268	118	1	Median
2/9 1230	9.9	48	11.5	99	690	8.1	48	2.09		0.0	222	3.64	46	1.30		0.0	435 ^e	28	270	88	1	Maximum
3/4 1630	56	53	9.5	87	554	7.7	27	1.17		0.0	180	2.95	25	0.70		0.1	350 ^e	21	295	77	4	Minimum
4/1 1100	19	60	10.0	100	638	7.9	36	1.57		6	203	3.33	36	1.02		0.1	403 ^e	27	215	39	2	
5/13 0855	7.7	59	9.6	94	708	7.9	42	1.83	3.9	0.0	241	104	51	1.44	0.2	0.1	444 ^f	25	274	76	0	
6/8 1145	4.3	64	9.3	97	713	7.9	47	2.04	12	0.0	222	3.64	65	1.73	0.0	0.0	450 ^e	27	276	74	1	
7/2 1010	4.3	66	10.6	113	721	7.9	50	2.18	1	0.0	240	3.73	51	1.52	0.1	0.1	455 ^e	28	281	78	1	
8/3 1100	2.5	68	10.9	118	696	7.9	44	1.91	6	0.20	232	3.80	61	1.79	0.2	0.2	435 ^e	27	252	52	4	
9/9 0715	2.5	65	9.6	101	699	7.9	46	2.0	5.3	0.0	246	75	63	0.7	0.4	0.1	442 ^f	26	270	68	3	
10/7 0900	6.1	62	9.9	101	803	7.9	58	2.52	0.0	0.00	262	4.73	92	2.59	0.2	0.2	507 ^e	29	309	102	2	
11/4 0845	5.2	52	9.5	86	810	7.5	56	2.44	0.0	0.00	262	4.59	76	2.11	0.2	0.2	511 ^e	29	299	84	3	
12/9 0810	4.1	45	10.8	89	810	7.7	67	2.91	0.0	0.00	261	4.28	80	2.35	0.1	0.1	511 ^e	30	303	89	8	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

0.00

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i Mineral analyses (USGCD), United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), District of Southern California (DSC), City of Los Angeles, Department of Water and Power (LADWP), City of Long Beach, Department of Public Health (LDBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

SPECIFIC CONDUCTANCE, $\mu\text{mhos/cm}$, 25°C

ANALYSES OF SURFACE WATER, 1964-1965

Date and time sampled P S Y	Discharge in cfs	Temp. in °F	Dissolved oxygen ppm	Specific conductance (microhm-cm at 25°C)	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent sed- iment in ppm	Hardness as CaCO ₃ Total N.C. ppm	Temp. Cor- rected Hardness ppm	Assigned By
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)					
				</															

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

AMERICAN RIVER AT HUBBARD DAM (SPN, 22a)

Date and time sampled P.S.T.	Discharge Temp in °C (est.)	Dissolved oxygen ppm	Specific Conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent calcium in ppm	Hardness in CaCO ₃ ppm	Turbidity in nephelometric units	Analyzed by
				Calcium (mg)	Sodium (mg)	Potassium (mg)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Other constituents					
1959																		USGS
1/15	1,100	12.5	120	8.4 ^a	2.8	0.6	0.0	33	3.8	6.2	0.3	0.0	10	50 ^f	16	31	4	
1/20	1,400	12.0	107	7.6 ^a	3.5	0.4	0.0	30	2.1	7.0	0.0	0.0	10	48 ^f	21	29	4	
2/5	1,350	12.0	106	7.2 ^a	2.4	0.5	0.0	30	1.9	4.8	0.0	0.0	11	46 ^f	15	28	3	
3/11	1,245	10.7	99	6.6 ^a	2.1	0.5	0.0	29	3.5	3.6	0.0	0.0	12	45 ^f	16	25	1	
4/8	1,400	10.7	99	6.6 ^a	2.1	0.5	0.0	29	3.5	3.6	0.0	0.0	12	45 ^f	16	25	1	
5/12	1,050	10.2	97	7.1 ^a	2.8	0.3	0.0	29	3.8	3.0	0.0	0.1	10	43 ^f	18	26	2	
6/10	1,570	10.2	108	13	4.5	1.0	0.0	40	5.8	12	0.2	0.1	9.6	69 ^f	18	44	11	
10/14	1,445	10.6	106	6.4 ^a	2.1	0.5	0.0	25	2.0	2.2	0.0	0.2	18	45 ^f	17	20	0	
7/3	3,360	10.6	106	6.4 ^a	2.1	0.5	0.0	25	2.0	2.2	0.0	0.2	18	45 ^f	17	20	0	
8/6	4,100	8.6	86	6.4 ^a	2.6	0.9	0.0	23	3.0	2.0	0.0	0.1	15	46 ^f	22	19	0	
9/7	1,850	7.1	85	6.4 ^a	2.1	0.5	0.0	27	2.6	2.5	0.0	0.0	10	40 ^f	16	23	1	
10/12	699	9.4	82	8.0	3.1	0.5	0.0	35	2.0	4.5	0.0	0.1	8.6	46 ^f	18	29	0	
11/15	698	10.1	102	8.0	2.9	0.8	0.0	33	2.0	4.8	0.4	0.1	8.5	46 ^f	17	30	3	
12/1	596	10.4	101	88.7	3.1	0.3	0.0	35	7.2	0.2	0.0	0.0	0.0	59 ^f	15	37	8	

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs. TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range.

i United States Geological Survey, Quality of Water Branch, USGS.

j United States Geological Survey, Quality of Water Branch, USGS.

k San Bernardino County Flood Control District, San Bernardino Water District of Southern California (WBSD).

l Los Angeles Department of Water and Power (LADWP).

m City of Los Angeles, Department of Public Health (LADPH).

n City of Long Beach, Department of Public Health (LBPH).

o Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE No. 4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY MOJOON (NO. 5)

AMERICAN RIVER AT SACRAMENTO (STA. 72)

Date and time of day P. S. T.	Discharge Temp. in cfs.	Dissolved oxygen in ppm % Sat.	Specific conductance (m-cmhos at 25°C)	Mineral constituents in equivalents per million							Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total Hardness as CaCO ₃ in ppm	Assigned by I		
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Barium (Ba)	Silica (SiO ₂)	Other constituents
1950																
1119	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1120	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1121	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1122	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1123	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1124	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1125	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1126	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1127	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1128	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1129	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1130	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1131	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1132	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1133	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1134	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1135	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1136	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1137	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1138	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1139	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1140	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1141	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1142	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1143	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1144	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1145	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1146	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1147	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1148	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1149	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1150	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1151	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1152	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1153	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1154	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1155	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1156	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1157	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1158	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1159	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1160	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1161	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1162	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1163	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1164	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1165	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1166	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1167	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1168	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1169	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1170	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1171	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1172	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1173	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1174	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1175	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1176	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1177	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1178	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1179	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1180	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1181	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1182	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1183	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1184	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1185	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1186	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1187	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1188	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1189	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1190	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1191	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1192	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1193	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1194	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1195	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1196	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1197	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1198	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1199	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8
1200	7.1	8.1	175	17.0	1.7	2.4		0.0	11.5	1.8	1.8	1.8	1.8	1.8	1.8	1.8

1. The analyses were made by the standard methods of the American Public Health Association, 1925, 1935, and 1945. The methods used were: (a) Dissolved oxygen, (b) Specific conductance, (c) Total dissolved solids, (d) Hardness, (e) Calcium, (f) Magnesium, (g) Sodium, (h) Potassium, (i) Carbonate, (

TABLE 3-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

AMERICAN RIVER, MIDDLE FORK, NEAR AUBURN (STA. 296)

Date and time sampled P.S.T.	Discharge in cfs	Temp. in °F	Dissolved oxygen in ppm	Specific conductance (microhm/cm at 25°C)	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent hardness as CaCO ₃ in ppm	Turbidity in NTU	Analyzed by		
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					Silicate (SiO ₂)	
1999																					
1/15 1005	876	53	12.9	119	38.8	7.0	4.8 0.24	0.5 0.04	2.0 0.09	0.1 0.00	1.6 0.06	2.9 0.08	2.6 0.07	0.3 0.00	0.0 0.00	0.0 0.00	30 ^f	24	14	1	UGS
2/12 1300	634	40	13.2	102	68.5	7.2	7.6 0.38	2.1 0.17	2.6 0.11	0.4 0.01	4.8 0.49	4.8 0.10	3.8 0.11	0.2 0.00	0.0 0.00	0.0 0.00	50 ^f	16	28	3	
3/10 0845	1,240	46	11.5	96	37.0	7.2	4.6 0.23	0.6 0.05	2.1 0.09	0.1 0.00	18 0.30	2.3 0.06	3.1 0.08	0.0 0.00	0.0 0.00	0.0 0.00	31 ^f	24	14	0	
4/13 0800	2,250	53	10.7	98	28.8	7.2	3.0 0.15	2.1 0.17	1.5 0.07	0.0 0.00	26 0.33	2.9 0.06	3.0 0.09	0.0 0.00	0.0 0.00	0.0 0.00	30 ^f	18	16	0	
5/14 0830	1,750	58	10.2	99	30.4	7.3	3.8 0.19	1.1 0.01	1.6 0.07	0.3 0.01	14 0.23	1.5 0.03	1.5 0.04	0.0 0.00	0.0 0.00	0.0 0.00	25 ^f	25	10	0	1
6/11 0810	535	64	9.0	94	42.6	7.3	4.6 0.23	1.1 0.09	2.2 0.10	0.6 0.02	20 0.33	2.9 0.06	3.0 0.08	0.2 0.00	0.0 0.00	0.0 0.00	36 ^f	23	16	0	
7/10 0900	127	75	8.3	97	60.2	7.3	3.0 0.15	2.0 0.17	1.0 0.04	0.0 0.00	26 0.43	2.9 0.06	3.0 0.11	0.0 0.00	0.0 0.00	0.0 0.00	46 ^f	21	25	4	1
8/7 0845	55	76	8.0	95	79.2	7.1	3.1 0.16	2.0 0.17	1.1 0.04	0.0 0.00	34 0.56	4.0 0.11	4.0 0.11	0.0 0.00	0.0 0.00	0.0 0.00	60 ^f	17	39	4	1
9/4 1130	42	73	8.4	96	90.8	7.4	9.8 0.49	2.3 0.19	3.6 0.16	0.8 0.02	36 0.59	5.0 0.16	5.8 0.16	0.1 0.00	0.1 0.00	0.0 0.00	60 ^f	19	34	4	0.5
10/14 1000	65	65	8.1	86	89.1	7.5	7.5 0.38	2.4 0.19	3.4 0.15	0.0 0.00	36 0.59	5.0 0.16	6.0 0.17	0.0 0.00	0.0 0.00	0.0 0.00	68 ^f	17	36	6	1
11/12 0900	53	54	10.1	94	94.0	7.4	7.4 0.38	2.4 0.19	3.4 0.15	0.0 0.00	36 0.59	5.0 0.16	5.5 0.16	0.0 0.00	0.0 0.00	0.0 0.00	72 ^f	19	37	7	1
12/3 0925	52	44	11.8	98	94.7	7.1	7.1 0.38	2.4 0.19	3.4 0.15	0.0 0.00	37 0.61	5.0 0.16	5.2 0.15	0.0 0.00	0.0 0.00	0.0 0.00	70 ^f	19	37	7	2

g Field pH
h Laboratory pH

i Sum of calcium and magnesium in ppm

j Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

k Derived from conductivity vs TDS curves

l Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analysis of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Annual analyses (USFCD), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (SPHS), San Bernardino County Flood Control District (SBCFD), Los Angeles County Flood Control District (LACFD), City of Long Beach, Department of Public Health (LADPH), Temescal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

ANTELOPE CREEK NEAR NORTH STATION, 86c1

Date and time of day sampled P.S.T.	Discharge in cfs in ft ³	Dissolved oxygen in ppm	Specific conductance in μ mhos at 25°C	Mineral constituents in parts per million												Total dissolved solids in ppm	Per cent hardness as CaCO ₃ Total N.C. ppm	Turbidity in ppm	Analyzed by		
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					Other constituents	
1959																		USGS			
1/6 1340	46	11.0	94	99.0	8.0 0.40	4.4 0.19	1.3 0.03	0.0 0.00	29 0.40	6.7 0.14	10 0.28	6.5 0.10	0.1 0.01	0.4 0.01	29		19	39	15		
2/2 1440	49	10.6	94	139	7.3 1.35	8.4 0.39	0.7 0.02	0.0 0.00	78 1.28	3.8 0.08	6.0 0.17	0.0 0.00	0.0 0.00	0.0 0.00	28		113 ^f	26	6		
3/13 1240	60	11.6	116	327	7.1 1.25	29 1.61	12 0.52	0.0 0.00	160 2.78	13 0.27	14 0.39	6.7 0.11	0.0 0.00	0.1 0.01	46		217 ^f	15	143	12	
4/10 0000	16	8.9	87	169	7.1 0.60	6.6 0.34	11 0.48	0.0 0.00	70 1.15	11 0.29	10 0.28	0.0 0.00	0.0 0.00	0.2 0.01	34		125 ^f	26	57	0	
5/11 1130	8.5 (est.)	66	8.3	88	176	7.3 0.74	7.6 0.38	0.0 0.00	67 1.10	15 0.31	12 0.34	1.0 0.02	0.0 0.01	0.4 0.01	34	Fe 0.13 Al 0.18 Pb 0.32	127 ^f	30	56	1	2
6/1 1445	8 (est.)	74	7.8	91	179	7.3 0.60	12 0.52	0.0 0.00	66 1.08	15 0.31	12 0.34	1.2 0.02	0.0 0.01	0.3 0.01	40		153 ^f	29	60	6	
7/15 0915	6 (est.)	75	6.5	76	255	7.3 0.95	19 0.81	0.0 0.00	113 1.85	14 0.29	17 0.48	0.8 0.01	0.2 0.01	0.4 0.01	43		193 ^f	28	88	0	
8/10 1245	4 (est.)	82	8.2	103	284	7.4 0.95	14 0.76	0.0 0.00	130 2.13	16 0.33	18 0.51	0.7 0.01	0.1 0.01	0.5 0.01	54		208 ^f	23	107	0	
9/1 1320	2 (est.)	74	8.2	95	250	7.4 0.75	16 0.79	0.0 0.00	104 1.70	12 0.25	20 0.68	0.3 0.00	0.2 0.01	0.6 0.01	44	Fe 0.02 Zn 0.01 Al 0.11 Cu 0.01 Pb 0.40	180 ^f	30	82	0	10
10/12 1440	4.5 (est.)	68	8.3	90	291	7.4 1.00	9.2 0.76	0.0 0.00	93 1.58	23 0.44	20 0.59	0.0 0.00	0.2 0.01	0.7 0.01	41		199 ^f	35	88	12	
11/3 0900	52	9.6	87	227	7.4 1.30 ^c	19 0.83	1.9 0.10	0.0 0.00	80 1.31	24 0.58	24 0.68	0.0 0.00	0.0 0.00	0.6 0.01	41		167 ^f	38	67	1	5
12/15 1215	2 (est.)	45	10.1	83	276	7.3 0.95	13 1.03	0.0 0.00	132 2.28	7.0 0.15	16 0.45	0.5 0.01	0.2 0.01	0.5 0.01	41		189 ^f	29	99	0	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in eqm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown 0.00

e Derived from conductivity vs TDS curves

f Determined by difference

g Gravimetric determination

h Annual median and range, respectively

i Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service, or United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFCD), Metropolitan Water District of Southern California (MWD), Los Angeles County Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE 2-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY ARBIDE (26, 5)

ANTILLOPE CREEK NEAR OLD MILLY (371, 484)

Date and time sampled P.S.T.	Discharge Temp in °F	Dissolved oxygen in ppm	Specific conductance at 25°C	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total Hardness in mg/l	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)	Other constituents	
1959																		
1/15	68	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
1/16	68	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
1/17	68	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/13	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/14	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/15	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/16	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/17	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/18	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/19	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/20	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/21	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/22	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/23	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/24	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/25	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/26	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/27	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/28	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/29	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/30	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
3/31	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/1	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/2	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/3	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/4	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/5	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/6	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/7	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/8	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/9	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/10	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/11	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/12	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/13	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/14	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/15	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/16	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/17	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/18	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/19	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/20	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/21	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/22	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/23	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/24	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/25	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/26	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/27	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/28	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/29	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
4/30	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/1	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/2	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/3	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/4	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/5	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/6	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/7	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/8	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/9	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/10	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/11	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/12	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/13	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/14	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/15	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/16	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/17	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/18	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/19	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/20	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
5/21	58	11.4	91	7.3	11.1	1.2	1.1	1.1	0.05	1.1								

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (WD, 5)
BOTTLE CREEK NEAR CORTWOOD (STA. 886)

Date and time of collection P.S.T.	Discharge Temp in cfs	Temp in °F	Dissolved oxygen in ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million											Total dis- solved solids in ppm	Per- cent solid in ppm	Hardness as CaCO ₃ Total ppm	Temp. in °F N.C. ppm	Analyzed by	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)						Silica (SiO ₂)
1/29/59																			USGS		
1/5 1/30	671			111	7.5 ^b	6.3 ^b	5.6 ^b	1.2 ^b	0.0 ^b	87 ^b	6.7 ^b	4.0 ^b	1.3 ^b	0.0 ^b	0.0 ^b	34	96 ^f	44	0		
2/9 1/100	475			134	7.8 ^b	7.1 ^b	5.3 ^b	0.33 ^b	0.0 ^b	79 ^b	2.2 ^b	3.0 ^b	0.0 ^b	0.0 ^b	0.0 ^b	44	116 ^f	23	54	0	
3/11 12/30	396	51	122.0	107	7.4 ^b	9.6 ^b	5.5 ^b	0.28 ^b	0.0 ^b	71 ^b	3.3 ^b	1.5 ^b	0.0 ^b	0.0 ^b	0.0 ^b	42	106 ^f	22	46	0	
4/15 1/310	361	57	101.6	101	7.7 ^a	8.8 ^b	6.3 ^b	1.4 ^a	0.0 ^a	69 ^a	3.8 ^a	1.0 ^a	0.0 ^a	0.0 ^a	0.0 ^a	43	106 ^f	22	48	0	
5/15 0/600	361	54	101.9	101	7.8 ^b	9.0 ^b	6.1 ^b	1.2 ^b	0.0 ^b	71 ^b	0.0 ^b	2.0 ^b	0.5 ^b	0.1 ^b	0.0 ^b	42	102 ^f	23	45	0	3
6/16 1/345	215	62	101.0	102	8.1 ^b	9.2 ^b	7.1 ^b	1.8 ^b	0.0 ^b	79 ^b	3.0 ^b	3.5 ^b	0.0 ^b	0.0 ^b	0.0 ^b	23	122 ^f	23	52	0	
7/9 1/125	196	64	8.7	91	8.1 ^b	12 ^b	2.8 ^b	2.6 ^b	0.0 ^b	84 ^b	4.0 ^b	1.8 ^b	0.0 ^b	0.0 ^b	0.0 ^b	48	124 ^f	25	54	0	
8/11 0/600	167	63	9.4	97	7.5 ^b	1.28 ^b	2.1 ^b	0.40 ^b	0.0 ^b	88 ^b	3.0 ^b	1.5 ^b	0.0 ^b	0.0 ^b	0.1 ^b	20	132 ^e	25	60	0	20
9/1 0/935	172	59	101.2	100	7.7 ^a	11 ^b	8.8 ^b	2.2 ^b	0.0 ^b	88 ^b	3.0 ^b	4.0 ^b	0.0 ^b	0.0 ^b	0.1 ^b	47	127 ^f	24	58	0	10
10/13 1/145	215	56	101.0	96	7.8 ^b	11.76 ^b	10 ^b	0.44 ^b	0.0 ^b	92 ^b	1.51 ^b	3.5 ^b	0.10 ^b	0.0 ^b	0.0 ^b	40	130 ^e	28	58	0	2
11/11 1/325	189	49	111.4	99	8.0 ^b	11.71 ^b	7.4 ^b	0.42 ^b	0.0 ^b	90 ^b	1.40 ^b	3.8 ^b	0.11 ^b	0.0 ^b	0.0 ^b	40	130 ^e	27	58	0	4
12/10 1/140	202	44	121.6	102	8.2 ^b	11.07 ^b	2.0 ^b	0.39 ^b	0.0 ^b	90 ^b	1.40 ^b	3.8 ^b	0.11 ^b	0.0 ^b	0.0 ^b	40	128 ^e	25	58	0	4

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water Resources (LADWP), City of Los Angeles, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (W. 1)

BEAR CREEK AT BUREAU (T₂, 114)

Date and time sampled PST	Discharge Temp. in cft in deg	Dissolved oxygen gram % Sat	Specific Conductance in ccm/cm at 25°C	pH	Metal constituents in parts per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Temp. in deg F	Accepted By:
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)				
1970	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1971	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1972	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1973	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1974	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1975	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1976	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1977	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1978	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1979	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1980	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1981	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1982	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1983	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1984	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1985	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1986	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1987	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1988	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1989	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1990	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1991	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1992	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1993	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1994	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1995	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1996	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1997	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1998	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
1999	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	
2000	6.1	9.4	160	6.1	1.60	1.70	1.40	0.47	0.00	3.00	2.0	0.00	0.00	0.00	0.00	0.00	0.00	

1970-1979

1980-1989

1990-1999

2000-2009

2010-2019

2020-2029

2030-2039

2040-2049

2050-2059

2060-2069

2070-2079

2080-2089

2090-2099

2100-2109

2110-2119

2120-2129

2130-2139

2140-2149

2150-2159

2160-2169

2170-2179

2180-2189

2190-2199

2200-2209

2210-2219

2220-2229

2230-2239

2240-2249

2250-2259

2260-2269

2270-2279

2280-2289

2290-2299

2300-2309

2310-2319

2320-2329

2330-2339

2340-2349

2350-2359

2360-2369

2370-2379

2380-2389

2390-2399

2400-2409

2410-2419

2420-2429

2430-2439

2440-2449

2450-2459

2460-2469

2470-2479

2480-2489

2490-2499

2500-2509

2510-2519

2520-2529

2530-2539

2540-2549

2550-2559

2560-2569

2570-2579

2580-2589

2590-2599

2600-2609

2610-2619

2620-2629

2630-2639

2640-2649

2650-2659

2660-2669

2670-2679

2680-2689

2690-2699

2700-2709

2710-2719

2720-2729

2730-2739

2740-2749

2750-2759

2760-2769

2770-2779

2780-2789

2790-2799

2800-2809

2810-2819

2820-2829

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

BEAR RIVER AT MOUTH (STA. 205)

Date and time of day and P.S.T.	Discharge Temp in cfs	Dissolved oxygen in %	Specific conductance at 25°C (microhm/cm)	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent solids from farm	Hardness as CaCO ₃ Total ppm	Turbidity in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)			
1959																			UNOS
1/9	Not Available	10.4	94	7.3	23	9.4	10	3.2	0.0	67	40	16	1.8	0.1	0.1	15	152 ^f	18	61
1/15					115	0.77	0.44	0.58	0.00	1.10	0.83	0.45	0.03	0.03					
2/9	Not Available	12.1	101	7.3	20	7.3	6.3	0.7	0.0	66	27	9.0	0.2	0.1	0.0	10	114 ^f	14	26
2/10					1700	0.50	0.27	0.08	0.00	1.08	0.56	0.25	0.00	0.01					
2/11	Not Available	11.6	104	84.0	7.3	8.8	2.2	0.4	0.0	30	11	4.2	0.0	0.0	0.0	11	56 ^f	17	34
11/10					0.41	0.24	0.14	0.01	0.00	0.49	0.23	0.12	0.00	0.00					
4/14	10 sec.	9.2	106	268	22	13	12	3.5	0.0	185	18	14	0.0	0.3	0.1	21	166 ^f	19	110
7/20					1.10	0.52	0.52	0.09	0.00	2.05	0.37	0.39	0.00	0.01					
5/12	6.4 sec.	7.9	99	477	26	18	41	6.2	0.0	163	12	66	0.4	0.2	0.1	13	296 ^f	37	143
7/15					1.40	1.46	1.78	0.16	0.00	2.27	0.25	1.06	0.01	0.01			Fe 0.15 Al 0.24 PO ₄ 0.10	9	3
6/9	5 sec.	8.9	102	505	34	22	42	5.4	0.0	208	13	60	0.2	0.2	0.1	25	334 ^f	34	174
13/20					1.76	1.78	1.83	0.14	0.00	3.41	0.27	1.03	0.00	0.01					
7/7 0740	Not Sampled - Dry																		
8/7	Not Sampled - Dry																		
9/3	1 sec.	8.3	102	513	34	23	35	8.1	0.0	213	15	52	0.7	0.2	0.1	28	335 ^f	29	180
13/20					1.76	1.90	1.97	0.21	0.00	3.49	0.31	1.47	0.01	0.01			Fe 0.04 Al 0.22 PO ₄ 0.15	5	3
10/14	Not Sampled - No Flow																		
18/15																			
11/12	Not Sampled - Dry																		
16/00																			
12/14	Not Sampled - Dry																		
09/40																			

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 2-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

1959-1976 Total-Dissolved Solids (TDS), pH

Date and time sampled PST	Geographic location in 1959	Discharge in cfs	Specific conductance (micro-mhos/cm at 25°C)	Major constituents in equivalents per million							Total dissolved solids in ppm	Percent solids in ppm	Hardness as CaCO ₃ in ppm	Total N.C. in ppm	Total Coliform in MPN/100 ml	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Other constituents		
1959	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1960	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1961	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1962	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1963	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1964	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1965	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1966	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1967	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1968	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1969	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1970	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1971	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1972	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1973	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1974	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1975	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1976	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1977	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1978	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1979	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1980	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1981	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1982	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1983	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1984	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1985	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1986	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1987	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1988	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1989	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1990	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1991	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1992	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1993	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1994	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1995	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1996	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1997	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1998	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
1999	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2000	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2001	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2002	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2003	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2004	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2005	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2006	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2007	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2008	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2009	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2010	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2011	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2012	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2013	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2014	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2015	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2016	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2017	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2018	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2019	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1
2020	10	15	120	1	1	1	1	1	1	1	1	1	1	1	1	1

1. 1959-1976

2. 1977-1978

3. 1979-1980

4. 1981-1982

5. 1983-1984

6. 1985-1986

7. 1987-1988

8. 1989-1990

9. 1991-1992

10. 1993-1994

11. 1995-1996

12. 1997-1998

13. 1999-2000

14. 2001-2002

15. 2003-2004

16. 2005-2006

17. 2007-2008

18. 2009-2010

19. 2011-2012

20. 2013-2014

21. 2015-2016

22. 2017-2018

23. 2019-2020

24. 2021-2022

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69. 2111-2112

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95. 2163-2164

96. 2165-2166

97. 2167-2168

98. 2169-2170

99. 2171-2172

100. 2173-2174

101. 2175-2176

102. 2177-2178

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)
B13 DETION CHECK AT 01230 (STA. 05A)

Date and time of day and P.S.T.	Discharge Temp in C's	Dissolved oxygen in %	Specific Conductivity at 25°C or 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total Chlorophyll in MPN/ml	Analyzed By
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)		
1/5/59																	1950S
1/7 11:15	102	11.5	99	8.8	1.1	6.2	0.7	0.0	52	3.8	5.2	0.0	0.0	0.1	27		
2/10 1:50	16	11.3	97	10.1	1.1	8.8	0.7	0.0	78	7.7	10.1	0.0	0.1	0.1	21		
3/12 0:50	109	11.5	101	10.1	1.1	6.2	0.7	0.0	64	2.3	6.4	0.0	0.0	0.1	22		
4/10 0:50	55	9.9	98	10.1	1.1	8.8	0.7	0.0	79	2.6	6.5	0.0	0.0	0.0	36		
5/8 10:30	29	9.3	98	11.0	1.1	11.0	0.7	0.0	98	1.8	11.0	0.0	0.0	0.1	28	Fe 0.02, Al 0.02, Mn 0.02	
6/11 1:30	19	9.0	98	10.1	1.1	11.0	0.7	0.0	101	1.8	11.0	0.0	0.0	0.1	27		
7/11 1:55	11	6.2	71	10.1	1.1	11.0	0.7	0.0	110	1.8	11.0	0.0	0.0	0.1	31		
8/10 10:40	18	7.5	91	10.1	1.1	11.0	0.7	0.0	112	1.8	11.0	0.0	0.0	0.1	27		
9/1 10:5	Not Sampled	Probed															
10/12 12:00	9.2	8.6	89	10.1	1.1	11.0	0.7	0.0	112	1.8	11.0	0.0	0.0	0.1	26		
11/2 12:00	6.0	10.3	95	10.1	1.1	11.0	0.7	0.0	112	1.8	11.0	0.0	0.0	0.1	26		
12/1 11:10	9.2	11.5	98	10.1	1.1	11.0	0.7	0.0	112	1.8	11.0	0.0	0.0	0.1	27		

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Tammam Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 8-4
ANALYSES OF SURFACE WATER
CHONTALA VALLEY REGION (NO. 5)

[illegible]

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4. *Journal of the American Medical Association*, 1997; 277: 1039-1043.

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TABLE 3-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

BUTTE, CREEK NEAR CHICO (STA. 111)

Date and time of day P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm %Sat	Specific conductance pH at 25 °C	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Hard- ness as CaCO ₃ ppm	Turbid- ity MPN/ml	Analyzed by		
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Chlor- ine (Cl)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Fluor- ide (F)	Barium (Ba)	Silic acid (SiO ₂)					Other constituents	
1959																				
1/8 0820	632	46	114.9	100	98.7	7.1 ^a	0.16 ^b	3.0 0.13	0.00	0.00	0.00	0.00	0.00	0.00	0.00	14	14.0	2	10	Median 23.
2/10 1300	932	40	124.8	98	97.5	7.2 ^a	0.08 ^b	3.5 0.15	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15	14.0	0	10	Maximum 620.
3/12 0920	656	46	114.6	97	76.1	7.6 ^a	0.07 ^b	4.8 0.12	0.00	0.00	0.00	0.00	0.00	0.00	0.00	14	36	0	10	Maximum 0.23
4/9 1600	1044	56	102.3	98	71.4	7.3 ^a	0.07 ^b	2.6 0.11	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15	31	0	2	
5/8 0825	323	50	101.1	93	77.2	7.3 ^a	0.08 ^b	2.8 0.12	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15	33	0	0	
6/11 3100	1284	64	92.2	96	93.5	7.3 ^a	0.06 ^b	4.0 0.17	0.00	0.00	0.00	0.00	0.00	0.00	0.00	18	40	0	12	
7/13 3130	140	72	87.7	99	105	8.1 ^a	0.06 ^b	4.1 0.19	0.00	0.00	0.00	0.00	0.00	0.00	0.00	17	48	0	1	
8/20 0920	134	68	84.5	93	105	7.4 ^a	0.07 ^b	3.8 0.17	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15	47	0	3	
9/1 0915	119	60	92.9	99	112	7.5 ^a	0.07 ^b	3.7 0.16	0.00	0.00	0.00	0.00	0.00	0.00	0.00	13	50	0	3	
10/12 2000	137	55	104.1	96	115	7.5 ^a	0.07 ^b	4.6 0.20	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15	56	0	1	
11/2 1100	76	51	104.6	95	124	7.5 ^a	0.07 ^b	3.9 0.17	0.00	0.00	0.00	0.00	0.00	0.00	0.00	13	56	0	1	
12/1 0945	113	41	124.2	96	119	7.3 ^a	0.06 ^b	4.8 0.21	0.00	0.00	0.00	0.00	0.00	0.00	0.00	17	50	0	20	

Field pH

Laboratory pH

Sum of calcium and magnesium in eqm.

Sum of calcium and magnesium in ppm.

0.00

Derived from conductivity vs TDS curves.

f. Determined by addition of analyzed constituents.

9. Gravimetric determination.

of Gram-negative determination.

^a Annual median and range, respectively. Calculated from analyses at separate locations. ^b California Department of Water Resources (CDWR). ^c United States Geological Survey, Quality of Water Branch (USGS). ^d United States Department of the Interior, Bureau of Reclamation (BORR). ^e United States Public Health Service (USPHS), San Bernardino County Field Office. ^f Metropolitan Water District of Southern California (MWD). ^g Los Angeles Department of Water and Power (LADWP). ^h City of Los Angeles, Department of Public Health (LAPDH). ⁱ City of Long Beach, Department of Public Health (LBPH). ^j California Department of Water Resources (CDWR); as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

GAOIE CREEK NEAR LAMER LAKE (37R, 3Q)

Date collected sampled P.S.T.	Discharge in cfs in Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Metal constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent dissolved solids	Hardness as CaCO ₃ Total N.C. ppm	Tox- icity by I ppm	Analyzed by I
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- dioxide (CO ₂)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)	Boron (B)	Silica (SiO ₂)			
1959																		USGS
1/	Not Sampled																	
2/6	144.5	59	10.0	104	199	7.7												
12/5																		
3/9	9.1	53	11.0	100	252	7.7												
12/30																		
4/2	9.4.5	59	9.8	96	301	7.6												
1500																		
5/12	34.1	68	7.9	86	242	7.7												
13/5																		
6/20	4.31	68	5.6	61	248	7.7												
1600																		
7/2	371	75	7.8	91	246	7.7												
1000																		
8/13	278	77	7.5	99	252	7.6												
1215																		
266																		
9/3																		
0945																		
10/14	30	65	8.2	87	201	7.7												
1300																		
11/4	1.6	59	9.3	90	292	7.7												
1300																		
12/3	1.9	47	10.5	89	292	7.7												
1045																		

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWDSC); Los Angeles Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

Analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWDSC); Los Angeles Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

CACHE SLOUGH BELOW LINDSEY SLOUGH (Sta. 110a)

Date and time sampled P.S.T.	Discharge in cfs in 4p	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C	pH	Mineral constituents in parts per million										Total dissolved solids in ppm	Per- cent total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Total Hardness ppm	Turbidity NTU	Color in PCU	Analyzed by
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Borate (B)	Silica (SiO ₂)					
1959	Tidal																					USBR
2/9				182				1.3					1.4						126	31		
1120								6.4					4.3						100	18		
4/13		64		155																		
1290																						
7/13		75		190				12					18						112	27		
1200																						
10/12		67		188				12					1.1						27			
1445																						

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses by the United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Water Pollution Control District of Southern California (WPCD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE 1-4

CENTRAL VALLEY REGION (NO. 5)

CALAVERAS RIVER AT JENNY LIND (Sta. 16.0)

[illegible]

1

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

CALAVERAS RIVER NEAR STOCKTON (STA. 16b)

Date and time sampled P.M.T.	Discharge Temp in cts	Dissolved oxygen ppm	Specific Conductance (microhm/cm at 25°C)	Mineral constituents in parts per million										Total dissolved in ppm	Per cent solid in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Turbidity MPN/ml	Coliforms	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)	Other constituents				
1/5/59																				11/2/58
1/1/59	Not Sampled - Dry																			
2/4	Not Sampled - Dry																			
3/10	Not Sampled - Pooled																			
4/5	Not Sampled - Dry																			
5/6	Not Sampled - Dry																			
6/10	0.2	7.2	83	7.4	26	1.30	2.6	0.00	1.95	13	8.2	0.1	0.1	0.0	11		137 ^f	14	101	3
11/20																				
7/3	Not Sampled - Dry																			
8/6	Not Sampled - Pooled																			
9/3	Not Sampled - Dry																			
10/7	Not Sampled - Dry																			
11/12	Not Sampled - Dry																			
12/1	Not Sampled - Dry																			

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

0.00

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analysis of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBOPH); Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

COLUSA TRIMBLE NEAR COLUSA (STA. RT)

Date and time sampled P.S.T.	Discharge in cfs in op	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ in ppm	Turbidity in ppm	Analyzed by	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)						
1959																						
1/12 1010	1,420	54	8.4	78	26	18	67	3.5	0.0	1.68	85	39	5.8	0.3	0.2	15		343 ^f	50	140	2	Median 196.
2/9 0900	148	46	9.4	79	5.4	4.3	14.5	1.8	0.0	3.27	206	93	0.7	0.5	0.3	1.6		741 ^f	50	312	44	Maximum 2,400
3/9 1015	222	56		1,120	8.0 ^b	4.1	1.8	1.8	0.0	3.95	290	68	1.3	0.4	0.2	1.3		701 ^f	50	300	33	Minimum 5.
4/6 0825	403	63	8.0	83	408	7.4 ^b	3.5	2.1	0.0	1.63	48	22	2.2	0.2	0.2	22		284 ^f	36	132	0	
5/4 0815	1,200	62	8.4	80	359	7.4 ^b	1.9	1.6	0.0	4.43	0.98	17	0.5	0.2	0.3	1.5		218 ^f	43	194	0	9
6/1 1550	703	74	7.0	81	427	7.6 ^b	2.2	1.5	0.0	1.68	57	21	1.4	0.2	0.2	19		269 ^f	46	121	0	
7/6 0810	617	73	6.8	78	392	7.4 ^b	1.6	0.0	1.79	49	16	0.1	0.1	0.2	0.2	30		237 ^f	40	106	0	
8/10 1430	997	79	6.9	84	378	7.4 ^b	2.1	0.0	1.80	30	14	0.4	0.2	0.2	0.2	20		231 ^f	38	104	0	
9/7 1330	1,340	73	7.2	83	379	7.4 ^b	2.1	1.9	0.0	1.80	21	22	0.0	0.2	0.1	21		290 ^f	39	118	0	40
10/5 1030	294	61	8.2	82	408	7.9 ^b	2.0	3.0	0.0	1.74	1.25	19	1.2	0.1	0.1	30		283 ^f	35	146	3	
11/13 1230	361	56	8.9	85	506	7.6 ^b	60	2.61	0.0	2.00	2.00	28	0.2	0.2				301 ^f	48	143	60	
12/3 0830	425	49	10.0	87	474	7.4 ^b	57	2.42	0.0	1.86	26	26	0.1	0.1				260 ^f	48	132	0	15

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in gpm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS).

j Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

k Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

l Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

m Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

n Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

o Mineral analyses made by United States Department of the Interior, Bureau of Reclamation (USBR).

Analyses made by United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWD), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

FERNAL VALLEY REGION (NO. 5)

OPEN CUTS CANAL AT 1st TPOD LIFT (Sta. 10+00)

Date and time of day P.S.T.	Discharge in cfs in pipe	Temp in pipe	Dissolved oxygen ppm	Specific (meq/l) at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total Coliforms in MPN/m	Accepted
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)		
1950	Tidal																	
1951				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1952				70.0	8.5	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1953				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1954				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1955				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1956				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1957				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1958				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1959				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1960				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1961				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1962				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1963				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1964				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1965				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1966				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1967				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1968				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1969				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1970				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1971				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1972				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1973				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1974				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1975				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1976				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1977				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1978				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1979				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1980				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1981				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1982				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1983				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1984				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1985				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1986				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1987				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1988				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1989				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1990				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1991				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1992				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1993				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1994				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1995				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1996				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1997				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1998				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
1999				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					
2000				71.4	7.1	1.5	7.8		1.5	7.1	1.25	6.1	6.1					

1990-2000

1990-2000

1990-2000

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1990-2000

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1990-2000

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1990-2000

1990-2000

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

COURTESY RIVER AT MCCONNELL (SVL, gha)

Date and time sampled P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Per- cent sul- fate in ppm	Hardness as CaCO ₃ Total in ppm	Tur- bid- ity in ppm	Conform- ity in ppm	Analyzed By
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)	Boron (B)	Silica (SiO ₂)						
1959																				Units	
1/6	184	11.0	97	101	7.3 ^b	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	69	18	40	6	Median 4.0
1/30	176	11.3	96	120	7.7 ^b	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	78	16	48	0	Maximum 4.0
2/4	108	11.3	96			0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	58	44	48	0	Minimum 0.23
3/10	402	10.5	98	71.7	7.3 ^a	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	46	19	42	0	
4/5	432	9.0	96	56.9	7.2 ^a	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	55	18	26	0	
5/6	177	9.1	100	70.1	7.4 ^a	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
7/30						0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
6/	Dry					0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
7/3	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
8/5	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
9/3	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
10/5	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
11/2	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					
12/1	Dry					0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0					

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analysis by USGS, United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Long Beach, Department of Public Health (LDBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

CANYON VALLEY AREA, N. W. 51

Oxygen at 20°C at 1000 ft. (N. 51)

Date sampled P. S. T.	Discharge Temp in °F	Dissolved oxygen ppm %Sat	Specific conductance at 25°C µmhos/cm	Major constituents in equivalents per million										Total dissolved solids in ppm	Per- cent acid- sol- in ppm	Hardness as CaCO ₃ ppm	Total Calcium ^a in ppm	Adjusted pH
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- dioxide (CO ₂)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlor- ide (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)					
1949																		
1/16 (1951)	80	11.1	176	7.7	1.2	1.2		0.56										
3/16 (1951)	58	10.1	171	7.7	1.2	1.2		0.56										
5/16 (1951)	68	11.1	176	7.7	1.2	1.2		0.56										
7/16 (1951)	80	11.1	176	7.7	1.2	1.2		0.56										
9/16 (1951)	80	11.1	176	7.7	1.2	1.2		0.56										
11/16 (1951)	80	11.1	176	7.7	1.2	1.2		0.56										
1/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
3/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
5/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
7/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
9/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
11/16 (1952)	80	11.1	176	7.7	1.2	1.2		0.56										
1/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
3/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
5/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
7/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
9/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
11/16 (1953)	80	11.1	176	7.7	1.2	1.2		0.56										
1/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
3/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
5/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
7/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
9/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
11/16 (1954)	80	11.1	176	7.7	1.2	1.2		0.56										
1/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										
3/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										
5/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										
7/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										
9/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										
11/16 (1955)	80	11.1	176	7.7	1.2	1.2		0.56										

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TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

COTTONWOOD CREEK NEAR COTTONWOOD (STA. 12b)

[illegible] α Field pH₂

b Laboratory off.

Sum of calcium and magnesium in ppm.

d. Iron (Fe), aluminum (Al), arsenic (As),

^a Derived from conductivity vs. TDS curves.

t. Determined by addition of analyzed constituents

Determined by addition of

g Gravimetric determination

^a Annual means and range, respectively. Calculated from analyses of duplicate monthly samples by California Department of Public Health, Division of Laboratories, and United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (N.).

WORLD OLD CRUEL BELOW NORTH YORK OFF NEW YORK 11

TABLE 2-A
ANALYSES OF SURFACE WATER

NATURAL VALLEY RESERVOIR (T.W. 17)

FOR FISHES NEAR MILVILLTOWN (T.W. 18)

Date and time sampled P.S.T.	Discharge Temp. in °F. in 2 ft.	Dissolved oxygen ppm % Sat	Specific conductance at 25°C. μ mhos/cm	Major constituents in parts per million										Total dissolved solids in ppm	Per cent total dissolved solids in ppm	Hardness as CaCO_3 ppm	Total N. ppm	Total P. ppm	Total S. ppm	Total dissolved solids in ppm	Other constituents	Sulfate (SO ₄)	Boron (B)	Fluoride (F)	Nitrate (NO ₃)	Chloride (Cl)	Sulfate (SO ₄)	Bicarbonate (HCO ₃)	Carbonate (CO ₃)	Sulfate (SO ₄)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)
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1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

1977-1999

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY BASIN (NO. 5)

DELTA CROSS CHANNEL NEAR VALUITY CROSS (TTL, 90)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in ppm % Sat	Specific conductance (micro-mhos/cm) at 25°C or pH ^a	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Percent sediment in ppm	Hardness as CaCO ₃ Total N.C. in ppm	Turbidity in nephelometric form	Analyzed by	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonates (CO ₃)	Bicarbonates (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)						Other constituents
1959	Total																			USGS		
1/14 1350	52	10.1	91	57.0	7.1		4.3 0.19		0.00	35 0.57		2.0 0.16			0.0		64 ^b	20	37	8	30	Median 11.2.
2/9 1120	48	10.5	91	159	7.3		2.4 0.41		0.0	70 1.21		8.5 0.26			0.0		105 ^b	26	58	0	1	Maximum 2,000.
4/11 1600	55	9.7	91	153	7.3		7.4 0.38		0.0	70 1.15		7.5 0.21			0.0		106 ^b	20	64	7	35	Minimum 6.2
4/1 0930	56	9.6	91	146	7.1		7.5 0.33		0.0	67 1.10		2.8 0.16			0.1		97 ^b	23	55	0	35	
5/11 0600	70	7.5	83	191	7.3		14 0.61	1.1 0.03	0.0	83 1.36	1.3 0.27	12 0.36	1.2 0.02		0.1		120 ^c	30	68	0	50	
6/8 0925	72	7.9	90	219	7.5		16 0.70		0.0	94 1.73		14 0.39			0.1		110 ^b	32	73	0	10	
7/1 0835	71	8.5	96	167	7.1		12 0.52		0.0	71 1.16		2.5 0.27			0.1		110 ^b	32	56	0	9	
8/10 0600	73	8.1	93	166	7.1		13 0.57		0.0	72 1.23		8.0 0.23			0.1		110 ^b	33	57	0	25	
9/7 0815	74	8.2	95	25.1	7.1		23 0.91	1.7 0.04	0.0	118 1.93	1.3 0.27	17 0.43	0.0 0.00	0.0 0.00	0.1	20	138 ^d	35	84	0	6	
10/5 0930	64	8.8	92	179	7.1		13 0.57		0.0	91 1.49		8.5 0.26			0.0		110 ^b	30	68	0	20	
11/2 0900	56	8.2	78	184	7.3		12 0.52		0.0	91 1.49		2.5 0.27			0.2		122 ^b	27	70	0	5	
12/7 0930	49	9.8	85	190	7.3		13 0.57		0.0	92 1.51		16 0.45			0.1		126 ^b	28	72	0	20	

^a Field pH.

^b Laboratory pH.

^c Sum of calcium and magnesium in ppm.

^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganous (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

^e Derived from conductivity vs TDS curves.

^f Determined by addition of analyzed constituents.

^g Gravimetric determination.

^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Public Health (LADPH); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 8-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY TIME (S)

DELTA-MENDOTA CANAL NEAR MENDOTA (STA. 92)

[illegible]I
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8

TABLE 1. *Continued*

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1990-1991

A 1980

TABLE B-4

ANALYSES OF SURFACE WATER

COSTA MOUNTAIN, CALIF. (NO. 5)

PREFORMULATED CANAL NEAR TEMPE (STA. 627)

Date and time of collection P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million											Total dis- solved solids in ppm	Per- cent hard- ness as CaCO ₃ ppm	Tota- l bio- bio- film MPN/ml	Analyzed by		
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Baron (B)					Silica (SiO ₂)	Other constituents
10:59																		15503			
1/13	0	53	4.5	90	7.07	3.09 ^c	7.0	0.0	0.0	1.0	1.0	1.0	0.1	0.1	0.1	398 ^d	52	150	55	1	Median 180.
1/15																					
2/13	865	51	11.2	100	8.05	3.12 ^c	1.07	0.0	0.0	1.0	1.0	1.0	0.1	0.1	0.1	453 ^d	57	173	63		Maximum 7,400.
3/12	1,660	61	8.9	90	6.11	2.69 ^c	6.5	0.0	0.0	1.0	1.0	1.0	0.1	0.1	0.1	316 ^d	52	135	46	20	Minimum 0.23
3/13																					
4/12	2,431	63	8.1	87	6.20	2.70 ^c	5.2	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	298 ^d	46	131	60	50	
5/13	1,620	70	7.0	78	3.37	1.70 ^c	3.2	0.0	0.0	0.1	0.1	0.1	0.1	0.1	0.1	195 ^d	42	92	26	110	
6/9	3,360	70	7.3	81	2.0 ^c	1.17 ^c	2.7	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	161 ^d	42	82	11	48	
10:00																					
7/9	3,428	75	7.1	63	1.1 ^c	0.7 ^c	5.0	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	216 ^d	51	103	23	35	
12:00																					
6/9	3,159	79	6.4	82	5.22 ^c	2.07 ^c	1.00	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	499 ^d	66	134	67	70	
9/7	3,270	76	7.0	83	6.0 ^c	3.27 ^c	7.5	1.6	0.0	0.0	0.0	0.0	0.1	0.1	0.1	319 ^d	56	122	42	15	
10:30																					
12/6	1,670	66	7.6	82	4.65	2.16 ^c	5.6	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	266 ^d	53	108	22	35	
12/6																					
11/2	860	60	12.3	123	8.1 ^c	3.64 ^c	9.0	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	457 ^d	53	142	59	50	
10:1																					
12/7	834	51	13.1	117	8.67	3.76 ^c	10.7	0.0	0.0	0.0	0.0	0.0	0.1	0.1	0.1	483 ^d	55	133	73	30	
1:40																					

a. Field pH

b. Laboratory pH.

c. Sum of calcium and magnesium in eqm.

d. Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e. Derived from conductivity vs. TDS curves.

f. Determined by addition of analyzed constituents.

g. Gravimetric determination.

h. Annual median and range.

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water Resources (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 1.1

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

LIMITA-MONDOTTA CANAL AT TRACY PUMPING PLANT (STA. 01)

[illegible]

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[illegible]

0.6798

A model of the technology, which is used by the company.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

DELTA-MENDOTA CANAL AT TRACEY PUMPING PLANT (STN. 93 continued)

Date and time sampled PST	Dissolved in air	Dissolved oxygen ppm	Specific imcrection pH at 25°C	Mineral constituents in parts per million										Total dissolved in ppm	Per- cent dissolved in ppm	Hardness as CaCO ₃ Total ppm	Tur- bid- ity in ppm	Coliform MPN/ml	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- tro- gen (NO ₃)	Fluo- ride (F)	Bor- on (B)	Sig- na (SiO ₂)				
1959																			
7/1-13			516	21	17	29	3.6	0.0	23	0.0	96	0.2	0.2	0.0	1.2	Fe 0.01			US28
7/14-24			760	22	18	102	4.4	0.0	73	0.0	188	1.7	0.2	0.0	1.8	Fe 0.02			
7/25-31			997	28	19	147	6.8	0.0	84	0.0	266	2.2	0.0	0.2	1.9	Fe 0.04			
8/1-9			1,030	23	23	147	6.4	0.0	86	0.0	248	1.5	0.2	0.1	2.0	Fe 0.00			
8/10-16			877	22	21	123	5.8	0.0	84	0.0	212	1.8	0.2	0.3	1.9				
8/17-31			694	7.9	16	94	4.4	0.0	88	0.0	131	0.8	0.0	0.2	2.1	Fe 0.02			
9/1-6			920	7.9	18	148	4.5	0.0	92	0.0	96	0.9	0.2	0.1	2.1	Fe 0.02			
9/9-18			1,110	8.2	19	136	6.4	0.0	188	0.0	204	2.8	0.0	0.5	2.5	Fe 0.01			
9/19-25																			
9/26-30			531	8.0	22	63	3.9	0.0	108	0.0	103	1.4	0.2	0.2	2.4	Fe 0.02			
10/1-14			667	8.0	34	75	4.7	0.0	136	0.0	114	0.8	0.1	0.2	2.3	Fe 0.00			
10/15-21			1,010	7.2	52	116	6.4	0.0	179	0.0	186	2.4	0.2	0.4	2.2	Fe 0.00			
10/22-31			770	7.2	13	90	4.8	0.0	145	0.0	133	1.3	0.2	0.2	2.0	Fe 0.00			

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analysis of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

ELDER CREEK AT ORFORD (STA. 95a)

Date and time sampled P.S.T.	Discharge Temp in cfs	Dissolved oxygen ppm	Specific conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ Total T.C. ppm	Total dissolved solids in ppm	Coliform bacteria MPN/ml	Analyzed by	
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Silica (SiO ₂)
1959																		USGS	
1/6	142	11.4	94	20	17	15	1.4	0.0	101	23	28	7.6	0.2	0.2	11	173	21	118	35
1/6				1.00	1.36	0.65	0.04	0.00	1.65	0.45	0.12	0.12	0.01	0.01					
2/9	39	11.7	102	22	19	14	0.8	1.0	140	16	28	9.9	0.1	0.0	10	195	17	151	12
1350				1.45	1.77	0.71	0.02	0.33	2.41	0.33	0.62	0.01	0.01	0.01					
3/13	87	10.6	104	23	16	9.0	1.1	0.10	1.26	11	12	0.0	0.0	0.0	11	154	14	123	10
1330				1.13	1.30	0.39	0.03	0.10	2.16	0.23	0.34	0.00	0.00	0.00					
4/14	47	9.2	89	24	16	9.3	0.6	0.17	1.38	4.8	14	0.0	0.1	0.1	14	156	14	126	5
0730				1.20	1.32	0.40	0.02	0.17	2.26	0.10	0.39	0.00	0.01	0.01					
5/12	12	8.0	85	26	23	15	1.3	0.0	186	19	24	9.6	0.2	0.1	18	212	17	161	8
0650				1.30	1.52	0.73	0.03	0.00	3.05	0.28	0.63	0.00	0.01	0.01					
6/1	2.6	7.4	84	36	27	17	1.3	6	244	12	29	9.9	0.2	0.1	24	258	16	200	15
1230				1.50	2.20	0.74	0.03	0.20	3.51	0.45	0.52	0.01	0.01	0.01					
7/15	Dry																		
8/10	Dry																		
9/1	Dry																		
10/12	Dry																		
11/3	Dry																		
12/1	Dry																		

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn) and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood

Control District (SBCFCD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of

Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE 2-1

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

RIVERS ENTREE NEAR FORTYFIVE (TFA, 1.5 mi)

Date and time of day of P.S.	Discharge Temp in °F	Specific Conductance at 25°C	Dissolved oxygen ppm	Mineral constituents in parts per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total dissolved solids in mg/L
				equivalents per million												
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)	Other constituents
1959																
1/7	45	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/10	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/13	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/16	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/19	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/22	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/25	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/28	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
1/31	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/3	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/6	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/9	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/12	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/15	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/18	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/21	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/24	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/27	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
2/30	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/3	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/6	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/9	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/12	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/15	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/18	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/21	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/24	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/27	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
3/30	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/2	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/5	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/8	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/11	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/14	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/17	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/20	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/23	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/26	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
4/29	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/2	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/5	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/8	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/11	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/14	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/17	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/20	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/23	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/26	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
5/29	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/1	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/4	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/7	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/10	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/13	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/16	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/19	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/22	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/25	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
6/28	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/1	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/4	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/7	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/10	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/13	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/16	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/19	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/22	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/25	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/28	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
7/31	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/3	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/6	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/9	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/12	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/15	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/18	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/21	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/24	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/27	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/30	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0
8/31	51	11.5	9.5	17.7	1.7	1.7	0	1.7	17.6	2.1	1.7	1.7	0	0	0	0

* Data not available.

* Data not available.

* Data not available.

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* Data not available.

* Data not available.

* Data not available.

* Data not available.

* Data not available.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

FALSE RIVER AT WEBB DAM (STA. 112a)

Date and time of sample P.S.T.	Discharge Temp in cfs	Dissolved oxygen ppm	Specific conductance (microhm/cm at 25°C)	Mineral constituents in equivalents per million							Total solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ ppm	Tur- bidity in ppm	Acidified by ppm
				Calcium (Ca) (Mg)	Sodium (Na) (Cl)	Potassium (K) (SO ₄)	Carbonate (CO ₃) (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl) (NO ₃)	Fluoride (F)	Barium (Ba)	Other constituents			
1959	Tidal														
5/14 11:50 ^a	66		203										160		
6/17 11:00	69		339										216		
7/13 10:35	74		1,518										899		
8/10 10:35			1,412										890		
9/15 10:30	68		405										268		
10/14 12:40	69		294										224		
11/26 11:55			330	23				26					176	30	
															USBR

^a Field pH^b Laboratory pH^c Sum of calcium and magnesium in ppm^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.^e Derived from conductivity vs TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control Agency (SBCFA); Los Angeles County Department of Public Health (LADPH); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

FEATHER RIVER NEAR ORVILLE (STA. 19)

Core numbers and P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25 °C µmhos/cm	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Hardness as CaCO ₃ ppm	Tur- bidity in ppm	Analyzed by ^h	
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- ium (K)	Carbon- ate (CO ₃)	Bicor- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)					Other constituents
1959																			UNOS		
1/8 0925	4,000	42	13.0	108	7.2	0.36 ^c	4.3 0.19		0.0 0.00	61 0.00	2.5 0.07					76 ^e	17	47	0	1	Median 2.3
2/10 1600	5,400	42	12.9	102	90.6	7.3	4.6 0.20		0.0 0.00	49 0.00	2.0 0.06					64 ^e	20	40	0	30	Maximum 230.
3/6 1310	7,330	47	11.9	101	91.0	7.5	3.5 0.15		0.0 0.00	48 0.00	2.5 0.07					64 ^e	17	37	0	0	Minimum 1.13
4/9 1455	5,680	52	11.1	100	72.8	7.3	2.1 0.09		0.0 0.00	38 0.00	1.5 0.04					51 ^e	12	34	3	20	
5/8 0700	4,360	55	10.2	96	84.7	7.0	3.2 0.13	0.7 0.02	0.0 0.00	46 0.00	1.0 0.03	0.4 0.01	0.0 0.00	15 ^d	Fe 0.03 Al 0.00 Pb 0.00	56 ^f	16	33	0	0	
6/10 1030	2,360	66	9.1	97	109	7.3	4.6 0.20		0.0 0.00	60 0.00	1.5 0.04					77 ^f	18	46	0	2	
7/17 1135	2,360	72	8.5	97	110	7.6	5.7 0.25		0.0 0.00	66 0.00	2.5 0.07					77 ^f	20	50	0	1	
8/13 1355	1,980	74	8.8	102	111	7.9	5.6 0.24		0.0 0.00	66 0.00	1.0 0.03					78 ^e	20	48	0	3	
9/10 1500	1,000	71	8.8	99	125	7.7	4.7 0.39	1.8 0.05	0.0 0.00	72 0.00	2.4 0.07	0.1 0.00	0.0 0.00	16 ^d	Al 0.13 Fe 0.00	82 ^f	19	52	0	2	
10/15 1730	1,990	60	9.7	97	122	7.5	6.6 0.29		0.0 0.00	70 0.00	4.0 0.11					86 ^e	21	54	0	2	
11/13 1200	1,650	48	11.5	99	121	7.4	6.1 0.27		0.0 0.00	70 0.00	2.0 0.06					85 ^e	21	50	0	1	
12/10 1600	1,380	42	12.0	95	127	7.5	6.5 0.28		0.0 0.00	77 0.00	2.2 0.06					89 ^f	20	56	0	1	

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁺⁶), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

[illegible]

Field pH

Laboratory pH.

Sum of calcium and magnesium in ppm.

Sum of calcium and magnesium in opit.

Iron (Fe), aluminum (Al), arsenic (As), cop-

Derived from conductivity vs TDS curves.

Determined by addition of a

g. Gravimetric determination.

[illegible]

TABLE B-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

INDIAN CREEK NEAR CRESCENT MILLS (STA. 176)

Date and time sampled P.S.T.	Discharge in cfs (m³)	Temp in air (°F)	Dissolved oxygen in ppm	Specific conductance at 25°C	Mineral constituents in ————— parts per million —————										Total dissolved solids in ppm	Percent calcium in ppm	Hardness as CaCO ₃ in ppm	Turbidity in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Cobalt (Co)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)			
1959																			
1/	Not Sampled	Not Sampled																	REGS
2/5	218	44	12.1	98	140	7.2													
3/6	856	40	11.2	86	97.3	7.2													
4/6	691	47	10.2	86	98.3	7.1													
5/7	264	69	8.3	84	121	7.1													
6/5	105	74	8.8	102	144	7.2													
7/17	15	67	6.9	75	240	6.9													
8/13	1280	11	7.8	89	286	7.0													
9/10	5.2	68	7.7	84	293	7.3													
10/5	42	60	8.6	86	183	7.3													
11/13	56	35	10.4	74	183	7.1													
12/10	62	36	11.2	93	178	7.1													
1400																			

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and inorganic chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

INDIAN CLOTH NEAR BENTWOOD (STA. 107)

[illegible]

F. 91.10.10.10

Laboratory of H

System of H_2O and H_2O in com

10000 (Fe) 10000 (Al) 10000 (As)

$\text{Fe}(\text{OH})_3$

Derived from on-farmity v&lt; IUS curv

Determined by addition of analysed on

Gravimetric determination

Averaging method and range respectively

Mineralocorticoid excess made by White House

Control	Distillate	SBCF-D	Monomer
Control	Distillate	SBCF-D	Monomer

P.O. Box 111, Health/LBDPH, Tel. 703-792-1000

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

ITALIAN SLUDGE NEAR MOUTH (STA. 106)

Dots and time of day P.S.T.	Discharge Temp in cfs	Ossolved oxygen ppm	Specific conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent total in ppm	Hardness as CaCO ₃ in ppm	Turbidity in ppm	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					
959	Total	54	9.3	86	67	2.91	0.0	101	108	3.05	0.5			365 ^e	52	40	11	Medley-62
1713		50	8.6	76	223	5.35	0.0	112	168	1.68	1.1			538 ^e	57	202	117	Maximum >7,000.
2710		66	7.6	81	95	4.13	0.0	116	137	3.86	1.1			448 ^e	58	148	53	Minimum 0.62
3112		66	8.7	93	49	2.13	0.0	88	73	2.78	0.2			305 ^e	44	138	66	
479		70	7.8	87	18	0.64	2.5	74	24	0.55	0.2	0.5	0.2	157 ^f	38	77	16	
5713		71	7.8	88	26	0.74	1.7	87	24	0.55	0.2	0.4	0.0	167 ^f	40	82	11	
679		76	7.2	85	13	1.87	0.0	93	58	1.64	0.1	0.0	0.0	294 ^e	49	96	20	
879		84	6.8	87	148	3.08	0.0	84	232	2.32	0.1	0.0	0.0	571 ^e	68	150	81	
979		80	6.7	82	70	1.22	4.6	84	115	1.38	0.1	0.2	0.2	314 ^f	58	106	37	
1079		66	8.2	87	57	2.06	0.0	97	81	2.88	0.2	0.0	0.0	264 ^e	54	103	23	
1179		62	9.0	92	71	3.09	0.0	184	27	0.27	0.2	0.0	0.0	330 ^e	53	137	35	
1277		51	8.7	78	113	4.56	0.0	113	181	1.81	0.6	0.0	0.0	561 ^e	52	228	135	
1379																		

^a Field pH

^b Laboratory pH

^c Sum of calcium and magnesium in eqm

^d Iron (Fe), aluminum (Al), organic (A), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{VI}), reported here as 0.0 except as shown.

^e Derived from conductivity vs TDS curves

^f Determined by addition of analyzed constituents

^g Gravimetric determination

^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

ⁱ Mineral analyses made by United States Geological Survey, Quality Assurance Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), California State Water Resources Control Board (CSWRB), California State Water Resources Control Board (CSWRB), City of Long Beach, Department of Public Health (LBPH), Terminal Tasting Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)
KEEN RIVER NEAR BACSFIELD (STA. 36)

Date discharge sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	% Sat	Specific conductance at 25°C µM	Mineral constituents in equivalents per million											Total dissolved solids in ppm	Per- cent sulfate from total	Hardness as CaCO ₃ ppm	Turbidity in ppm	Coliform bacteria per 100 ml	Analyzed by
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)						
1959																						
1/21 0830	439	53	9.7	89	7.2 ²⁴	17 ²⁰ 0.71	1.28 ²	78 0.00	0.00	1.28 ²	10 0.28 ²	0.2 ²	0.2	122 ⁰	36	66	2	25	Median 2+3	USGS		
2/3 1225	364	48	12.3	106	7.4 ¹²	15 ² 0.65	1.38 ²	81 ² 0.00	0.00	1.38 ²	7.5 ² 0.21 ²	0.2	0.2	116 ⁶	38	54	0	0	Maximum 25			
3/9 1345	601	55	10.2	91	7.2 ²⁴	15 ² 0.65	1.39 ²	85 ² 0.00	0.00	1.39 ²	9.8 ² 0.25 ²	0.2	0.2	116 ⁶	37	56	0	14	Minimum 0+23			
4/6 1600	575	64	9.1	95	7.4 ¹²	15 ² 0.65	1.36 ²	82 ² 0.00	0.00	1.36 ²	7.0 ² 0.20 ²	0.2	0.2	111 ⁶	38	53	0	0				
5/6 1335	431	64	9.7	101	7.6 ⁴	15 ² 0.65	1.31 ²	80 ² 0.00	0.00	1.31 ²	0.0 ² 0.00	0.2	0.2	95 ¹	37	53	0	0				
6/2	490	72	9.0	102	7.6 ⁴	15 ² 0.65	1.28 ²	78 ² 0.00	0.00	1.28 ²	7.0 ² 0.20 ²	0.1	0.1	107 ⁶	39	50	0	14				
7/7 1050	1,732	62	8.7	89	7.8 ⁴	15 ² 0.65	1.28 ²	101 ² 0.00	0.00	1.28 ²	6.0 ² 0.17 ²	0.1	0.1	104 ⁶	34	64	0	5				
8/5 1320	550	77	8.5	101	7.6 ⁴	15 ² 0.65	1.06 ²	78 ² 0.00	0.00	1.06 ²	6.2 ² 0.21 ²	0.2	0.2	109 ⁶	39	50	0	25				
9/3 0730	231	78	8.3	100	7.8 ⁴	2.7 0.22 ²	2.1 ² 0.87 ²	20 ² 0.00	0.00	2.1 ² 0.87 ²	12 ² 0.31 ²	0.0	0.3	100 ⁶	43	56	0	15				
10/6 1100	145	68	10.4	113	7.7 ⁴	23 ² 1.08 ²	1.38 ²	84 ² 0.00	0.00	1.38 ²	11 ² 0.23 ²	0.1	0.1	116 ⁶	46	59	0	3				
11/21 1100	146	53	10.7	96	7.9 ⁴	25 ² 1.09 ²	1.51 ²	92 ² 0.00	0.00	1.51 ²	13 ² 0.37 ²	0.2	0.2	150 ⁶	46	64	0	3				
12/																						

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY DIVISION (NO. 53)

KODJ RIVER, BELOW ISABELLA DAM (STA. 34)

Date and time sampled PST	Discharge in cfs m ³ /s	Temp. in air °F/°C	Dissolved oxygen ppm % Sat.	Specific conductance in air at 25°C	pH	Mineral constituents in equivalents per million										Total solids in ppm	Per- cent total solids in ppm	Hardness in CaCO ₃ ppm	Total Dissolved Solids in ppm	Total Dissolved Solids in ppm
						Calcium (Ca) (mg)	Sodium (Na) (mg)	Potassium (K) (mg)	Carbonate and bicarbonate (CO ₃) (mg)	Bicarbonate (HCO ₃) (mg)	Chloride (Cl) (mg)	Iron (Fe) (mg)	Fluoride (F) (mg)	Boron (B) (mg)	Other constituents					
1/5/59																				
1/7/59	2	68	10.40	86	7.4	1.22 ^b	11	0.51	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/10/59																				
2/7/59	2	61	10.40	89	7.1	1.78 ^b	11	0.51	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/16/59	11.6	52	10.1	71	7.2	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/17/59	3	58	9.2	69	7.5	1.18 ^b	16	0.70	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/18/59	2	58	9.1	69	7.5	1.18 ^b	16	0.70	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/19/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/20/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/21/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/22/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/23/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/24/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/25/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/26/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/27/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/28/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/29/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/30/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
1/31/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/1/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/2/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/3/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/4/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/5/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/6/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/7/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/8/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/9/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/10/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/11/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/12/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/13/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/14/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280
2/15/59	1.09	61	9.2	93	7.4	0.77 ^b	13	0.57	0.0	1.78	6.5	0.28		0.2		11.0 ^a	33	63	20	Maximum 280

a. F = 1.8

b. Laboratory pH

c. Sum of calcium and magnesium in ppm

d. Iron, Fe; aluminum, Al; arsenic, As; copper, Cu; lead, Pb; manganese, Mn; zinc, Zn; and hexavalent chromium, Cr⁶⁺, reported here as 0.0 except as shown

e. Derived from conductivity via TDS curves

f. Determined by addition of analyzed constituents

g. Gravimetric determination

h. Annual median and range, respectively

i. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

j. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (ADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (CDWR) as indicated

San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (ADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (CDWR) as indicated

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (No. 5)
KEEN RIVER NEAR KEENTILLE (ST4, 36b)

Date and time of day P.S.T.	Overscore Temp in °F	Dissolved oxygen in ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Pit- tied in ppm	Hedest- ium in ppm	Tur- bidity in ppm	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- ium (K)	Carbon- ate (CO ₃)	Bicor- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Flo- ride (F)	Boro- Silica (B, SiO ₂)				
1959																		
2/9	280	12	11.4	90	217	7.1												
10/3				0.90 ^b		0.57		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	10525
2/19	140	14	11.2	91	132	7.1												
2/19				0.81 ^b		0.18		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
3/6	507	50	10.2	90	112	7.5												
11/30				0.68 ^c		0.10		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
1/1	450	52	10.1	91	117	7.5												
5/6	602	52	10.1	91	82.6	7.5												
5/15				0.77 ^c		0.11		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
6/15	566	59	9.6	94	68.5	7.2												
8/30				0.12 ^c		0.26		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
7/1	285	66	8.5	91	98.1	7.4												
12/15				0.52 ^c		0.37		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
8/3	166	70	8.2	91	127	7.4												
9/1	111	72	8.2	93	168	7.2												
10/1				0.75	0.17	0.71		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
10/1				0.90 ^b		0.78		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
11/2	152	53	10.0	92	150	7.4												
10/30				1.00 ^b		0.85		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
12/1	132	43	11.5	102	182	7.3												
10/5				1.00 ^b		0.78		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	

a Field pH

b Laboratory pH

c Sum of sodium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination

h Actual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of

Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY DESHORE (NO. 5)

CINCO MILES BEHIND NORTH PARK (STA. 340)

Date and time of day and P. M. T.	Discharge Temp. in C. in 6 ft. of water	Dissolved oxygen in C. in 6 ft. of water	Specific conductance in C. in 6 ft. of water	Mineral constituents in equivalent per million										Total dissolved solids in ppm	Calc. as % of total solids	Hardness as CaCO ₃ in ppm	Total in C. in ppm	Temp. in C.	Total in C. in ppm	Ascorbic acid in ppm
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonates (CO ₃)	Bicarbonates (HCO ₃)	Sulfates (SO ₄)	Chlorides (Cl)	Nitrates (NO ₃)	Fluorides (F)	Boron (B)	Other constituents					
1959																				
1/12	56.5	11.8	103	53.5	7.0			0.0	22		3.2			0.0			17	17	1	10.120
11/18				0.21 ^c		0.13 ^c		0.00	0.76		0.09			0.0						0.0
2/9	57.3	12.2	96	31.7	6.7			0.0	16		1.2			0.0			19	12		0.0
1/30				0.25 ^c		0.06		0.00	0.26		0.03			0.0						0.0
3/9	53	11.3	96	33.3	6.4			0.0	11		0.8			0.0			22	10		0.0
1/20				0.10 ^c		0.10 ^c		0.00	0.23		0.02			0.0						0.0
4/7	56	11.4	108	32.1	7.0			0.0	16		2.0			0.0			23	12	2	0.0
12/10				0.29 ^c		0.09		0.00	0.26		0.06			0.0						0.0
5/2	52	10.6	96	23.0	6.7			0.0	8		0.5			0.0			11	9	2	0.0
11/30				0.18 ^c		0.12 ^c		0.00	0.13		0.04			0.0						0.0
6/2	54.0	9.4	96	22.4	6.6			0.0	8		1.0			0.0			16	9	1	0.0
1/20				0.07 ^c		0.07 ^c		0.00	0.13		0.03			0.0						0.0
7/2	56.0	10.0	100	22.2	6.6			0.0	9		0.5			0.0			16	9	1	0.0
1/20				0.16 ^c		0.06		0.00	0.13		0.03			0.0						0.0
8/	Not sampled																			
9/1	13.6	71	12.3	138	57.4	7.1		0.0	23		2.8			0.0			17	20	1	0.0
9/1				0.21 ^c		0.13 ^c		0.00	0.76		0.09			0.0						0.0
10/1	256	62	15.0	153	54.3	6.9		0.0	22		1.2			0.0			18	19	2	0.0
11/50				0.36 ^c		0.10		0.00	0.30		0.03			0.0						0.0
11/1	13.7	56	11.5	109	66.7	6.8		0.0	28		1.0			0.0			13	22	1	0.0
11/10				0.28 ^c		0.08		0.00	0.16		0.03			0.0						0.0
12/1	11.5	45	12.0	99	71.2	7.1		0.0	28		1.8			0.0			11	19	1	0.0
11/20				0.56 ^c		0.25		0.00	0.16		0.03			0.0						0.0

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺) reported here as 0.0 except as shown

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Amel medium and range

i Amel medium and range

j Amel medium and range

k Amel medium and range

l Amel medium and range

m Amel medium and range

n Amel medium and range

o Amel medium and range

p Amel medium and range

q Amel medium and range

r Amel medium and range

s Amel medium and range

t Amel medium and range

Analyses made by United States Geological Survey, Quality of Water Branch, U.S. Geological Survey, Department of the Interior, Bureau of Reclamation, USBR, Los Angeles District, California Division of Water and Power, LADWP, City of Los Angeles, Department of Public Health, LADPH, City of Long Beach, Department of Public Health, LDBPH, Terminal Testing Laboratories, Inc. (TTL) or California Departmental Water Resources (DWR) as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

XINDS RIVER BELOW PROPOSED WPP (SITE 24)

Date and time of sample P.S.T.	Dissolved Temp in °C	Dissolved oxygen ppm	Specific conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Per- cent total solids in ppm	Hardest as CaCO ₃ Total N.C. ppm	Tur- bid- ity in ppm	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Corban- ium (CO ₃)	Bicarb- onate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)	Boron (B)	Silica (SiO ₂)			
1959																		1959
1/13	35.4	9.0	87	106	6.5													
1530																		
2/1	31.2	10.2	90	94.8	6.8													
0815																		
3/9	35.3	9.4	90	58.3	7.1													
1630																		
4/6	21.1	8.7	96	149	7.2													
1555																		
5/5	16.7	8.8	94	139	7.2													
0700																		
6/2	9.30	9.8	95	41.7	7.1													
0630																		
7/6	9.72	8.7	95	39.9	7.1													
1415																		
8/5	7.27	8.5	83	39.8	7.0													
0925																		
9/1	7.33	8.4	93	38.0	7.0													
1030																		
10/6	5.9	8.4	91	30.8	7.1													
0905																		
11/10	5.7	9.5	95	222	7.7													
1530																		
12/1	6.2	9.9	90	203	7.5													
1500																		

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination

h Mineral analysis made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWD), Los Angeles County Flood Control District (LAFCD), Los Angeles County Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBOPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

i Gravimetric determination

Mineral analysis made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWD), Los Angeles County Flood Control District (LAFCD), Los Angeles County Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBOPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE 9-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

LEED RY CLOUGH WITH RED VENTS (TTL, 1953)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Oxidized oxygen ppm	Specific conductance at 25°C	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent solid in ppm	Hordas on CaCO ₃ Total N.C. ppm	T _{eq} - Corform [®] in ppm	Analyzed by 1
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)					
1/4	1,000																	USGS	
1/10	1,000	2	10.0	26			10 0.78		22 1.71	12 0.93	11 0.83			0.2		146 ^a	1/1 2 60		
1/10	1,000	4	10.0	27			10 0.78		20 1.48	12 0.93	12 0.93			0.1		133 ^a	1/1 0 12		
1/4	1,000	10	9.8	26			22 1.67		112 8.6	20 1.48	20 1.48			0.2		177 ^a	1/1 0 100		
1/1	1,000	6	9.6	26			10 0.78		88 6.8	12 0.93	11 0.83			0.1		119 ^a	1/1 0 80		
1/1	1,000	11	8.8	26			12 0.93		87 6.8	12 0.93	12 0.93			0.1		130 ^a	1/1 0 15		
1/1	1,000	10	8.0	26			18 1.36		104 7.9	16 1.13	16 1.13			0.1		150 ^a	1/1 0 70		
1/4	1,000	10	8.4	26			17 1.27		98 7.4	16 1.13	16 1.13			0.1		144 ^a	1/1 0 20		
1/10	1,000	10	8.2	26			18 1.36		104 7.9	16 1.13	16 1.13			0.1		108 ^a	1/1 0 100		
9/1 09:40	1,000	77	9.0	29	221	16 1.13	22 1.68	2.4 0.18	102 7.4	16 1.13	16 1.13			0.2	Fe 0.13 PO ₄ 0.20 Al 0.20	115 ^a 31 78 0	1/1 0 40		
10/1 10:00	1,000	64	8.2	27	220	17 1.27	10 0.78	2.4 0.18	102 7.4	16 1.13	16 1.13			0.2		138 ^a 31 81 0	1/1 0 30		
11/1 10:16	1,000	56	9.0	27	260	17 1.27	10 0.78	2.4 0.18	98 7.4	13 0.93	13 0.93			0.1		123 ^a 29 74 0	1/1 0 30		
12/1 11:00	1,000	50	9.2	24	197	17 1.27	10 0.78	2.4 0.18	97 7.4	13 0.93	13 0.93			0.1		120 ^a 30 71 0	1/1 0 32		

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (W. 4)

LIMIT: PPB 20 10000 AT TEMPERATURE (°C. 92)

Date and time sampled PST	Discharge temp in C in 10 min	Dissolved oxygen in % Sat	Specific conductance at 25°C	Mercury constituents in — parts per million —						parts per million			Total solids in ppm	Per cent suspended solids in ppm	Hardness in ppm	Turbidity in NTU	Accepted by
				Cadmium (µg)	Mercury (µg)	Sodium (µM)	Potassium (µM)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Calcium (Ca)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents	
10/27	71.4	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
10/28	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
10/29	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
10/30	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
10/31	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/1	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/2	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/3	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/4	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/5	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/6	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/7	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/8	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/9	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/10	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/11	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/12	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/13	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/14	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/15	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/16	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/17	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/18	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/19	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/20	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/21	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/22	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/23	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/24	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/25	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/26	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/27	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/28	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/29	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
11/30	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/1	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/2	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/3	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/4	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/5	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/6	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/7	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/8	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/9	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/10	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/11	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/12	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/13	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/14	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/15	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/16	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/17	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/18	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/19	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/20	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/21	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/22	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/23	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/24	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/25	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/26	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/27	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/28	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/29	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/30	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1
12/31	10.0	10	100	—	—	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1

P. 4. 11

1. Analyzed

2. Analyzed

3. Analyzed

4. Analyzed

5. Analyzed

6. Analyzed

7. Analyzed

8. Analyzed

9. Analyzed

10. Analyzed

11. Analyzed

12. Analyzed

13. Analyzed

14. Analyzed

15. Analyzed

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

MCCLOD RIVER ABOVE BRANTA LAKE (STA. 18)

Date of sample P.S.T.	Discharge Temp in cfs in deg F	Dissolved oxygen in ppm % Sat	Specific conductance in $\mu\text{mhos/cm}$ at 25°C	pH ^a	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Hardness as CaCO ₃ Total N.C. ppm	Temp in deg F	Caliform ^b MPN/ml	Analyzed by ^h	
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)						Boron (B)
1959																		UFOS		
1/ Not Sampled																				
2/ Not Sampled																				
3/2 1615	2,720	47	11.5	98	88.4	7.5	0.76 ^c	3.6 0.16	0.0	0.77 ^c	2.5 0.07 ^c	0.0	0.0	0.0	79 ^e	10	95	0	0	Median n 60
4/7 0830	2,350	46	11.4	95	90.7	7.3	0.80 ^c	3.1 0.13	0.0	0.85 ^c	1.2 0.03	0.0	0.0	0.0	81 ^e	14	40	0	90	Maximum 330
5/5 1245	1,790	51	11.0	98	88.5	7.3	0.40	3.6 0.36	0.0	0.90 ^c	2.5 0.07 ^c	0.4	0.1	0.0	73 ^f	19	35	0	0	Minimum 60-165
6/3 0900	1,440	55	10.8	108	7.3	0.78 ^c	5.6 0.24	0.0	0.61 ^c	1.00 ^c	3.0 0.08 ^c	0.0	0.0	0.0	96 ^e	21	44	0	11	
7/13 1240	1,190	57	9.9	95	91.0	7.7	0.88 ^c	3.6 0.36	1.5 0.04	0.89 ^c	1.0 0.03 ^c	0.0	0.0	0.0	85 ^f	21	37	0	0	
8/10 1130	1,130	54	10.1	94	92.2	7.4	0.76 ^c	5.7 0.25	0.0	0.93 ^c	1.0 0.03 ^c	0.0	0.0	0.0	80 ^g	25	38	0	9	
9/9 0850	1,080	51	10.8	96	92.6	7.4	0.84 ^c	3.5 0.29	1.2 0.03 ^c	0.99 ^c	0.6 0.01 ^c	0.5	0.0	0.0	86 ^f	24	36	0	7	
10/12 1400	1,070	56	10.3	94	129	7.4	1.05 ^c	8.6 0.37	0.0	1.15 ^c	4.8 ^c	0.0	0.0	0.0	115 ^h	27	51	0	13	
11/9 1200	1,030	48	11.5	96	93.2	7.5	0.78 ^c	5.3 0.23	0.0	0.99 ^c	1.0 0.03 ^c	0.0	0.0	0.0	83 ^e	24	36	0	1	
12/7 1335	998	42	11.9	98	94.7	7.5	0.76 ^c	5.5 0.24	0.0	0.99 ^c	1.2 0.03 ^c	0.0	0.0	0.0	80 ^g	24	38	0	9	

^a Field pH^b Laboratory pH^c Sum of calcium and magnesium in ppm^d Non (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.^e Derived from conductivity vs TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively. Calculated from analysis of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBDPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

WENTWORTH RIVER BELOW ERIKSDALE DAM (STA. 324)

Date and time of day and P.S.T.	Discharge in cfs in day	Temp in deg F	Dissolved oxygen in %	Specific conductance in micro mhos/cm at 25°C	Mineral constituents in parts per million												Total dissolved solids in ppm	Percent solids in ppm	Hardness as CaCO ₃ in ppm	Total calcium in ppm	Assigned to CaCO ₃ in ppm
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					
1/1/4	53	53	14.2	94																	
1/15	50	50	15.4	90																	
1/16	50	50	15.4	90																	
1/17	40	40	10.5	91																	
1/18	51	51	9.8	87																	
1/19	50	50	9.4	80																	
1/20	56	56	14.8	79																	
1/21	59	59	9.5	94																	
1/22	72	72	7.5	85																	
1/23	77	77	7.0	84																	
1/24	68	68	9.0	80																	
1/25	68	68	10.0	84																	
1/26	68	68	10.0	84																	
1/27	68	68	10.0	84																	
1/28	68	68	10.0	84																	
1/29	68	68	10.0	84																	
1/30	68	68	10.0	84																	
1/31	68	68	10.0	84																	
1/1/5	68	68	10.0	84																	
1/2/5	68	68	10.0	84																	
1/3/5	68	68	10.0	84																	
1/4/5	68	68	10.0	84																	
1/5/5	68	68	10.0	84																	
1/6/5	68	68	10.0	84																	
1/7/5	68	68	10.0	84																	
1/8/5	68	68	10.0	84																	
1/9/5	68	68	10.0	84																	
1/10/5	68	68	10.0	84																	
1/11/5	68	68	10.0	84																	
1/12/5	68	68	10.0	84																	
1/13/5	68	68	10.0	84																	
1/14/5	68	68	10.0	84																	
1/15/5	68	68	10.0	84																	
1/16/5	68	68	10.0	84																	
1/17/5	68	68	10.0	84																	
1/18/5	68	68	10.0	84																	
1/19/5	68	68	10.0	84																	
1/20/5	68	68	10.0	84																	
1/21/5	68	68	10.0	84																	
1/22/5	68	68	10.0	84																	
1/23/5	68	68	10.0	84																	
1/24/5	68	68	10.0	84																	
1/25/5	68	68	10.0	84																	
1/26/5	68	68	10.0	84																	
1/27/5	68	68	10.0	84																	
1/28/5	68	68	10.0	84																	
1/29/5	68	68	10.0	84																	
1/30/5	68	68	10.0	84																	
1/31/5	68	68	10.0	84																	
1/1/6	68	68	10.0	84																	
1/2/6	68	68	10.0	84																	
1/3/6	68	68	10.0	84																	
1/4/6	68	68	10.0	84																	
1/5/6	68	68	10.0	84																	
1/6/6	68	68	10.0	84																	
1/7/6	68	68	10.0	84																	
1/8/6	68	68	10.0	84																	
1/9/6	68	68	10.0	84																	
1/10/6	68	68	10.0	84																	
1/11/6	68	68	10.0	84																	
1/12/6	68	68	10.0	84																	
1/13/6	68	68	10.0	84																	
1/14/6	68	68	10.0	84																	
1/15/6	68	68	10.0	84																	
1/16/6	68	68	10.0	84																	
1/17/6	68	68	10.0	84																	
1/18/6	68	68	10.0	84																	
1/19/6	68	68	10.0	84																	
1/20/6	68	68	10.0	84																	
1/21/6	68	68	10.0	84																	
1/22/6	68	68	10.0	84																	
1/23/6	68	68	10.0	84																	
1/24/6	68	68	10.0	84																	
1/25/6	68	68	10.0	84																	
1/26/6	68	68	10.0	84																	
1/27/6	68	68	10.0	84																	
1/28/6	68	68	10.0	84																	
1/29/6	68	68	10.0	84																	
1/30/6	68	68	10.0	84																	
1/31/6	68	68	10.0	84																	
1/1/7	68	68	10.0	84																	
1/2/7	68	68	10.0	84																	
1/3/7	68	68	10.0	84																	
1/4/7	68	68	10.0	84																	
1/5/7	68	68	10.0	84																	
1/6/7	68	68	10.0	84																	
1/7/7	68	68	10.0	84																	
1/8/7	68	68	10.0	84																	
1/9/7	68	68	10.0	84																	
1/10/7	68	68	10.0	84																	
1/11/7	68	68	10.0	84																	
1/12/7	68	68	10.0	84																	
1/13/7	68	68	10.0	84																	
1/14/7	68	68	10.0	84																	
1/15/7	68	68	10.0	84																	
1/16/7	68	68	10.0	84																	
1/17/7	68	68	10.0	84																	
1/18/7	68	68	10.0	84																	
1/19/7	68	68	10.0	84																	
1/20/7	68	68	10.0	84																	
1/21/7	68	68	10.0	84																	
1/22/7	68	68	10.0	84																	
1/23/7	68	68	10.0	84																	
1/24/7	68	68	10.0	84																	
1/25/7	68	68	10.0	84																	
1/26/7	68	68	10.0	84																	
1/27/7	68	68	10.0	84																	
1/28/7	68	68	10.0	84																	
1/29/7	68	68	10.0	84																	
1/30/7	68	68	10.0	84																	
1/31/7	68	68	10.0	84																	
1/1/8	68	68	10.0	84																	
1/2/8	68	68	10.0	84																	
1/3/8	68	68	10.0	84																	
1/4/8	68	68	10.0	84																	
1/5/8	68	68	10.0	84																	

Field PM

In Laboratory PM

In Central and local agencies in

1. Central and local agencies in

2. Central and local agencies in

3. Central and local agencies in

4. Central and local agencies in

5. Central and local agencies in

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TABLE 2-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

MERCED RIVER NEAR STOKELAND (STA. 32)

Date and time sampled P.S.T.	Oscillograph Temp in °C	Dissolved oxygen in ppm % sat	Specific conductance (microhm/cm at 25°C)	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Per cent suspended in ppm	Hardness as CaCO ₃ Total PPM	Turbidity in Nephelometric Units	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					
1954																			UNBS	
1/12 1600	28.5	8.6 84	222	6.6	1.50 ^c		1.9 0.83		0.0 0.00	1.02 0.46		9.0 0.38					143 ^e 37	70 0 25	Median 25.	
2/2 1330	15.7	9.5 89	272		1.56 ^c		3.1 1.35		0.0 0.00	1.23 0.52		18 0.73					175 ^e 46	78 0 2	Maximum 7,000	
3/9 1145	15.7	9.0 91	359	7.7	1.72 ^c	36 1.57			0.0 0.00	1.10 0.49		30 0.85					210 ^e 48	86 0 2	Minimum 2.3	
4/6 0845	13.8	8.6 91	247	7.6	1.32 ^c	24 1.00			0.0 0.00	0.98 0.43		20 0.75					150 ^e 44	66 0 0		
5/4 1235	22.3	9.5 104	265	7.4	1.0 0.95	8.1 0.27	25 1.09	2.3 0.86	0.0 0.00	1.16 0.51	17 0.75	18 0.73	2.2 0.90	0.0 0.00	Fe 0.15 Al 0.07 ^d Pb 0.30 Mn 0.03				0 15	
6/1 1050	17.2	8.0 93	266	7.3	1.22 ^c	28 1.22			0.0 0.00	1.22 0.52		18 0.73					171 ^e 45	76 0 25		
7/6 0830	9.7	8.3 96	256	7.3	1.32 ^c	34 1.42			1.4 0.57	0.9 0.41		19 0.73			Tot. Alx. 117				165 ^e 53	0 2
8/4 1225	9.6	8.2 102	255	7.4	1.40 ^c	29 1.27			0.0 0.00	1.14 0.51		19 0.73			0.2				164 ^e 47	0 35
9/10 1400	7.7	8.4 101	385	7.4	1.27 0.25	8.4 0.27	43 1.87	2.8 0.97	0.0 0.00	1.69 0.75	11 0.43	38 1.57	2.8 0.95	0.1 0.01	Pd 0.25 Al 0.01 ^d Cu 0.01				248 ^e 47	0 4
10/8 1345	10.8	9.1 100	332	7.3	1.06 ^c	34 1.42			0.0 0.00	1.47 0.65		25 0.95			0.0				214 ^e 44	0 10
11/5 1300	10.9	10.0 101	345	7.5	1.40 ^c	36 1.57			0.0 0.00	2.41 1.05		23 0.95			0.0				228 ^e 46	0 6
12/10 1430	13.1	11.1 102	316	7.5	1.38 ^c	34 1.42			0.0 0.00	1.44 0.65		19 0.73			0.0				203 ^e 45	0 5

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-3

ANALYSES OF SURFACE WATER

CENTRAL VALLEY RIVER (NO. 5)

MILL CREEK NEAR LOS ANGELES (PPM, PH)

Date and time sampled P.S.T.	Dissolved Temp. in °F	Dissolved oxygen ppm %Sat	Specific conductance at 25°C pH	Major constituents in equivalents per million										Total dissolved solids in ppm	Percent as CaCO ₃ ppm	Total Hardness as CaCO ₃ ppm	Total Softness as CaCO ₃ ppm	Analyzed by		
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Other constituents							
1/6/4	4.1	45	11.5	95	109	7.3	7.1	0.11	0.0	0.0	0.0	0.0	0.0	0.0	0.0	61 ^a	28	7	1	Meridian
1/6/4	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	1/6	Meridian
2/2/4	4.5	46	11.1	94	154	7.3	12	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	114 ^a	36	2	10	Meridian
3/13	20.1	53	10.9	100	133	7.7	11	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	92 ^a	34	2	11	Meridian
4/10	14.8	44	10.5	97	126	7.3	8.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	94 ^a	32	3	11	Meridian
5/11	14.7	63	9.5	98	131	7.3	8.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	97 ^a	33	3	11	Meridian
6/3	13.0	70	8.9	96	139	7.3	11	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	103	34	3	11	Meridian
7/15	18.0	89	11.1	140	201	7.8	14	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	140 ^a	32	2	3	Meridian
8/17	20.0	93	9.4	130	227	7.3	16	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	146 ^a	32	1	1	Meridian
9/1	20.0	82	10.6	133	247	7.4	17	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	148 ^a	30	2	1	Meridian
10/12	18.0	65	9.3	98	214	7.5	20	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	174 ^a	36	2	1	Meridian
11/1	18.0	62	10.0	143	209	7.4	19	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	144 ^a	31	2	1	Meridian
12/11	18.0	45	11.5	95	212	7.3	19	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	146 ^a	31	2	1	Meridian

Footnote

a. Laboratory test

b. Field test

c. Data from

d. Data from

e. Data from

f. Data from

g. Data from

h. Data from

i. Data from

j. Data from

k. Data from

l. Data from

m. Data from

n. Data from

o. Data from

p. Data from

q. Data from

r. Data from

1. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

2. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

3. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

4. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

5. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

6. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

7. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

8. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

9. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

10. All data were obtained from the California Department of Public Health, Division of Laboratories, and the United States Public Health Service, San Francisco, California.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

NIKELOMIE RIVER BELOW COSUMES RIVER (STA. 235)

Date and time sampled P.S.T.	Discharge Temp in cfs	Dissolved oxygen in %Sat ppm	Specific conductance in cmhos/cm at 25°C	pH	Mineral constituents in equivalents per million													Total dis- solved solids in ppm	Per- cent sod- ium in ppm	Hardness as CaCO ₃ Total N.C. ppm	Tot- als- Caliform MPN/ml water	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- dioxide (CO ₂)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Nit- rate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)	Other constituents					
1959	Tidal																		USBR			
1/13			147					4.6					3.6					128	14			
1050								2.3					1.8					54	17			
3/19	56		60															68	17			
1270																						
4/11	66		77					3.0					2.8					88	14			
1125																						
4/18	67		95					3.0					2.1					112	25			
1115																						
6/15			159					9.0					18					136	28			
1140																						
7/13	75		216					14					21					132	45			
1120													11									
8/11	75		183					1.5										152	35			
1250																						
9/14	71		213					17					17					100	25			
1110																						
10/12	69		115					6.7					4.3					76	27			
1315																						
11/9	56		110					6.9					5.7					88	15			
1145																						
12/18			130					4.6					5.0									
1140																						

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses by U.S. Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE No. 4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

MARIPOSA RIVER BEYOND CROCKER-HARRIS DAM (TPT. 236)

Date sampled P. S. T.	Discharge in cfs in ft ³	Dissolved oxygen in ppm % sat.	Specific conductance at 25°C in micromhos/cm	Mineral constituents in equivalents per million							Total dissolved solids in ppm	Hardness as CaCO ₃ Total N.C. in ppm	Total Hardness as CaCO ₃ in ppm	Applied by
				Calcium (Ca) (mg)	Sodium (Na) (mg)	Potassium (K) (mg)	Bicarbonate (CO ₃) (HCO ₃) (mg)	Sulfate (SO ₄) (mg)	Chloride (Cl) (mg)	Nitrate (NO ₃) (mg)	Fluoride (F) (mg)	Boron (B) (mg)	Silica (SiO ₂) (mg)	
1969	7164													
11.11			188	7.4					7.8					18
11.12			188	7.4					7.8					18
11.13			114	5.5					7.8					18
11.14			74	4.4					5.2					18
11.15			209	3.3					21					18
11.16			264	5.5					21					18
11.17			71	3.3					11					18
11.18			258	18					18					18
11.19			189	12					7.8					18
11.20			229	12					18					18
11.21			81	12					18					18
11.22				12					18					18

P. S. T. M

L. C. C. C. C. C.

S. C. C. C. C. C.

A. C. C. C. C. C.

D. C. C. C. C. C.

E. C. C. C. C. C.

F. C. C. C. C. C.

G. C. C. C. C. C.

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J. C. C. C. C. C.

K. C. C. C. C. C.

L. C. C. C. C. C.

M. C. C. C. C. C.

N. C. C. C. C. C.

O. C. C. C. C. C.

P. C. C. C. C. C.

Q. C. C. C. C. C.

R. C. C. C. C. C.

S. C. C. C. C. C.

T. C. C. C. C. C.

U. C. C. C. C. C.

V. C. C. C. C. C.

W. C. C. C. C. C.

X. C. C. C. C. C.

Y. C. C. C. C. C.

Z. C. C. C. C. C.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

MORTLAND RIVER NEAR LANTANA PLANA (STA. 23a)

Date and time analyzed P.S.T.	Dissolved Temp. in °C	Dissolved oxygen % Sat	pH	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Hardness as CaCO ₃ Total Hardness in ppm	Turbid- ity in m.p.m.	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)				
1959 1/14 0940	67	54	11.0	4.8 0.20	2.1 0.17	2.0 0.09	0.2 0.01	0.0 0.00	16 0.76	9.6 0.28	3.0 0.08	0.0 0.00	0.0 0.00	8.6 0.01	38 ^a	18	20	7	Median 0.68
2/11 1115	111	48	11.3	7.6 ^c 0.31 ^c	3.6 0.16	3.6 0.16	0.0 0.00	0.0 0.00	15 0.65	4.0 0.11	4.0 0.11	0.0 0.00	0.0 0.00	0.1 0.01	81 ^e	17	38	26	Maximum 500.
3/10 1430	690	51	10.9	7.6 ^c 0.31 ^c	2.2 0.10	2.2 0.10	0.0 0.00	0.0 0.00	20 0.90	1.4 0.04	1.4 0.04	0.0 0.00	0.0 0.00	0.0 0.00	37 ^f	24	16	0	Minimum 0.06
4/15 1410	169	55	10.5	4.0 ^c 0.17 ^c	1.7 0.07	1.7 0.07	0.0 0.00	0.0 0.00	18 0.80	0.0 0.00	0.0 0.00	0.0 0.00	0.0 0.00	0.0 0.00	37 ^f	15	20	5	0
5/15 1445	669	53	10.7	5.6 0.24	0.5 0.04	2.8 0.12	1.1 0.05	0.0 0.00	20 0.90	3.8 0.15	3.8 0.15	0.0 0.00	0.0 0.00	0.1 0.01	35 ^f	26	16	0	5
6/5 0835	675	53	10.9	7.6 ^c 0.31 ^c	2.5 0.11	2.5 0.11	0.0 0.00	0.0 0.00	20 0.90	3.5 0.10	3.5 0.10	0.0 0.00	0.0 0.00	0.0 0.00	38 ^f	23	18	2	15
7/3 1015	690	53	11.0	4.8 0.20	2.4 0.10	2.4 0.10	0.0 0.00	0.0 0.00	19 0.85	3.0 0.08	3.0 0.08	0.0 0.00	0.0 0.00	0.0 0.00	38 ^f	24	16	0	0
8/12 1000	712	59	9.8	7.6 ^c 0.31 ^c	2.5 0.11	2.5 0.11	0.0 0.00	0.0 0.00	22 0.95	1.5 0.04	1.5 0.04	0.0 0.00	0.0 0.00	0.1 0.01	36 ^f	26	16	0	10
9/1 0930	318	69	9.6	5.6 0.24	1.2 0.10	2.7 0.12	0.8 0.04	0.0 0.00	22 0.95	4.0 0.15	4.0 0.15	0.0 0.00	0.0 0.00	0.1 0.01	41 ^f	23	19	1	1
10/8 0930	108	59	9.9	4.8 0.20	2.3 0.10	2.3 0.10	0.0 0.00	0.0 0.00	18 0.80	3.0 0.08	3.0 0.08	0.0 0.00	0.0 0.00	0.0 0.00	36 ^f	25	15	0	2
11/9 1015	61	56	10.1	7.6 ^c 0.31 ^c	2.5 0.11	2.5 0.11	0.0 0.00	0.0 0.00	16 0.76	3.5 0.10	3.5 0.10	0.0 0.00	0.0 0.00	0.0 0.00	31 ^f	27	15	2	3
12/1 1300	68	57	10.0	7.6 ^c 0.31 ^c	2.3 0.10	2.3 0.10	0.0 0.00	0.0 0.00	15 0.65	2.8 0.08	2.8 0.08	0.0 0.00	0.0 0.00	0.0 0.00	35 ^f	26	14	2	1

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Metropolitan Water Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

OLD RIVER AT CLIFTON COURT FERRY (Sta. 114)

Date and time of day and P.S.T.	Discharge Temp in °C	Dissolved oxygen ppm	Specific conductance at 25°C $\mu\text{mhos/cm}^2$	pH	Mineral constituents in										ports per million					Total dissolved solids in ppm	Penetration as CaCO_3 ppm	Hardness as CaCO_3 ppm	Temp. stability in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonates (CO_3)	Bicarbonates (HCO_3)	Sulfates (SO_4)	Chlorides (Cl)	Nitrate (NO_3)	Fluoride (F)	Boron (B)	Silica (SiO_2)	Other constituents							
7/1/61	56	5.7	81	6.61	7.2	27.76	7.1	0.0	115	11.5	11.5	11.0	0.1	0.1	0.1	376 ^e		51	1.3	51	20	Median 5%	USPS	
10/1/61	51	5.7	79	7.4	7.05	1.05	3.74	0.05	178	17.8	17.8	11.0	0.1	0.1	0.1	534 ^e		53	1.71	53	Maximum >7 000			
2/2/62	51	5.7	79	7.4	7.05	1.05	3.74	0.05	178	17.8	17.8	11.0	0.1	0.1	0.1	534 ^e		53	1.71	53	Maximum >7 000			
5/12/62	50	5.7	83	6.17	7.3	2.55	6.9	0.05	180	18.0	18.0	9.7	0.2	0.2	0.2	351 ^e		51	1.40	52	70	Minimum 5%		
6/7/62	49	5.7	84	5.89	7.1	2.65	4.4	0.05	86	8.6	8.6	6.6	0.3	0.3	0.3	778 ^e		49	1.30	61	8%			
9/13/62	70	7.5	83	4.78	7.2	2.8	4.2	0.05	111	11.1	11.1	7.2	0.1	0.1	0.1	273 ^f		45	1.21	50				
6/9/63	70	7.5	83	2.89	7.3	1.75	2.7	0.05	88	8.8	8.8	31	0.0	0.0	0.0	164 ^g		40	86	14	60			
7/2/63	76	7.1	84	4.92	7.4	2.31	5.6	0.05	107	10.7	10.7	7.6	0.1	0.1	0.1	288 ^g		51	11.5	27	20			
8/9/63	89	6.9	87	8.70	7.5	2.78	12.1	0.05	89	8.9	8.9	19.2	0.1	0.1	0.1	483 ^h		67	131	64	70			
9/8/63	80	6.8	84	6.08	7.3	2.4	7.5	0.05	97	9.7	9.7	13.2	0.0	0.0	0.0	330 ^f		56	129	42	10			
11/30/63	66	8.1	86	4.29	7.3	2.05	3.27	0.12	159	15.9	15.9	13.2	0.0	0.0	0.0	244 ^g		52	100	17	30			
10/5/65	57	12.2	118	8.11	8.1	3.78	2.18	0.05	166	16.6	16.6	7.5	0.2	0.2	0.2	461 ^g		59	186	63	20			
11/2/65	57	12.2	118	8.11	8.1	3.78	2.18	0.05	166	16.6	16.6	7.5	0.2	0.2	0.2	461 ^g		59	186	63	20			
13/40	50	11.9	104	8.9	7.7	3.52	1.03	0.05	233	23.3	23.3	16.0	0.2	0.2	0.2	486 ^g		56	178	62	35			
19/7/66																								

Field OH.

Laboratory pH

Sum of calcium and magnesium in ppm.

Al (Fe) aluminum (As) arsenic (As) copper (C

Derived from conductivity vs TDS curves.

Determined by addition of analyzed constituents.

Calimetric determination

6. Actual median and range respectively. Calculated

United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY BASIN (No. 5)

OLD RIVER AT MARSHVILLE ISLAND (STA. 212)

Date and time of day and P.S.T.	Oscenorge Temp in cfs	Dissolved oxygen in %	Specific conductance ^a in %Sat	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent of solids in ppm	Hydrazine or CCl ₄ in ppm	Total N.C. in ppm	Total Turbidity in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Barium (Ba)	Other constituents				
1959	Tidal																			USGS
1/12	53	10.1	92	462	7.2	2.56 ^c	14.23	14.23	0.0	7.5	1.23	60	1.69		0.1		4.2	1.5	63	1.5
1/13																				
2/9	49	10.1	88	536	7.3	2.80 ^c	2.44	2.44	0.0	8.7	1.75	72	2.03		0.1		4.0	14.4	71	3
3/30																				
3/12	60	8.2	82	497	7.2	2.42 ^c	1.91	1.91	0.0	9.0	1.48	66	1.06		0.2		4.2	1.4	56	30
4/2	60	8.9	89	253	7.3	1.60 ^c	1.00	1.00	0.0	7.4	1.18	27	0.70		0.1		4.0	38	80	4.0
6/9																				
5/12	68	8.1	88	208	7.3	1.5	1.0	1.0	0.0	7.3	1.00	30	0.56		0.1	Fe 0.15, Cu 0.03, d	3.2	30	66	4.0
11/30																				
6/9	67	8.0	77	249	7.3	1.40 ^c	0.96	0.96	0.0	9.1	1.09	24	0.68		0.1		3.9	74	0	4.5
8/15																				
7/2	74	8.1	94	631	7.4	2.22 ^c	3.65	3.65	0.0	9.1	1.49	36	1.55		0.1		6.2	111	36	4.5
8/30																				
8/9	78	7.7	93	573	7.3	1.96 ^c	3.18	3.18	0.0	7.9	1.29	38	1.61		0.0		3.34	62	94	4.0
10/6																				
9/11	79	7.5	91	801	7.3	1.90	3.0	3.0	0.0	3.9	0.79	33	1.5		0.1	Fe 0.03, PO ₄ 1.4, d	4.57 ^f	54	172	13
10/6	66	8.0	85	379	7.3	1.88 ^c	4.35	4.35	0.0	10.1	1.66	60	1.09		0.2		21.9 ^e	50	94	11
11/6	58	9.4	94	289	7.5	1.78 ^c	3.13	3.13	0.0	10.1	1.66	45	0.96		0.1		16.7 ^e	39	88	5
12/11	50			331	7.3	1.90 ^c	3.2	3.2	0.0	10.1	1.72	49	1.10		0.1		13.4 ^e	44	95	9
0955																				17

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in gm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn) and hexavalent chromium (Cr⁶⁺) reported here as 0.0 except as shown.

e Derived from conductivity vs. TDS curves

f Determined by analysis of analyzed constituents

g Gravimetric determination

h Gross median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); United States Department of the Interior, Bureau of Reclamation (USBR); Los Angeles Department of Water and Power (LADWP); City of Long Beach, Department of Public Health (LBPDH); Tammam Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE 3-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO.)

THE UNIVERSITY OF CHICAGO

[illegible]

F. J. O'Hara

100

[illegible]

	As	Cu	Pb	Mn	Zn	Cr
iron						
Fe						

Derive from and by by TOS uses

Date received by the publisher: 11/11/2011

Conclusions and future research

Analysed in 1998. ^aAnalysed in 1999. ^bAnalysed in 2000. ^cAnalysed in 2001. ^dAnalysed in 2002. ^eAnalysed in 2003. ^fAnalysed in 2004. ^gAnalysed in 2005. ^hAnalysed in 2006. ⁱAnalysed in 2007. ^jAnalysed in 2008. ^kAnalysed in 2009. ^lAnalysed in 2010. ^mAnalysed in 2011. ⁿAnalysed in 2012. ^oAnalysed in 2013. ^pAnalysed in 2014. ^qAnalysed in 2015. ^rAnalysed in 2016. ^sAnalysed in 2017. ^tAnalysed in 2018. ^uAnalysed in 2019. ^vAnalysed in 2020. ^wAnalysed in 2021. ^xAnalysed in 2022. ^yAnalysed in 2023. ^zAnalysed in 2024. ^{aa}Analysed in 2025. ^{ab}Analysed in 2026. ^{ac}Analysed in 2027. ^{ad}Analysed in 2028. ^{ae}Analysed in 2029. ^{af}Analysed in 2030. ^{ag}Analysed in 2031. ^{ah}Analysed in 2032. ^{ai}Analysed in 2033. ^{aj}Analysed in 2034. ^{ak}Analysed in 2035. ^{al}Analysed in 2036. ^{am}Analysed in 2037. ^{an}Analysed in 2038. ^{ao}Analysed in 2039. ^{ap}Analysed in 2040. ^{aq}Analysed in 2041. ^{ar}Analysed in 2042. ^{as}Analysed in 2043. ^{at}Analysed in 2044. ^{au}Analysed in 2045. ^{av}Analysed in 2046. ^{aw}Analysed in 2047. ^{ax}Analysed in 2048. ^{ay}Analysed in 2049. ^{az}Analysed in 2050. ^{ba}Analysed in 2051. ^{bb}Analysed in 2052. ^{bc}Analysed in 2053. ^{bd}Analysed in 2054. ^{be}Analysed in 2055. ^{bf}Analysed in 2056. ^{bg}Analysed in 2057. ^{bh}Analysed in 2058. ^{bi}Analysed in 2059. ^{bj}Analysed in 2060. ^{bk}Analysed in 2061. ^{bl}Analysed in 2062. ^{bm}Analysed in 2063. ^{bn}Analysed in 2064. ^{bo}Analysed in 2065. ^{bp}Analysed in 2066. ^{bq}Analysed in 2067. ^{br}Analysed in 2068. ^{bs}Analysed in 2069. ^{bt}Analysed in 2070. ^{bu}Analysed in 2071. ^{bv}Analysed in 2072. ^{bw}Analysed in 2073. ^{bx}Analysed in 2074. ^{by}Analysed in 2075. ^{bz}Analysed in 2076. ^{ca}Analysed in 2077. ^{cb}Analysed in 2078. ^{cc}Analysed in 2079. ^{cd}Analysed in 2080. ^{ce}Analysed in 2081. ^{cf}Analysed in 2082. ^{cg}Analysed in 2083. ^{ch}Analysed in 2084. ^{ci}Analysed in 2085. ^{cj}Analysed in 2086. ^{ck}Analysed in 2087. ^{cl}Analysed in 2088. ^{cm}Analysed in 2089. ^{cn}Analysed in 2090. ^{co}Analysed in 2091. ^{cp}Analysed in 2092. ^{cq}Analysed in 2093. ^{cr}Analysed in 2094. ^{cs}Analysed in 2095. ^{ct}Analysed in 2096. ^{cu}Analysed in 2097. ^{cv}Analysed in 2098. ^{cw}Analysed in 2099. ^{cx}Analysed in 2100. ^{cy}Analysed in 2101. ^{cz}Analysed in 2102. ^{da}Analysed in 2103. ^{db}Analysed in 2104. ^{dc}Analysed in 2105. ^{dd}Analysed in 2106. ^{de}Analysed in 2107. ^{df}Analysed in 2108. ^{dg}Analysed in 2109. ^{dh}Analysed in 2110. ^{di}Analysed in 2111. ^{dj}Analysed in 2112. ^{dk}Analysed in 2113. ^{dl}Analysed in 2114. ^{dm}Analysed in 2115. ^{dn}Analysed in 2116. ^{do}Analysed in 2117. ^{dp}Analysed in 2118. ^{dq}Analysed in 2119. ^{dr}Analysed in 2120. ^{ds}Analysed in 2121. ^{dt}Analysed in 2122. ^{du}Analysed in 2123. ^{dv}Analysed in 2124. ^{dw}Analysed in 2125. ^{dx}Analysed in 2126. ^{dy}Analysed in 2127. ^{dz}Analysed in 2128. ^{ea}Analysed in 2129. ^{eb}Analysed in 2130. ^{ec}Analysed in 2131. ^{ed}Analysed in 2132. ^{ee}Analysed in 2133. ^{ef}Analysed in 2134. ^{eg}Analysed in 2135. ^{eh}Analysed in 2136. ^{ei}Analysed in 2137. ^{ej}Analysed in 2138. ^{ek}Analysed in 2139. ^{el}Analysed in 2140. ^{em}Analysed in 2141. ^{en}Analysed in 2142. ^{eo}Analysed in 2143. ^{ep}Analysed in 2144. ^{eq}Analysed in 2145. ^{er}Analysed in 2146. ^{es}Analysed in 2147. ^{et}Analysed in 2148. ^{eu}Analysed in 2149. ^{ev}Analysed in 2150. ^{ew}Analysed in 2151. ^{ex}Analysed in 2152. ^{ey}Analysed in 2153. ^{ez}Analysed in 2154. ^{fa}Analysed in 2155. ^{fb}Analysed in 2156. ^{fc}Analysed in 2157. ^{fd}Analysed in 2158. ^{fe}Analysed in 2159. ^{ff}Analysed in 2160. ^{fg}Analysed in 2161. ^{fh}Analysed in 2162. ^{fi}Analysed in 2163. ^{fj}Analysed in 2164. ^{fk}Analysed in 2165. ^{fl}Analysed in 2166. ^{fm}Analysed in 2167. ^{fn}Analysed in 2168. ^{fo}Analysed in 2169. ^{fp}Analysed in 2170. ^{fq}Analysed in 2171. ^{fr}Analysed in 2172. ^{fs}Analysed in 2173. ^{ft}Analysed in 2174. ^{fu}Analysed in 2175. ^{fv}Analysed in 2176. ^{fw}Analysed in 2177. ^{fx}Analysed in 2178. ^{fy}Analysed in 2179. ^{fz}Analysed in 2180. ^{ga}Analysed in 2181. ^{gb}Analysed in 2182. ^{gc}Analysed in 2183. ^{gd}Analysed in 2184. ^{ge}Analysed in 2185. ^{gf}Analysed in 2186. ^{gg}Analysed in 2187. ^{gh}Analysed in 2188. ^{gi}Analysed in 2189. ^{gj}Analysed in 2190. ^{gk}Analysed in 2191. ^{gl}Analysed in 2192. ^{gm}Analysed in 2193. ^{gn}Analysed in 2194. ^{go}Analysed in 2195. ^{gp}Analysed in 2196. ^{gq}Analysed in 2197. ^{gr}Analysed in 2198. ^{gs}Analysed in 2199. ^{gt}Analysed in 2200. ^{gu}Analysed in 2201. ^{gv}Analysed in 2202. ^{gw}Analysed in 2203. ^{gx}Analysed in 2204. ^{gy}Analysed in 2205. ^{gz}Analysed in 2206. ^{ha}Analysed in 2207. ^{hb}Analysed in 2208. ^{hc}Analysed in 2209. ^{hd}Analysed in 2210. ^{he}Analysed in 2211. ^{hf}Analysed in 2212. ^{hg}Analysed in 2213. ^{hh}Analysed in 2214. ^{hi}Analysed in 2215. ^{hj}Analysed in 2216. ^{hk}Analysed in 2217. ^{hl}Analysed in 2218. ^{hm}Analysed in 2219. ^{hn}Analysed in 2220. ^{ho}Analysed in 2221. ^{hp}Analysed in 2222. ^{hq}Analysed in 2223. ^{hr}Analysed in 2224. ^{hs}Analysed in 2225. ^{ht}Analysed in 2226. ^{hu}Analysed in 2227. ^{hv}Analysed in 2228. ^{hw}Analysed in 2229. ^{hx}Analysed in 2230. ^{hy}Analysed in 2231. ^{hz}Analysed in 2232. ^{ia}Analysed in 2233. ^{ib}Analysed in 2234. ^{ic}Analysed in 2235. ^{id}Analysed in 2236. ^{ie}Analysed in 2237. ^{if}Analysed in 2238. ^{ig}Analysed in 2239. ^{ih}

P. Hedges, LBDPH, Testing Laboratories Inc. (TL), Coliform Department of Water Resources (WR), as indicated

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (REL. 5)
OLD METRIC BASIC ANALYSIS (T.T. 401)

Date sampled P.S.T.	Discharge in cfs in 1970	Specific conductance ^a at 25°C	Mineral constituents in equivalents per million							Total solids in ppm	Per- cent solids in ppm	Total N.C. in ppm	Tur- bidity in ppm	Analyzed by ^b
			Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Other constituents	
1959	Tidal													
5/13	56	669	7.1	2.76c	3.09		0.00	3.20	1.48	1.07				
10/6	50	763	3.18e	4.09	3.00		0.00	4.25	1.79	1.04				
10/5	62	676	3.18e	3.75	3.00		0.00	1.93	1.07	1.04				
3/1/7	66	950	4.48e	3.02	4.14		0.00	2.52	1.24	1.04				
4/2	70	965	3.24	1.72	3.00		0.00	3.02	1.76	1.04				
5/13	68	1,070	5.30e	5.31	3.00		0.00	3.18	1.79	1.04				
6/9	75	1,100	5.76e	5.06	3.00		0.00	3.16	1.76	1.04				
7/2	80	788	3.76e	4.45	3.00		0.00	4.62	1.54	1.04				
8/10	79	1,180	6.7	3.62	3.00		0.00	3.49	1.76	1.04				
9/11	68	1,110	5.08e	5.61	3.00		0.00	3.52	1.76	1.04				
10/8	55	838	3.88e	4.00	3.00		0.00	3.52	1.76	1.04				
11/6	46	890	4.70	4.70	3.00		0.00	2.69	1.52	1.04				
12/11														
08/50														

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Actual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood

Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of

Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE 1-A
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (MC, S)
FAYES CREEK NEAR RVD BLUFF (STA. 064)

Date and time analyzed P.S.T.	Discharge in cfs	Temp in °F	Oxidized oxygen ppm	Specific conductance at 25°C µmhos/cm	Metal constituents in equivalents per million												Total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Temp in °C	Analytical method
					Cadmium (Cd)	Manganese (Mn)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)				
1/5/59																		US/20		
1/6 1/60	48	11.1	97	7.2 ^a	7.2	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			8 f	71	1
2/1 1/60	34	11.1	98	7.2 ^a	7.2	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.8 f	90	0
3/13 11/5	57	11.4	110	7.0 ^b	6.7	6.1	1.2	0.0	0.71	4.2	7.1	5.1	0.1	0.2	45			1.32 f	11	0
4/15 1/60	9.7	10.1	104	7.5 ^a	7.5	6.7	1.2	0.0	0.71	4.2	7.1	5.0	0.1	0.2	45			1.32 f	99	0
5/15 10/5	2.5	10.0	104	7.8 ^b	7.8	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.67 f	28	0
6/16 11/60	0.6	7.0	100	8.2 ^b	8.2	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.75 f	11	0
7/9 09/60	0.2	7.6	83	8.2 ^b	8.2	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.76 f	22	0
8/11 07/60	0.1	8.6	89	7.3 ^a	7.3	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.79 f	82	0
9/1 08/60	1.2	9.1	95	7.3 ^a	7.3	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.76 f	94	0
10/11 10/60	0.9	9.5	90	7.3 ^b	7.3	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.86 f	82	0
11/11 1/60	2.5	9.3	91	7.3 ^a	7.3	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.85 f	79	0
12/10 10/60	2.5	9.4	84	7.3 ^a	7.3	6.5	1.2	0.0	0.70	3.8	6.5	5.0	0.1	0.1	21			1.84 f	76	0

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺) reported here as 0.0 except as shown

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range

i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County, General District SEPCD, Metropolitan Water District of Southern California (MWD), Los Angeles Department of Public Health (LADPH), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), and California Departmental Paper Resources (CDPR) as indicated

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

FTT RIVER NEAR BISHOP (STA. 174)

PIT RIVER NEAR BERTHR (STA. 174)																				
Date and time sampled P.S.T.	Discharge Temp. in cts. in °F.	Dissolved oxygen ppm % Sol.	Specific Conductance at 25°C.	pH	Mineral constituents in parts per million												Total dissolved solids in ppm	Hardness as CaCO ₃ Total N.C. ppm	Turbidity in MPN/m	Analyzed by I
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)				
1959																				
1/	Not Sampled																			
2/	Not Sampled																			
3/5	53.0	46	155	7.4	13.0	5.0	1.4	2.6	0.0	80	17	3.5	0.2	0.2	0.0	27	129	35	0	
1310																				
4/	Not Sampled																			
5/7	30	56	295	7.7	22	9.5	28	4.6	0.0	164	13	8.0	0.7	0.2	0.3	26	193	38	0	
0900																				
6/4	11	69	336	7.8	23	11	37	6.6	0.0	180	27	0.83	1.3	0.2	0.2	32	235	42	109	
1450																				
7/16	10	76	359	8.3	28	9.7	41	6.3	16	178	13	9.0	0.8	0.2	0.0	30	241	43	110	
1425																				
8/12	0.2	71	339	8.3	16	4.9	55	7.6	26	129	29	1.4	0.3	0.2	0.2	25	299	63	0	
1310																				
9/9	0.2	77	383	8.2	14	5.6	69	8.8	53	90	15	23	0.0	0.1	0.3	10	243	68	48	
1445																				

a. Field pH.

b. Laboratory pH.

c. Sum of calcium and magnesium in ppm.

d. Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

0.00

e. Derived from conductivity vs. TDS curves.

f. Determined by addition of analyzed constituents.

g. Gravimetric determination.

h. Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Mineral analyses made by United States Geological Survey, County of Kern, Beech, (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (CDFD); Mendocino County (MCD); California State Water Resources Control Board (CSWRCB); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-14
ANALYSES OF SURFACE WATER

PIT MEVER NEAR CANY (ATA. 170)

[illegible][illegible]

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TABLE 3-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

STP RIVER NEAR MONTEREY PENNY (NO. 1)

Date and time sampled PST	Discharge Temp in deg F	Dissolved oxygen ppm % Sat	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent sod- ium in ppm	Hardness as CaCO ₃ Total N.C. ppm	Tot- al alkalinity in ppm	Analyzed by
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)					
1050																		TDS	
1/	Not Sampled - Inacceptable																		
2/1	Not Sampled - Inacceptable																		
3/5	4,950	48	11.4	98	7.5														
1045						8.6 0.37													
4/8	4,740	56	10.3	98	7.9														
1355						7.0 0.31													
5/5	2,870	55	10.0	94	7.5														
6/9						8.2 0.36													
6/9	2,680	57	10.1	97	7.1														
6/9						7.4 0.30													
7/16	760	66	8.8	94	7.7														
8810						13 0.62													
8/12	3,780	66	9.3	99	8.1														
0930						12 0.55													
9/9	10,460	65	9.0	95	8.0														
1110						12 0.55													
10/	Not Sampled																		
11/11	4,080	49	10.9	95	7.5														
1555						12 0.58													
12/9	4,030	43	11.9	96	7.5														
1125						12 0.58													

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Long Beach, Department of Public Health (LBPH); National Testing Laboratories, Inc. (NTLI); or California Department of Water Resources (CDWR), as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

PIT RIVER, ROUTE 90 NEAR LIPLEY (Sta. 104)

Date and time sampled P.S.T.	Discharge temp. in cfs in day	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Major constituents in equivalents per million										Total dissolved solids in ppm	Per- cent total dissolved solids as CaCO ₃ in ppm	Hardness as CaCO ₃ in ppm	Total Hardness as CaCO ₃ in ppm	Adapted by method
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)					
1969	Not Sampled																	1995	
1/ 1/30	35	34	12.2	87	7.6	11.55	4.5	5.2	1.9	0.0	1.13	1.0	1.8	0.2	0.1	0.0	15	Median	
1/5	12	51	9.4	84	8.3	8.4	5.4	6.4	2.4	0.0	1.09	0.0	2.1	0.6	0.1	0.1	16	Maximum	
1/15	91	41	11.0	86	107	7.3	9.5	4.4	2.4	0.0	1.03	1.3	1.3	0.3	0.0	0.0	31	Minimum	
1/21	66	57	8.9	80	114	8.1	17.75	6.4	2.4	0.0	1.16	1.2	1.0	0.8	0.1	0.1	31	Minimum	
1/23	104	62	8.5	86	131	7.9	17.05	8.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	
1/29	37	78	7.3	88	138	8.1	13.75	1.9	4.1	0.0	1.28	1.0	1.0	1.5	0.2	0.0	38	Minimum	
1/31	162	85	7.4	89	162	7.7	15.75	5.5	4.7	0.0	1.08	1.0	1.0	0.8	0.01	0.0	38	Minimum	
2/5	3	67	7.8	78	172	8.3	15.75	6.1	4.4	0.0	1.03	1.3	1.3	0.3	0.0	0.0	31	Minimum	
2/10	98	10	9.9	107	7.9	17.75	6.4	6.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	
2/11	74	1	10.4	107	7.5	17.75	6.4	6.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	
2/17	18	18	10.4	107	7.4	17.75	6.4	6.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	
2/18	18	18	10.4	107	7.4	17.75	6.4	6.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	
2/19	18	18	10.4	107	7.4	17.75	6.4	6.4	2.4	0.0	1.08	1.0	1.0	0.8	0.01	0.0	31	Minimum	

Notes:

1. Sampled at 10:00 a.m.
2. Sampled at 10:00 a.m.
3. Sampled at 10:00 a.m.
4. Sampled at 10:00 a.m.
5. Sampled at 10:00 a.m.
6. Sampled at 10:00 a.m.
7. Sampled at 10:00 a.m.
8. Sampled at 10:00 a.m.
9. Sampled at 10:00 a.m.
10. Sampled at 10:00 a.m.
11. Sampled at 10:00 a.m.
12. Sampled at 10:00 a.m.
13. Sampled at 10:00 a.m.
14. Sampled at 10:00 a.m.
15. Sampled at 10:00 a.m.

TABLE B-4

ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

CALIFORNIA REGIONAL WATER QUALITY CONTROL BOARD

Date and time sampled P.S.T.	Dissolved oxygen in air	Specific conductance at 25°C	Major constituents in parts per million							Total solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ Total N.C. ppm	Tur- bidity in ppm	Analyzed by
			Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trogen (NO ₃)	Fluo- ride (F)	Silic- ate (SiO ₂)	
1/5/59														
1/6/59	55	10.0	92	263	7.3	1.7	0.0	0.0	0.0	1.0	0.0	0.0	0.0	158°
1/7/59	56	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/8/59	57	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/9/59	58	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/10/59	59	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/11/59	60	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/12/59	61	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/13/59	62	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/14/59	63	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/15/59	64	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/16/59	65	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/17/59	66	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/18/59	67	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/19/59	68	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/20/59	69	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/21/59	70	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/22/59	71	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/23/59	72	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/24/59	73	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/25/59	74	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/26/59	75	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/27/59	76	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/28/59	77	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/29/59	78	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/30/59	79	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/31/59	80	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/32/59	81	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/33/59	82	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/34/59	83	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/35/59	84	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/36/59	85	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/37/59	86	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/38/59	87	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/39/59	88	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/40/59	89	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/41/59	90	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/42/59	91	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/43/59	92	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/44/59	93	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/45/59	94	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/46/59	95	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/47/59	96	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/48/59	97	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/49/59	98	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/50/59	99	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
1/51/59	100	1.45	100	27	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively.

i Mineral analysis made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

ROCK SLUDGE NEAR NORTREYSEN (STA. 10-1)

Date and time of day and P.S.T.	Onshore Temp in cts	Dissolved oxygen in %	Specific conductance (microhm/cm at 25°C)	pH ^a	Mineral constituents in parts per million										Total dissolved solids in ppm	Per cent solids	Hardness as CaCO ₃ Total in ppm	Tur- bid- ity in fpm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)			
1959																			USGS
1/13 1959	55	8.2	77	7.1	3.74 ^c		51	25.91	0.0	86		96					1.87	116	20
2/10 1959	49	9.2	80	7.2	3.99 ^c		89	31.87	0.0	127		130					1.96	92	3
3/11 1959	61	8.3	84	7.2	3.44 ^c		80	31.48	0.0	130		106					1.72	74	40
4/3 1959	64	6.7	70	7.1	3.28 ^c		36	1.37	0.0	68		54					1.14	42	30
5/12 1959	70	6.3	70	7.2	3.5	7.0	39	2.6	0.0	74	23	21	0.6	0.2	0.1	16			
6/10 1959	70	6.0	67	7.2	3.58 ^c	0.45	22	0.96	0.0	81	0.94	25	0.6	0.2	0.1				
7/2 1959	78	7.2	87	7.3	3.23 ^c		72	3.13	0.0	93	1.52	108							
8/10 1959	79	7.0	85	7.2	3.28 ^c		171	1.44	0.0	86		293							
9/8 1959	80	6.8	84	7.2	3.28	1.60	101	2.8	0.0	108	35	168	2.2	0.2	0.1	19			
10/6 1959	68	7.5	82	7.3	3.99 ^c		20	0.15	0.0	103	0.73	74	0.6	0.2	0.1				
11/3 1959	58	8.5	83	7.3	3.99 ^c		33	3.18	0.0	106	2.09	47							
12/3 1959	46	9.0	75	7.2	3.58 ^c		40	1.44	0.0	119	1.55	74							
6/85																			

^a Field pH^b Laboratory pH^c Sum of calcium and magnesium in ppm^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.^e Derived from conductivity vs TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.ⁱ Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 4)

SACRAMENTO RIVER AT BEED (STA. 124)

Date and time of sample and P.S.T.	Discharge Temp. in cfs	Dissolved oxygen in cfs	Specific conductance (micro-mhos at 25°C)	Mineral constituents in parts per million												Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total N.C. in ppm	Coliforms per 100 ml	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Barium (Ba)	Silica (SiO ₂)					
Average Daily Mean																			THM90	
1/1-11	11,600	7.4	139	12 0.65	5.8 0.48	8.4 0.37	1.6 0.04	0.0	69	173	4.0 0.11	0.3 0.00	0.2 0.01	0.0	34	Fe 0.04 Cu 0.01 Zn 0.11	110	25	44	
1/12-25	16,000	7.2	136	12 0.60	5.8 0.48	8.6 0.37	1.8 0.06	0.0	67	170	4.0 0.11	0.2 0.00	0.2 0.01	0.0	33	Fe 0.03 Cu 0.01 Zn 0.17	111	26	44	
1/26-31	24,700	6.8	120	10 0.50	5.8 0.48	7.1 0.31	1.6 0.04	0.0	49	111	4.0 0.11	0.3 0.00	0.1 0.01	0.1	30	Fe 0.03 Cu 0.01 Zn 0.15	99	23	44	
2/1-15	11,300	7.0	121	11 0.55	4.7 0.39	6.9 0.30	1.4 0.04	0.0	48	133	3.0 0.09	0.0 0.00	0.1 0.01	0.0	24	Fe 0.06 Cu 0.01 Zn 0.10	93	23	47	
2/16	65,000	73.2	6.2	0.65		3.2	1.3	0.06	28	10	2.0	1.7	0.06	16		Zn 0.19	48	18	30	
2/17-28	28,500	6.9	118	10 0.50	5.1 0.48	6.7 0.29	1.3 0.03	0.0	44	14	2.0	0.0	0.0	0.1	23	Fe 0.08 Cu 0.01 Zn 0.12	93	22	46	
3/1-15	8,440	6.7	136	13 0.65	6.7 0.47	7.2 0.31	1.3 0.03	0.0	66	14	3.0	0.0	0.0	0.2	27	Fe 0.03 Cu 0.01 Zn 0.11	104	21	56	
3/16-31	6,790	6.9	134	12 0.50	5.8 0.48	7.8 0.34	1.3 0.03	0.0	64	16	3.0	0.0	0.0	0.1	25	Fe 0.03 Cu 0.01 Zn 0.23	103	23	54	
4/1-16	5,300	7.8	125	12 0.50	4.6 0.38	6.8 0.30	1.2 0.03	0.0	63	5.2	3.8	0.9	0.0	0.0	25	Fe 0.01 Cu 0.00 Zn 0.11	91	23	49	
4/17-30	8,980	7.8	118	10 0.50	4.6 0.38	6.8 0.30	1.4 0.04	0.0	61	5.2	3.8	0.7	0.0	0.0	27	Zn 0.10 Fe 0.00 Cu 0.00	89	25	44	
5/1-16	8,390	7.7	116	10 0.50	4.9 0.40	6.6 0.29	1.4 0.04	0.0	69	4.4	2.8	0.8	0.0	0.0	26	Zn 0.07 Fe 0.00 Cu 0.00	84	24	45	
5/18-31	8,160	7.8	117	11 0.55	4.0 0.33	7.1 0.31	1.9 0.06	0.0	69	4.0	3.3	0.7	0.0	0.0	26	Fe 0.07 Fe 0.00 Cu 0.00	89	25	44	
6/1-12	8,500	7.6	122	11 0.55	4.3 0.35	7.4 0.32	1.8 0.06	0.0	64	3.8	2.9	0.7	0.0	0.0	26	Fe 0.01 Zn 0.07 Cu 0.00	91	25	45	
6/13-30	9,650	7.7	119	11 0.55	4.3 0.35	7.3 0.32	1.7 0.04	0.0	66	4.2	2.5	0.9	0.0	0.0	26	Fe 0.01 Zn 0.07 Cu 0.00	91	25	45	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWD), City of Los Angeles, Department of Water and Power (LADWP), City of Long Beach, Department of Public Health (LBPH), Lemoore Tasting Laboratories, Inc. (TTL) for California Department of Water Resources (DWR), as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (B-1)

SACRAMENTO RIVER AT BEND (774, 10, continued)

Date and time analyzed P.S.T.	Discharge in cfs	Temp in deg F	Dissolved oxygen in ppm	Specific conductance in micromhos at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total Calcium in ppm	Applied by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)				
10/9/54																		
7/13/57	11,000	7.7	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/14/57	11,400	7.7	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/15/57	11,400	7.2	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/16/57	11,400	7.4	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/17/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/18/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/19/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/20/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/21/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/22/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/23/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/24/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/25/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/26/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/27/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/28/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/29/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/30/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
7/31/57	11,400	7.1	11.0	11	7.2	0.12	1.4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01

1. Field pH

2. Laboratory pH

3. Sample temperature in deg

4. Dissolved oxygen in ppm (A) (B) (C) (D) (E) (F) (G) (H) (I) (J) (K) (L) (M) (N) (O) (P) (Q) (R) (S) (T) (U) (V) (W) (X) (Y) (Z) (AA) (AB) (AC) (AD) (AE) (AF) (AG) (AH) (AI) (AJ) (AK) (AL) (AM) (AN) (AO) (AP) (AQ) (AR) (AS) (AT) (AU) (AV) (AW) (AX) (AY) (AZ) (BA) (BB) (BC) (BD) (BE) (BF) (BG) (BH) (BI) (BJ) (BK) (BL) (BM) (BN) (BO) (BP) (BQ) (BR) (BS) (BT) (BU) (BV) (BW) (BX) (BY) (BZ) (CA) (CB) (CC) (CD) (CE) (CF) (CG) (CH) (CI) (CJ) (CK) (CL) (CM) (CN) (CO) (CP) (CQ) (CR) (CS) (CT) (CU) (CV) (CW) (CX) (CY) (CZ) (DA) (DB) (DC) (DD) (DE) (DF) (DG) (DH) (DI) (DJ) (DK) (DL) (DM) (DN) (DO) (DP) (DQ) (DR) (DS) (DT) (DU) (DV) (DW) (DX) (DY) (DZ) (EA) (EB) (EC) (ED) (EE) (EF) (EG) (EH) (EI) (EJ) (EK) (EL) (EM) (EN) (EO) (EP) (EQ) (ER) (ES) (ET) (EU) (EV) (EW) (EX) (EY) (EZ) (FA) (FB) (FC) (FD) (FE) (FF) (FG) (FH) (FI) (FJ) (FK) (FL) (FM) (FN) (FO) (FP) (FQ) (FR) (FS) (FT) (FU) (FV) (FW) (FX) (FY) (FZ) (GA) (GB) (GC) (GD) (GE) (GF) (GG) (GH) (GI) (GJ) (GK) (GL) (GM) (GN) (GO) (GP) (GQ) (GR) (GS) (GT) (GU) (GV) (GW) (GX) (GY) (GZ) (HA) (HB) (HC) (HD) (HE) (HF) (HG) (HH) (HI) (HJ) (HK) (HL) (HM) (HN) (HO) (HP) (HQ) (HR) (HS) (HT) (HU) (HV) (HW) (HX) (HY) (HZ) (IA) (IB) (IC) (ID) (IE) (IF) (IG) (IH) (II) (IJ) (IK) (IL) (IM) (IN) (IO) (IP) (IQ) (IR) (IS) (IT) (IU) (IV) (IW) (IX) (IY) (IZ) (JA) (JB) (JC) (JD) (JE) (JF) (JG) (JH) (JI) (JJ) (JK) (JL) (JM) (JN) (JO) (JP) (JQ) (JR) (JS) (JT) (JU) (JV) (JW) (JX) (JY) (JZ) (KA) (KB) (KC) (KD) (KE) (KF) (KG) (KH) (KI) (KJ) (KK) (KL) (KM) (KN) (KO) (KP) (KQ) (KR) (KS) (KT) (KU) (KV) (KW) (KX) (KY) (KZ) (LA) (LB) (LC) (LD) (LE) (LF) (LG) (LH) (LI) (LJ) (LK) (LM) (LN) (LO) (LP) (LQ) (LR) (LS) (LT) (LU) (LV) (LW) (LX) (LY) (LZ) (MA) (MB) (MC) (MD) (ME) (MF) (MG) (MH) (MI) (MJ) (MK) (ML) (MM) (MN) (MO) (MP) (MQ) (MR) (MS) (MT) (MU) (MV) (MW) (MX) (MY) (MZ) (NA) (NB) (NC) (ND) (NE) (NF) (NG) (NH) (NI) (NJ) (NK) (NL) (NM) (NN) (NO) (NP) (NQ) (NR) (NS) (NT) (NU) (NV) (NW) (NX) (NY) (NZ) (OA) (OB) (OC) (OD) (OE) (OF) (OG) (OH) (OI) (OJ) (OK) (OL) (OM) (ON) (OO) (OP) (OQ) (OR) (OS) (OT) (OU) (OV) (OW) (OX) (OY) (OZ) (PA) (PB) (PC) (PD) (PE) (PF) (PG) (PH) (PI) (PJ) (PK) (PL) (PM) (PN) (PO) (PP) (PQ) (PR) (PS) (PT) (PU) (PV) (PW) (PX) (PY) (PZ) (QA) (QB) (QC) (QD) (QE) (QF) (QG) (QH) (QI) (QJ) (QK) (QL) (QM) (QN) (QO) (QP) (QQ) (QR) (QS) (QT) (QU) (QV) (QW) (QX) (QY) (QZ) (RA) (RB) (RC) (RD) (RE) (RF) (RG) (RH) (RI) (RJ) (RK) (RL) (RM) (RN) (RO) (RP) (RQ) (RR) (RS) (RT) (RU) (RV) (RW) (RX) (RY) (RZ) (SA) (SB) (SC) (SD) (SE) (SF) (SG) (SH) (SI) (SJ) (SK) (SL) (SM) (SN) (SO) (SP) (SQ) (SR) (SS) (ST) (SU) (SV) (SW) (SX) (SY) (SZ) (TA) (TB) (TC) (TD) (TE) (TF) (TG) (TH) (TI) (TJ) (TK) (TL) (TM) (TN) (TO) (TP) (TQ) (TR) (TS) (TT) (TU) (TV) (TW) (TX) (TY) (TZ) (UA) (UB) (UC) (UD) (UE) (UF) (UG) (UH) (UI) (UJ) (UK) (UL) (UM) (UN) (UO) (UP) (UQ) (UR) (US) (UT) (UU) (UV) (UW) (UX) (UY) (UZ) (VA) (VB) (VC) (VD) (VE) (VF) (VG) (VH) (VI) (VJ) (VK) (VL) (VM) (VN) (VO) (VP) (VQ) (VR) (VS) (VT) (VU) (VV) (VW) (VX) (VY) (VZ) (WA) (WB) (WC) (WD) (WE) (WF) (WG) (WH) (WI) (WJ) (WK) (WL) (WM) (WN) (WO) (WP) (WQ) (WR) (WS) (WT) (WU) (WV) (WW) (WX) (WY) (WZ) (XA) (XB) (XC) (XD) (XE) (XF) (XG) (XH) (XI) (XJ) (XK) (XL) (XM) (XN) (XO) (XP) (XQ) (XR) (XS) (XT) (XU) (XV) (XW) (XX) (XY) (XZ) (YA) (YB) (YC) (YD) (YE) (YF) (YG) (YH) (YI) (YJ) (YK) (YL) (YM) (YN) (YO) (YP) (YQ) (YR) (YS) (YT) (YU) (YV) (YW) (YX) (YZ) (ZA) (ZB) (ZC) (ZD) (ZE) (ZF) (ZG) (ZH) (ZI) (ZJ) (ZK) (ZL) (ZM) (ZN) (ZO) (ZP) (ZQ) (ZR) (ZS) (ZT) (ZU) (ZV) (ZW) (ZX) (ZY) (ZZ)

5. Dissolved oxygen in ppm

6. Dissolved oxygen in ppm

7. Dissolved oxygen in ppm

8. Dissolved oxygen in ppm

9. Dissolved oxygen in ppm

10. Dissolved oxygen in ppm

11. Dissolved oxygen in ppm

12. Dissolved oxygen in ppm

13. Dissolved oxygen in ppm

14. Dissolved oxygen in ppm

15. Dissolved oxygen in ppm

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

SACRAMENTO RIVER AT EMERY CITY (STA. 87A)

Date of analysis PST	C charge in cts in g	Temp. in °C	Dissolved oxygen in ppm	Specific conductance at 25°C	Mineral constituents in parts per million										Total dissolved solids in ppm	Per- cent sulfate in sum	Hardness as CaCO ₃ total in ppm	Tur- bidity in ptm	Caliform ^a MPN/ml	Analyzed by
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)				
1959																				
8/11	10,200	65	9.3	117	7.4 ^a		6.2 0.29		0.0 0.00	64 1.05		2.8 0.08			0.0		85 ^c	46	0	Median 23
9/8	6,220	68	8.6	94	127	7.7 ^b	7.4 0.38	2.2 0.06	0.0 0.00	70 1.15	11 0.23	3.5 0.10	0.3 0.00	0.1 0.01			104 ^d	56	0	Maximum 62
9/95	5,010	63		140	7.4 ^a		8.6 0.37		0.0 0.00	74 1.21		5.5 0.16			0.1		100 ^e	60	0	Minimum 2-3
10/13	4,890	55	10.2	96	133	7.5 ^b	9.6 0.42		0.0 0.00	86 1.41		5.5 0.16			0.0		109 ^f	60	0	8
12/9	4,080	53	10.7	98	157	7.4 ^b	9.7 0.42		0.0 0.00	84 1.39		4.5 0.13			0.0		111 ^g	64	0	15
1340																				USGS

^a Field pH

^b Laboratory pH

^c Sum of calcium and magnesium in ppm

^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb),

manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0, except as shown.

^e Derived from conductivity vs TDS curves.

^f Determined by addition of analyzed constituents.

^g Gravimetric determination

^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 8-4

ANALYSES OF SURFACE WATER

THE UNIVERSITY OF CHICAGO

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Journal of Management Studies

A... ..

More information on the
Dow Chemical Company

[illegible]

TABLE B-4
ANALYSES OF SURFACE WATER

DATE: 10/15/2010 10:21 AM

PROJECT: 62706 AT SITE C-75 (ETA, PTA, and TPA)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	pH	Major constituents in parts per million										Total dis- solved solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ in ppm	Tot- al N.C. in ppm	Total Calcium in ppm	Total Magnesium in ppm
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)						
7/1/11	7,357	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	7,715	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	7,266	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	8,240	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/2/11	9,329	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/2/11	8,084	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/2/11	6,402	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	5,692	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	9,189	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/2/11	6,183	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	4,824	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	4,679	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	4,890	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	
7/1/11	4,046	7.1	13.2	7.1	1.1	5.2	7.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	

a. Field pH.

b. Laboratory pH.

c. Sum of calcium and magnesium in ppm.

d. Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown 0.00

e. Derived from conductivity vs TDS curves

f. Determined by addition of analyzed constituents.

g. Gravimetric determination

h. Annual median and range, respectively. Calculated from all years of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Manual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBOPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

CENTRAL VALLEY REGION (NO. 11)

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$$S_{\text{max}} = 1 - \frac{1}{n} \sum_{i=1}^n \log \frac{1}{p_i}$$

1100 Fe 1500000 At 1500000 At

the Derivative of \ln vs \ln curves

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Figure 1. The effect of the concentration of the polymer on the rate of polymerization.

(continued)

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the fact that the system is not yet fully operational.

Р. М. ИВАНОВ. Тестирование

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (RD. 1)

SACRAMENTO RIVER AT DELTA (STA. 11)

Date and time sampled P.S.T.	Discharge Temp in cts in deg F	Dissolved oxygen ppm	Specific conductance (microhm at 25°C)	Mineral constituents in equivalents per million										Total solid in ppm	Per- cent solid in ppm	Hardness as CaCO ₃ Total N.C. ppm	Turb- idity MPN/ml	Analyzed by 1
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- sum (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Flu- oride (F)					
11/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	1072
2/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
3/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
4/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
5/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
6/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
7/1	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
8/10	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
9/10	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
10/10	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
11/9	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
12/7	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		
1500	20.6	11.2	165	7.3	3.5	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25		

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown, 0.00

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (CDWR), as indicated.

TABLE 10-16
ANALYSES OF SURFACE WATER

[illegible]

H. J. G.

E

M. J. Griffin et al.

TABLE B-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (SG, C)

SACRAMENTO RIVER AT KESTICK (296, 174)

Date and time of sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in % Sat	Specific Conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent solid in ppm	Hardness as CaCO ₃ in ppm	Total N.C. in ppm	Total hardness as CaCO ₃ in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Silica (SiO ₂)					
10/9																				
1/6/90	4,240	56	83	134	10	6.1	7.7	1.4	0.0	59	9.6	3.0	0.5	0.3	21	102	24	1	1	1100
2/11/90	11,300	50	84.5	109	7.1	0.33	6.1	0.31	0.0	54	1.13	0.08	0.01	0.0	0.0	74	24	44	0	3
3/2/90	7,430	50	84.8	128	7.0	0.33	6.8	0.31	0.0	58	0.95	3.2	0.0	0.0	0.0	24	44	0	80	11.4 mm
4/6/90	3,940	52	84.8	115	7.1	0.33	5.0	0.31	0.0	52	0.85	2.8	0.0	0.0	0.0	43	11	43	1	3
5/5/90	7,530	50	84.8	107	7.7	0.33	5.4	0.31	1.1	55	4.8	2.5	0.6	0.1	24	102	24	44	0	3
6/2/90	7,540	50	84.8	109	7.7	0.33	5.6	0.31	0.0	58	0.95	3.2	0.0	0.0	0.0	74	24	44	0	3
7/7/90	10,600	50	84.8	112	7.1	0.33	6.2	0.31	0.0	60	0.95	3.2	0.0	0.0	0.0	74	24	44	0	3
8/4/90	13,800	52	84.8	112	7.3	0.33	6.2	0.31	0.0	60	0.95	3.2	0.0	0.0	0.0	74	24	44	0	3
9/8/90	7,720	56	84.8	113	7.3	0.33	6.0	0.31	0.0	63	1.03	3.0	1.0	0.1	24	102	24	44	0	3
10/13/90	5,190	57	84.8	114	7.1	0.33	3.9	0.31	0.0	69	1.02	3.0	0.0	0.0	0.0	82	11	42	0	3
11/10/90	4,350	59	84.8	130	7.1	0.33	8.5	0.31	0.0	73	1.20	4.0	0.0	0.0	0.0	96	26	52	0	3
12/6/90	3,690	54	84.8	134	7.2	0.33	8.4	0.31	0.0	76	1.24	3.0	0.0	0.0	0.0	102	26	52	0	3

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), organic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (BSFCD), Metropolitan Water District of Southern California (MWD), Los Angeles County Flood Control District (LACFCD), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

SACRAMENTO RIVER AT KNIGHTS LANDING (STA. 14)

Date sample supplied P.S.T.	Discharge Temp in cfe in °F	Dissolved oxygen ppm	Specific conductance in 25°C pH	b	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- cent sulfate in ppm	Hardness on CaCO ₃ in ppm	Turbid- ity - in ppm	Analyzed by I					
					Other constituents																			
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)	Silica (SiO ₂)			
1/5-9			162	7.0	12.0	6.8	11.0	1.7	0.0	82	1.9	2.5	2.1	0.1	0.0	22	Fe 0.05	98	0	106	29	58	0	0255
1/5-10					12.0	6.8	11.0	1.7	0.0	82	1.9	2.5	2.1	0.1	0.0	22	Fe 0.05	98	0	106	29	58	0	
1/21-16			111	7.3	8.4	5.0	5.0	3.1	0.0	144	11	4.4	2.6	0.0	0.1	20	Fe 0.12	144	8	81	19	144	8	
1/21-31			131	7.4	11.0	6.0	6.3	1.0	0.0	61	5.8	5.0	1.0	0.0	0.1	26	Fe 0.10	50	0	93	20	52	0	
2/1-20			134	7.3	11.0	6.0	6.4	0.9	0.0	62	7.7	5.5	0.9	0.0	0.0	25	Fe 0.06	50	1	94	21	52	1	
2/25-3/3			162	7.6	11.0	7.5	7.2	1.0	0.0	78	8.6	5.8	0.8	0.0	0.0	24	Fe 0.03	66	2	107	19	66	2	
3/4-11			175	8.0	16.0	7.8	8.6	1.0	0.0	84	12	7.8	0.6	0.0	0.1	24	Fe 0.01	72	3	119	20	72	3	
3/12-23			203	7.7	16.0	9.3	11.0	1.0	0.0	92	17	9.5	0.9	0.0	0.1	26	Fe 0.01	77	2	136	23	77	2	
3/24-31			178	7.6	16.0	8.3	8.5	1.1	0.0	90	9.6	7.0	0.9	0.0	0.1	26	Fe 0.01	74	0	122	20	74	0	
4/1-5			194	7.3	16.0	8.3	11.0	0.7	0.0	94	12	7.8	0.8	0.0	0.0	30	Fe 0.02	74	0	133	24	74	0	
4/6	Not Sampled																							
5/4	Not Sampled																							
7/1	Not Sampled																							
7/1-18			214	8.0	14.0	8.8	17.0	0.2	0.0	110	15	9.0	0.2	0.1	0.1	32	Fe 0.01	71	0	146	34	71	0	
7/19-8/1			193	7.6	14.0	7.5	15.0	0.0	0.0	11	13	7.2	0.1	0.1	0.1	28	Fe 0.01	66	0	131	33	66	0	

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively.

i Manual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Public Health (LADPH); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Tammam Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 3-4

THE NATIONAL ARCHIVES

AGREEMENT TO VISIT AT KNIGHT LANDING "A. M. CONTINUED"

[illegible]

Index

b. Laboratory of M

Sum of calcium and magnesium is equal to 1.00

	Copper	Manganese	Zinc	Chromium	
(Cu)	(Mn)	(Zn)	(Cr ⁺⁶)	reported here as	except as shown
0.0	0.0	0.0	0.0	0.0	0.0

and iron Fe) aluminum Al, organic Al), cobalt Co, copper Cu, gallium Ga, germanium Ge, gold Au, hydrogen H, iodine I, iron Fe, lead Pb, lithium Li, magnesium Mg, manganese Mn, mercury Hg, nickel Ni, nitrogen N, oxygen O, phosphorus P, potassium K, rubidium Rb, selenium Se, silver Ag, sodium Na, strontium Sr, tellurium Te, tin Sn, titanium Ti, vanadium V, zinc Zn, zirconium Zr, and yttrium Y).

e Derived from conductivity vs. IUS curves

† Determined by addition of analyzed constituents

g Gravimetric determination

^b Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories or United States Public Health Service.

^a Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of Interior, Bureau of Reclamation (BOR), United States Public Health Service (USPHS). San Bernardino County, Field Control District 38BFCU. Metropolitan water district of Southern California (MWD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health LADPH, City of Long Beach, Department of Public Health LBPH. Tarrant, Tarrant Irrigation Districts, Inc. TITL or California Department of Water Resources (CDR) as indicated.

TABLE B-14
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 2)

CIPMENTO RIVER NE'P MALLARD CLOUGH (77, 132)

Date sample was collected P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in parts per million												Total dis- solved solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Turbid- ity MPN/ml	Analyzed By
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- ium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)					
1960	Tidal																				
1/14 12:45	56		364			32							50			248	61				
2/11 13:00	51		258			22							36			190	37				
5/19 11:15	60		395			29							41			141	52				
6/14 12:00	69		333			36							50			198	60				
5/15 12:45			2,034			253							176			1,068	54				
7/14 11:00	72		15,294			2,530							4,288			10,804	72				
8/13 13:00			9,076			1,465							2,932			6,070	72				
9/16 13:05	69		4,254			649							1,271			2,694	86				
10/25 11:45	69		2,454			536							203			2,008	65				
11/13 09:05	58		4,391			644							1,218			2,560	64				
12/16 13:15	46		9,941			1,384							2,549			5,368	58				

Field pH

Laboratory of H

Sum of calcium and magnesium in ppm.

Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{+6}), reported here as $\frac{0.0}{0.0}$ except as shown.

Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

Determined by addition of
C₁₂H₂₂O₁₁ to the solution.

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District, San Bernardino County, California.

City of Long Beach, Department of Public Health (LADPH); City of Los Angeles, Department of Public Health (LADWP); City of Los Angeles, Department of Water and Power (LADWP); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water Resources (DWR) are indicated

TABLE 3-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NW, S)

SACRAMENTO RIVER AT SACRAMENTO (STA. 15 continued)

Date and time of day and day of year	Discharge Temp in °F in air	Dissolved oxygen in ppm	Specific Conductance at 25°C in %S _{cl}	Mineral constituents in equivalents per million										f Total dissolved solids in ppm	g Percent solids in ppm	Hardness as CaCO ₃ in ppm (100)	Total H.C. in ppm (100)	Type - big - in ppm	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)				
1959																			USGS
1/13-24		14.7	7.1	11.0	1.5	8.4	1.5	0.0	66	11.0	6.5	0.5	0.0	0.1	21	Fe 0.01	54	0	
1/25-30		16.1	7.8	12.0	6.8	11.0	1.6	0.0	70	11.0	9.0	0.3	0.0	0.1	21	Fe 0.02	58	1	
5/2-8		19.0	7.5	14.0	7.5	14.0	1.6	0.0	80	16.0	10.0	0.5	0.0	0.1	22	Fe 0.01	66	0	
5/11-20		20.5	6.9	16.0	7.8	16.0	2.1	0.0	92	13.0	13.0	0.3	0.1	0.1	24	Fe 0.00	72	0	
5/21-31		21.3	7.0	16.0	8.3	16.0	1.6	0.0	96	15.0	15.0	0.2	0.2	0.1	24	Fe 0.00	74	0	
6/3-10		22.4	7.0	17.0	8.4	17.0	1.6	0.0	98	18.0	18.0	0.0	0.1	0.1	24	Fe 0.00	77	0	
6/11-19		21.5	7.3	15.0	8.6	16.0	1.6	0.0	92	11.0	11.0	0.0	0.1	0.1	25	Fe 0.00	73	0	
6/20-30		19.8	7.0	14.0	7.1	15.0	1.5	0.0	89	13.0	12.0	0.3	0.2	0.1	24	Fe 0.00	65	0	
7/1-10		17.9	7.4	12.0	7.5	14.0	1.6	0.0	78	10.0	11.0	0.2	0.0	0.0	24	Fe 0.00	61	3	
7/11-20		18.1	7.4	13.0	6.9	14.0	1.4	0.0	78	9.8	11.0	0.2	0.0	0.0	24	Fe 0.00	61	0	
7/21-31		16.0	7.2	11.0	7.2	12.0	1.4	0.0	72	8.0	13.0	0.3	0.2	0.0	22	Fe 0.02	57	0	
8/1-2		14.4	7.1	12.0	5.4	12.0	1.3	0.0	68	9.0	8.0	0.2	0.1	0.0	21	Fe 0.00	52	0	
8/3-9		17.1	7.1	14.0	6.6	13.0	1.5	0.0	76	16.0	8.5	0.8	0.1	0.0	25	Fe 0.00	62	0	
8/10		15.9	7.4	14.0	6.6	15.0	1.3	0.0	80	11.0	11.0	0.2	0.1	0.0	21	Fe 0.00	62	0	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively.

i Mineral analyses by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

United States Public Health Service.
 San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE 2-2

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (RP, 5)

SACRAMENTO RIVER AT SACRAMENTO STA. 15 continued

Date and time of day and P.S.T.	Dissolved oxygen in %	Temperature in °C	Specific conductance at 25°C in %S/cm	Major constituents in mg/l or µM							Total dissolved solids in g/l		Percent total dissolved solids in g/l	Hardness as CaCO ₃ in g/l	Total dissolved solids in mg/l	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents	
1959																
8/11-17	176	7.9		11.7	1.1	11.7	1.1	0.0	8.2	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
8/11-17	217	8.1		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
9/1-10	240	7.1	18	10	1.0	10	1.0	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
9/1-10	248	7.4	18	10	1.0	10	1.0	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
9/11-18	211	7.2		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
10/1-10	192	6.9		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
11/1-10	179	7.1		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
11/21-30	162	7.1		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
11/21-30	278	7.9		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
11/21-30	181	7.9	16	11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
11/21-30	180	7.8	16	11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
12/1-9	203	6.7		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
12/11-20	190	7.1		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0
12/11-31	198	6.8		11.7	1.1	11.7	1.1	0.0	11.7	11.7	1.1	0.1	0.1	0.1	Fe 1.0	1.0

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in g/l

d Iron (Fe), aluminum (Al), uranium (U), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn) and hexavalent chromium (Cr⁶⁺) reported here as 0.0 except as shown

e Derived from conductivity x 1.05 factor

f Determined by addition of unfiltered constituents

g Gross-matrix determination

h Annual median and range, respectively

i Calculated from analyses of duplicate samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

j Mineral analyses made by United States Geological Survey, Quality Assurance Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), California State Water Resources Control Board (CSWRCB), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (ADPH), City of Long Beach, Department of Public Health (LBPH), Lemoore Testing Laboratories, Inc. (TLI), or California Department of Water Resources (CDWR), as indicated

TABLE 3a-1
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 5)

SACRAMENTO RIVER AT SINDIGASS CLOVER (STA. 07)

Date and time sampled P.S.T.	Discharge Temp in cfs	Dissolved oxygen ppm	Specific conductance at 25°C	Mineral constituents in equivalents per million												Total ^k dis- solved solids in ppm	Hardness as CaCO ₃ ppm	Tur- bid- ity in ppm	Analyzed by ^h
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Silic- ic acid (SiO ₂)	Other constituents				
1990	Tidal																	UTSP	
1/1			125	6.6	13	4.5	5.3	2.0	0.0	39	13	7.8	1.9		18				
1/4			144	7.6	14	5.1	8.7	0.0	0.0	69	13	6.4	0.6		106	79			
7/6			160																
10/17			160	4.2	7.4	8.7	6.9	2.0	5.1	54	7.2	6.7	0.6		113	21			
12/1	1.0	1.0	162	7.0	13	6.3	8.5	0.0	0.0	60	21	5.7	0.6		144	24			
5/12	70	70	223	6.8	16	7.2	15	0.0	0.0	74	30	11	0.6		200	32			
7/15	0.0	0.0	217	7.0	14	8.2	14	0.0	0.0	79	15	17	0.0		134	30			
7/19	13.3	13.3	165	7.3	13	6.0	10	0.0	0.0	60	6.2	9.2	0.6		102	28			
12/20			187	7.4	14	5.2	11	1.2	0.0	66	9.6	7.8	1.2		144	29			
2/11			260	7.8	16	12	16	0.0	0.0	106	15	15	0.0		156	28			
7/14			208	7.6	13	9.6	12	1.6	0.0	82	11	5.7	0.6		176	25			
10/12	68	68	179	8.1	13	6.1	14	2.0	0.0	84	19	8.5	0.6		120	33			
12/15	58	58	204	8.5	17	7.4	11	1.6	4.2	79	7.2	7.8	0.0		120	24			
12/14	53	53																	
12/6																			

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY PROJECT (NO. 5)

SACRAMENTO RIVER AT TOLAND LANDING (STA. 15+)

Date and time analyzed P.S.T.	Overcast Temp. in air	Dissolved oxygen ppm	Specific conductance in open air at 25°C µmhos/cm	Mineral constituents in equivalents per million							Total dissolved solids in ppm	Total suspended solids in ppm	Total solids in ppm	Turbidity in nephelometric units	Accepted by
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Silica (SiO ₂)	Other constituents
1959	Tidal														
1/13			162										130		
1/30															
8/10	40		152										120		
3/17			197												
1/30			154										140		
4/13	63												120		
5/18	65		206												
1/30			160												
6/15	66												176		
9/45													320		
7/13	72		4,760										3,000		
12/40															
8/11	71		1,268										720		
1/30															
9/14	69		318										96		
12/50															
10/18	67		775												
1/40															
11/9	59		760												
1/45															

a. Field pH

b. Laboratory pH

c. Sum of calcium and magnesium in ppm

d. Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{VI}), reported here as 0.0 except as shown

e. Derived from conductivity vs. TDS curves

f. Determined by addition of analyzed constituents

g. Gravimetric determination

h. Annual median and range respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County, Food Control District (SBCFD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBOPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (CDWR) as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

[illegible]

a Field OH

b. Laboratory off.

Sum of calcium and magnesium in ppm.

4. Iron (Fe) aluminum (Al), arsenic (As),

Derived from conductivity vs TDS curves.

(f) Determined by addition of analyzed constituents.

Gravimetric determination

6. *Journal of the American Medical Association* 2000;283:2689-2694.

Annual income and usage reported by respondents. Water analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Control District (SCCD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LDBPH), Temoak, Temoak Department of Water Resources (DWR), as indicated.

TABLE 3-A

ANALYSES OF SURFACE WATER

CENTRAL WATER DISTRICT (C)

SALT SLUDGE AT SAN LUIS BAY (STA. 90-1)

Date and time of day and P.S.T.	Discharge in cfs in day	Temp. in deg. fahr.	Dissolved oxygen, ppm	Specific conductance at 25°C, μ mhos/cm	Major constituents in equivalents per million										Total dissolved solids in ppm	Percent solids in ppm	Temperature of water, °C	Temperature of water, °F	Amount of water, ml.
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					
1/10	94	56	1.2	30	1800	7.1	250	6.1	0.3	296	273	268	8.1	0.2		60	35	185	1350
1/11	99	56	1.2	30	2370	6.9	250	6.1	0.3	296	273	268	8.1	0.2		61	35	185	1360
1/12	99	56	1.2	30	1700	7.1	250	6.1	0.3	296	273	268	8.1	0.2		59	35	185	1360
1/13	90	64	0.3	66	1320	6.2	170	5.6	0.3	296	273	268	8.1	0.2		71	35	185	1360
1/14	69	67	0.3	61	1320	6.3	170	5.6	0.3	296	273	268	8.1	0.2		71	35	185	1360
1/15	126	76	5.3	68	962	7.3	260	7.2	0.3	296	273	268	8.1	0.2		59	35	185	1360
1/16	99	76	5.8	69	973	7.2	250	7.1	0.3	296	273	268	8.1	0.2		56	35	185	1360
1/17	75	76	7.4	47	1290	7.1	250	7.1	0.3	296	273	268	8.1	0.2		59	35	185	1360
1/18	54	78	7.2	87	1330	7.1	270	7.1	0.3	296	273	268	8.1	0.2		71	56	268	1113
1/19	38	78	7.4	88	1360	7.1	270	7.1	0.3	296	273	268	8.1	0.2		71	56	268	1113
1/20	24	76	7.2	46	1300	7.2	250	7.2	0.3	296	273	268	8.1	0.2		71	56	268	1113
1/21	62	78	6.6	71	938	7.1	250	7.1	0.3	296	273	268	8.1	0.2		71	56	268	1113

TABLE B-5

ANALYSES OF SURFACE WATER

11 JULY 1968 (P. 1)

in 100 ml of 10% NaOH (pH 12.0)

Date and time sampled PST	Discharge temp in °F	Dissolved oxygen in ppm % Sat	Specific conductance at 25°C	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total N, C, S in ppm	As analyzed by I
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Bromine (Br)	Silica (SiO ₂)	Other constituents	
10																	
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99																	
100																	

F-10 (M)

Continued on

Analyses for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

Water, Sediment, Air, and Soil (P) - Analyze for constituents in

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 1)

SAN JOAQUIN RIVER AT CHOWS LANDING BRIDGE (STA. 26.1)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance (at 25°C) µmhos/cm	Mineral constituents in equivalents per million											Total dissolved solids in ppm	Hardness as CaCO ₃ Total N.C. ppm	Turbidity MPN/ml 100 ml	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)				
1959																		USBR	
1/15		56		948			122						120			736	14		
1/25		54		786			94						114			592	52		
2/17																			
4/15		62		1,665			211						274			672	44		
11/30		68		1,195			69						212			760	26		
4/14																			
1200		70		1,288			143						188			112	50		
5/13																			
1955		68		949			22						165			616	24		
1310		76		1,379			155						244			994	40		
7/16		42		1,108			133						192			716	52		
1115																			
8/11		63		1,265			154						212			894	53		
1210		70		1,288			192						214			752	70		
1295																			

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown

0.00

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Geological Survey, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

SAN JOAQUIN RIVER AT BRECKENT RIVER BRIDGE (SPA. 25C-)

Date of sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance in 25°C	pH ^a	Mineral constituents in parts per million										Total dis- solved solids in ppm	Res- sidual solids in ppm	Hardness on CaCO ₃ basis ppm	Tur- bid- ity in ppm	Analyzed by ^h		
						Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlor- ide (Cl)	Nit- rate (NO ₃)	Phos- phate (P)						Boron (B)	Silica (SiO ₂)
1959																						
8/4	76	82	10.9	137	8.3	7.02 ^e	225	9.79		0.0	202	172			0.5		1060 ^e	58	352	186	20	Mean
1300										0.0	3.31	13.31										6.2
9/10	77	80	8.5	104	8.2	66	208	7.0	0.0	0.0	212	298	1.9	0.2	0.7	21	91.1 ^f	57	336	162	30	Maximum
1230										0.0	3.13	17.03	0.03				1630 ^e					>7,000.
10/8	38	68	9.1	99	8.0	11.20 ^e	26.9	15.05	0.0	0.0	218	68.0			1.0		1630 ^e	58	590	377	20	Maximum
1300										0.0	1.75	15.75										0.23
11/5	26	58	11.7	114	7.3	13.20 ^e	199	12.71	0.0	250	276	72			0.8		1960 ^e	62	660	455	20	
1260										0.0	4.20	20.40										
12/10	66	67	12.44	105	7.7	8.38 ^e	304	13.22	0.0	246	265	44.5			1.0		1880 ^e	61	419	217	30	
1230										0.0	1.03	15.55										

^a Field pH.^b Laboratory pH.^c Sum of calcium and magnesium in ppm.^d Iron (Fe), aluminum (Al), arsenic (As).^e Derived from conductivity vs TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY SYSTEM (C.V.S.)

SAN JOAQUIN RIVER 300 YDS. UP-STREAM FROM INTAKE (1000 YDS. UP-STREAM)

Date and time sampled P.S.T.	Discharge in cfs in day	Temp. in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Per- cent soli- dity in ppm	Hardness as CaCO ₃ Total N.C. ppm	Temp. Dis- sol- ution in °F	Analyzed by
						Calcium (Ca) (mg)	Magne- sium (Mg) (mg)	Sodium (Na) (mg)	Potash- ium (K) (mg)	Carbon- dioxide (CO ₂) (mg)	Bicar- bonate (HCO ₃) (mg)	Sul- fate (SO ₄) (mg)	Chlor- ide (Cl) (mg)	Ni- trate (NO ₃) (mg)	Fluor- ide (F) (mg)	Boron (B) (mg)	Silica (SiO ₂) (mg)					
1/10/54					7.4	124	66	17	1	236	1	1	68	1	2	6	144	1	1			
5/18/54																						

C.V.S.

1/10/54

5/18/54

1/10/54

5/18/54

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5/18/54

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1/10/54

5/18/54

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

SAN JOAQUIN RIVER AT FRIANT (SP. 24)

Date and time of day P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in ppm	pH (membrane or 25°C)	Major constituents in equivalents per million										Total dissolved solids in ppm	Percent calcium in ppm	Hardness as CaCO ₃ in ppm	Temp by NTC in ppm	Caliform by MPN/ml	Analyzed by									
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)							Boron (B)	Silica (SiO ₂)	Other constituents						
1959																			USBR										
1/15 0830	175	50	9.7	85	44.1	6.5	0.28 ^c	2.9	0.13	0.0	0.00	16	0.26	4.8	0.34	0.0	0.0	35	12	0	Median	PS							
2/3 1300	180	53	10.3	94	43.4	6.8	0.28 ^c	3.7	0.16	0.0	0.00	16	0.26	4.0	0.11	0.0	0.0	40	12	0	Maximum	PS							
3/10 9:30	186	46	11.0	99	56.6	6.8	0.30 ^c	4.5	0.20	0.0	0.00	20	0.33	6.0	0.17	0.1	0.0	40	15	0	Minimum	PS							
4/7 0645	108	47	10.4	88	46.6	6.8	0.33 ^c	3.2	0.14	0.0	0.00	18	0.30	4.5	0.13	0.0	0.0	20	17	2	5								
5/6 0640	137	47	10.9	92	45.9	6.8	0.24	3.5	0.15	0.6	0.00	18	0.30	3.8	0.11	0.0	0.0	31	16	1	7								
6/3 1535	133	60	11.4	124	44.0	7.7	0.28 ^c	4.1	0.18	0.0	0.00	16	0.26	4.0	0.11	0.0	0.0	41	13	0	6								
7/8 0745	181	48	11.2	96	44.8	7.0	0.28 ^c	3.9	0.17	0.0	0.00	18	0.30	3.2	0.09	0.0	0.0	40	13	0	3								
8/6 0635	175	48	10.5	90	44.4	6.8	0.25 ^c	3.6	0.16	0.0	0.00	18	0.30	2.5	0.07	0.0	0.0	39	12	0	5								
9/ 10/7 1600	Sample Broken in Transit																												
11/4 1700	148	48	10.6	91	72.2	6.8	0.30 ^c	6.6	0.29	0.0	0.00	20	0.32	7.5	0.21	0.0	0.0	38	24	0	2								
12/9 1545	106	51	10.9	97	45.0	6.8	0.28 ^c	3.9	0.17	0.0	0.00	16	0.26	4.0	0.11	0.0	0.0	41	19	0	6								
	78	51	11.0	98	50.3	6.9	0.32 ^c	4.4	0.19	0.0	0.00	15	0.25	2.5	0.07	0.0	0.0	37	16	4	2								

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 4)

SAN JOAQUIN RIVER NEAR GRAYSON (STA. 26)

Date and time sampled P.S.T.	Discharge Temp in cfs in °F	Dissolved oxygen ppm	% Sat	Specific pH (microbore at 25°C)	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent sodium Total N.C. ppm	Hardness as CaCO ₃ ppm	Turbid- ity - Coliform ^b MPN/ml	Analyzed by 1	
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Flu- oride (F)						Bor- on (B)
1959																				
1/20	960	50	7.2	63	1,030	7.9	145	6.31	0.0	226	1.60	4.51				58	956	41	60	Median 130.
11/40	660	52	9.3	84	1,460	8.1	280	9.57	0.0	222	2.90	7.05		0.8		61	301	119	0	Maximum 2,400
2/10	1100						215	9.35	0.0	234	2.77	7.81		0.7		48	333	131	20	Minimum 23
3/11	665	60	8.8	88	1,660	8.4	172	9.35	0.0	208	2.18	6.15		0.5		56	209	121	35	
3/14	440	70	10.9	121	1,340	8.4	172	7.43	0.0	216	1.64	3.31		0.4	23	54	208	111	40	
5/11	525	75	12.4	145	1,310	8.4	79	26	163	4.8	220	1.0	0.1	0.1	0.16	773	54	208	111	40
13/30	340	74	11.0	127	1,280	8.5	167	7.26	0.0	214	2.48	6.43		0.5		56	283	8	13	
6/8	1240						167	7.26	0.0	214	2.48	6.43		0.5		59	348	145		
7/30	130	80	12.6	124	1,490	8.1	167	7.26	0.0	208	1.64	3.31		0.5		59	348	145		
13/45							167	7.26	0.0	208	1.64	3.31		0.5		59	348	145		
8/	Not Sampled																			
9/	Sample broken in transit																			
10/9	385	68	9.2	100	1,070	8.2	134	5.78	0.0	178	1.81	5.75		0.4		56	230	84	15	
11/14	315	59	9.0	88	1,380	8.0	190	8.26	0.0	230	2.25	6.34		0.4		58	304	115	20	
12/16	345	49	9.9	86	1,550	8.1	211	9.13	0.0	220	2.80	7.30		0.6		48	333	153	10	
12/10																				

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFCD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER
CENTRAL VALLEY REGION (NO. 1)

SAN JOAQUIN RIVER AT JERRY POINT (33° 42' N, 120° 42' W)

Date sample P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm %Sat	Specific conductance at 25°C µmhos/cm	Mineral constituents in —equivalents per million—										Total dissolved solids in ppm	Per- cent sulfate in total	Hardness as CaCO ₃ in ppm	Turbidity in NTU	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Silica (SiO ₂)	Other constituents			
1959	Tidal																		USER
1/14		53		304													268		
1/14		49		317													299		
2/9		57		321													215		
4/14		66		183													354		
5/14		66		293													256		
10/35		70		484													304		
6/17		75		2,820													1,692		
7/13				2,394													1,308		
8/10		68		426													972		
9/15		68		288													236		
10/16		59		342													208		
11/12																			
11/10																			

a. Field pH.

b. Laboratory pH.

c. Sum of calcium and magnesium in ppm.

d. Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e. Derived from conductivity vs TDS curves.

f. Determined by addition of analyzed constituents.

g. Gravimetric determination.

h. Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

SAN JUANITO RIVER NEAR MONROVIA (STA. 25)

Date and time of sample P.S.T.	Discharge in cfs	Temp. in °F	Dissolved oxygen ppm	Specific conductance (microhm/cm at 25°C)	pH	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Total Solids by gravimetry MPN/ml	Analyzed by	
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)						Other constituents
1959																			UPDS				
1/13	26*	54	9.7	200	6.8	1.04 ^c	1.9	0.83	0.0	0.0	60	22	0.0	0.1			116 ^e	44	50	3	55	Median %	
1/60																							
2/2	10*	56	10.7	101	6.9	2.26 ^c	56	2.11	0.0	1.01	1.76	67	1.95	0.1			278 ^e	53	110	27	2	Maximum 2.4%	
16/5																							
3/9	16.4*	63	9.3	96	7.5	2.26 ^c	59	2.57	0.0	1.02	1.67	76	2.11	0.3			208 ^e	51	123	29	9	Maximum 0.2%	
1/65																							
4/6	30.2*	67	9.1	98	7.5	2.78 ^c	49	2.13	0.0	0.94	1.94	77	2.17	0.3			208 ^e	43	139	62	5		
11/5																							
5/4	26.3*	67	8.7	94	7.7	2.10	79	3.11	3.4	1.18	65	148	0.0	0.1	0.2	1.2	Fe 0.07 PO ₄ 0.30 d	445 ^e	44	214	101	15	
16/40																							
6/1	32.4*	74	8.0	93	7.3	1.78 ^c	30	1.36	0.0	0.87	1.43	38	0.0	0.1			191 ^e	42	91	20	7%		
7/6	38*	78	8.5	104	7.6	2.11 ^c	27	2.09	0.0	0.99	1.69	78	2.25	0.1			276 ^e	53	108	27	20		
12/45																							
8/5	40.9	74	7.7	89	7.5	3.13 ^c	131	5.70	0.0	1.48	90	235	6.63	0.1			596 ^e	65	156	82	130		
9/10	30.2	75	7.6	89	6.6	1.15	75	3.26	0.0	0.91	24	120	0.6	0.2	0.2	1.7	Fe 0.08 Al 0.10 d	222 ^e	58	113	38	60	
9/30																							
10/8	80	68	8.5	93	7.5	2.55 ^c	64	2.78	0.0	1.13	1.85	97	2.74	0.3			354 ^e	53	125	32	25		
11/5	74	55	9.9	93	7.5	3.76 ^c	111	4.83	0.0	1.50	1.41	141	3.08	0.3			600 ^e	56	188	65	25		
9/50																							
12/10	53	47	12.3	104	7.5	3.92 ^c	213	4.62	0.0	0.0	1.43	140	1.25	0.4			504 ^e	56	196	70	5		
9/51																							

• Last 2 years

* Lat. 37° 39' N.

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs. 103 curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWD), Department of Water Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBDPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (RD. 5)

SAN JOAQUIN RIVER ABOVE NORTHER RIVER (STA. 3+0)

Date and time sampled P.S.T.	Discharge Temp. in cfs in day	Dissolved oxygen ppm % Sat	Specific conductance at 25°C	pH	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Total suspended solids in ppm	Total hardness as CaCO ₃ ppm	Total hardness as CaCO ₃ ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					
1/26/54	54		1,260	7.9	51	27	112	3.5		217	175	178	5.3						UTPR		
1/27/54																					
1/28/54	54		1,420	8.1	55	27	114	2.1	0.0	169	271	205	5.3						UTPR		
1/29/54																					
1/30/54	68		2,731	7.7	77	64	184	4.3	0.0	243	659	325	5.6						UTPR		
1/31/54	68		1,814	7.7	69	57	170	5.3	0.0	189	234	141	1.0						UTPR		
2/1/54	71		1,307	7.6	71	54	21	3.3	0.0	178	242	24	1.0						UTPR		
2/2/54	68		1,114	7.7	70	16	171	2.7	0.0	186	134	268	1.6						UTPR		
2/3/54	68		1,391	7.4	53	35	131	3.0	0.0	143	162	234	1.4						UTPR		
2/4/54	68		1,463	7.7	47	41	242	4.5	0.0	146	199	21	1.0						UTPR		
2/5/54	68		1,510	7.8	54	48	271	2.4	0.0	150	240	172	1.1						UTPR		
2/6/54	71		1,334	8.1	57	71	264	2.6	0.0	168	341	135	1.1						UTPR		
2/7/54	74		1,401	7.6	54	71	269	2.1	0.0	204	144	115	1.1						UTPR		
2/8/54	74		1,361	8.1	50	77	14	3.9	1.2	20	300	200	1.0						UTPR		

B-13 PM

Continued on p. 1

Data are unrounded except as noted.

1. For Fe, Mn, Ni, Cu, Zn, Pb, Cd, Hg, and P, the minimum value is 1.0 mg/l unless otherwise noted. For all other constituents, the minimum value is 0.1 mg/l unless otherwise noted.

2. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

3. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

4. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

5. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

6. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

7. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

8. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

9. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

10. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

11. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

12. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

13. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

14. For all constituents, the minimum value is 0.1 mg/l unless otherwise noted.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

SAN JOAQUIN RIVER AT MENDOTA BRIDGE (STA. 102)

Date and time of day P.S.T.	Discharge Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C pH	Mineral constituents in equivalents per million												Total dissolved solids in ppm	Percent total N.C. in ppm	Hardness on CaCO ₃ ppm	Turbidity in Nephelometric Units	Coliform MPN/ml	Analyzed by 1
				Calcium (Ca) (mg)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)						
1999	71.6A1																			1992G	
5/13 0950	58	8.4	82	2.32 ^c		60	2.61	0.0	103	1.59	82	2.31		0.2		304 ^e	53	116	32	Median 23.	
2/10 0940	50	10.5	93	3.28 ^c		87	3.76	0.0	186	2.07	122	3.47		0.3		423 ^f	54	161	58	0	
3/12 1700	60	8.8	705	3.26 ^c		79	3.44	0.0	124	2.03	114	3.21		0.2		399 ^g	51	162	60	40	
4/3 1020	70	13.6	151	3.70 ^c		96	4.78	0.0	162	2.66	156	4.40		0.3		494 ^e	51	202	69	40	
5/13 1230	74	8.4	97	2.69 ^c	2.09	133	5.10	0.0	185	3.05	180	3.40	0.2	0.3	Al 0.15 Fe 0.704	656 ^g	54	239	86	85	
6/10 1000	70	10.3	114	3.78 ^c		115	5.00	0.0	176	2.92	200	5.64		0.2		572 ^g	52	232	86	30	
7/3 0700	74	4.4	51	3.78 ^c		129	5.61	0.0	188	3.08	205	5.78		0.3		600 ^g	53	246	92	15	
8/12 0800	82	6.5	82	3.78 ^c		131	5.70	0.0	184	3.02	232	6.54		0.2		638 ^g	54	246	95	30	
9/11 0530	79	6.8	83	3.78 ^c	2.01	132	6.40	0.0	203	3.33	210	6.80	0.0	0.2	Al 0.02 Zn 0.05 d Fe 0.152	633 ^g	53	245	79	10	
10/8 1615	68	7.7	84	3.78 ^c		127	5.52	0.0	192	3.20	198	5.58		0.2		594 ^g	54	237	77	40	
11/7 0815	57	10.4	98	3.70 ^c		83	3.61	0.0	141	2.31	130	3.67		0.3		434 ^e	51	173	57	20	
12/11 0700	48	11.1	95	3.70 ^c		88	3.98	0.0	130	2.13	142	4.00		0.2		436 ^g	53	174	67	35	

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles County Sanitation Department (LASAN); Los Angeles County Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CERNAL VALLEY RIVER AT PATTERSON WATER TREATMENT PLANT (TIA, 1964)

Date sampled PST	Discharge in cfs in 99	Dissolved oxygen ppm	Specific conductance in 25°C ppm %Sol	pH	Mineral constituents in parts per million						Total dissolved solids in ppm	Total hardness as CaCO ₃ ppm	Total TDS ppm	Temp. in °C	Temp. in °F	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Bromine (Br)	Other constituents
1-8-64																
1-11-64	76		1,008	8.0	48	70	133	2.3	—	19	116	135	—	—	—	224
1-15-64	86		84	8.2	4	15	66	—	6.0	171	97	—	3.1	—	—	464
1-16-64	80		1,000	7.7	18	70	228	1.2	—	23	—	209	—	—	—	3,000
2-10-64	800		1,084	7.1	39	4	135	2	—	87	132	—	—	—	—	2,700
2-11-64	775		1,111	8.1	10	86	129	2.7	0.0	138	133	130	—	—	—	3,400
2-12-64	—		1,280	7.7	8	—	127	1.6	—	126	160	200	—	—	—	4,000
2-13-64	—		1,111	7.3	31	42	177	1.2	—	138	86	270	—	—	—	3,000
2-14-64	—		1,111	7.1	4	—	155	—	—	72	100	—	—	—	—	2,000
2-15-64	—		1,000	7.0	70	46	130	4.5	—	110	26	263	—	—	—	3,300
2-16-64	—		1,400	8.4	—	—	102	—	—	171	172	162	—	—	—	4,000
3-10-64	—		816	7.0	—	—	136	1.0	—	103	96	203	—	—	—	2,000
3-20-64	—		1,110	8.1	6	46	111	6.1	—	112	203	—	—	—	—	3,300

1. Data left

2. Quantity not

3. Quantity not

4. Quantity not

5. Quantity not

6. Quantity not

7. Quantity not

8. Quantity not

9. Quantity not

10. Quantity not

11. Quantity not

12. Quantity not

13. Quantity not

14. Quantity not

15. Quantity not

16. Quantity not

17. Quantity not

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 4)

SAN JOAQUIN RIVER ABOVE SALT SLURRY (STA. 1133)

Date and time sampled P.S.T.	Discharge Temp in °C in cfs	Dissolved oxygen ppm %S _{at}	Specific (micromhos at 25°C)	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent total solids in ppm	Hardness on CaCO ₃ Total N.C. ppm 50°C	Turbidity - big - in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)			
1959																			USBR
1/25	56		248	7.3	21	70	17	1.6	0.0	54	58	22	0.6				156	31	
1/30																			
2/17	54		376	8.2	22	12	41	0.0	0.0	134	27	33	6.2				252	46	
2/25																			
3/16	62		946	7.7	50	15	154	1.2	0.0	213	73	173	1.0				608	64	
12/15																			
4/14	68		941	7.5	46	20	107	2.0	0.0	179	72	158	3.1				546	54	
12/40																			
5/13	72		863	7.7	50	19	89	3.5	0.0	154	89	132	2.5				530	48	
11/15																			
6/16	69		1,056	7.7	59	23	99	2.7	0.0	174	93	167	0.6				700	47	
1/45																			
7/16	74		1,168	7.4	63	26	118	3.9	0.0	187	102	204	0.6				682	49	
10/25																			
8/11			1,401	7.5	60	30	161	5.5	0.0	186	107	279	0.6				896	56	
13/05																			
9/16	69		1,357	8.0	64	29	144	5.9	0.0	186	92	271	2.5				812	59	
13/30																			
10/14	70		1,076	8.1	56	23	112	5.5	0.0	176	75	189	2.5				628	50	
13/30																			
11/16	60		1,468	8.0	57	24	195	2.3	0.0	262	88	263	0.6				872	62	
1/400																			
12/14			1,885	8.1	66	35	253	4.3	0.0	300	115	373	0.6				1,124	64	
13/15																			

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFCD), Metropolitan Water District of Southern California (MWD), Los Angeles County Flood Control District (LACFCD), and the Central Valley Regional Water Control District (CVRWD).
j Water analyses made by the California Department of Public Health (LADPH), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Departmental Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 4)

SAN JOAQUIN RIVER NEAR YERMALES (STA. 27)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in ppm	pH	Mineral constituents in equivalents per million										Total dissolved solids in ppm	Percent sodium in ppm	Hardness on CaCO ₃ in ppm	Total bicarbonate in ppm	Analyzed by	
					Specific constituents in															
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)
1/5																			USGS	
1/12 1340	2,560	58	8.4	82	477	6.5	5.4	2.35		0.0	1.57	70	1.97		0.2	270 ^d	53	104	25	Median 20
2/9 1110	2,020	50	10.5	93	794	7.1	108	4.70		0.0	1.02	127	3.78		0.4	446 ^d	58	170	54	Maximum 1,000
3/9 1015	2,930	55	9.4	88	609	7.3	69	3.00		0.0	1.09	96	2.71		0.3	344 ^d	54	130	41	Minimum 0.62
4/5 1350	780	70	8.8	98	1,010	8.1	100	4.35		0.0	1.66	190	5.36		0.3	570 ^d	49	226	90	0
5/6 1030	1,070	64	10.4	108	801	7.8	81	3.61		0.0	1.64	51	1.96		0.1	Pe 0.04, Pb 0.55, Zn 0.01, Al 0.11 ^d	47	200	66	25
6/1 0645	715	70	10.2	114	971	8.1	111	4.83		0.0	1.89	185	5.82		0.2	548 ^d	52	227	78	30
7/3 0745	392	72	11.0	125	976	8.1	125	5.11		0.0	1.41	198	5.55		0.3	552 ^d	57	203	87	10
8/12 0940	323	82	7.8	98	1,140	8.0	132	5.74		0.0	1.89	260	7.33		0.3	644 ^d	53	250	101	60
9/10 1500	554	79	8.0	98	1,130	8.1	26	5.83		0.0	1.95	62	2.12		0.3	Pe 0.05, Zn 0.01, Pb 0.03, Al 0.09, Cu 0.02 ^d	53	259	99	20
10/8 1500	1,000	68	9.0	98	1,030	8.1	132	5.74		0.0	1.69	190	5.76		0.3	582 ^d	55	252	77	10
11/6 1510	1,010	59	10.6	104	783	7.5	86	3.70		0.0	1.33	120	3.61		0.1	420 ^d	53	164	55	5
12/10 1540	1,210	51	12.1	108	743	7.7	86	3.36		0.0	1.22	133	3.75		0.1	420 ^d	51	168	68	10

^a Field pH.^b Laboratory pH.^c Sum of calcium and magnesium in gm.^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.^e Derived from conductivity vs. TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively.ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-3 (Continued)
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (Ch. 2)

SAN JOAQUIN RIVER NEAR VISALIA - (Ch. 2)

Date and time sampled P.S.T.	Dissolved in ch. Average Daily	Temp in air °F	Oxidized oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million												Total ⁶ dissolved solids in ppm	Percent solids from Total ⁶ N.C.	Hardness as CaCO ₃ Total ⁶ N.C. ppm	Total ⁶ dissolved solids in ppm	Analyzed by 1
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium + Lithium (K+Li)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)					
5/19-21	Mean	766		9.1	20	25	113	6.6	0	1.6	16	168	2.2	0.2	0.1	21	Fe 0.00	340	4	11.2	
6/1-4		664		9.8	20	27	129	7.0	0.00	1.80	104	100	1.0	0.2	0.2	21	Fe 0.00	6.0	15	50	
6/10-30		471		1.110	24	24	135	7.6	0.2	1.86	76	205	2.6	0.2	0.1	26	Fe 0.01	6.9	44	189	
7/1-8		428		1.130	28	31	146	7.4	0.00	1.92	100	230	3.3	0.2	0.1	26	Fe 0.00	707	11	11.4	
7/19-20		302		1.260	31	35	155	8.6	0.00	1.88	81	261	4.1	0.2	0.1	26	Fe 0.01	748	13	286	
7/21-31		279		1.140	60	27	137	8.0	0.00	1.64	61	372	2.0	0.2	0.1	30	Fe 0.00	707	76	262	
8/1-19		304		1.210	31	36	140	7.8	0.00	1.94	119	270	2.0	0.2	0.1	42	Fe 0.01	738	32	272	
8/20-31		518		1.090	24	28	120	6.4	0.00	1.94	66	210	2.0	0.2	0.2	28	Fe 0.01	641	51	242	
9/1-17		563		1.120	23	26	118	6.6	0.00	2.04	73	229	2.0	0.2	0.1	37	Fe 0.01	675	51	225	
9/19-28		1,260		735	19	16	77	6.9	0.00	1.56	39	119	3.8	0.2	0.1	22	Fe 0.01	433	49	165	
10/7-14		783		1,010	25	28	113	5.4	0.00	1.90	77	189	3.1	0.2	0.1	28	Fe 0.01	600	51	230	
10/16-31		977		781	20	19	86	4.6	0.00	1.44	65	137	3.6	0.2	0.1	31	Fe 0.01	464	50	178	
11/1-15		1,011		764	19	18	86	4.0	0.00	1.36	64	155	3.6	0.2	0.2	30	Fe 0.02	451	52	171	
11/16-29		1,100		783	19	19	81	4.6	0.00	1.40	69	143	3.7	0.2	0.2	29	Fe 0.03	475	52	177	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Water Department (LAWD); City of Los Angeles, Department of Public Health (ADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CYPRESS VALLEY DRAIN (NO. 3)

SAN JOAQUIN RIVER WIDE-STREAM LAKE I, D. INTAKE (NO. 275)

Date sampled P.S.T.	Discharge Temp in cfs in sq ft	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Major constituents in equivalents per million					gparts per million			Total dissolved solids in ppm	Total hardness as CaCO ₃ in ppm	Total hardness as CaCO ₃ in ppm	Total hardness as CaCO ₃ in ppm	Analyzed as l
				Calcium (Ca) (M)	Magnesium (Mg) (M)	Potassium (K) (M)	Sodium (Na) (M)	Carbonate (CO ₃) (M)	Bicarbonate (HCO ₃) (M)	Sulfate (SO ₄) (M)	Chloride (Cl) (M)	Nitrate (NO ₃) (M)	Fluoride (F) (M)	Boron (B) (M)	Silica (SiO ₂) (M)	
1/29			1,112	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
1/30	56		796	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
2/17	54		1,050	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
3/16	55		1,050	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
4/16	56		1,155	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
5/13	57		1,114	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
6/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
7/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
8/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
9/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
10/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
11/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
12/13	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
1/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
2/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
3/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
4/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
5/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
6/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
7/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
8/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
9/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
10/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
11/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				
12/14	58		1,080	7.7			1.42	1.4	1.87	1.28	1.08	3.1				

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TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

SAN JOAQUIN RIVER AT WHITEHOUSE (STA. 246)^a

Data date sampled P.S.T.	Oxygen Temp in cte in deg F	Specific conductance in cte at 25°C ppm	pH	Mineral constituents in equivalents per million							Total dissolved solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ in ppm	Turbidity in NTU in ppm	Analyzed by
				Calcium (Ca) (Mg)	Sodium (Na) (K)	Potash- sum (K)	Carbon- ate (CO ₃) (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Bar- ium (Ba)	Other constituents		
9/59															USBR
1/27 1:50	51	81													
2/10 11:15	48	101													
2/24 1:50	56	144													
3/10 7:30	58	141													
3/24 9:30	58	118													
4/7 11:00	68	106	6.9	9.8	0.7	8.5	0.0	15	6.4	0.6				60	
4/28 11:45	67	78												60	
5/12 11:35	75	82												60	
5/26 10:00	68	77	6.6	6.4	1.8	7.4	0.0	32	2.4	0.0				50	
6/9 10:00	70	78												66	
6/23 9:45	80	80												60	
7/14 11:20	80	116												72	
7/28 12:30	82	76	7.5	4.8	1.1	6.9	1.2	20	5.8	0.6				52	
8/11 12:30	81	80												60	

^a Field pH.

^b Laboratory pH.

^c Sum of calcium and magnesium in eqm.

^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

^e Derived from conductivity vs TDS curves.

^f Determined by addition of analyzed constituents.

^g Gravimetric determination.

^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

ⁱ Mineral analyses by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

STANISLAUS RIVER NEAR MONTE CRESTA, 290

Date and time of day and P.S.T.	Discharge Temp in cfs	Dissolved oxygen in ppm	Specific conductance in microhm/cm at 25°C	pH ^a	Mineral constituents in parts per million					Total dissolved solids in ppm	Percentum	Hardness as CaCO ₃ in ppm	Total hardness as CaCO ₃ in ppm	Turbidity in NTU	Coliforms by MPN/ml	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents
1999																
1/12	604	8.5	81	7.3	170 ^b		5.4		0.0	60		4.8			0.0	
1/15							0.23		0.0	0.78		0.11				
2/2	287	9.7	88	7.1	204 ^b		15		0.0	134		9.1			0.0	
1/135							0.65		0.00	270		0.25				
3/9	1,120	9.8	90	7.2			3.5		0.0	48		3.5			0.0	
0845							0.13		0.00	0.79		0.10				
4/6	132	9.9	111	7.5	216 ^b		16		0.0	146		9.5			0.0	
2/30							0.76		0.00	279		0.27				
5/4	139	8.8	90	7.3	25	9.8	15		0.0	149		8.5			0.0	
1000							0.75		0.00	273		0.16				
6/1	76	7.8	87	7.3	294 ^b		15		0.0	153		9.0			0.0	
0950							0.65		0.00	251		0.24				
7/3	82	7.5	75	7.3	270 ^b		17		5	147		8.5			0.0	
0815							0.74		0.17	241		0.24				
8/12	29	6.6	85	7.3	212 ^b		16		0.0	150		8.0			0.0	
1030							0.76		0.00	246		0.23				
9/10	47	7.0	85	7.3	286	10	18		2.9	158		13			0.0	
1600							0.78		0.07	259		0.37				
10/8	145	7.7	85	7.3	279 ^b		19		0.0	163		14			0.1	
1350							0.83		0.00	267		0.29				
11/5	109	10.0	100	7.5	278 ^b		18		0.0	156		8.8			0.0	
12/5							0.78		0.00	256		0.25				
12/10	160	11.3	101	7.4	212 ^b		17		0.0	136		8.5			0.0	
1600							0.74		0.00	223		0.24				

^a Field pH^b Laboratory pH^c Sum of calcium and magnesium in ppm^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.^e Derived from conductivity vs TDS curves.^f Determined by addition of analyzed constituents.^g Gravimetric determination.^h Annual median and range, respectively.ⁱ Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR); San Bernardino County Flood^j Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR); City of Long Beach, Department of^k Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

STOCKTON SHIP CHANNEL ON FINDER ISLAND (STA. 100)

Date sampled P.S.T.	Discharge in cfs in op.	Temp in op. °F	Dissolved oxygen in op. ppm	Specific conductance at 25°C µmhos/cm	pH ^a	Miscellaneous constituents in equivalents per million											Total dissolved solids in ppm	Percent calcium in ppm	Hardness Total in ppm CaCO ₃ (2.5 x 2.5)	Tur- bidity in ptl per ft	Conform with MPL/ml	Analyzed by
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)						
12/29	Total																					UGGS
1/12		54	8.8	82	6.94	7.2	2.95 ^c		6.1	0.0	10.1	1.69	105			0.1		352 ^e	48	148	64	15
12/45										0.0	1.04		2.56									Median 230.
2/9		50	8.8	78	7.18	7.3	3.25 ^c		8.1	0.0	1.33		116			0.1		399 ^e	52	160	51	1
1/35										0.0	2.15		3.27									Maximum 27,000.
3/12		60	7.5	75	5.21	7.2	2.45 ^c		2.28	0.0	1.22		71			0.2		290 ^e	48	124	24	70
6/30										0.0	2.00		2.12									Minimum 4.5
4/1		65	12.5	132	5.97	7.9	2.78 ^c		6.1	0.0	1.28		92			0.2		332 ^e	50	139	47	55
1/45										0.0	2.00		2.59									
5/12		70	7.0	78	4.51	7.3	2.6	1.3	3.4	0.0	1.06	4.0	72	1.1	0.2	0.2	5.4	263 ^f	47	120	33	85
6/20							1.30	1.10	2.18	0.09			2.03	0.02	0.01							
6/8		74	10.1	117	4.88	8.1	2.59 ^c		5.0	0.0	1.23		74			0.2		271 ^e	47	125	24	27
1/50										0.0	2.78		2.09									
7/1		78	7.0	85	3.66	7.3	2.08 ^c		4.7	0.0	1.04		46			0.1		203 ^e	44	102	17	15
1/40										0.0	1.70		1.50									
8/12		80	6.8	84	4.08	7.3	2.38 ^c		3.5	0.0	1.08		61			0.1		227 ^e	40	116	27	35
12/20										0.0	1.77		1.72									
9/11		79	6.0	73	81.0	7.3	2.00	1.6	10.7	0.0	1.98	26	132	1.6	0.0	0.2	12	461 ^f	57	166	4	5
11/00							1.33	4.65	0.22	0.0	3.25	0.54	4.29	0.03	0.00							
10/6		68	7.3	80	7.87	7.3	3.74 ^c		2.8	0.0	1.66		142			0.2		438 ^e	54	182	29	30
7/30										0.0	3.05		4.00									
11/6		60	8.3	83	7.70	7.5	3.48 ^c		8.9	0.0	1.69		132			0.1		438 ^e	53	174	41	25
1/40										0.0	2.00		3.72									
12/11		52			72.2	7.4	3.18 ^c		8.8	0.0	1.42		135			0.2		418 ^e	52	174	58	45
1/45										0.0	2.33		3.01									

a Field pH.

b Laboratory pH.

c Sum of sodium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr^{VI}), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWSD); Los Angeles Department of Water and Power (LADWP); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-4

ANALYSES OF SURFACE WATER

TERRELL VALLEY BASIN IN (FIG. 5)

SPRING CHERRY AT BEAVER DAM, TIME 10:00 A.M., 11-5

Date sampled P.S.T.	Dissolved in the temp of 25°C	Specific conductance at 25°C ppm %S ₀	Mineral constituents in equivalents per million										Total diss. solids in ppm	Hardness as CaCO ₃ in ppm	Total CaCO ₃ in ppm	Analyzed by J.	
			Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)					Boron (B)
1963																	
1-17	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
1-20	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
1-23	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
1-26	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
1-29	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
2-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
3-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
4-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
5-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
6-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
7-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
8-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-10	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-13	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-16	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-19	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-22	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-25	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
9-28	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
10-1	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
10-4	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
10-7	10	100	10	24	10	1.3	0.0	1.7	6	8	1	0.2	0.2	1.2			
10-1																	

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (CVR)

TORY CREEK NEAR HAMILTON CITY, CALIF., 1981

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	% Sat	Specific conductance at 25°C µmhos/cm	pH	Mineral constituents in equivalents per million											Total dis- solved solids in ppm	Per- cent sod- ium in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Total hard- ness MPN/ml	Analyzed by ¹
							Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)						
1989																							
1/7	4.6	70	11.4	0.6	282	7.4	39.9 ^c		0.0	152		64											
1/13							1.22		0.00	2.10 ^c		1.30 ^c											
2/10	7.4	46	11.8	1.0 ^b	370	7.5	3.20 ^c		0.0	166 ^c		24											
3/17	40 ^b	56	14.5	1.2 ^b	289	8.1	2.10 ^c		0.0	130		17											
4/4							0.57 ^c		0.00	2.13 ^c		0.18											
4/24	22.2	66	8.8	0.4	308	7.7	2.00 ^c		0.0	152		17											
10/00							0.57 ^c		0.00	2.49 ^c													
5/11	5.58	71	6.9	77	339	7.5	36	13	0.0	163	14	18											
10/05							1.20	1.06	0.01	2.27 ^c	0.51	0.29	0.51										
6/11	Not Sampled	Dry																					
7/14	Not Sampled	Dry																					
8/11	Not Sampled	Dry																					
9/1	Not Sampled	Dry																					
10/13	Not Sampled	Dry																					
11/4	Not Sampled	Dry																					
12/2	Not Sampled	Dry																					

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWDSC), Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBDPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 11)

TOMMASO CHEK AT PASKEWITA (ETA 12-1)

Date and time of day P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen in ppm	Specific conductance at 25°C µmhos/cm	Mercurial constituents in equivalents per million												Total diss- olved solids in ppm	Per- cent solid in ppm	Hardness on CaCO ₃ Total in ppm	Total N.C. in ppm	Coliform ^b MPN/ml	Analyzed by I
					Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- ium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluo- ride (F)	Boron (B)	Silica (SiO ₂)						
4/4	14	42	19.2	132	7.5	11	1.55	8.6	0.6	0.3	120	27	10	0.6	0.1	0.2	14	175	13	124	26	1752
4/5	1776	45	11.1	98	160	7.7	1.05	3.7	0.3	0.05	80	12	3.0	0.2	0.1	0.0	12	27	15	21	11	
4/6	479	42	10.8	98	127	7.7	0.95	2.8	0.3	0.05	66	7.7	2.4	0.0	0.0	0.0	10	77	9	68	10	
4/11	1646	48	9.6	93	116	7.5	0.95	3.2	0.6	0.0	61	3.8	2.0	0.0	0.3	0.1	8.7	67	9	53	7	
5/11	1553	72	8.2	93	114	7.7	1.05	3.9	0.4	0.05	75	3.8	4.2	0.0	0.0	0.1	9.7	86	9	66	4	
6/1	99	78	8.0	96	180	7.7	0.95	4.7	0.9	0.0	93	14	5.0	0.0	0.0	0.1	17	116	13	99	8	
7/14	1095	81	8.4	104	282	8.0	1.05	7.2	1.2	2	110	21	16	0.4	0.0	0.0	11.4	157	17	137	24	
8/11	1080	1.2	8.0	98	362	7.8	1.05	17	2.0	0.05	117	35	36	0.0	0.1	0.2	17	216	20	145	40	
9/2	1330	2.2	10.1	127	390	8.1	1.05	19	1.7	0.05	116	39	40	0.0	0.0	0.2	14	297	21	159	57	
10/13	1310	7.1	9.6	107	410	8.1	0.95	13	1.0	0.05	150	40	36	0.0	0.0	0.3	11	240	20	170	47	
11/3	1230	5.1	10.3	100	434	8.0	0.95	21	0.9	0.05	150	35	35	0.0	0.0	0.1	1	246	20	170	46	
12/2	1315	4.9	11.5	106	463	7.9	0.95	17	1.1	0.05	188	34	38	0.0	0.0	0.3	12	267	18	108	44	

0 Field pH.

b. Laboratory only.

Sum of calcium and magnesium in eqm.

Sum of calcium and magnesium in epm.

d Iron (Fe), aluminum (Al), organic (As), cop

^e Derived from conductivity vs TDS curves.

f Determined by addition of a

g. Gravimetric determination.

Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

TUOLUMNE RIVER BELOW DON PEDRO DAM (STA. 31A)

Date and time sampled P.S.T.	Discharge in cfs	Temp in °F	Dissolved oxygen ppm	Specific conductance at 25°C (µmhos/cm)	pH	Mineral constituents in parts per million												Total dissolved solids in ppm	Percent solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. ppm	Turbidity in ppm	Analyzed by
						elements per million																	
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)						
1959																							
1/20	985	50	9.0	79	57.8	6.9		2.4		0.0	26	0.13	2.0	0.76				44 ^c	17	28	7	Median 2.3	
1/30								0.10		0.0	13	0.21	2.0	0.76		0.0		30 ^c	20	16	5	Maximum 251	
2/10	1,340	48	9.0	85	39.4	6.9		1.0		0.0	0.0	0.21	2.0	0.76		0.0		30 ^c	20	16	5	Maximum 251	
3/11	1,500	48	8.4	72	60.1	7.1		2.3		0.0	28	0.42	0.8	0.76		0.0		46 ^c	18	23	0	Minimum 0.21	
3/10								0.10		0.0	0.0	0.42	1.2	0.76		0.0		33 ^c	21	17	1	2	
4/14	2,480	53	9.4	86	41.0	6.7		0.0		0.0	20	0.33	1.2	0.76		0.0		33 ^c	21	17	1	2	
5/11	2,190	58	8.3	81	39.3	6.8	0.5	2.2	0.7	0.0	15	0.76	0.8	0.2	0.0	0.0	2.0	36 ^c	28	12	0	2	
5/11	2,190	58	8.3	81	39.3	6.8	0.5	2.2	0.7	0.0	15	0.76	0.8	0.2	0.0	0.0	2.0	36 ^c	28	12	0	2	
6/8	2,190	59	7.9	79	25.8	6.8		1.8		0.0	12	0.20	0.5	0.0		0.0		20 ^c	29	10	0	2	
6/8	2,190	59	7.9	79	25.8	6.8		1.8		0.0	12	0.20	0.5	0.0		0.0		20 ^c	29	10	0	2	
7/14	2,220	64	7.5	76	24.9	6.7		2.7		0.0	12	0.20	2.5	0.0		0.0		10 ^c	43	8	0	10	
7/14	2,220	64	7.5	76	24.9	6.7		2.7		0.0	12	0.20	2.5	0.0		0.0		10 ^c	43	8	0	10	
8/	Not Sampled																						
9/	Broken in Transit																						
10/8	1,500	67	7.1	77	40.9	6.8		1.8		0.0	14	0.23	2.5	0.0		0.1		31 ^c	21	15	4	3	
10/8	1,500	67	7.1	77	40.9	6.8		1.8		0.0	14	0.23	2.5	0.0		0.1		31 ^c	21	15	4	3	
11/14	1,500	61	7.6	77	33.3	6.8	1.5	1.9		0.0	15	0.25	1.5	0.0		0.1		29 ^c	21	15	3	10	
11/14	1,500	61	7.6	77	33.3	6.8	1.5	1.9		0.0	15	0.25	1.5	0.0		0.1		29 ^c	21	15	3	10	
12/17	1,500	59	8.7	78	36.8	6.8		1.7		0.0	16	0.26	2.2	0.0		0.0		36 ^c	21	13	0	17	
12/17	1,500	59	8.7	78	36.8	6.8		1.7		0.0	16	0.26	2.2	0.0		0.0		36 ^c	21	13	0	17	

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Mineral analysis (SCEC) made by the State Water Resources Control Board, San Bernardino County Flood Control District, San Bernardino, California

i Mineral analysis (SCEC) made by the State Water Resources Control Board, San Bernardino County Flood Control District, San Bernardino, California

Public Health (LBDPH), Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); as indicated.

TABLE B-1

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (EO. 5)

TUCUMCURI RIVER AT RICHMAN-VALENTINE BRIDGE (479, 70)

Date and time sampled PST	Discharge in cfs	Temp in °F	Dissolved oxygen in ppm	Specific conductance at 25°C in µmhos/cm	Mineral constituents in equivalents per million										Total solids in ppm	Per- centage of solids in ppm	Colloidal solids in ppm	Analyzed by
					Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate equivalents (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)				
10/9	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/9	
10/10	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/10	
2/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/11	
3/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/11	
4/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/11	
5/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/11	
6/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/11	
7/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/11	
8/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/11	
9/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/11	
10/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/11	
11/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	11/11	
12/11	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	12/11	
1/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	1/12	
2/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/12	
3/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/12	
4/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/12	
5/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/12	
6/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/12	
7/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/12	
8/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/12	
9/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/12	
10/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/12	
11/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	11/12	
12/12	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	12/12	
1/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	1/13	
2/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/13	
3/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/13	
4/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/13	
5/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/13	
6/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/13	
7/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/13	
8/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/13	
9/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/13	
10/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/13	
11/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	11/13	
12/13	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	12/13	
1/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	1/14	
2/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/14	
3/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/14	
4/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/14	
5/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/14	
6/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/14	
7/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/14	
8/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/14	
9/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/14	
10/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/14	
11/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	11/14	
12/14	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	12/14	
1/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	1/15	
2/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/15	
3/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/15	
4/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/15	
5/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/15	
6/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/15	
7/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/15	
8/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/15	
9/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/15	
10/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/15	
11/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	11/15	
12/15	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	12/15	
1/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	1/16	
2/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	2/16	
3/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	3/16	
4/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	4/16	
5/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	5/16	
6/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	6/16	
7/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	7/16	
8/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	8/16	
9/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	9/16	
10/16	1100	72	8.9	80	125	7.3	0.78	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	0.79	10/16	
11/16	1100	72																

TABLE B-4

ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

FIVE-MILE RIVER AT THERMOS CITY (STA. 13)

Date sampled P.S.T.	Discharge in cfs in of	Dissolved oxygen ppm	Specific conductance micromhos at 25°C	Mineral constituents in parts per million										Total dissolved solids in ppm	Per- cent dissolved in ppm	Hardness as CaCO ₃ in ppm	Tem- per- ature in ppm	Calorific value in ppm	Applied by ¹
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potas- sium (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Flu- oride (F)	Boron (B)	Silica (SiO ₂)	Other constituents			
1/26/51	1,470	5.2	74	380	7.2	1,100	1.00	0.0	71	75	3.12	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/20/51	1,200	5.2	74	380	7.2	1,100	1.00	0.0	71	75	3.12	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
2/2/51	1,200	5.2	74	380	7.2	1,100	1.00	0.0	71	75	3.12	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/15/51	1,200	5.2	74	380	7.2	1,100	1.00	0.0	71	75	3.12	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
3/11/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/20/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
7/1/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/25/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
7/11/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
12/15/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
6/8/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
7/21/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/20/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
3/7/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
9/2/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
1/25/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
10/9/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
11/14/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0
12/16/51	1,400	6.0	73	351	7.3	1,200	1.00	0.0	73	67	1.80	0.0	0.0	0.0	0.0	0.0	1.0	1.0	100.0

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

i Mineral analyses (S&P) by United States Geological Survey, Quality Control Division, Office of Reclamation (USGR), United States Public Health Service (USPHS), San Bernardino County Flood Control District, California State Water Project (CSWPD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR); are indicated.

TABLE B-4
ANALYSES OF SURFACE WATER

CENTRAL VALLEY REGION (NO. 5)

TYRA RIVER NEAR SHASTAVILLE (STA. 91-3)

Date and time of day and P.S.T.	Discharge temp in °C ^a	Dissolved oxygen in ppm % Sol ^a	Specific conductance in micromhos/cm at 25°C ^a	Major constituents in equivalents per million										Total dissolved solids in ppm	Percent dissolved solids in ppm	Hardness as CaCO ₃ in ppm	Total N.C. in ppm	Turbidity in MPN/ml	Analyzed by ^h
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)	Other constituents			
1959																			TPDS
1/8 1150	49 ^b	11.8	100	7.2	3.5	2.9	0.16	0.0	65	1.07	3.5	0.10		0.0			56	3	10 Median 0.05
2/9 1330	46	12.1	101	7.3	2.9	0.13	0.0	0.0	41	0.67	2.5	0.07		0.0			18	4	45 Maximum 0.0
3/10 1800	2,200	11.4	99	71.5	2.5	0.11	0.0	0.0	34	0.56	2.5	0.07		0.0			30	2	0 Minimum 0.06
4/14 1400	57	10.3	99	65.4	1.9	0.08	0.0	0.0	32	0.55	1.2	0.03		0.0			28	2	0
5/19 1345	1,720	9.8	100	69.3	1.9	0.06	0.0	0.2	35	0.57	2.7	0.08	0.1	0.0	1.3	Fe 0.01 Al 0.03 d	31	2	1
6/10 0930	776	9.3	101	75.9	2.5	0.11	0.0	0.0	38	0.68	1.5	0.06	0.5	0.0		PO ₄ 0.026	34	3	8
7/7 1145	440	9.1	101	87.5	3.7	0.15	0.0	0.0	42	0.69	1.8	0.05	0.0	0.0			39	6	1
8/7 1135	460	9.2	109	105	3.3	0.14	0.0	0.0	58	0.95	1.0	0.03	0.0	0.0			47	0	2
9/1 1500	420	8.9	105	118	3.6	0.15	0.6	0.0	64	1.05	3.5	0.2	0.0	0.0	1.6	Al 0.05 PO ₄ 0.00 d	52	0	2
10/14 1345	100	8.7	94	129	2.8	0.17	0.0	0.0	66	1.08	3.2	0.09	0.1	0.0			63	9	1
11/2 1430	200	10.0	95	130	4.6	0.20	0.0	0.0	66	1.08	3.5	0.10	0.1	0.0			57	3	1
12/3 1410	150	11.9	108	116	3.9	0.17	0.0	0.0	58	0.95	3.2	0.09	0.1	0.0			52	4	2

^a Field pH

^b Laboratory pH

^c Sum of calcium and magnesium in ppm

^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.6 except as shown.

^e Derived from conductivity vs. TDS curves

^f Determined by addition of analytical constituents.

^g Gravimetric determination.

^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.

ⁱ Manual analyses made by United States Geological Survey, Quality Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBCFD); California Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBDPH); Terminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-5
ANALYSES OF SURFACE WATER
LANGRISH BASIN (NO. 6)

CANON RIVER, NEAR FISH AT MOUNTAIN (STA. 1194)

Date sample collected P.S.T.	Discharge in cfs in SF	Temp in °F	Dissolved oxygen in ppm % Sat	Specific conductance in micromhos at 25°C	pH ^a	Mineral constituents in equivalents per million										Total solids in ppm	Per- cent solids in ppm	Hardness in ppm	Tur- bidity in ppm	Tur- bidity in ppm	Analyzed by ^h
						Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Bromine (Br)	Silica (SiO ₂)				
1959																					
1/22 1430	38	38	11.0	87	72.6	7.2	1.8	2.7	1.2	0.2	13.7	0.0	0.0	1.5	0.0	0.0	0.0	0.0	0	0	1959 ^c
2/	Not sampled																				
2/10 0810	38	34	11.5	81	88.7	7.3	1.5	5.4	0.9	0.3	17.0	1.8	0.0	1.0	0.0	0.0	0.0	0.0	0	0	1959 ^c
4/14 1430	188	44	10.1	81	49.0	7.3	0.34	1.6	0.0	0.0	28.0	0.0	0.0	1.0	0.0	0.0	0.0	0.0	0	0	1959 ^c
5/15 0800	182	43	10.3	83	45.5	7.3	0.0	0.9	0.0	0.0	29.0	2.9	0.0	1.0	0.0	0.0	0.0	0.0	0	0	1959 ^c
6/16 0830	85	60	8.2	82	51.2	7.3	0.33	2.2	1.1	0.1	28.0	4.8	0.0	0.2	0.0	0.0	0.0	0.0	0	0	1959 ^c
7/9 0750	32	55	8.6	81	63.6	7.3	0.0	1.1	1.7	0.0	29.0	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0	0	1959 ^c
8/14 0900	20	60	8.2	82	80.2	7.3	0.8	2.6	1.6	0.0	35.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0	0	1959 ^c
9/3 1000	9.4	54	8.4	78	87.2	7.4	1.0	2.2	1.8	0.0	31.0	1.0	0.0	1.3	0.0	0.0	0.0	0.0	0	0	1959 ^c
10/13 1000	16	45	9.4	78	79.0	7.8	0.8	1.2	1.6	0.0	46.0	1.0	0.0	0.2	0.0	0.0	0.0	0.0	0	0	1959 ^c
11/10 0905	Broken in throat																				
12/2	34	11.5	100	78.1	7.1	0.36	5.0	4.2	0.0	0.0	42.0	1.2	0.0	1.2	0.0	0.0	0.0	0.0	0	0	1959 ^c

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i Mineral analyses made by United States Geological Survey, Office of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District, Southern California (SMC), Los Angeles Department of Water and Power (LADWP), City of Long Beach, Department of Public Health (LSDPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-5

ANALYSES OF SURFACE WATER
LAKEWATER SECTION (M1, 6)

LAB. TIME AT BLDG (7A, 39)

Date sampled PST	Discharge temp in cfs	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million							Total dissolved solids in ppm	Total solids in ppm	Hardness as CaCO ₃ in ppm	Total hardness as CaCO ₃ in ppm	Assigned By 1
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Other constituents
2-59															
1-22	44	9.9	81	94.5	7.3			0.00	0.82		4.0	0.08			
1-23															
2-5															
2-6															
2-16	Not sampled														
3-6	46	9.7	81	91.4	7.7			0.00	0.70		4.1	0.07			
3-16															
4-7	5	8.6	81	75.4	7.5			0.00	0.69		4.2	0.07			
4-13															
4-14	54	8.6	80	76.1	7.6			0.00	0.69		4.3	0.08			
4-16															
4-17	48	7.5	77	90.4	7.5			0.00	0.68		4.4	0.08			
4-18															
4-19	61	8.4	83	93.5	7.5			0.00	0.68		4.5	0.08			
4-20															
4-21	63	8.1	83	94.4	7.6			0.00	0.68		4.6	0.08			
4-22															
4-23	74	7.5	81	95.4	7.6			0.00	0.68		4.7	0.08			
4-24															
4-25	58	7.7	81	96.4	7.7			0.00	0.68		4.8	0.08			
4-26															
4-27	50	7.9	77	95.6	7.7			0.00	0.68		4.9	0.08			
4-28															
4-29	48	9.4	113	96.9	7.7			0.00	0.68		5.0	0.08			
4-30															
4-31															

From pH

Laboratory pH

Laboratory temperature in °C

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

Laboratory pH (at 25°C) (at 25°C) (at 25°C)

TABLE B-4
ANALYSES OF SURFACE WATER
LABORTON REGION (NO. 6)

Date sampled PST	Oxygen Temp in cis in °F	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million										Total dis- solved solids in ppm	Per- centage of Total N.C. ppm	Hardness as CaCO ₃ ppm	Tur- bidity in FTU	Coliform ^h MPN/ml	Analyzed by
				Calcium (Ca)	Magnesium (Mg)	Sodium sum (Na)	Potassium sum (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Silica (SiO ₂)					
1959																			USBR
1/6 1959	51	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	3	Median 0.23
2/3 1959	5	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
3/10 1959	5	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
4/1 1959	5	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
5/2 1959	52	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
5/21 1959	52	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
7/15 1959	58	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
7/8 1959	60	8.5	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
8/13 1959	62	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
10/12 1959	65	7.5	79	94.8	2.2 0.1	6.2 0.25	1.0 0.05	0.0	0.0	0.0	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
11/9 1959	59	9.1	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
12/1 1959	51	8.8	81.1	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23
12/15	1.9	9.1	95.7	7.65 ^c	5.9 0.25	5.9 0.25	0.0	0.0	50 0.75	3.5 0.1	0.0	0.0	0.0	0.0	63 ^d	28	0	0.8	Median 0.23

a Field pH.

b Laboratory pH.

c Sum of calcium and magnesium in ppm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves.

f Determined by addition of analyzed constituents.

g Gravimetric determination.

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i Annual analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR); United States Public Health Service (USPHS); San Bernardino County Flood Control District (SBFCFD); Metropolitan Water District of Southern California (MWD); Los Angeles Department of Water and Power (LADWP); City of Los Angeles, Department of Public Health (LADPH); City of Long Beach, Department of Public Health (LBPH); Teminal Testing Laboratories, Inc. (TTL); or California Department of Water Resources (DWR), as indicated.

TABLE B-5
ANALYSES OF SURFACE WATER
LABONTIAN REGION (NO. 6)

SISAN RIVER AT SERRAVALLE (SPN, 176)

Date and time sampled PST	Discharge Temp in °F in °C	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Major constituents in parts per million										Total solved solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ ppm	Tur- bidity in ppm	Sulfate in mg/l	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash sum (K)	Carbon- ate (CO ₃)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trate (NO ₃)	Fluor- ide (F)	Boron (B)	Other constituents				
1959																			
1/	Not Sampled	Not Sampled																	
2/5	19.4	31	11.9	88	1.39	7.3													
3/10																			
3/6	87	36	11.9	86	1.08	7.4													
4/9																			
4/9																			
5/7	163	56	8.7	89	98.2	7.5													
6/6	21.6	58	7.9	77	122	7.4													
7/17	3.9	64	7.3	76	178	7.5													
8/13																			
9/10	4.3	61	8.3	84	198	7.6													
10/15	8.6	48	9.6	82	173	7.6													
11/12	9.3	37	11.0	81	171	7.3													
12/10	11	33	11.8	95	175	7.3													
12/15																			

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in eqm.

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Amol median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service.
i. Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS); United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBFCFD), Metropolitan Water District of Southern California (MWSD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL) or California Department of Water Resources (DWR), as indicated.

TABLE B-5

ANALYSES OF SURFACE WATER

LOS ANGELES REGION (WC - 6)

THICKER ALYER NEAR FAHAD (STA. 417)

Date and time of day, P.S.T.	Discharge Temp. in cts in air	Dissolved oxygen ppm % Sat.	Specific Conductance at 25°C gm/cm. Sec.	Major constituents in equivalents per million										Total dissolved solids in gram	Percent total dissolved solids in gram	Hardness as CaCO ₃ ppm	Total N. ppm	Assigned by	
				Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)						Boron (B)
1959 1/23 (30)	40.3	38	10.9	8.											6.0	1.6	0	34	Residue
1/	Not available																		
3/9 4:15	4.98	5	1.9	86												4.	0	0	Residue
3/11 4:00	5.1	4.7	8.3	1.0												3.9	34	1	Residue
3/11 11:30	5.0	5.2	4.	44.5												4.7	0	0	Residue
6/15 10:00	47.6	60	8.1	81												4.7	0	0	Residue
7/6 1:00	5.0	63	80.0	63												4.7	0	0	Residue
8/13 1:00	54.8	6	7.6	82	103	7.7										4.7	0	0	Residue
9/15 1:00	51.9	40	6.3	8												4.7	0	0	Residue
10/13 1:00	53.6	53	8.	74	97.9	7.9										4.7	0	0	Residue
11/19 1:00	54.3	40	4.7	65												4.7	0	0	Residue
12/11 1:00	36.9	40	10.9	106	7.4											4.7	0	0	Residue

J. F. and J. H.

b. Laboratory pH

c. Laboratory analyses in mg/l

d. Total dissolved solids in mg/l

e. Total dissolved solids in mg/l

f. Total dissolved solids in mg/l

g. Total dissolved solids in mg/l

h. Total dissolved solids in mg/l

i. Total dissolved solids in mg/l

j. Total dissolved solids in mg/l

k. Total dissolved solids in mg/l

l. Total dissolved solids in mg/l

m. Total dissolved solids in mg/l

n. Total dissolved solids in mg/l

o. Total dissolved solids in mg/l

p. Total dissolved solids in mg/l

q. Total dissolved solids in mg/l

r. Total dissolved solids in mg/l

s. Total dissolved solids in mg/l

t. Total dissolved solids in mg/l

u. Total dissolved solids in mg/l

v. Total dissolved solids in mg/l

w. Total dissolved solids in mg/l

x. Total dissolved solids in mg/l

y. Total dissolved solids in mg/l

z. Total dissolved solids in mg/l

TABLE B-5

ANALYSES OF SURFACE WATER

LABORATORY REGION (30, 4)

TRUCKEE RIVER NEAR TRUCKEE (STA. 52)

Date and time sampled PST	Discharge in cfs m ³ /s	Observed oxygen in % ppm	Specific conductance (microhm) at 25°C µmS/cm	Mineral constituents in parts per million										Total dissolved solids in ppm	Per- cent solid in ppm	Hardness on CaCO ₃ Total N C ppm	Tur- bidity in MPN/ml	Analyzed by I			
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potash- ium (K)	Carbon- dioxide (CO ₂)	Bicar- bonate (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Ni- trogen (N)	Fluo- ride (F)						Boron (B)	Silica (SiO ₂)	Other constituents
10/6																					
1/12	41 ^a		66.8	7.1 ^b	3.1 ^c 0.13	0.0	0.0	30	0.0	2.0	0.06				48 ^e	19	27	2	170	Median 5-6	
8/2			46.5	7.7 ^b	2.6 ^c 0.08	0.0	0.0	51	0.0	1.5	0.04				63 ^d	26	34	7	1	Maximum 230	
3/3	212	10.1	91.4	7.9 ^b	4.7 ^c 0.20	0.0	0.0	47	0.0	3.6	0.10					65 ^e	23	34	0	1	Maximum 0.08
4/10 ^a	141	0.6	90	73.3	3.7 ^c 0.16	0.0	0.0	37	0.0	2.0	0.06					58 ^e	20	32	2	2	
10/16 ^a					1.2 ^c 0.03	0.0	0.0	56	0.0	1.5	0.04					44 ^f	19	22	1	30	
5/14	170	0.8	83	54.9	3.1 ^c 0.13	0.0	0.0	30	0.0	2.0	0.06					48 ^e	21	94	0	0.3	
13/6	84	7.5	76	63.9	2.6 ^c 0.08	0.0	0.0	51	0.0	1.5	0.04					71 ^e	27	37	0	1	
6/15	68				4.7 ^c 0.20	0.0	0.0	47	0.0	3.6	0.10										
7/8	65	8.7	90	99.5	3.7 ^c 0.16	0.0	0.0	37	0.0	2.0	0.06					69 ^e	27	36	0	10	
11/95					1.2 ^c 0.03	0.0	0.0	51	0.0	1.5	0.04										
8/13	492	8.0	85	96.2	3.1 ^c 0.13	0.0	0.0	30	0.0	2.0	0.06					65 ^d	30	33	0	0.4	
9/23	480	7.6	78	97.2	2.6 ^c 0.08	0.0	0.0	51	0.0	1.5	0.04										
11/15	277	96.5	7.9 ^b	98.5	4.7 ^c 0.20	0.0	0.0	47	0.0	3.6	0.10					70 ^e	26	42	1	1	
10/12	315	9.9	78	94	3.7 ^c 0.16	0.0	0.0	37	0.0	2.0	0.06					67 ^e	30	37	0	2	
11/9	310	10.1	102	102	1.2 ^c 0.03	0.0	0.0	57	0.0	1.5	0.04					73 ^e	32	35	0	3	
12/1					3.1 ^c 0.13	0.0	0.0	30	0.0	2.0	0.06										
11/05					2.6 ^c 0.08	0.0	0.0	51	0.0	1.5	0.04										

^a Field pH^b Laboratory pH^c Sum of calcium and magnesium in ppm^d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown 0.00^e Derived from conductivity vs TDS curves^f Determined by addition of analyzed constituents^g Gravimetric determination^h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Serviceⁱ Manual analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood^j Control District (SBCFCD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water and Power (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of

Public Health (LBDPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-5
ANALYSES OF SURFACE WATER

LABORATORY NO. 100

WATER, RIVER, LAKE, RESERVOIR, etc.

Date and time sampled PST	Storage Temp. in °C in 24 hr	Dissolved Oxygen in ppm	Specific Conductance at 25°C in $\mu\text{mhos/cm}$	Major constituents in equivalents per million										Total Solids in mg/l	Total Solids in ppm	Turbidity in NTU	Color in Pt-Co	Analyze By
				Calcium (Ca)	Magnesium (Mg)	Potassium (K)	Sodium (Na)	Chloride (Cl)	Sulfate (SO ₄)	Carbonate (CO ₃)	Bicarbonate (HCO ₃)	Silica (SiO ₂)	Fluoride (F)					
1/1/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/2/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/3/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/4/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/5/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/6/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/7/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/8/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/9/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/10/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/11/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/12/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/13/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/14/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/15/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/16/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/17/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/18/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/19/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/20/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/21/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/22/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/23/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/24/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/25/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/26/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/27/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/28/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/29/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/30/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10
1/31/55	15	10	150	100	50	10	10	10	10	10	10	10	10	100	100	10	10	10

TABLE B-5

ANALYSES OF SURFACE WATER

LAKOTA RIVER, WEST NEAR GONZALEZ (STA. 116)

WALKER RIVER, WEST NEAR GONZALEZ (STA. 116)

Date and time analyzed P.S.T.	Discharge Temp. in °F.	Dissolved oxygen ppm	Specific conductance at 25°C µmhos/cm	Mineral constituents in equivalents per million												Total dis- solved solids in ppm	Per- cent solids in ppm	Hardness as CaCO ₃ ppm	Total N.C. in ppm	Turbid- ity in ppm	Conform- ity in ppm	Analyzed by
				Calcium (Ca)	Magne- sium (Mg)	Sodium (Na)	Potassium (K)	Carbon- dioxide (CO ₂)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Nitrate (NO ₃)	Fluoride (F)	Boron (B)	Silica (SiO ₂)							
1/9/9																						
1/22	48	33	11.6	81	111	7.3																
1/23																						
2/	Not sampled																					
3/1	8	36	11.1	81	152	7.7																
10/40																						
4/16	320	42	10.6	44	52.4	7.4																
11/00																						
5/15	634	44	10.2	83	31.3	7.5																
10/30																						
6/16	47	54	8.6	82	33.4	7.4																
11/15																						
7/8	126	67	7.0	76	89.0	7.7																
16/00																						
8/14	50	65	7.3	77	119	7.5																
11/30																						
9/3	34	61	8.3	84	137	7.8																
11/45																						
10/13	40	52	9.0	81	180	8.2																
13/15																						
11/10	36	44	10.1	82	110	7.8																
11/30																						
12/2	44	35	11.4	104	174	7.3																
11/45																						

a Field pH

b Laboratory pH

c Sum of calcium and magnesium in ppm

d Iron (Fe), aluminum (Al), arsenic (As), copper (Cu), lead (Pb), manganese (Mn), zinc (Zn), and hexavalent chromium (Cr⁶⁺), reported here as 0.0 except as shown.

e Derived from conductivity vs TDS curves

f Determined by addition of analyzed constituents

g Gravimetric determination

h Annual median and range, respectively. Calculated from analyses of duplicate monthly samples made by California Department of Public Health, Division of Laboratories, or United States Public Health Service

i Mineral analyses made by United States Geological Survey, Quality of Water Branch (USGS), United States Department of the Interior, Bureau of Reclamation (USBR), United States Public Health Service (USPHS), San Bernardino County Flood Control District (SBCFD), Metropolitan Water District of Southern California (MWD), Los Angeles Department of Water (LADWP), City of Los Angeles, Department of Public Health (LADPH), City of Long Beach, Department of Public Health (LBPH), Terminal Testing Laboratories, Inc. (TTL), or California Department of Water Resources (DWR), as indicated.

TABLE B-6

RADIOASSAY OF SURFACE WATERS

NORTH COASTAL REGION (NO. 1)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
1	Klamath River	Copco	5-17 9-11	6.18 ± 3.90 7.93 ± 4.70	2.05 ± 4.13 0.79 ± 4.48	0.00 ± 0.50 0.90 ± 0.34	0.36 ± 0.55 0.37 ± 0.30
1a	Shasta River	Yreka	5-6 9-8	14.07 ± 4.12 0.00 ± 4.56	0.96 ± 4.11 0.00 ± 4.56	0.30 ± 0.50 0.00 ± 0.21	0.45 ± 0.69 0.28 ± 0.34
1b	Scott River	Fort Jones	5-6 9-8	1.73 ± 3.73 0.00 ± 4.51	1.63 ± 4.38 0.00 ± 4.46	0.10 ± 0.55 0.90 ± 0.25	0.45 ± 0.65 0.00 ± 0.20
1c	Klamath River	Hamburg Reservoir Site	5-13 9-8	3.59 ± 4.00 8.95 ± 4.71	1.74 ± 4.39 2.18 ± 4.51	0.10 ± 0.55 0.00 ± 0.29	0.27 ± 0.47 0.00 ± 0.29
1d	Butte Creek	MacDoel	5-6 9-7	6.18 ± 4.05 0.00 ± 4.56	13.16 ± 4.06 0.00 ± 4.41	0.72 ± 0.58 0.00 ± 0.34	0.45 ± 0.65 0.17 ± 0.34
1e	Antelope Creek	Tennant	5-6 9-7	3.36 ± 4.00 0.00 ± 4.51	13.16 ± 4.06 0.00 ± 4.56	0.41 ± 0.48 0.17 ± 0.40	0.54 ± 0.64 0.00 ± 0.37
2	Klamath River	Somesbar	5-6 9-10	1.59 ± 4.10 3.54 ± 4.41	12.85 ± 3.95 0.00 ± 4.31	0.10 ± 0.34 0.27 ± 0.34	0.36 ± 0.50 0.09 ± 0.26
2a	Salmon River	Somesbar	5-6 9-10	1.59 ± 4.12 0.00 ± 4.16	17.67 ± 4.12 0.14 ± 4.11	0.10 ± 0.34 0.00 ± 0.26	0.00 ± 0.38 0.00 ± 0.21
2b	Klamath River	Seiad Valley	5-13 9-8	15.61 ± 4.60 7.61 ± 4.31	19.08 ± 4.17 0.00 ± 4.11	0.20 ± 0.37 0.09 ± 0.15	0.45 ± 0.60 0.56 ± 0.48
3	Klamath River	Klamath	5-5 9-1	15.56 ± 4.55 0.00 ± 4.26	10.88 ± 3.90 0.00 ± 4.26	0.10 ± 0.34 0.19 ± 0.31	0.64 ± 0.65 0.65 ± 0.40
3a	Smith River	Crescent City	5-5 9-2	3.87 ± 4.00 3.46 ± 3.57	2.48 ± 4.43 2.67 ± 3.57	0.20 ± 0.37 0.18 ± 0.29	0.54 ± 0.64 0.09 ± 0.25

TABLE B-6

RADIOASSAY OF SURFACE WATERS

NORTH COASTAL REGION (NO. 1)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
3b	Redwood Creek	Orick	5-6 9-1	0.00 \pm 5.30 0.00 \pm 3.81	4.51 \pm 4.50 1.88 \pm 3.86	0.20 \pm 0.31 0.00 \pm 0.33	0.00 \pm 0.50 0.27 \pm 0.44
4	Trinity River	Hoopa	5-6 9-10	0.31 \pm 5.35 1.09 \pm 4.06	2.73 \pm 3.75 0.00 \pm 4.01	0.92 \pm 0.55 0.00 \pm 0.35	0.54 \pm 0.50 0.27 \pm 0.43
4a	Trinity River	Lewiston	5-4 9-10	1.88 \pm 4.00 9.05 \pm 4.11	9.27 \pm 4.18 3.98 \pm 4.01	0.00 \pm 0.37 0.08 \pm 0.25	0.36 \pm 0.50 0.26 \pm 0.37
4b	Trinity River	Burnt Ranch	5-6 9-10	15.56 \pm 4.80 4.14 \pm 4.02	5.04 \pm 4.13 6.02 \pm 4.06	0.41 \pm 0.45 0.18 \pm 0.42	0.09 \pm 0.42 0.09 \pm 0.45
5	Eel River	McCann	5-5 9-8	2.08 \pm 4.55 0.00 \pm 3.86	7.72 \pm 3.66 1.58 \pm 3.97	0.72 \pm 0.45 0.00 \pm 0.36	0.00 \pm 0.32 0.09 \pm 0.45
5a	Van Duzen River	Bridgeville	5-7 9-8	0.80 \pm 3.90 6.30 \pm 3.86	5.35 \pm 3.58 4.20 \pm 3.81	0.61 \pm 0.66 0.00 \pm 0.41	0.63 \pm 0.73 0.19 \pm 0.49
5b	Outlet Creek	Longvale	5-14 9-15	1.08 \pm 3.90 0.00 \pm 3.59	3.49 \pm 4.60 0.00 \pm 3.59	0.61 \pm 0.66 0.19 \pm 0.38	0.63 \pm 0.65 0.38 \pm 0.49
5c	Eel River, Middle Fork	Dos Rios	5-14 9-15	4.24 \pm 4.00 2.23 \pm 3.59	12.03 \pm 4.80 0.00 \pm 3.50	0.10 \pm 0.56 0.27 \pm 0.49	0.85 \pm 0.64 0.09 \pm 0.39
5d	Eel River	Dos Rios	5-13 9-15	0.00 \pm 3.90 0.00 \pm 3.76	0.00 \pm 4.40 0.21 \pm 3.76	0.51 \pm 0.58 0.09 \pm 0.25	0.63 \pm 0.74 0.36 \pm 0.46
6	Eel River	Scotia	5-5 9-8	5.15 \pm 4.16 7.00 \pm 4.60	3.94 \pm 4.38 0.00 \pm 4.43	0.51 \pm 0.56 0.17 \pm 0.17	0.10 \pm 0.56 0.25 \pm 0.37
6a	Mad River	Arcata	5-5 9-9	4.61 \pm 4.00 0.00 \pm 4.15	8.62 \pm 4.37 0.00 \pm 4.15	0.20 \pm 0.37 0.09 \pm 0.24	0.42 \pm 0.58 0.26 \pm 0.31

TABLE B-6

RADIOASSAY OF SURFACE WATERS

NORTH COASTAL REGION (NO. 1)
(continued)

Sto. No.	Stream	Neor	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
7	El River, South Fork	Miranda	5-5 9-8	4.84 ± 3.90 0.00 ± 4.26	6.40 ± 4.29 0.00 ± 4.20	0.20 ± 0.31 0.09 ± 0.39	0.21 ± 0.58 0.36 ± 0.42
7a	Mattole River	Petrolia	5-7 9-9	2.65 ± 4.10 0.00 ± 4.20	12.91 ± 4.29 0.00 ± 4.32	0.10 ± 0.34 0.27 ± 0.39	0.00 ± 0.31 0.00 ± 0.46
8a	Russian River	Hopland	5-15 9-3	0.00 ± 3.90 0.00 ± 4.26	6.82 ± 4.30 0.00 ± 4.32	0.51 ± 0.56 0.19 ± 0.31	0.21 ± 0.53 0.36 ± 0.31
8b	Navarro River	Navarro	5-4 9-7	2.16 ± 4.12 0.00 ± 4.15	6.73 ± 3.97 0.00 ± 4.08	0.00 ± 0.37 0.09 ± 0.16	0.42 ± 0.52 0.09 ± 0.27
8c	Big River	Mouth	5-4 9-7	3.59 ± 4.15 0.00 ± 3.90	8.82 ± 4.25 0.00 ± 3.96	0.20 ± 0.34 0.17 ± 0.36	0.63 ± 0.57 0.09 ± 0.15
9	Russian River	Healdsburg	5-11 9-4	0.08 ± 4.70 0.00 ± 3.90	0.93 ± 4.11 0.00 ± 3.84	0.20 ± 0.31 0.09 ± 0.33	0.42 ± 0.48 0.09 ± 0.38
9a	Gualala River, South Fork	Annapolis	5-4 9-7	0.00 ± 4.35 0.00 ± 3.77	4.37 ± 4.14 0.00 ± 3.64	0.51 ± 0.51 0.00 ± 0.24	0.63 ± 0.48 0.00 ± 0.28
10	Russian River	Guerneville	5-4 9-7	0.00 ± 4.40 0.00 ± 4.08	4.00 ± 4.12 0.00 ± 4.04	0.30 ± 0.41 0.17 ± 0.20	0.21 ± 0.37 0.17 ± 0.28
10a	Russian River, East Fork	Potter Valley Powerhouse	5-13 9-3	21.11 ± 4.60 13.82 ± 3.91	2.45 ± 4.05 10.63 ± 8.81	0.41 ± 0.48 0.09 ± 0.25	0.31 ± 0.53 0.18 ± 0.42
10c	Noyo River	Fort Bragg	5-4 9-7	12.05 ± 4.40 9.13 ± 4.06	0.36 ± 4.08 8.15 ± 4.01	0.10 ± 0.41 0.00 ± 0.29	0.10 ± 0.42 0.18 ± 0.29

TABLE B-7

RADIOASSAY OF SURFACE WATERS

SAN FRANCISCO BAY REGION (NO. 2)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
71	Arroyo del Valle	Livermore	5-13	5.10 \pm 4.15	3.86 \pm 4.11	0.52 \pm 0.45	0.62 \pm 0.61
72	Napa River	St., Helena	5-11	1.88 \pm 3.59	4.36 \pm 4.40	0.21 \pm 0.60	0.29 \pm 0.47
73	Alameda Creek	Niles	5-13	2.25 \pm 4.00	0.31 \pm 4.10	0.41 \pm 0.50	0.62 \pm 0.73
74	Los Gatos Creek	Los Gatos	5-13	0.60 \pm 4.16	3.46 \pm 4.21	0.52 \pm 0.50	0.62 \pm 0.68
			9-8	3.57 \pm 3.88	6.92 \pm 3.96	0.00 \pm 0.44	0.19 \pm 0.38
82	Coyote Creek	Madrone	5-12	0.00 \pm 4.10	2.95 \pm 4.43	0.62 \pm 0.48	0.62 \pm 0.56
			9-9	8.56 \pm 3.71	11.46 \pm 3.27	0.19 \pm 0.31	0.19 \pm 0.44

TABLE B-8

RADIOASSAY OF SURFACE WATERS

CENTRAL COASTAL REGION (NO. 3)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
43	Salinas River	Spreckels	5-12	0.00 \pm 4.12	2.39 \pm 4.41	0.20 \pm 0.53	0.41 \pm 0.65
75	San Lorenzo River	Big Trees	5-13 9-8	3.11 \pm 4.23	5.80 \pm 4.45	0.20 \pm 0.37	0.20 \pm 0.43
				7.74 \pm 3.97	0.00 \pm 3.22	0.17 \pm 0.29	0.17 \pm 0.35
76	Soquel Creek	Soquel	5-13 9-9	0.00 \pm 4.34	3.38 \pm 4.47	0.41 \pm 0.53	1.04 \pm 0.61
				5.56 \pm 4.06	7.09 \pm 4.06	0.00 \pm 0.35	0.35 \pm 0.35
77	Pajaro River	Chittenden	5-12 9-9	5.55 \pm 4.90	1.83 \pm 4.35	0.00 \pm 0.48	0.41 \pm 0.48
				4.00 \pm 4.83	9.29 \pm 4.93	0.00 \pm 0.21	0.18 \pm 0.28
83	Carmel River	Carmel	5-12	4.98 \pm 4.42	2.11 \pm 4.40	0.30 \pm 0.45	0.31 \pm 0.40
96	Uvas Creek	Morgan Hill	5-12 9-9	0.00 \pm 4.35	6.65 \pm 4.76	0.10 \pm 0.40	0.00 \pm 0.31
				4.40 \pm 4.78	0.00 \pm 4.63	0.90 \pm 0.26	0.00 \pm 0.30

TABLE B-9

RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
11	Sacramento River	Delta	5-5 9-9	0.00 ± 3.82 1.06 ± 4.42	6.73 ± 3.60 2.46 ± 4.42	0.58 ± 0.51 0.36 ± 0.36	0.59 ± 0.40 0.18 ± 0.41
11a	Cottonwood Creek	North Fork Cottonwood Creek	5-4 9-2	4.27 ± 4.16 6.24 ± 4.42	3.41 ± 4.13 4.92 ± 4.42	0.34 ± 0.51 0.09 ± 0.25	0.82 ± 0.51 0.62 ± 0.43
11b	Cottonwood Creek	South Fork Cottonwood Creek	5-4 9-2	3.30 ± 4.18 2.18 ± 4.37	0.98 ± 4.15 2.43 ± 4.37	0.11 ± 0.40 0.00 ± 0.33	0.47 ± 0.38 0.09 ± 0.25
12	Sacramento River	Kearwick	5-5 9-8	2.42 ± 4.45 8.26 ± 4.42	0.70 ± 4.41 3.30 ± 5.39	0.58 ± 0.36 0.00 ± 0.21	0.47 ± 0.32 0.29 ± 0.39
12b	Cottonwood Creek	Cottonwood	5-4 9-1	0.00 ± 4.35 0.33 ± 3.91	2.87 ± 4.45 1.82 ± 3.91	0.97 ± 0.70 0.00 ± 0.41	0.89 ± 0.60 0.28 ± 0.52
12c	Sacramento River	Bend	9-1	8.26 ± 3.91	4.81 ± 3.86	0.00 ± 0.36	0.27 ± 0.45
12d	Clear Creek	Igo	5-14 9-1	4.44 ± 3.54 23.21 ± 4.32	6.54 ± 3.91 4.62 ± 3.71	0.58 ± 0.66 0.10 ± 0.35	0.20 ± 0.32 0.47 ± 0.56
13	Sacramento River	Hamilton City	5-8 9-1	1.99 ± 3.52 9.85 ± 4.12	1.24 ± 4.42 8.20 ± 4.07	0.09 ± 0.43 0.09 ± 0.39	0.30 ± 0.36 0.27 ± 0.49
13a	Stony Creek	Hamilton City	5-12	0.00 ± 3.50	5.46 ± 4.47	0.48 ± 0.43	0.40 ± 0.45
13b	Sacramento River	Colusa	5-4 9-7	3.44 ± 3.48 3.08 ± 4.07	7.86 ± 4.03 2.18 ± 4.07	0.68 ± 0.70 0.00 ± 0.29	0.20 ± 0.45 0.00 ± 0.29
13c	Stony Creek	Black Bluff Dam Site	5-12 9-2	5.21 ± 3.50 3.94 ± 4.12	12.12 ± 4.19 1.54 ± 4.02	0.58 ± 0.70 0.00 ± 0.29	0.40 ± 0.50 0.26 ± 0.32
13d	Thames Creek	Packenta	5-11 9-2	3.50 ± 3.57 5.29 ± 3.96	9.27 ± 4.41 5.29 ± 3.96	0.58 ± 0.60 0.00 ± 0.15	0.60 ± 0.45 0.29 ± 0.39

TABLE B-9
RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
13e	Elder Creek	Paskenta	5-11 9-2	8.52 ± 3.80 0.00 ± 3.45	5.41 ± 4.38 2.27 ± 3.61	0.68 ± 0.62 0.18 ± 0.21	0.20 ± 0.32 0.27 ± 0.33
14	Sacramento River	Knights Landing	5-4 9-7	6.86 ± 3.75 0.00 ± 3.56	3.83 ± 4.10 0.00 ± 3.50	0.38 ± 0.56 0.18 ± 0.21	0.40 ± 0.38 0.09 ± 0.15
14a	Sacramento Slough	Knights Landing	5-4 9-7	0.00 ± 3.83 3.64 ± 3.91	2.36 ± 4.14 2.77 ± 3.86	0.48 ± 0.54 0.09 ± 0.15	0.60 ± 0.45 0.37 ± 0.29
15	Sacramento River	Sacramento	5-11 9-7	5.50 ± 3.98 2.41 ± 3.97	12.96 ± 3.99 3.64 ± 4.13	0.09 ± 0.43 0.00 ± 0.30	0.30 ± 0.47 0.00 ± 0.33
16	Sacramento River	Rio Vista	5-11 9-7	2.28 ± 3.93 1.56 ± 3.96	11.41 ± 4.47 1.03 ± 3.96	0.09 ± 0.43 0.09 ± 0.33	0.20 ± 0.50 0.27 ± 0.26
16a	Calaveras River	Jenny Lind	5-15	0.00 ± 4.11	7.04 ± 3.45	0.28 ± 0.54	0.40 ± 0.32
17	Pit River	Montgomery Creek	5-5 9-9	0.00 ± 4.15 8.82 ± 4.07	0.28 ± 3.64 0.64 ± 6.47	0.09 ± 0.48 0.09 ± 0.32	0.50 ± 0.42 0.17 ± 0.32
17a	Pit River	Canby	5-7 9-9	4.64 ± 4.30 6.72 ± 4.02	8.25 ± 3.91 0.00 ± 3.82	0.19 ± 0.51 0.17 ± 0.17	0.40 ± 0.50 0.09 ± 0.25
17d	Indian Creek	Crescent Mills	5-7 9-10	6.49 ± 3.90 0.00 ± 4.42	2.28 ± 4.00 0.70 ± 4.47	0.51 ± 0.63 0.00 ± 0.35	0.00 ± 0.40 0.17 ± 0.41
17e	Pit River	Bieber	5-7 9-9	6.15 ± 4.35 13.33 ± 4.58	7.16 ± 3.83 2.66 ± 4.27	0.00 ± 0.33 0.00 ± 0.25	0.10 ± 0.42 0.09 ± 0.25
18	McCloud River	Shasta Lake	5-5 9-9	3.07 ± 3.52 2.10 ± 4.07	0.50 ± 3.94 0.11 ± 4.01	0.51 ± 0.56 0.00 ± 0.21	0.22 ± 0.40 0.55 ± 0.52

TABLE B-9

RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
18a	Pit River, South Fork	Likely	5-7 9-10	6.35 \pm 3.55 9.32 \pm 4.27	0.00 \pm 3.91 0.00 \pm 4.02	0.51 \pm 0.41 0.09 \pm 0.34	0.44 \pm 0.50 0.27 \pm 0.45
19	Feather River	Oroville	5-8 9-10	6.61 \pm 3.55 8.06 \pm 4.07	1.80 \pm 3.98 0.00 \pm 3.86	0.20 \pm 0.37 0.27 \pm 0.54	0.10 \pm 0.38 0.09 \pm 0.39
20	Feather River	Nicolaus	5-12 9-8	0.23 \pm 3.90 6.66 \pm 3.71	9.38 \pm 4.65 0.00 \pm 3.45	0.82 \pm 0.70 0.09 \pm 0.40	0.10 \pm 0.32 0.45 \pm 0.53
20a	Feather River	Shanghai Bend	5-12 9-1	5.53 \pm 4.12 4.28 \pm 3.46	1.18 \pm 5.02 0.64 \pm 3.35	0.61 \pm 0.62 0.17 \pm 0.35	0.00 \pm 0.21 0.35 \pm 0.35
20b	Bear River	Mouth	5-12 9-1	13.05 \pm 4.10 15.76 \pm 3.81	0.00 \pm 4.02 0.86 \pm 3.40	0.51 \pm 0.58 0.09 \pm 0.32	0.10 \pm 0.24 0.43 \pm 0.48
21	Yuba River	Marysville	5-12 9-1	11.25 \pm 4.14 0.00 \pm 3.66	0.76 \pm 3.72 0.00 \pm 3.66	0.20 \pm 0.53 0.00 \pm 0.30	0.22 \pm 0.38 0.72 \pm 0.62
21a	Yuba River	Smartville	5-12 9-1	3.62 \pm 3.85 0.00 \pm 3.87	0.25 \pm 3.30 3.80 \pm 3.96	0.30 \pm 0.56 0.18 \pm 0.36	0.77 \pm 0.52 0.18 \pm 0.30
22	American River	Sacramento	5-13 9-7	0.20 \pm 3.75 0.56 \pm 3.66	3.41 \pm 3.30 1.23 \pm 3.66	0.00 \pm 0.62 0.00 \pm 0.21	0.44 \pm 0.40 0.36 \pm 0.52
22a	American River	Nimbus Dam	5-13 9-7	3.05 \pm 3.64 0.70 \pm 3.81	3.66 \pm 4.04 0.00 \pm 3.66	0.30 \pm 0.51 0.09 \pm 0.40	0.77 \pm 0.52 0.27 \pm 0.33
22b	American River, Middle Fork	Auburn	5-14 9-4	2.96 \pm 3.64 1.84 \pm 3.35	0.28 \pm 4.12 0.00 \pm 3.76	0.61 \pm 0.37 0.00 \pm 0.29	0.44 \pm 0.45 0.27 \pm 0.47
22c	American River, South Fork	Lotus	5-14 9-4	2.42 \pm 3.62 1.56 \pm 3.96	0.00 \pm 4.06 2.71 \pm 3.96	0.20 \pm 0.46 0.00 \pm 0.36	0.22 \pm 0.35 0.27 \pm 0.39

TABLE B-9
RADIOASSAY OF SURFACE WATERS
CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
23	Mokelumne River	Woodbridge	5-6 9-3	6.41 ± 4.04 4.75 ± 4.06	0.00 ± 4.08 2.46 ± 3.96	0.41 ± 0.43 0.09 ± 0.35	1.10 ± 0.52 0.43 ± 0.52
23a	Mokelumne River	Lancha Plana	5-15 9-1	9.49 ± 4.15 3.38 ± 4.17	0.96 ± 4.12 0.05 ± 4.06	0.20 ± 0.48 0.00 ± 0.35	0.44 ± 0.45 0.26 ± 0.32
24	San Joaquin River	Friant	5-6 9-9	15.01 ± 4.25 6.85 ± 4.27	0.81 ± 4.20 2.77 ± 4.17	0.51 ± 0.58 0.89 ± 0.33	0.22 ± 0.45 0.27 ± 0.49
25	San Joaquin River	Mendota	5-4 9-10	5.18 ± 4.00 2.32 ± 4.12	0.93 ± 4.25 0.00 ± 4.12	1.13 ± 0.81 0.89 ± 0.44	0.21 ± 0.46 0.00 ± 0.54
25b	San Joaquin River	Hills Ferry Bridge	5-4 9-10	11.79 ± 4.40 10.52 ± 4.32	6.42 ± 4.17 0.64 ± 4.06	0.61 ± 0.48 0.00 ± 0.37	0.86 ± 0.63 0.00 ± 0.45
25c	San Joaquin River	Fremont Ford Bridge	9-10	14.21 ± 4.27	7.27 ± 4.06	0.09 ± 0.26	0.18 ± 0.36
26	San Joaquin River	Grayson	5-14 9-2	6.86 ± 3.78 4.25 ± 3.73	5.49 ± 4.14 3.58 ± 3.73	0.00 ± 0.31 0.18 ± 0.36	0.00 ± 0.39 0.18 ± 0.41
26a	San Joaquin River	Maze Road Bridge	5-14 9-2	0.00 ± 3.64 5.75 ± 4.08	6.96 ± 4.19 4.69 ± 4.08	0.51 ± 0.46 0.27 ± 0.42	0.64 ± 0.51 0.27 ± 0.44
27	San Joaquin River	Vernalis	5-4 9-10	0.00 ± 3.92 8.44 ± 4.23	4.45 ± 4.14 3.61 ± 4.65	0.51 ± 0.88 0.26 ± 0.43	0.53 ± 0.44 0.17 ± 0.41
28	San Joaquin River	Antioch	5-12 9-7	0.00 ± 3.93 7.58 ± 4.23	2.48 ± 4.43 2.50 ± 4.08	0.61 ± 0.88 0.00 ± 0.25	0.31 ± 0.44 0.26 ± 0.43
29	Stanislaus River	Mouth	5-4 9-10	2.59 ± 4.08 2.52 ± 4.29	2.67 ± 4.44 0.66 ± 0.42	0.30 ± 0.66 0.09 ± 0.25	0.53 ± 0.53 0.44 ± 0.38

TABLE B-9

RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
29a	Stanislaus River	Tulloch Dam	5-6 9-1	0.00 \pm 3.98 5.33 \pm 3.93	1.24 \pm 4.32 3.08 \pm 3.88	0.41 \pm 0.53 0.00 \pm 0.29	0.21 \pm 0.42 0.26 \pm 0.43
30	Tuolumne River	Hickman-Waterford Bridge	5-14 9-2	1.16 \pm 4.03 4.05 \pm 3.28	1.29 \pm 4.23 5.10 \pm 3.18	0.41 \pm 0.62 0.26 \pm 0.38	0.31 \pm 0.42 0.00 \pm 0.20
31	Tuolumne River	Tuolumne City	5-14 9-2	6.14 \pm 4.68 2.47 \pm 3.38	0.00 \pm 4.13 1.97 \pm 3.38	1.23 \pm 0.66 0.28 \pm 0.26	0.42 \pm 0.42 0.09 \pm 0.26
31a	Tuolumne River	Don Pedro Dam	5-14 9-1	9.57 \pm 4.90 6.05 \pm 4.09	0.00 \pm 4.18 4.22 \pm 4.03	0.51 \pm 0.46 0.00 \pm 0.25	1.07 \pm 0.56 0.35 \pm 0.50
32	Merced River	Stevinson	5-4 9-10	3.79 \pm 5.27 3.33 \pm 4.24	0.25 \pm 4.19 0.00 \pm 4.09	0.41 \pm 0.43 0.27 \pm 0.43	0.53 \pm 0.53 0.26 \pm 0.32
32a	Merced River	Exchequer Dam	5-6 9-1	4.04 \pm 4.22 0.00 \pm 3.89	3.21 \pm 4.19 0.36 \pm 3.90	0.10 \pm 0.51 0.09 \pm 0.32	0.21 \pm 0.42 0.60 \pm 0.51
33b	Kings River	Pine Flat Dam	5-1 9-1	7.52 \pm 4.34 4.77 \pm 4.00	6.65 \pm 4.31 6.11 \pm 4.03	0.41 \pm 0.63 0.00 \pm 0.40	1.07 \pm 0.55 0.09 \pm 0.38
33c	Kings River	North Fork	5-1 9-1	8.29 \pm 4.67 5.80 \pm 4.29	2.99 \pm 4.21 4.00 \pm 4.24	0.31 \pm 0.51 0.00 \pm 0.36	0.42 \pm 0.36 0.97 \pm 0.56
34	Kings River	Peoples Weir	5-5 9-3	11.88 \pm 4.89 0.00 \pm 4.34	6.42 \pm 4.37 0.00 \pm 4.23	0.61 \pm 0.63 0.17 \pm 0.41	0.64 \pm 0.51 3.26 \pm 0.48
35	Kaweah River	Three Rivers	5-5 9-2	8.37 \pm 6.03 3.77 \pm 4.54	6.73 \pm 4.38 0.97 \pm 4.44	0.31 \pm 0.61 0.09 \pm 0.27	0.20 \pm 0.45 0.37 \pm 0.44
36	Kern River	Bakersfield	5-5 9-3	2.05 \pm 5.92 5.16 \pm 4.39	2.87 \pm 4.35 0.00 \pm 4.19	0.42 \pm 0.70 0.56 \pm 0.58	0.40 \pm 0.50 0.84 \pm 0.60

TABLE B-9

RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
36a	Kern River	Isabella Dam	5-6	0.00 ± 3.50	8.11 ± 4.43	0.53 ± 0.61	0.60 ± 0.45
36b	Kern River	Kernville	5-6	0.00 ± 3.55	0.93 ± 4.53	0.53 ± 0.40	0.40 ± 0.38
41	Clear Lake	Lakeport	5-13 9-3	0.91 ± 3.70 8.91 ± 3.98	0.00 ± 4.50 6.50 ± 4.09	0.10 ± 0.64 0.00 ± 0.61	0.20 ± 0.45 0.26 ± 0.48
42	Cache Creek	Lower Lake	5-13 9-3	1.54 ± 3.59 2.55 ± 3.53	4.08 ± 4.49 2.50 ± 3.53	0.85 ± 0.84 0.17 ± 0.35	0.40 ± 0.50 0.35 ± 0.46
78	Bear River	Wheatland	5-11	0.00 ± 3.48	0.00 ± 4.21	0.00 ± 0.49	0.00 ± 0.45
79	Cache Creek, North Fork	Lower Lake	5-13 9-3	0.00 ± 3.37 0.19 ± 3.73	4.51 ± 4.11 2.55 ± 3.78	0.85 ± 0.56 0.09 ± 0.25	0.20 ± 0.53 0.00 ± 0.14
80	Cache Creek	Capay	5-11 9-4	2.22 ± 3.30 7.58 ± 4.09	3.52 ± 4.10 7.38 ± 4.04	0.21 ± 0.47 0.26 ± 0.25	0.50 ± 0.60 0.35 ± 0.29
81	Putah Creek	Winters	5-11 9-4	5.81 ± 3.40 0.67 ± 5.17	3.80 ± 4.01 0.84 ± 4.18	0.10 ± 0.44 0.00 ± 0.20	0.29 ± 0.47 0.26 ± 0.38
84	Butte Creek	Chico	9-1	5.94 ± 4.39	0.00 ± 4.23	0.00 ± 0.32	0.00 ± 0.20
85	Big Chico Creek	Chico	9-1	5.25 ± 4.44	0.50 ± 4.34	0.37 ± 0.37	0.18 ± 0.30
85a	Big Chico Creek	Chico	5-8	2.22 ± 3.47	2.56 ± 4.03	0.15 ± 0.36	0.60 ± 0.55
87	Colusa Trough	Colusa	5-4 9-7	5.35 ± 3.58 3.87 ± 4.49	1.88 ± 4.00 0.00 ± 4.39	0.63 ± 0.60 0.28 ± 0.34	0.10 ± 0.36 0.00 ± 0.21

TABLE B-9

RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)

(continued)

Sta. No.	Stream	Nea.	Date 1959	Micro-micro curies per liter		
				Dissolved Beta	Solid Beta	Solid Alpha
87a	Sacramento River	Butte City	9-8	1.68 ± 4.64	0.00 ± 4.49	0.00 ± 0.20
88	Mill Creek	Los Molinos	9-1	7.44 ± 4.64	0.00 ± 4.44	0.34 ± 0.35
88a	Cow Creek	Millville	5-15 9-1	5.84 ± 3.50 0.00 ± 3.98	0.00 ± 4.01 0.00 ± 3.93	0.10 ± 0.68 0.09 ± 0.31
88b	Battle Creek	Cottonwood	5-15 9-1	2.79 ± 3.95 14.94 ± 4.39	1.41 ± 4.05 5.20 ± 4.08	0.00 ± 0.55 0.17 ± 0.34
88c	Antelope Creek	Mouth	5-11 9-1	0.31 ± 4.70 1.03 ± 4.03	0.00 ± 4.01 0.00 ± 3.93	0.00 ± 0.47 0.00 ± 0.20
88e	Antelope Creek	Red Bluff	5-11	9.49 ± 4.85	3.49 ± 4.43	0.10 ± 0.15
88g	Paynes Creek	Red Bluff	5-15 9-1	12.05 ± 5.00 0.00 ± 3.98	0.28 ± 4.21 2.61 ± 4.08	0.50 ± 0.36 0.09 ± 0.14
92	Delta-Mendota Canal	Mendota	5-4 9-10	9.12 ± 6.20 0.00 ± 3.93	1.21 ± 4.00 0.00 ± 3.90	0.30 ± 0.36 0.00 ± 0.18
92a	Salt Slough	San Luis Ranch	5-4 9-10	4.90 ± 6.08 3.37 ± 3.72	2.14 ± 4.15 0.78 ± 3.64	0.40 ± 0.38 0.09 ± 0.33
93	Delta-Mendota Canal	Tracy	9-7	7.58 ± 4.29	1.61 ± 4.13	0.27 ± 0.25
94	Cosumnes River	Michigan Bar	9-1	1.32 ± 4.39	0.59 ± 4.39	0.09 ± 0.33
94a	Cosumnes River	McConnell	5-6	3.07 ± 6.00	2.53 ± 4.10	0.40 ± 0.50

TABLE B-9
RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
95a	Elder Creek	Gerber	5-12	0.62 ± 5.95	0.70 ± 4.12	0.30 ± 0.47	0.37 ± 0.53
95b	Thomes Creek	Mouth	5-12	2.08 ± 5.30	3.69 ± 3.35	0.30 ± 0.52	0.00 ± 0.48
98	Delta Cross Channel	Walnut Grove	9-7	3.79 ± 4.21	5.42 ± 4.27	0.43 ± 0.37	0.17 ± 0.35
99	Little Potato Slough	Terminus	9-7	6.51 ± 4.39	0.00 ± 4.18	0.17 ± 0.28	0.17 ± 0.28
100	Stockton Ship Channel	Rindge Island	9-11	6.43 ± 4.35	1.68 ± 4.23	0.09 ± 0.39	0.00 ± 0.33
101	San Joaquin River	Garwood Bridge	9-11	2.21 ± 3.77	0.00 ± 3.67	0.27 ± 0.44	0.09 ± 0.39
102	San Joaquin River	Mossdale Bridge	9-11	3.25 ± 3.37	0.00 ± 3.21	0.44 ± 0.33	0.54 ± 0.46
103	Old River	Tracy	9-11	9.21 ± 3.52	6.88 ± 3.18	0.09 ± 0.14	0.71 ± 0.41
103a	Grant Line Canal	Tracy Road Bridge	9-11	6.99 ± 5.15	5.36 ± 4.34	0.26 ± 0.32	0.09 ± 0.32
104	Old River	Clifton Court Ferry	9-8	0.84 ± 4.34	0.06 ± 4.28	0.00 ± 0.52	0.53 ± 0.46
106	Italian Slough	Mouth	9-8	4.52 ± 4.55	0.42 ± 4.49	0.17 ± 0.44	0.34 ± 0.34
107	Indian Slough	Brentwood	9-8	4.13 ± 4.79	0.00 ± 4.59	0.17 ± 0.28	0.34 ± 0.44
108	Old River	Orwood Bridge	9-8	0.00 ± 4.54	7.86 ± 4.80	0.09 ± 0.25	0.61 ± 0.48

TABLE B-9
RADIOASSAY OF SURFACE WATERS

CENTRAL VALLEY REGION (NO. 5)
(continued)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter			
				Dissolved Beta	Solid Beta	Dissolved Alpha	Solid Alpha
109	Rock Slough	Knightesen	9-8	3.68 ± 4.52	5.03 ± 4.54	0.09 ± 0.32	0.26 ± 0.32
110	Lindsey Slough	Rio Vista	9-7	0.00 ± 4.32	0.00 ± 4.35	0.18 ± 0.30	0.28 ± 0.40
111a	Bear Creek	Merced	5-6 9-2	12.36 ± 4.00 3.73 ± 4.23	6.39 ± 3.45 0.56 ± 4.13	0.20 ± 0.50 0.09 ± 0.40	0.37 ± 0.57 0.18 ± 0.43
112	Old River	Mandeville Island	9-11	10.95 ± 4.53	0.00 ± 4.18	0.26 ± 0.48	0.09 ± 0.38
113	Fresno River	Daulton	5-6	9.49 ± 3.90	8.45 ± 3.45	0.70 ± 0.65	0.55 ± 0.53
114	Chowchilla River	Buchanan Dam Site	5-6	2.90 ± 3.85	11.55 ± 3.50	0.40 ± 0.55	0.64 ± 0.55

TABLE B-10
RADIOASSAY OF SURFACE WATERS
LAHONTAN REGION (NO. 6)

Sta. No.	Stream	Near	Date 1959	Micro-micro curies per liter		
				Dissolved Beta	Solid Beta	Dissolved Alpha
17b	Susan River	Susanville	5-7 9-10	4.16 ± 3.50 7.64 ± 4.63	4.03 ± 3.00 0.00 ± 4.41	0.30 ± 0.52 0.26 ± 0.32
37	Lake Tahoe	Tahoe Vista	5-14 9-2	3.64 ± 3.57 1.07 ± 4.04	7.55 ± 3.25 2.81 ± 4.09	0.70 ± 0.58 0.00 ± 0.43
38	Lake Tahoe	Tahoe City	5-14 9-2	5.10 ± 3.52 5.56 ± 3.87	3.55 ± 3.55 2.67 ± 3.78	0.10 ± 0.38 0.00 ± 0.27
39	Lake Tahoe	Bljau	5-14 9-2	7.23 ± 3.60 2.64 ± 3.23	3.77 ± 4.45 0.00 ± 3.15	0.10 ± 0.42 0.00 ± 0.42
52	Truckee River	Truckee	5-14 9-2	9.49 ± 4.15 1.03 ± 2.90	8.54 ± 3.72 1.26 ± 2.90	0.30 ± 0.36 0.26 ± 0.42
53	Truckee River	Parad	5-14 9-2	2.76 ± 4.03 6.37 ± 3.47	7.78 ± 3.90 3.09 ± 3.37	0.30 ± 0.36 0.00 ± 0.25
115	Carson river, East Fork	Markleeville	9-3	0.84 ± 3.72	2.81 ± 3.78	0.00 ± 0.40
115a	Carson River, West Fork	Woodfords	5-15 9-3	4.16 ± 3.50 8.79 ± 3.79	0.96 ± 3.25 6.04 ± 3.66	0.00 ± 0.31 0.33 ± 0.43
116	Walker River, West	Coeville	5-15 9-3	4.16 ± 3.55 0.64 ± 3.45	6.54 ± 3.44 0.00 ± 3.42	0.20 ± 0.22 0.17 ± 0.27
116a	Walker River, East	Bridgeport	9-3	0.36 ± 3.83	0.00 ± 3.82	0.33 ± 0.43

TABLE B-11
RADIOASSAY OF SNOW
CENTRAL VALLEY REGION (NO. 5)

Stream basin	Snow survey course	Date 1959	Gross radioactivity in micro-micro curies per liter
American River	Phillips	1-30	980 ± 15
		2-27	240 ± 8
		3-31	1080 ± 15
Feather River	Harkness Flat	2-28	890 ± 15
		3-30	625 ± 10
		4-29	570 ± 10
	Lower Lake Helen	1-29	900 ± 15
		2-26	410 ± 9
		4-1	900 ± 15
		4-30	290 ± 9
Kings River	Sand Meadow	2-2	700 ± 10
		2-26	1270 ± 15
		3-27	940 ± 15
		4-29 ¹	1090 ± 15
Mokelumne River	Lumberyard Ranger Station	2-3	1530 ± 15
		3-2	470 ± 10
		4-3	580 ± 10
San Joaquin River	Kaiser Pass Meadow	2-3	640 ± 10
		2-26	1020 ± 15
		3-26	490 ± 10
		4-30	180 ± 7
Tuolumne River	Gin Flat	1-29	1390 ± 45
		2-25	1010 ± 15
		3-30	570 ± 15
		4-23	1440 ± 25
	Horse Camp Lodge	1-30	470 ± 10
		2-27	210 ± 8
		3-27	710 ± 10

¹ Collected from Dodson's Meadow - Sand Meadow under water of Courtright Reservoir.

TABLE B-11
 RADIOASSAY OF SNOW
 LAHONTAN REGION (NO. 6)

Stream basin	Snow survey course	Date	Gross radioactivity
:	:	:	:
:	:	1959	in micro-micro
:	:	:	curies per liter
Owens River	Upper Minarettas	2-6	720 \pm 10
		3-4	640 \pm 10
		3-31	640 \pm 10
		5-1	1180 \pm 15



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